Synthesis of Alkynes and New Transformations Catalyzed by Gold(I) Complexes
Igor Jurberg

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SYNTHESIS OF ALKYNES AND NEW TRANSFORMATIONS CATALYZED
BY GOLD(I) COMPLEXES

Presented on December 6\textsuperscript{th}, 2010 to a committee composed of:

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CHAPTER 1: UNUSUAL APPROACH TO BRANCHED 3-ALKYNYLAMIDES AND TO 1,5-DIHYDROPYRROL-2-ONES

The work described in this chapter was performed individually and was reported in the journal Organic Letters: Jurberg, I. D.; Gagosz, F.; Zard, S.; Org. Lett., 2010, 12, 3, 416.
CHAPTER 1: UNUSUAL APPROACH TO BRANCHED 3-ALKNYLAMIDES AND TO 1,5-DIHYDROPYRROL-2-ONES

1.1. INTRODUCTION: Alkyne synthesis

Because alkynes are possibly one of the most versatile functions in organic chemistry, being reported to participate in a countless number of transformations\(^1\), new and original methodologies to assemble these molecules represent a constantly important endeavour.

Despite the rich variety of conceivable routes leading to acetylenic compounds, a comprehensive discussion of each method available in the literature is outside the scope of this introduction. The focus here will be given on some of the most representative procedures to synthesize these molecules, followed by a section with a special emphasis on the methodology developed in our laboratory, based on the use of isoxazolones.

1.1.1 Alkylation of terminal alkyne through a S\(_{N}2\)-type reaction

Terminal alkynes (pKa = 24)\(^2\) can be deprotonated by a strong base, thus generating an acetylide anion that can displace primary alkyl halides in a S\(_{N}2\) fashion (scheme 1.1).

\[
\text{R}^1 \equiv - + \, \text{X} \text{R}^2 \rightarrow \text{R}^1 \equiv \text{X} \text{R}^2 \quad \text{via:} \quad \text{Nu} \quad \sigma^* \\
\text{R}^1 \equiv \text{X} \text{R}^2
\]

Scheme 1.1: Reaction of S\(_{N}2\) displacement of a primary halide.


\(^2\) All pKa values were extracted from Evans pKa table, available from the website: http://evans.harvard.edu/pdf/evans_pKa_table.pdf
This reaction of displacement allows a small level of variety\(^3\): one can perform simple alkylations, to attain terminal or internal alkynes (scheme 1.2). This reaction can be performed in liquid ammonia or in a polar aprotic solvent, such as HMPT (hexamethylphosphorous triamide, \(\text{P(NMe}_2\text{)}_3\)). Typical strong bases employed are sodium amide \((pK_a\text{conjugate acid (NH}_3\text{)} = 38)\) or \(n\)-buthyllithium \((pK_a\text{conjugate acid (n-BuH)} = 48\).

\[
\begin{align*}
\equiv & \quad \begin{array}{c}
1) \text{NaNH}_2/ \text{NH}_3 \text{ (liq)} \\
-33 \degree C
\end{array} \\
\equiv & \quad \begin{array}{c}
2) n\text{-C}_4\text{H}_9\text{Br}
\end{array} \\
\equiv & \quad \begin{array}{c}
1) 2 \text{NaNH}_2/ \text{NH}_3 \text{ (liq)} \\
-33 \degree C
\end{array} \\
\equiv & \quad \begin{array}{c}
2) 2 n\text{-C}_3\text{H}_7\text{Br}
\end{array} \\
\equiv & \quad \begin{array}{c}
1) n\text{-BuLi/ HMPT} \\
25 \degree C
\end{array} \\
\equiv & \quad \begin{array}{c}
2) \text{Cl}
\end{array} \\
\equiv & \quad 89\% \\
\equiv & \quad 66\% \\
\equiv & \quad 90\%
\end{align*}
\]

Scheme 1.2: Examples of application of different bases and electrophiles used in the synthesis of simple alkynes (extracted from ref. 3).

This method is useful as a general procedure for the synthesis of certain types of alkynes. It is also one among a not very large number of methods for the homologation of a carbon chain, another one being notably organometallic coupling reactions, which will be discussed in the next section.

A great limitation of this method comes from the fact that acetylides are very basic, thus being also capable of efficiently participating in elimination reactions (E2). For this reason, the reaction of \(S_n2\)-displacement can only be envisaged for the synthesis of alkynes using primary halides which do not have branches close to the reaction center (scheme 1.3).

---

Scheme 1.3: Examples of reactions of type E2 taking place instead of the desired substitution pathway (extracted from ref. 3).

1.1.2 Organometallic additions and coupling reactions\textsuperscript{3}

Bunsen was the first chemist to discover an organometallic coupling reaction in the 1850’s, obtaining tetramethyldiarsane from cacodyl oxide, [(CH\textsubscript{3})\textsubscript{2}As]\textsubscript{2}O, and thus opening the doors for the work of Wanklin and Frankland in the following ten years. In the 1910’s, Reformatzky, Grignard and Schrigin, and in the 1920’s, Schlenk were responsible for confirming the power of the organometallic approach and to motivate many other chemists to that growing new domain.

The main strength of this method consists on the “umpolung”\textsuperscript{4} of RX to RM species, which considerably increases reactions scope (scheme 1.4).

Scheme 1.4: Organometallic chemistry opens access to countless possibilities through the umpolung of previous electrophilic compounds (scheme 1.4 was extracted from ref. 3).

\textsuperscript{4} Expression created by G. Wittig and popularized by D. Seebach, it means “inverse poles”. For a review on the subject, see: Gröbel, B.-T.; Seebach, D.; Synthesis, 1977, 357.
A brief list of methods associated to alkyne synthesis using organometallic reagents will be presented below according to the metal employed.

### 1.1.2.1 Aluminium

The utilisation of organoalanes allows the Michael addition of acetylenic moieties to \( \alpha,\beta \)-enones\(^5\). In contrast to the use of organocuprates, which are often employed in 1,4-additions, acetylenic cuprates can hardly participate, because the alkyne moiety binds very strongly to copper making further transfer difficult to proceed\(^6\).

Organoalanes can be prepared by converting the terminal alkyne into its lithiated derivative with \( n \)-BuLi and by treating the resulting mixture with diethylaluminium chloride (scheme 1.5).

\[
\begin{align*}
R & \quad \text{1) } n\text{-BuLi} \\
& \quad \text{2) Et}_2\text{AlCl} \\
\text{R-AlEt}_2 & + \text{LiCl}
\end{align*}
\]

Scheme 1.5: Synthesis of alkynyl alanes from transmetallation of lithium derivatives.

The mixture containing the alane is treated with the \( \alpha,\beta \)-unsaturated ketone to furnish after work-up, the \( \gamma,\delta \)-alkynyl-ketone (scheme 1.6). The reaction is restricted to ketones which can adopt a s-cis conformation. Cyclic ketones, where the enone is found in a s-trans rigid conformation, such as 2-cyclohexenone, react with the alane to give only the tertiary carbinol derivative from the 1,2-addition, rather than the 1,4-addition adduct (not shown).

---


\(^6\) This led the use of acetylides as dummy ligands or place holders on cuprates, allowing the selective transfer of an alkyl or alkenyl group in the course of a Michael addition. Nevertheless, exceptions do exist: (a) Knöpfel, T. F.; Carreira, E.; *J. Am. Chem. Soc.*, 2003, 125, 6054. (b) Knöpfel, T. F.; Zarotti, P.; Ichikawa, T.; Carreira, E. M.; *J. Am. Chem. Soc.*, 2005, 127, 9682. For acetylides as cuprate dummy ligands, see: (c) Corey, E. J.; Beames, D. J.; *J. Am. Chem. Soc.*, 1972, 94, 7210.
Scheme 1.6: An example of conjugate addition using an alane derivative.\(^5\)\(^a\)

One way of correcting this limitation, thus allowing 1,4-additions on s-trans forms\(^7\) is the addition of Ni-catalysts and DiBAl-H in addition to the previous alane, which are capable of increasing considerably the yields.

### 1.1.2.2 Copper

#### 1) Alkynyl (phenyl) iodonium tosylates\(^8\).

Iodonium groups are excellent leaving groups that allow, through the loss of neutral iodobenzene, the feasibility of nucleophilic acetylenic substitutions (S\(_{nA}\)), which are otherwise, commonly reported as unfavorable.

In this context, many nucleophiles were reported to add to alkynyl(phenyl)iodonium 1.1 affording the corresponding alkynes. Among them, Gilman cuprates \(R_2CuLi\)\(^9\) are one of these viable nucleophiles, giving access to internal alkynes (scheme 1.7).

### Scheme 1.7: Examples of Gilman cuprate additions to alkynyl(phenyl) iodonium compounds.

In this regard, these species serve as masked alkynyl cations \(R\equiv C^+\), allowing umpolung

---

reactions with nucleophiles, in contrast to the better known reaction of acetylides \( \text{RC} \equiv \text{C}^- \) with electrophiles.

2) Transmetallation with alkynyl silanes\(^{10}\)

The transmetallation between a trimethylsilylalkyne and catalytic amounts of CuCl is a simple way of generating alkynylcopper reagents capable of participating in acylation reactions with acyl chlorides. The use of CuCl has been demonstrated to be superior compared to its analogues CuBr, CuI or CuCN, which produced poor yields. The choice of the solvent proved to play an important role in the ligand exchange, with the best performances being found to be with 1,3-dimethylimidazolidinone (DMI) and dimethylformamide (DMF).

The metal exchange is sensitive to steric hindrance associated to the group bounded to the alkyne extremity, which suggests the existence of a \( \pi \)-complexation with CuCl in the reaction mechanism. Groups sensitive to acidic conditions, such as acetate and tert-butyl(dimethyl)sililoxy groups are tolerated (scheme 1.8).

\[
\begin{align*}
\text{R}^1 \equiv \text{SiMe}_3 & \xrightarrow{\text{CuCl}} \text{DMI} \quad \text{R}^1 \equiv \text{Cu} \\
\text{R}^1 = \text{Ph, } n\text{-C}_6\text{H}_{13}, \text{ AcO, } \text{tBuMe}_2\text{SiO, EtO}_2\text{C} \\
\text{R}^2 = \text{Ph, } p\text{-MePh, } p\text{-ClPh, } \text{tBu}
\end{align*}
\]

Scheme 1.8: Example of a transmetallation reaction between an alkynylsilane species and copper chloride followed by nucleophilic addition to acyl compounds.

1.1.2.3 Boron:

Treatment of lithium 1-alkynyltriorganoborates with iodine produces the corresponding alkynes with essentially quantitative yields under mild conditions\(^{11}\). The reaction is applicable to a good variety of alkynyl derivatives bearing primary and secondary alkyl groups and aryl and organoboranes as well (scheme 1.9).

---


Trialkyl boranes can be readily prepared via hydroboration. They react with lithium acetylides to produce the corresponding lithium 1-alkynyltrialkylborates, which upon treatment with iodine results in the migration of the alkyl group from boron to the more electron poor carbon atom, producing a putative β-iodovinylborane intermediate. This intermediate follows a further dehaloboration step to provide the corresponding alkyne (scheme 1.10).

Scheme 1.10: Supposed β-iodovinylborane as intermediate in the iodine mediated cleavage of alkynylborate complexes.

The reaction of B-allenyl-9-boracyclo[3.3.1]nonane (B-allenyl-9-BBN) with compounds having a C=X group (X = O or N), such as aldehydes, ketones and imines, followed by the oxidation of the adduct thus obtained with hydrogen peroxide in an alkaline medium produces the corresponding homopropargylic alcohols and amines in good yields.
and without the formation of the allene isomer in most of the cases, and only trace amounts in the case of imines\textsuperscript{12} (scheme 1.11).

![Scheme 1.11: Selected examples of homopropargyl alcohols and amines derived from allenylbore complexes.](image)

1.1.2.4 Zinc

1) Addition of zinc acetylides to aldehydes

Some of the modern methods available in the literature use zinc acetylides in nucleophile additions with different degrees of complexity. Two examples are the (+)-N-methylephedrine ligand \textbf{1.2} reported by Carreira’s group\textsuperscript{13} and the elaborate proline-derived dinuclear zinc catalyst system \textbf{1.3} reported by Trost’s group\textsuperscript{14} (scheme 1.12).

---


Scheme 1.12: Two examples of zinc acetylide additions to aldehydes, in a) using Carreira’s (+)-N-methylephedrine ligand and in b) the proline-derived dinuclear zinc catalyst system reported by Trost.

2) Conjugate addition of zinc acetylides.

Conjugate addition reactions of organometallic species are well-established approaches for the stereoselective formation of carbon-carbon bonds. Besides the use of aluminium acetylides, zinc acetylides are also a worth-mentioning metal species useful in additions to α,β-enone derivatives. One example of such addition to an ephedrine derived Michael acceptor 1.4 provides enantiomerically enriched β-alkynyl acids in good yields\textsuperscript{15} (scheme 1.13).

Scheme 1.13: Example of a conjugate addition of zinc acetylides using an ephedrine-derived Michael acceptor.

1.1.2.5 Palladium

The cross-coupling reactions of organic halides and organometallic species catalyzed by palladium follow canonically the three-steps cycle: oxidative addition, transmetallation and reductive elimination. In the first step, Pd\(^{0}\)\(\text{L} \text{n}\) is inserted in the R-\(X\) bond to form the Pd\(^{2+}\) species through an oxidative addition, which is generally the rate-limiting step of the catalytic cycle. In the following step, the transmetallation, one of the substituents carried by the more electronegative metal, such as B, Zn, Si, Sn or Cu, is transferred to the more electropositive palladium species, in exchange for an halogen atom (\(X = \text{Cl}, \text{Br}, \text{I}\)). The final step is the reductive elimination, which liberates the coupled product \(R^1\)-\(R^2\) and regenerates the active palladium species, Pd\(^{0}\)\(\text{L} \text{n}\) (scheme 1.14).

\[
\begin{align*}
\text{oxidative} & \quad \text{addition} \\
\text{transmetallation} & \quad \text{reductive} \\
\text{elimination} & \quad \text{Scheme 1.14: The canonical catalytic cycle for palladium catalyzed cross-coupling reactions.}
\end{align*}
\]

Many catalysts can be employed, a few examples are Pd\((\text{PPh}_3)_4\), (\(\text{Ph}_3\text{P}\))\(_2\)PdCl\(_2\), and (\(\text{MeCN}\))\(_2\)PdCl\(_2\). Depending on the nature of the transmetalating species, the palladium mediated cross-coupling reaction receives different names. Among the well established transmetalating agents\(^{16}\) are RMgX (Kumada), RZnX (Negishi), \(R^1\text{SnR}_3\) (Stille), RSiX\(_3\) (Hiyama), \(R^1\text{B(OR)}_3\) (Suzuki) and RC\(=\text{CCu}\) (Sonogashira).

Organocopper and organoboron compounds mediated coupling reactions are maybe placed in a superior position to access alkynes; the Sonogashira reaction for obvious reasons, and the Suzuki coupling, due to the great advances this reaction has experienced in the past years. For this reason, these two cross-coupling reactions will receive more attention in this section.

1) Cross-coupling reaction of aryl and alkenyl halides with terminal alkynes catalyzed by Pd(0)/ Cu(I) (Sonogashira coupling)\textsuperscript{17}.

The coupling reactions between aryl/alkenyl halides and terminal alkynes were reported independently in 1975 for the first time by three groups\textsuperscript{18}. The first two methods\textsuperscript{18a,b} can be seen as an extension of the Heck reaction, applied to terminal alkynes. The third one\textsuperscript{18c}, on another hand, can be seen as an application of palladium catalysts to the reaction of Stephens-Castro\textsuperscript{19}. Disconsidering these historical details, one usually refers to this reaction as it is most known, the Sonogashira reaction.

Terminal alkynes cross-couple with aryl and alkenyl halides in the presence of catalytic amounts of palladium and copper(I) iodide and a secondary or tertiary amine. The reaction conditions are mild and not very sensitive to moisture, precluding the necessity of distilled solvents and reagents. The most common employed catalysts are Pd(PPh\textsubscript{3})\textsubscript{4} and PdCl\textsubscript{2}(PPh\textsubscript{3})\textsubscript{2} which are quickly transformed into Pd(PPh\textsubscript{3})\textsubscript{2} and [Pd(PPh\textsubscript{3})\textsubscript{2}X]\textsuperscript{1-}, respectively, being the active species in the catalytic cycle. In many cases, Pd(OAc)\textsubscript{2} or (CH\textsubscript{3}CN)\textsubscript{2}PdCl\textsubscript{2} and two equivalents of a tertiary phosphine, L, and a terminal alkyne are employed to reduce the Pd(II) complexes in situ to the catalytic active species [Pd\textsuperscript{0}].

Today, it is generally accepted that alkynyl copper derivatives are formed and transmetallated to RPdX species derived from the oxidative addition of the palladium active catalyst to the halide (scheme 1.15).


Scheme 1.15: The generally accepted catalytic cycle of the Sonogashira cross-coupling reaction.

Some examples of alkynes are shown below (scheme 1.16).

\[
\begin{align*}
&\text{RX-amine} & \text{HX-amine} \\
&\begin{array}{c}
R^2 \equiv \text{Cu} \\
\text{CuX}
\end{array} & \begin{array}{c}
\begin{array}{c}
\text{transmetallation} \\
\text{oxidative addition}
\end{array} \\
\text{reductive elimination}
\end{array}
\end{align*}
\]

\[
\begin{align*}
\text{[Pd}^0\text{]} & \xrightarrow{\text{oxidative addition}} \text{[Pd}^\text{II}\text{]}X \\
& \xrightarrow{\text{transmetallation}} \text{R}^1[\text{Pd}^\text{II}]X \\
& \xrightarrow{\text{reductive elimination}} \text{R}^1\text{X}
\end{align*}
\]

\[
\begin{align*}
\text{R}^1[\text{Pd}^\text{II}] & \text{CC} \equiv \text{CR}^2 \xrightarrow{\text{[Pd}^0\text{]}} \text{R}^1\text{X} \\
& \text{[Pd}^0\text{]} \xrightarrow{\text{reductive elimination}} \text{R}^1 \equiv \text{R}^2
\end{align*}
\]

Scheme 1.16: Examples of application of the Sonogashira cross-coupling reaction.

Modifications of the original method were developed. Some examples are the use of Pd/C as a catalyst\textsuperscript{20}, the use of water as reaction solvent\textsuperscript{21} and even catalyst systems which do not need the use of copper(I) iodide\textsuperscript{22}.

2) Cross-coupling of an aryl or alkenyl halide with boronic acid, borate esters or trifluoroborates catalyzed by Pd(0) (Suzuki coupling)\textsuperscript{23}


The Suzuki cross-coupling reaction or Suzuki-Miyaura, as it is also known, consists of a cross-coupling reaction between an aryl or alkenyl halide (or triflate) and a boronic acid, boronate ester or trifluoroborates catalyzed by palladium in the presence of a base (some examples are aqueous solutions of NaOH, K$_3$PO$_4$, Na$_2$CO$_3$, or KF, CsF, etc.). An excellent compatibility with numerous functional groups is one of the most remarkable features of this transformation (scheme 1.17).

![Scheme 1.17: A typical catalytical cycle of a Suzuki cross coupling reaction (extracted from ref. 23c).](image)

The transmetallation between the Pd center and the organoboron compound (BL$_3$) does not proceed readily. Despite this fact, the nucleophilicity of the organoboron derivative can be enhanced due to the _in situ_ generation of an “ate complex” (BL$_4$) from the reaction with a base.

The reactivity of the X groups employed decrease in the order: I > OTf > Br >> Cl. Aryl and 1-alkenyl halides activated by the proximity of an electron-withdrawing group are more

---


reactive towards the oxidative addition, than those compounds bearing an electron donating groups. This allows the use of chlorides in this coupling reaction.

Numerous catalysts can be employed, but the catalyst Pd(PPh₃)₄ is probably the most common. PdCl₂(PPh₃)₂ and Pd(OAc)₂ + PPh₃ or other phosphine ligands are also efficient, since they are stable to air and easily reduced to the active Pd(0) species.

The extension of the Suzuki reaction to alkynes was developed from the observation that B-methoxy-9-BBN and a organometallic polar reagent of type RM (with R = alkynyl, TMSCH₂ or Me), allowed access to borate complexes having this R group which could be further transferred to functionalized aryl or alkenyl halides, thus furnishing compounds which were not envisaged before through the traditional Suzuki coupling reaction.

Scheme 1.18: Two complementary ways of preparing a borate complex to be employed in the Suzuki coupling reaction R = alkynyl, TMSCH₂ or Me. M = K, Na, Li.

The use of stoichiometric amounts of B-methoxy-9-BBN is not even necessary, because the partial regeneration of the borate complex after the transmetallation step is possible in the presence of the organometallic species R²—M(scheme 1.19).

Scheme 1.19: Suzuki cross-coupling using sub-stoichiometric quantities of B-methoxy-9-BBN.

Other reactions using borate esters were later developed for examples using 
B(O\textsubscript{iPr})\textsubscript{3}\textsuperscript{26} and B(OMe)\textsubscript{3}\textsuperscript{27} (scheme 1.20).

\[
\begin{align*}
\text{R} & \quad \text{Li} \quad \text{B(O\textsubscript{iPr})}_3 \quad \text{DME} \quad \text{Li}^+ \quad \text{Pd(PPh}_3\text{)}_4 \\
\text{ArBr} \quad \text{DME/THF} \quad \text{reflux} \\
\text{R} & \quad \text{Ar}
\end{align*}
\]

Scheme 1.20: Examples using B(O\textsubscript{iPr})\textsubscript{3} in the Suzuki cross-coupling reactions to synthesize alkynes.

**1.1.2.6 The Corey-Fuchs reaction**

The Corey-Fuchs\textsuperscript{28} reaction consists of the homologation of aldehydes to either terminal alkynes, if the lithiated intermediate is quenched with a proton source or to an internal alkyne, if a suitable electrophile is added. This is a two-steps sequence, where a vinyldibromo compound can be isolated as a first intermediate upon treatment of the aldehyde with triphenyl phosphine and carbon tetrabromide. The second step uses two equivalents of \textit{n}-BuLi, the first equivalent deprotonates the vinylhydrogen and force the elimination of \textit{n}-BuH and LiBr, thus generating a bromoalkyne. The second equivalent of \textit{n}-BuLi makes the exchange between bromo and lithium (scheme 1.21).

\[
\begin{align*}
\text{R} & \quad \text{Ph}_3\text{P}/\text{CBr}_4 \quad \text{DCM, rt} \\
\text{Br} & \quad \text{Br} \quad \text{2 \textit{n}-BuLi} \quad \text{THF, -78°C} \\
\text{R} & \quad \text{H}^+ \\
\text{R} & \quad \text{E}^+ \\
\text{via} & \\
\end{align*}
\]

Scheme 1.21: The Corey-Fuchs reaction.


One problem of this method is the use of the strong base $n$-BuLi, which is going to limit the application of this route to more sensitive substrates.

1.1.2.7 Homologation of Seyferth-Gilbert and variations of Bestmann and Ohira

Another possible route to access alkynes is by using the Seyferth-Gilbert reagent, the diazocompound 1.5. This method consists in the conversion of aldehydes and ketones into the corresponding alkyne, which will have its carbon chain increased by one carbon. This method can be seen as a response to the strong basic conditions employed in the Corey-Fuchs reaction: the milder conditions used here will grant a wider generality of application. The mechanism of this reaction is believed to follow a Horner-Wadsworth-Emmons olefination and then a Fritsch-Buttenberg-Wiechell rearrangement of the alkylidene carbene formed (scheme 1.22):

![Scheme 1.22: The homologation of aldehydes to alkynes using the Seyferth-Gilbert reagent.](image)

Although the Seyferth-Gilbert reagent has become popular in synthesis\textsuperscript{31} and that the one-carbon homologation works smoothly, the multi-step procedure to synthesize this reagent is laborious\textsuperscript{29a}. In an effort to circumvent this, Bestmann\textsuperscript{32} exploited a first example reported by Ohira\textsuperscript{33}, developing an efficient way of generating the Seyferth-Gilbert reagent \textit{in situ} from the cleavage of a β-keto group present in 1-diazo-2-oxopropylphosphonate (scheme 1.23).

\begin{equation}
\text{RCHO} + \begin{array}{c}
\text{POMe} \\
\text{OMe}
\end{array} \xrightarrow{K_2CO_3, \text{MeOH}, \text{rt}} \text{R} \xrightarrow{72-97 \%}
\end{equation}

\textbf{Scheme 1.23: Examples of alkynes synthesized using the Bestmann-Ohira modification.}

From the \(R\) groups showed in scheme 1.23, one can see that the chiral center in the \(\alpha\)-position to the carbonyl is not racemized during the reaction. The significant advantage of this reagent is that it can be prepared from the commercial products dimethyl-2-oxopropylphosphonate and the \(p\)-toluenosulfonylazide\textsuperscript{34} in one step (not shown). A one-pot route for the same alkynes showed previously was also tested and produced the same products as the two-step sequence, with only a slight decrease in yields, 65-89%.

\section*{1.2 ISOXAZOLONES: general properties, synthesis and reactivity}

Although, today, excellent methodologies are available in the literature to access alkynes, the continuing saga for new methodologies is a constant worth-while endeavor. In this regard, we will present now our efforts to assemble such molecules based on previous


\textsuperscript{32} Müller, S.; Liepold B.; Roth, G. J.; \textit{Synlett}, \textbf{1996}, 521.


\textsuperscript{34} Roth, G. J.; Liepold, B.; Müller, S. G.; Bestmann, H. J.; \textit{Synthesis}, \textbf{2004}, 1, 59.
work in our laboratory using isoxazolones. This presentation will be preceded by a brief introduction of this heterocycle and some of the chemistry that can be performed from him.

1.2.1 General properties of isoxazolones

Isoxazolones do not constitute a new class of heterocyclic compounds. Their reactivity and tautomeric properties have been investigated since the beginning of the 20th century35 and employed in many studies concerning the synthesis of other heterocycles, such as 1,3-oxazin-6-ones36, pyrroles37, imidazoles38, tetrahydropyridines39 and pyridopyrimidines40. Additionally, human41 and anti-fungal42 activities have also been identified in isoxazolone derivatives. Two such compounds are presented below (Figure 1.1).

![Figure 1.1: Examples of an isoxazolone based inhibitor of a tumor necrosis factor-alpha (TNF-α) 1.6, and an arylhydrazonoisoxazolone derivative employed as fungicide, Drazoxolon 1.7.](image-url)
Isoxazolones can exist in three possible tautomeric forms, often called the OH form 1.8, CH form 1.9, and the NH form 1.10. The population of each tautomer in solution will depend on the nature of the substituents $R^1$, $R^2$ and the solvent they are dissolved in. Simple 3- and/or 4-substituted 5-isoxazolones tend to exist in the CH form in solvents of low polarity, such as CDCl$_3$. Depending on the substituents, the NH form might become more or less pronounced in solvents capable to supporting hydrogen bonds, such as (CD$_3$)$_2$SO. In strongly basic media (pyridine, piperidine and aqueous NaOH) the OH form and the ionized O$^-$ form also enter the equilibrium (scheme 1.24)$^{37}$.

Scheme 1.24: Three tautomeric forms are possible in isoxazolone compounds.

### 1.2.2 Isoxazolone synthesis

Although the main method for the synthesis of isoxazolones is the condensation of hydroxylamine with $\beta$-ketoesters$^{43}$ (scheme 1.25a), other procedures less often exploited, such as the [3+2]-cycloaddition of nitrile oxides with esters$^{44}$ (scheme 1.25b) and the gold catalyzed rearrangement of O-propiolyl oxymes$^{45}$ (scheme 1.25c) exist.

---


Scheme 1.25: Examples of possible routes to isoxazolones derivatives. Method exhibited in a) is the most often reported in the literature.

Scheme 1.25a shows the condensation of hydroxylamine with the β-ketoester, which occurs in general on the more electrophilic carbonyl group from the ketone\(^{46}\). Nevertheless, contamination with 3-isoxazolones coming from the cyclization of a hydroxamic acid formed from the attack of hydroxylamine to the ester group of the β-ketoester can also be present. It is even possible, under controlled pH (around 10), to revert the selectivity for these compounds, instead of 5-isoxazolones\(^{47}\) (not shown).

Scheme 1.25b shows the use of nitrile oxides, which can be synthesized \textit{in situ} typically by two well-established methods\(^{48}\). The first one derives from the condensation of aldehydes with hydroxylamines, followed by reaction of NaOX or NXS (X = Cl, Br, I), generating thus an hydroximinoyl halide, which upon treatment with a base, is capable of furnishing the nitrile oxide \textit{in situ}\(^{49}\). The second method is the Mukaiyama procedure, which derives from the reaction of a nitro alkane, phenyl isocyanate, triethylamine and DCC (\(N,N'\)-dicyclohexylcarbodiimide) or similar dehydration agent\(^{50}\) (not shown).


\(^{48}\) For an example of a third procedure recently reported in the literature using O-silylated hydroxamic acids, triflic anhydride and Et₃N, see: Carreira, E. M.; Bode, J. W.; Muri, D.; \textit{Organ. Lett.}, \textbf{2000}, 2, 539.


Scheme 1.25c concerns a recent method based on the gold catalyzed cycloisomerization of O-propioloyl oximes. Although not as practical as the condensation of hydroxylamine to β-ketoesters, it is mentioned here as an anecdotal reference to gold catalyzed processes that will be more deeply discussed in the next chapters of this thesis.

1.2.3 Isoxazolone reactivity

Great richness of the chemistry demonstrated by isoxazolones can be attributed to the fact the α-hydrogens to the carbonyl group in position 4 have a pKa in the range of 4-6\(^{51}\). This important feature allows isoxazolones to participate as nucleophiles in many transformations, such as Michael additions\(^{52}\), chlorinations\(^{53}\), condensations to aldehydes in Knoevenagel-type reactions\(^{54}\) and Mannich reactions\(^{55}\) (Scheme 1.26).

From the reactions listed above, Knoevenagel-type condensations with ketones and aldehydes are of particular interest, because the alkylidene isoxazolones formed serve as a Michael acceptor to a plethora of nucleophiles. Examples of such nucleophiles are hydrides\(^{56}\), Grignard reagents\(^{57}\), organozinc compounds\(^{57c}\), cyanides\(^{54}\), phosphonates\(^{58}\), aromatic compounds\(^{57a,b}\) and isoxazolones\(^{59}\) (Scheme 1.27).

---

\(^{51}\) See ref. 43a.


Scheme 1.27: Scope of possible nucleophiles that can be added to alkylidene isoxazolones.

Hydride addition to alkylidene isoxazolones is of special importance, because of the lack of selectivity between C- and N- alkylation. For instance, allylation of 3-phenyl-4-methyl-5-isoxazolone in the presence of sodium ethoxide affords a 2:1 mixture of NH-alkylated: CH-alkylated products (Scheme 1.28). Interestingly enough, in some cases, under high temperature, it is possible to force allyl chains to migrate following a 3,3-sigmatropic shift in favor of the CH-alkylated product\(^{60}\) (scheme 1.28). Nevertheless, as a general trend, direct alkylation on the 4-position of 5-isoxazolones remains a difficult and unsolved problem.

Scheme 1.28: Example of the difficulty of direct selective alkylation of 5-isoxazolones on position 4.

It is worth mentioning that alkylation of such compounds can be beautifully observed in light of the Curtin–Hammet principle\textsuperscript{61}, which states that for a reaction which has a pair of reactive intermediates or reactants that interconvert rapidly, each going irreversibly to a different product, the product ratio will depend only on the difference in the free energy of the transition state (i. e. the reaction rates) going to each product and not on the equilibrium constant between the intermediates (scheme 1.29).

Scheme 1.29: Alkylation of isoxazolones can be seen as a perfect example falling in the Curtin-Hammet principle scenario.

1.2.4 Isoxazolones as masked alkynes: Literature background

1.2.4.1 Flash Vacuum Pyrolysis

Thermal decomposition of alkylidene Isoxazolones under flash vacuum pyrolysis (FVP) conditions have been previously described to furnish acetylene derivatives, which decompose to nitriles, CO\textsubscript{2} and vinylidenes\textsuperscript{62}. Although excellent yields can be attained for the synthesis of simple acetylenes and this approach being also applicable to the synthesis of aminoacetylenes (Z = R\textsuperscript{2}NHCH or R\textsuperscript{2}R\textsuperscript{3}NCH)\textsuperscript{63}, isocyanides (Z = R\textsuperscript{2}N)\textsuperscript{64}, isocyanoamines (Z =


RNHN)\textsuperscript{65} and organic fulminates (Z = RON)\textsuperscript{66}, this procedure is clearly inappropriate for more sensitive substrates, due to the harsh conditions (scheme 1.30).

![Scheme 1.30](image)

Scheme 1.30: a) An example of a simple alkyne synthesis through flash vacuum pyrolysis using alkylidene isoxazolones. b) Application of the principle of decomposition of isoxazolones through FVP is also possible to similar substrates, giving rise for example to aminoacetylenes (Z = R\textsuperscript{2}R\textsuperscript{3}NCH), isocyanides (Z = R\textsuperscript{2}N), isocyanoamines (Z = R\textsuperscript{2}NHN) and organic fulminates (Z = RON).

\textbf{1.2.4.2 Nitrosative Cleavage}

Based on the earlier work of Abidi\textsuperscript{67}, our laboratory identified the evolution of isoxazolones as a possible entry to the late intermediate of this alkynylation process\textsuperscript{68}. Optimization of the reaction conditions implicated the use of sodium nitrite, ferrous sulfate, acetic acid and water\textsuperscript{69}. Sodium nitrite in aqueous acetic acid hydrolyzes to produce NO\textsuperscript{+}, which can react either on the CH- or NH-isoxazolone tautomers, 1.9 or 1.10, respectively, producing C-nitroso 1.11 or N-nitroso 1.12 adducts, respectively. Under reaction conditions,
these intermediates can undergo homolytic cleavage forming the persistent radical nitric oxide 1.13 and the resonance stabilized tertiary radical 1.14. Nitric oxide is a gas and eventually escapes the reaction medium, leaving no other choice to radicals 1.14 but to dimerize (1.15). In order to suppress this competitive pathway, ferrous sulfate is added to the reaction mixture, because it turns out that FeSO₄ in the presence of sodium nitrite and acetic acid react to produce nitric oxide 1.13 in situ⁶⁹b (Scheme 1.31).

Because this is a reaction with radical intermediates, it goes without saying about the importance of thoroughly degassing the solutions beforehand. Otherwise, oxidation of nitric oxide 1.13 to nitrogen dioxide radical 1.16 by remaining oxygen can afford by-products, such as 1.17. In the same way, the N-nitro compound 1.18 is also expected to form, but because the N-N bond is reasonably weak (dissociation energy of ca. 60 kcal.mol⁻¹, vs 105 kcal.mol⁻¹ for O-H, 108 kcal.mol⁻¹ for N-H, 105 kcal.mol⁻¹ for C-H bonds)⁷⁰, the reaction is expected to be reversible, due to the ease of the homolytic cleavage⁷¹ (this consideration is also valid for compound 1.12).

Formation of the C-nitroso compound 1.11 should also be anticipated to be present in the reaction medium, albeit isolation of such a compound or derivatives thereof were never confirmed experimentally. This is probably because the formation of tertiary nitroso compounds is expected to be reversible⁷².

---

Scheme 1.31: Mechanism proposed for the nitrosative cleavage of isoxazolones with the formation of possible by-products avoided
From what we discussed above, one can see all isoxazolone derivatives shown in schemes 1.26 and 1.27 as potential precursors to alkynes. In light of these findings, further studies were carried in our laboratory employing different strategies and nucleophiles to alkylidene isoxazolones. Some examples extracted from previous reports\(^73,74\) can be seen below (scheme 1.32):

Scheme 1.32: Examples of alkynes synthesized in previous works performed in our laboratory.

### 1.3 RESULTS AND DISCUSSION

#### 1.3.1 The Initial idea

During the work of this thesis, we were interested in the use of isocyanides 1.20 as nucleophiles to alkylidene isoxazolones 1.19, allowing in this way, upon hydrolysis of the expected intermediate 1.21, the access to isoxazolones bearing an amide moiety 1.22.

\(^73\) (a) See ref. 53 (b) See ref. 54 (c) See ref. 57c  
Further nitrosative cleavage of the isoxazolone ring would then afford the desired 3-alkynylamide 1.23 (scheme 1.33).

\[
\begin{align*}
\text{R}^1, \text{R}^2, \text{R}^3, \text{R}^4 & \quad \text{R}^4 \text{NC} \quad 1.20 \\
\text{N-O} & \quad \text{N-O} \\
1.19 & \quad 1.20 \\
\text{H}_2\text{O} & \quad \text{H}_2\text{O} \\
1.21 & \quad \text{via: FeSO}_4, \text{NaNO}_2, \text{AcOH, H}_2\text{O, rt} \\
1.22 & \quad \text{1.19a-j, 48-97\% (see table 1.1 for substituents)} \\
\text{N-O} & \quad \text{N-O} \\
\text{R}^1, \text{R}^2, \text{R}^3, \text{R}^4 & \quad \text{R}^1, \text{R}^2, \text{R}^3, \text{R}^4 \\
1.23 & \quad 1.24 \text{a-d} \\
1.24a, \text{R}^1 = \text{Ph} & \quad 1.24c, \text{R}^1 = \text{CH}_2\text{OBn} \\
1.24b, \text{R}^1 = \text{CH}_2\text{CO}_2\text{Me} & \quad 1.24d, \text{R}^1 = \quad \text{H}_2\text{O, 50 or 83°C} \\
\end{align*}
\]

Scheme 1.33: Envisioned synthesis of 3-alkynylamides.

The synthesis of alkylidene isoxazolones 1.19 was accomplished through a Knoevenagel condensation of 5-isoxazolones 1.24 with ketones and aldehydes (scheme 1.34). Different aldehydes, such as cyclopropyl carboxaldehyde, thiophene 2-carboxaldehyde and p-tolualdehyde; and ketones, such as cyclopentanone and 2-propanone could be condensed to isoxazolones bearing a phenyl ring, an ester group, a benzylated alcohol and a homoallyl chain.

\[
\begin{align*}
\text{N-O} & \quad \text{N-O} \\
\text{R}^1, \text{R}^2, \text{R}^3 & \quad \text{R}^1, \text{R}^2, \text{R}^3 \\
1.19a-j, \text{48-97\% (see table 1.1 for substituents)} & \quad \text{via: FeSO}_4, \text{NaNO}_2, \text{AcOH, H}_2\text{O, rt} \\
\end{align*}
\]

Scheme 1.34: The Knoevenagel condensation of isoxazolones and aldehydes or ketones.

As for the isocyanides, commercially available tert-butyl isocyanide 1.20a gave generally the best results. The addition of freshly prepared isocyanooacetate 1.20b and
benzyl isocyanide 1.20c tended to afford significant lower yields. Isocyanides were synthesized in a two steps sequence\(^7\) (scheme 1.35):

![Scheme 1.35](image)

Scheme 1.35: Two-steps sequence employed for the synthesis of isocyanides. The reaction mechanism is believed to pass by an iminium-chloride intermediate typical for other dehydration reactions using POCl\(_3\), such as the Vilsmeier-Haack reaction.

All the intermediates alkylidene isoxazolones 1.19a-j, amides 1.22a-n and \(\alpha\)-branched alkynes 1.23a-n synthesized in this work are indicated in table 1.1.

<table>
<thead>
<tr>
<th>isoxazolone 1.19</th>
<th>isocyanide 1.20</th>
<th>amide 1.22</th>
<th>alkynylamide 1.23</th>
</tr>
</thead>
<tbody>
<tr>
<td><img src="image" alt="Image" /></td>
<td><img src="image" alt="Image" /></td>
<td><img src="image" alt="Image" /></td>
<td><img src="image" alt="Image" /></td>
</tr>
<tr>
<td>1.19a 85%</td>
<td>1.20a</td>
<td>1.22a 100%</td>
<td>1.23a 92%</td>
</tr>
<tr>
<td><img src="image" alt="Image" /></td>
<td><img src="image" alt="Image" /></td>
<td>1.22a</td>
<td>1.23b R = 'Bu-, 87%</td>
</tr>
<tr>
<td>1.20a</td>
<td>1.20b</td>
<td>1.22b R = 'Bu-, 58%</td>
<td>1.23c R = EtO,C(_2)H, 81%</td>
</tr>
<tr>
<td>1.20b</td>
<td>1.20c</td>
<td>1.22c R = EtO,C(_2)H, 38%</td>
<td>1.23d R = PhCH(_2), 90%</td>
</tr>
<tr>
<td>1.19b 97%</td>
<td>1.20c</td>
<td>1.23c R = EtO,C(_2)H, 81%</td>
<td></td>
</tr>
<tr>
<td>1.19c 64%</td>
<td>1.20a</td>
<td>1.23d R = PhCH(_2), 90%</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>isoxazolone 1.19</th>
<th>isocyanide 1.20</th>
<th>amide 1.22</th>
<th>alkynylamide 1.23</th>
</tr>
</thead>
<tbody>
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<td><img src="image2.png" alt="isocyanide 1.20" /></td>
<td><img src="image3.png" alt="amide 1.22" /></td>
<td><img src="image4.png" alt="alkynylamide 1.23" /></td>
</tr>
<tr>
<td>1.19d 56%</td>
<td>1.20a</td>
<td>1.22f 58%</td>
<td>1.23f 76%</td>
</tr>
<tr>
<td><img src="image1.png" alt="isoxazolone 1.19" /></td>
<td><img src="image2.png" alt="isocyanide 1.20" /></td>
<td><img src="image3.png" alt="amide 1.22" /></td>
<td><img src="image4.png" alt="alkynylamide 1.23" /></td>
</tr>
<tr>
<td>1.19e 52%</td>
<td>1.20a</td>
<td>1.22g 51%</td>
<td>1.23g 71%</td>
</tr>
<tr>
<td><img src="image1.png" alt="isoxazolone 1.19" /></td>
<td><img src="image2.png" alt="isocyanide 1.20" /></td>
<td><img src="image3.png" alt="amide 1.22" /></td>
<td><img src="image4.png" alt="alkynylamide 1.23" /></td>
</tr>
<tr>
<td>1.19f 48%</td>
<td>1.20a</td>
<td>1.22h 100%</td>
<td>1.23h 9%</td>
</tr>
<tr>
<td><img src="image1.png" alt="isoxazolone 1.19" /></td>
<td><img src="image2.png" alt="isocyanide 1.20" /></td>
<td><img src="image3.png" alt="amide 1.22" /></td>
<td><img src="image4.png" alt="alkynylamide 1.23" /></td>
</tr>
<tr>
<td>1.19g 72%</td>
<td>1.20a</td>
<td>1.22i 56%</td>
<td>1.23i 15%</td>
</tr>
<tr>
<td><img src="image1.png" alt="isoxazolone 1.19" /></td>
<td><img src="image2.png" alt="isocyanide 1.20" /></td>
<td><img src="image3.png" alt="amide 1.22" /></td>
<td><img src="image4.png" alt="alkynylamide 1.23" /></td>
</tr>
<tr>
<td>1.19h 77%</td>
<td>1.20a</td>
<td>1.22j 72%</td>
<td>1.23j 47%</td>
</tr>
<tr>
<td><img src="image1.png" alt="isoxazolone 1.19" /></td>
<td><img src="image2.png" alt="isocyanide 1.20" /></td>
<td><img src="image3.png" alt="amide 1.22" /></td>
<td><img src="image4.png" alt="alkynylamide 1.23" /></td>
</tr>
<tr>
<td>1.19i 70%</td>
<td>1.20a</td>
<td>1.22k R = tBu-, 89%</td>
<td>1.23k R = tBu-, 94%</td>
</tr>
<tr>
<td><img src="image1.png" alt="isoxazolone 1.19" /></td>
<td><img src="image2.png" alt="isocyanide 1.20" /></td>
<td><img src="image3.png" alt="amide 1.22" /></td>
<td><img src="image4.png" alt="alkynylamide 1.23" /></td>
</tr>
<tr>
<td>1.19j 61%</td>
<td>1.20a</td>
<td>1.22l R = PhCH2-, 30%</td>
<td>1.23l R = PhCH2-, 60%</td>
</tr>
</tbody>
</table>

Table 1.1: 3-Alkynylamides synthesized during this work
Concerning the alkynylation step, we anticipated the possible formation of \(N\)-nitrosoamides by the nitrosation of the amide group under reaction conditions\(^{76}\). Nevertheless, this was expected to be reversible\(^{77}\), and it proved to have no consequence on reaction yields. For all compounds tested, yields were generally good, except for compounds \textbf{1.23h} and \textbf{1.23i}. This is possibly caused by the isomerization of the alkyne moiety to the allene form, due to the increased acidity of \(\alpha\)-hydrogens to the carbonyl group. This ultimately generates a highly reactive Michael acceptor and creates the opportunity to diverse degradation pathways (scheme 1.36).

Scheme 1.36: Examples of possible degradation pathways for compounds \textbf{1.23h} and \textbf{1.23i}, hypothesized from a common allene precursor, which can ultimately even culminate in oligomerization products.

Also noteworthy is the fact that to all alkylidene isoxazolones derived from ketones \textbf{1.23f}, -\textit{g}, -\textit{i}, -\textit{m} and -\textit{n}, nitro compounds \textbf{1.17} were observed and isolated in some cases. This is a mechanistically intriguing observation, since the corresponding nitro products coming from the alkylidene isoxazolones derived from aldehydes were never observed. If


\(^{77}\) See ref. 72, page 59.
oxygen is still present at the medium to a minor extent, promoting the oxidation of nitrous oxide 1.13 to nitro radical 1.16, this would be expected to be approximately true to all other experiments as it would be associated to the inefficiency of the method chosen to degas the reaction solvent. Consequently, one can assume that the radical formed in the α-position to a quaternary carbon in intermediates of type 1.14 are more nucleophilic than tertiary ones (derived from aldehyde condensed isoxazolones) and the hindrance promoted by the α-quaternary carbon does not prevent the by-product formation of 1.17f, -g, -i, -m and -n (scheme 1.37).

Scheme 1.37: By-products isolated during the alkyne synthesis of 1.23f, -g and -m.

Once we have synthesized these alkynylamides 1.23a-n, we were interested in typical gold cyclization products78. It was with great disappointment, that we observed the inertness of these substrates to all the reaction conditions tried, with only starting material recovered (Scheme 1.38)79.

Scheme 1.38: Reaction conditions tried for gold catalyzed cyclo-isomerizations.

---

78 See chapter 2 of this thesis for references on gold chemistry.
Next, we turned our attention to the use of N-iodosuccinimide (NIS), whose strong electrophilicity has already been taken advantage of in many past opportunities. Under classic conditions, namely 1.23a with NIS (1.1 equiv.) and acetone at room temperature, we found the clean but incomplete conversion of substrate 1.23a to iodo-cyclized product 1.25a, with the reaction crude containing a 1:1 mixture of 1.23a:1.25a (isolated yield of 45% for 1.25a). In our next attempt, 2.2 equiv. of NIS were used and we were delighted to observe 1.25a as the only product formed and an isolated yield of 72%. Nevertheless, the $^1$H NMR signals of the aromatic ring from 1.25a were splitted, suggesting a chiral environment in the proximity of the phenyl ring, which was not in agreement with the initially expected cyclized product 1.26. The X-ray crystal structure of 1.25a revealed actually that a molecule of water had been added, undoubtedly coming from the wet acetone employed (scheme 1.39).

Scheme 1.39: a) NIS-promoted cyclization reaction with 3-alkynylamides. The obtained product 1.25a, and the original imagined cyclization product 1.26. b) X-ray crystal structure of compound 1.25a. c) $^1$H NMR splitted signals of the aromatic ring of compound 1.25a (the

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broad signals are due to the congested environment and probably the reason for the inaccurate integrations, which should be 1:3:1).

Next, we envisioned that other nucleophiles such as methanol and allyl alcohol could also be used. To compete with the water dissolved in the acetone, a large excess of alcohol was employed (10 equiv.). In cases where the utilization of a not readily available alcohol is envisaged, further optimization is necessary (such as changing the reaction solvent) but this was not attempted. Furthermore, other nucleophiles can also prove suitable, but were not investigated. The results are summarized below (table 1.2).

<table>
<thead>
<tr>
<th>alkynylamide</th>
<th>nucleophile</th>
<th>cyclized product</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.23a</td>
<td>MeOH</td>
<td><img src="image" alt="cyclized product" /></td>
</tr>
<tr>
<td>1.23a</td>
<td>allylOH</td>
<td><img src="image" alt="cyclized product" /></td>
</tr>
<tr>
<td>1.23d</td>
<td>H2O</td>
<td><img src="image" alt="cyclized product" /></td>
</tr>
<tr>
<td>1.23l</td>
<td>MeOH</td>
<td><img src="image" alt="cyclized product" /></td>
</tr>
</tbody>
</table>

Table 1.2: Synthesis of 1,5-dihydropyrrol-2-ones based on the NIS-cyclization reaction of 3-alkynylamides.

The proposed mechanism for this cyclization reaction involves alkyne activation through iodonium intermediate 1.27, followed by ring closure, which affords intermediate 1.26. But the reaction does not stop at this point. As already mentioned, 1 equiv. of NIS
furnished a 1:1 mixture of 1.23:1.25, which shows the greater nucleophilicity of 1.26 when compared to 1.23 toward NIS. Thus, the reaction goes further and a second iodonium intermediate, represented by the mesomeric form 1.28, is rapidly quenched by a nucleophile present in the medium, i.e. either by water or by an alcohol molecule, to after HI elimination, produce the final compound 1.25 (scheme 1.40).

![Scheme 1.40: Proposed mechanism for the NIS-promoted ring closure.](image)

Because NIS is such a strong electrophile, an alternative mechanism could also be imagined with the iodination of the nitrogen atom\(^81\) of 1.23, followed by nucleophilic attack from the alkyne moiety, which would also culminate in intermediate 1.26. But this seems unlikely since alkynes tend to behave as electrophiles and not nucleophiles (due to both relatively low LUMO and HOMO orbitals). Furthermore, formation of a dihalide molecule followed by HX (X = halogen) elimination such as the one suggested in 1.29, has already been described in the literature in a similar Pd-mediated cyclization event from 2,2-difluoropropargylic amides to give the corresponding fluoro hemiaminals\(^82\).

It is difficult to say at this point if this nitrogen promoted ring-closure is under kinetic or thermodynamic control (or both). Thermodynamic control would mean reversibility of the

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cyclization step. It is also interesting to note that iodolactonization of unsaturated amides often leads to lactones rather than lactams.\(^{83}\)

Furthermore, 5-hydroxypyrrol-2(5H)-ones of type 1.30 and their alkylidenepyrrolinones derivatives 1.31 exhibit a wide range of intriguing biological activities. Compound 1.30a for example was found in the urine of patients with acute intermittent porphiria and its presence has been linked to psychiatric disorders. Compound 1.30 (with \(R^5 = H,\) alkyl) can be dehydrated upon reaction with a base (e.g. EtONa) or acid (e.g. TFA) to afford 1.31. Such compounds, as 1.31a for example, are remarkably efficient as sun-screens, as their maximum UV absorbances are in the range of sun protecting compounds 290-346 nm (Figure 1.2).

A noteworthy aspect of this transformation is the chemoselectivity for the N-cyclization product, without any trace of O-cyclized compound. This selectivity can possibly be explained using the HSAB (hard-soft acid-base) theory. NH is a softer nucleophile which should react with the softer \(I^+\)-alkyne complex. Conversely, the oxygen from the amide carbonyl is a harder nucleophile and an alkyne moiety activated by a harder electrophile, such as a \(M^+\)-alkyne complex, should afford the O-cyclized compound.

Figure 1.2: Examples of biological active 5-hydroxypyrrol-2(5H)-ones 1.30 and their derivatives alkylidenepyrrolinones 1.31.

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In this regard, two very appealing examples from the literature are the I$_2$-mediated N-cyclization$^{86}$ and the Ag$^+$-mediated O-cyclization$^{87}$ of o-(1-alkynyl)benzamides (scheme 1.41).

![Chemical structures](image)

Scheme 1.41: Almost identical compounds are capable of producing chemoselective different E$^+$-cyclized products depending on the nature of the electrophilic compound employed to activate the alkyne.

Among an enlisting number of transition metal-catalyzed transformations amenable to vinyl iodide species, one example of a Heck reaction was carried out with molecule 1.25c, generating a third cycle, thus demonstrating the power of coupling reactions to create complexity (scheme 1.42).

![Chemical structures](image)

Scheme 1.42: The intramolecular Heck reaction of compound 1.25c affords the tricycle 1.32 corresponding to a 5-exo trig cyclization.


1.4 CONCLUSION

Alkylidene isoxazolones are commonly employed in Michael additions using multiple nucleophiles. In the work described in this chapter, we reported the first addition of isocyanides as such nucleophiles. The addition product is hydrolyzed to generate amide moieties. Nitrosative cleavage of the isoxazolones formed affords 3-alkynylamides, which can be cyclized in the presence of N-iodosuccinimide to give access to 1,5-dihydropyrrol-2-ones.

The methodology developed here not only represents a new way of synthesizing 3-alkynylamides, but also further extends the scope of possible nucleophiles to alkylidene isoxazolones, whereas others may still prove suitable.

1.5 PERSPECTIVES AND FURTHER DEVELOPMENTS

Further development of the chemistry of isoxazolones to access various substituted alkynes, whose majority are $\alpha$-branched ones, can be envisaged. For instance, the addition of aromatics to alkylidene isoxazolones has already been described$^{57a,b}$ but very few examples are reported. In addition, methods to propargylate arenes are also relatively scarce in the literature$^{88}$. Furthermore, the biological relevance and the rich chemistry of indoles$^{89}$ make these molecules important building blocks, whose new propargylation methods should constitute a worth-while endeavor (scheme 1.43).

\[ \text{P} = \text{H or protecting group} \]

Scheme 1.43: A possible route envisioned to construct propargylated-arenes.

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Some of the Lewis acids described in the conjugate addition to $\alpha$, $\beta$-enones and possibly applicable to Michael additions to alkylidene isoxazolones are $\text{BF}_3\cdot\text{OEt}_2$, $\text{Yb(OTf)}_3$, $\text{InBr}_3$, $\text{AuCl}_3^{89}$ and $\text{ScCl}_3^{90}$ or $\text{HOTf}$ (Brønsted acid)$^{91}$. The zinc dinuclear catalyst developed by Trost’s group presented in one of the introductory sections of this chapter (scheme 1.12b) can also be envisaged for an asymmetric version of conjugate addition$^{92}$.

Other interesting nucleophiles are silyl enol derivatives$^{93}$ and xanthates$^{94}$, which to the best of our knowledge, have not been previously reported in the literature, and could be of great synthetic value (Figure 1.3).

![Figure 1.3: Possible nucleophiles in conjugate addition to alkylidene isoxazolones: silyl enol ethers and xanthates.](image)

The formation of quaternary carbons upon conjugate additions also represents a great opportunity as a testing ground for asymmetric 1,4-additions. Due to the already well-established success of certain ligands and protocols to asymmetric conjugate additions, their usage can also be possibly envisioned to be effective in the asymmetric catalyzed addition of nucleophiles to alkylidene isoxazolones. Two examples of privileged ligands useful in conjugate additions (and other important organic reactions, see below) are bis(oxazoline) and Salen complexes (Figure 1.4)$^{95}$.

---


$^{92}$ The selectivity obtained for the pyrrol ring by Trost was the C-2 alkylation (and not C-3 which is generally the expected product): Trost, B. M.; Müller, C.; *J. Am. Chem. Soc.*, 2008, 130, 2438.


Figure 1.4: Examples of two possible competent ligands in the Michael addition to alkylidene isoxazolones, bis(oxazoline) and Salen complexes.

Azabis(oxazolines) ligands, a variant of the bis(oxazoline) ligands (see below), can be also potentially useful in Michael additions of indoles to alkylidene isoxazolones\(^\text{96}\).

Scheme 1.44: A potential route to enanto-enriched substituted 3-propargyl indole precursors.

Another option to a conjugate asymmetric addition is the use of diorganozinc reagents due to their compatibility with many functional groups. Only a few diorganozinc R\(_2\)Zn reagents are commercially available, being Et\(_2\)Zn by far the most used. Nevertheless, these reagents can be readily prepared by RI/Et\(_2\)Zn exchange or by hydroboration/transmetalation sequence (scheme 1.45)\(^\text{97}\).

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Scheme 1.45: A potential route to access α-branched chiral alkyne precursors.

Another possible extension of the alkyne synthesis developed in this chapter can be possibly envisaged to allenes. As previously seen in the introduction of this chapter, the Mannich reaction has already been reported with isoxazolones. One could possibly think that under the nitrosative conditions of the alkynylation step, further isomerization could take place, possibly mediated by the iron salts (Fe²⁺ and Fe³⁺) present in the reaction mixture or by an additive eventually added, in analogy to the Crabbé reaction and variations thereof. Hopefully, the nitrosation of the amine moiety in the case of propargylamines showed in scheme 1.46, would also activate them as leaving groups towards isomerization. Subsequent extension to the asymmetric synthesis of allenes would directly follow using proline derivatives in the place of the diisopropylamine (scheme 1.46).

Scheme 1.46: **a)** Use of isoxazolones as a possible entry to allenes. **b)** Other successful secondary amines employed in related processes with an asymmetric version being described using the amino-alcohol derived from proline\textsuperscript{100} can be envisioned as suitable amines to the Mannich reaction depicted in **a**).
CHAPTER 2: INTRODUCTION TO GOLD CATALYSIS

This chapter is a brief overview of the gold chemistry reported in the literature. No results are presented in this chapter.
CHAPTER 2: INTRODUCTION TO GOLD CATALYSIS

2.1 Gold chemistry

Gold chemistry has known a fulgurous development in the past thirty years. This can be easily observed if one considers the increasing number of publications of gold-catalyzed reactions (Figure 2.1)\textsuperscript{101}.

Figure 2.1: Number of publications in gold catalysis from the beginning of the 20\textsuperscript{th} century until May 2006. In a) the number of publications concerning gold catalyzed reactions in general, in b) the number of publications concerning only homogeneous gold catalyzed reactions. By comparing both pictures, one can see that the initial scenario of gold catalysis was largely dominated by heterogeneous processes. (Figure 2.1 was extracted from ref. 101)

In this section, some of the most representative features exhibited by the unique and powerful reactivity of gold catalysts are going to be discussed with a main focus on Au(I)-

catalyzed reactions. It is not our intention, by any means, to make a comprehensive
digression on the rich plethora of gold reactions already reported in the literature\(^\text{102}\), but
rather to present trends, properties and reactivity patterns, from where, we believe, most of
gold-catalyzed transformations can be derived from.

### 2.1.1 Gold catalysts

Great interest has been paid to gold complexes in the past years due to two main aspects: 1) gold strong \(\pi\)-Lewis acidity and 2) gold’s good potential to stabilize cationic reaction intermediates. The origin of these properties is now widely accepted to be relativistic effects (see paragraph 2.1.4).

Gold compounds exist in two oxidation states, \(\text{Au}^+\) and \(\text{Au}^{3+}\) (not considering \(\text{Au}(0)\) species). Complexes in both oxidation states are reported to be carbophilic, but \(\text{Au}(\text{III})\) exhibits a thermodynamic preference for heteroatoms complexation, rather than carbon-carbon multiple bonds. In theoretical grounds, this was shown by Straub\(^\text{103}\), who calculated the complexation preference of \(\text{AuCl}_3\) to a carbonyl oxygen to be superior by 5.5 kcal/ mol over a triple bond, and Yamamoto\(^\text{104}\), who compared the computed heat of formation of complexes derived from \(\text{AuCl}, \text{AuCl}_3\) and other representative Lewis acids with different electrophilic compounds (see table 2.1).

<table>
<thead>
<tr>
<th>Lewis Acid</th>
<th>BCl(_3)</th>
<th>MgCl(_2)</th>
<th>AlCl(_3)</th>
<th>CuCl</th>
<th>CuCl(_2)</th>
<th>AgCl</th>
<th>AuCl</th>
<th>AuCl(_3)</th>
<th>PtCl(_2)</th>
</tr>
</thead>
<tbody>
<tr>
<td><img src="image" alt="Chemical Structure 1" /></td>
<td><img src="image" alt="Chemical Structure 2" /></td>
<td>18.9</td>
<td>34.5</td>
<td>40.7</td>
<td>37.4</td>
<td>25.4</td>
<td>26.4</td>
<td><strong>33.1</strong></td>
<td><strong>35.9</strong></td>
</tr>
<tr>
<td><img src="image" alt="Chemical Structure 3" /></td>
<td><img src="image" alt="Chemical Structure 4" /></td>
<td>42.1</td>
<td>44.2</td>
<td>55.1</td>
<td>51.8</td>
<td>41.2</td>
<td>39.6</td>
<td>53.6</td>
<td>60.3</td>
</tr>
</tbody>
</table>


### Table 2.1: Computed heats of formation (B3LYP/SDD, kcal.mol\(^{-1}\))\(^a\) of substrates 1-8 with common Lewis acids (Table 2.1 was extracted from ref. 104).

<table>
<thead>
<tr>
<th>Substrate</th>
<th>3</th>
<th>4</th>
<th>5</th>
<th>6</th>
<th>7</th>
<th>8</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>0.9</td>
<td>15.2</td>
<td>19.1</td>
<td>33.1</td>
<td>14.3</td>
<td>22.6</td>
</tr>
<tr>
<td></td>
<td>0.4</td>
<td>15.7</td>
<td>19.2</td>
<td>33.6</td>
<td>18.1</td>
<td>24.4</td>
</tr>
<tr>
<td></td>
<td>17.2</td>
<td>33.1</td>
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<td>36.6</td>
<td>23.9</td>
<td>26.0</td>
</tr>
<tr>
<td></td>
<td>42.6</td>
<td>44.4</td>
<td>55.4</td>
<td>52.2</td>
<td>41.9</td>
<td>40.4</td>
</tr>
<tr>
<td></td>
<td>8.8</td>
<td>25.6</td>
<td>26.4</td>
<td>43.6</td>
<td>25.2</td>
<td>34.7</td>
</tr>
</tbody>
</table>

\(^a\) in the gas phase The heats of formation were calculated by substraction of the absolute energies of the starting compounds from the absolute energy of the optimized complex between the substrate and the Lewis acid

In practical grounds, possibly the most expressive example reported in the literature showing this dichotomy between Au(I) and Au(III) species was reported by Gevorgyan et al.\(^{105}\). Starting from the same bromo allenyketone and using the same reaction conditions, but changing the gold catalyst, either (PEt\(_3\))AuCl or AuCl\(_3\), two different regioisomers of bromofuran were obtained. The explanation for this regioselectivity was believed to be the different gold affinities, Au(I) for the carbon-carbon multiple bond, and Au(III) for the oxygen from the ketone moiety.

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Scheme 2.1: Different regioisomers are obtained in the cycloisomerization of bromo allenyl ketones, depending on the nature of the gold catalyst employed. This is also one of the rare examples in the literature where a Au(I) neutral complex exhibited catalytic activity.

2.1.2 Isolobal analogy of Au(I)

(R₃P)Au⁺ is isolobal to H⁺ and LHg²⁺ fragments. At first glance, the isolobal relationship between (R₃P)Au⁺, LHg²⁺ and H⁺ and parallels in their reactivities could suggest that such metal templates can be considered as expensive equivalents of a proton with increased carbophilic character. This is not true. It is common knowledge that addition reactions to olefins or alkynes with a Brønsted acid usually require harsh conditions and potentially have numerous side reactions of the carbocation intermediate formed (Wagner Meerwein rearrangements, elimination reactions, oligomerization, etc.).

Replacement of the proton by LHg²⁺ represents a typical solution to this aforementioned problem. LHg²⁺ is a large and polarizable cation, and as such a “soft” ion, LHg²⁺ has great affinity for the substrate and allows access to mild reaction conditions and high yields for the desired addition products. The concern of using mercury salts is their

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108 For a general review concerning activation of π-systems with different metals, including Hg²⁺, see: Freeman, F.; Chem. Rev.; 1975, 75, 439.
toxicities. Although many addition reactions using alkynes\textsuperscript{109} are catalytic in Hg\textsuperscript{2+}, the analogous reactions with alkenes are not, because the C(sp\textsuperscript{3})-Hg bond formed in the resulting intermediate is kinetically stable\textsuperscript{110}. Thus, addition reactions to alkenes generally require an extra step to release the mercury appendage (generally NaBH\textsubscript{4} + OH\textsuperscript{-}) and have the great inconvenient of being stoichiometric in toxic mercury salts.

On the other hand, (R\textsubscript{3}P)Au\textsuperscript{+} is essentially nontoxic, it has high affinity for carbon-carbon $\pi$-bonds and after each catalytic cycle, furnishes a kinetic labile C(sp\textsuperscript{3}/sp\textsuperscript{2})-Au bond that can be readily cleaved under the reaction conditions.

\subsection*{2.1.3 Au(I) geometry and enantioselective reactions}

Au(I) species predominantly adopt a linear, bicoordinate geometry\textsuperscript{111}, and because of the limited number of coordination sites, generally one must remove a ligand from the neutral bicoordinate Au(I) complex in order to allow the access of the substrate to the coordination sphere of the catalyst.

\begin{figure}[h]
\centering
\includegraphics[width=0.5\textwidth]{AuCl(PPh\textsubscript{3}).png}
\caption{A typical Au(I) neutral complex, AuCl(PPh\textsubscript{3}) and its X-ray crystal structure with $\alpha = 179.6^\circ$, $d_{\text{Au-P}} = 2.235 \text{ Å}$ and $d_{\text{Au-Cl}} = 2.279 \text{ Å}$. The considerably long distance between Au and Cl indicates a substantial weakening of this bond\textsuperscript{102f}. (Figure 2.2 was extracted from ref. 102f)}
\end{figure}


It is interesting to compare the previous structure of \( \text{AuCl}(\text{PPh}_3)_2 \) to the three-coordinate homologue \( \text{AuCl}(\text{PPh}_3)_2 \). \(^{112}\)

![Figure 2.3: X-ray structure of \( \text{AuCl}(\text{PPh}_3)_2 \) with \( \alpha = 115.1^\circ \), \( \beta = 108.1^\circ \), \( d_{\text{Au-P1}} = 2.230 \text{ Å} \), \( d_{\text{Au-P2}} = 2.313 \text{ Å} \), \( d_{\text{Au-Cl}} = 2.526 \text{ Å} \). (Figure 2.3 was extracted from ref. 102f)](image)

The greater distance observed in figure 2.3, with \( d_{\text{Au-P2}} = 2.313 \text{ Å} \) means that the third ligand \( \text{PPh}_3 \) is only weakly coordinated to the gold center. This evidence supports the apparent aversion of \( \text{Au}(I) \) complexes to adopt more than two ligands. This was also recognized in a theoretical study on the close-related structure \( \text{AuCl}(\text{PH}_3)_2 \), which is practically a linear species (\( \alpha = 178.2^\circ \)) and has the second phosphane bound only very loosely. \(^{113}\) Curiously enough, in this three-coordinate gold complex, the second phosphane directs its lone electron pair toward the \( \text{H} \) atoms of the first \( \text{PH}_3 \) and not to the gold center.

One important consequence probably derived from this linear geometry is the difficulty of \( \text{Au}(I) \) complexes to participate in enantioselective processes (because the chiral ancillary ligand is diametrically opposed to the substrate). To circumvent this problem, a clever solution was proposed by Toste et al. \(^{114}\) based on the principle of \textit{ion-pairing}. \(^{115}\)


alternative approach takes advantage of the fact that gold(I) catalysts are positively charged. Therefore, the use of a negative charged chiral counter-ion would be closer to the substrate and would possibly be more efficient in inducing enantioselective transformations or “help” an eventual chiral ligand already bound to gold (it is important to note that in this new scenario with a chiral ligand and a chiral counter anion, “match” and “mismatch” situations are expected to be present, see figure 2.4). This concept represented not only a landmark in gold asymmetric catalysis, but also a further general principle that could be potentially expanded to other metal-ions based catalysis such as Pd, Rh, Ru, Ir, etc..

Figure 2.4: Schematic drawing of the concept of enantioselective induction through ion-pairing developed by Toste et al. (Figure 2.4 was extracted from ref. 115c).

2.1.4 Relativistic effects

As mentioned earlier, in section 2.1.1, relativistic effects were recognized as being the responsible factors for gold properties and reactivity. Indeed, Toste et al.\textsuperscript{102e} and others before him\textsuperscript{102h,116} identified strong relativistic manifestations on gold as being the reason for the contraction of orbitals s and p\textsuperscript{117}. This contraction explains the increased ionization


\textsuperscript{117} This can be seen as a consequence of the theory of special relativity applied to the Bohr atomic model, where the Bohr radius is given by $r_n = \frac{n^2 \hbar^2}{2k_y e^2 m_e}$ and the velocity of an electron orbiting a nucleus of charge $q = Ze$ is given by $v_n = \frac{2k_y e^2}{n \hbar} = \frac{Zac}{n}$. The value $n$ is the principal quantum number, $\hbar = \frac{h}{2\pi} = 1.05\times10^{-34}$. $s$ is the
energy of Au when compared to other group 11 elements, Cu and Ag, or Pt (group 10). As a consequence, electrons occupying outer orbitals of type d and f are better shielded by those electrons of inner s and p orbitals, and they feel a less effective attraction to the positive nuclear charge. This results in an expansion of d and f orbitals. Even if these relativistic effects tend to be present in all heavier atoms of the periodic table\textsuperscript{118}, computational calculations show that they are more pronounced in elements which have their orbitals 5d and 4f fully occupied, and culminates with a maximum effect on gold (figure 2.5). Furthermore, it must be also mentioned that, even if all orbitals s, p, d and f exhibits to a certain extent the contraction and expansion effects mentioned above, in light of the frontier molecular orbital theory\textsuperscript{119}, greater attention must be paid to the highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO) of gold, i. e., the 5d and 6s orbitals, respectively\textsuperscript{120}.

\begin{equation}
\text{\textit{\textsuperscript{118}Pyykkö, P. & Desclaux, J. P.; Acc. Chem. Res., 1979, 12, 276.}}
\end{equation}

\begin{equation}
\end{equation}

\begin{equation}
\text{\textsuperscript{120}For a much deeper discussion of relativistic effects on periodic elements, see ref. 116.}
\end{equation}
Figure 2.5: Pronounced relativistic effects accounts for the contraction of the 6s (LUMO) orbital in gold complexes, as well for its closest neighbors in the 6th line of the periodic table, Hg and Pt. (Figure 2.5 was extracted from ref. 118).

The contraction of the 6s orbital (LUMO) is responsible for a stabilization of this orbital when compared to the same orbital without relativistic considerations and this ultimately accounts for a greater Lewis acidity of Au(I) cationic complexes. This result also correlates with the strong electronegativity of gold (2.4 for Au, compared to 1.9 of Ag and 1.9 for Cu), which can also be seen as a consequence of the contraction of the 6s and 6p orbitals\(^{102e}\). On the other hand, the expansion of the 5d orbital (HOMO) is responsible for a destabilization of this orbital when compared to the same orbital without relativistic considerations and this ultimately results in a greater potential of Au(I) to stabilize carbocationic intermediates (Figure 2.6).
Figure 2.6: Using the Hartree-Fock model, calculated molecular orbital energies for AgH and AuH are compared above taking relativistic effects into consideration (R) and neglecting these effects (NR). From this picture, a third effect can be noted due to relativistic effects: the splitting of energy levels due to spin-orbit coupling\textsuperscript{102e} (Figure 2.6 was extracted from ref. 118).

Additionally, the expansion of the 5\textit{d} orbital is also implicated in the poor nucleophilicity observed for gold metal species, because electrons in this orbital suffer a lower electronic repulsion from electrons in inner \textit{s} and \textit{p} orbitals. This makes electrons in the HOMO orbital of gold better hold to the nucleus and less available to nucleophilic additions, which is also mirrored in the inertness of Au(I) species to undergo oxidative additions\textsuperscript{121}. Computational and experimental studies also reveals that reductive elimination from R\textsubscript{3}LAu(III) complexes is equally unfavorable\textsuperscript{122}. These observations are consistent with the broad reactivity exhibited for Au(I) and Au(III) complexes, which are observed to not easily interchange between oxidation states\textsuperscript{123}.

### 2.1.5 Au(I)-catalyzed nucleophilic addition to C-C multiple bonds: elementary steps.

The general scheme of a gold catalyzed nucleophilic addition is represented in scheme 2.2. This scheme gives an overview of the two main pathways envisaged when planning a gold catalyzed reaction.

![Scheme 2.2](image)

Scheme 2.2: General plan when envisaging a gold catalyzed reaction, with two possibilities for trapping an electrophile, either in a “1,1-fashion” (path A) or in a “1,2-fashion” (path B).

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2.1.5.1 Nucleophilic addition to $\pi$-systems

The first step in scheme 2.2 consists of the activation of a $\pi$-system (alkene, alkyne or allene) by the important lowering of the LUMO orbital of the C-C multiple bond due to the contraction (stabilization) of the 6s orbital of gold. To this activated complex follows the nucleophilic addition, which is now widely accepted to proceed via a trans mechanism\textsuperscript{124}.

At this point, it is important to note that gold-alkyne complexes normally exhibit superior reactivity towards nucleophiles than gold-alkene complexes. This fact contrasts with studies\textsuperscript{125} realized on Au$^+$-ethylene and Au$^+$-ethyne bondings, which indicates ca. 10 kcal.mol$^{-1}$ greater stabilization for the ethylene complex over the ethyne complex. The reason for this apparent contradiction is the fact that an alkyne has intrinsically lower LUMO and HOMO (by ca. 12 kcal.mol$^{-1}$) than the corresponding alkene (therefore being less nucleophilic and more electrophilic). Therefore, even if a gold-alkene is more stabilized than the corresponding gold-alkyne system, gold-alkyne systems are overall lower in energy than gold alkene-systems. Another way of describing this energy situation is saying that gold catalysts do not distinguish between alkenes and alkynes, being able to form activated complexes with both entities. Conversely, nucleophiles do distinguish between activated complexes, and they give priority to gold-alkyne ones (which is probably a kinetic phenomenon in origin).

Independent of which $\pi$-system is bounded to the metal catalyst, “activation” involves slippage $\eta^2 \rightarrow \eta^1$ of the metal fragment along the axis of the $\pi$-ligand, which in turn, leads to enhanced electrophilicity of C-C multiple bond. Nucleophilic addition proceeds due to the relaxation of symmetry in bonding orbitals, which allows mixing of previously orthogonal orbitals and facilitates charge transfer from the nucleophile to the $\pi$-ligand, and finally to the metal center. It is possible that such distortion is already present in the ground state structure (and not only in the transition state).


Metal “slippage” has already been deeply studied through theoretical calculations\textsuperscript{126}, but just recently such $\eta^2 \rightarrow \eta^1$ displacement was experimentally investigated by the coordination of the gold complex (PPh\textsubscript{3})Au\textsuperscript{+} to alkene 2.1\textsuperscript{127}, which resulted in the complete regioselectivity for the end-on coordination of the $\pi$-system and the simultaneous formation of an imidazolium cation, as it was observed from the X-ray crystal structure of the resulting molecule, 2.2 (scheme 2.3). This reaction shows that significant charge density can be donated to a $\pi$-bond upon coordination to a suitable carbophilic metal fragment.

Scheme 2.3: Elementary steps of a gold(I)-catalyzed reaction of a $\pi$-system with a nucleophile (Scheme 2.3 was extracted from ref. 102a).

2.1.5.2 Path A: trapping gold intermediates with electrophiles in a “1,1-fashion” and proto-deauration.

Once the nucleophilic attack is done to the $\pi$-system, the new organo-gold obtained can evolve in contact with a suitable electrophile, either through path A, giving a 1,1-electrophilic trapped product or through path B, giving a 1,2-electrophilic trapped product (Scheme 2.2).

In the case of 1,1-trapping, the most common electrophiles are sources of halogens, such as Cl, Br, I or even F\textsuperscript{128}. Hashmi et al.\textsuperscript{128d} tried to extend the scope of suitable

electrophiles to enones, but no fruitful results were observed. Also worth mentioning are other strategies for trapping the organogold species using carbon\textsuperscript{129}, silicon\textsuperscript{130} and sulfonyl\textsuperscript{131} electrophiles or via palladium catalyzed cross coupling reactions\textsuperscript{132}. An impressive example is the reaction cascade leading to a final 1,1-electrophilic trapping by an iminium species, thus forming a new C-C bond (scheme 2.4).

Despite the successful cases mentioned above\textsuperscript{128-132}, it is not easy to trap organogold compounds. The difficulty in doing so arises from the great potential of these species to protodeaurate. Proto-deauration (or proto-demetallation as it is used when referring to a general metal) consists on the displacement of the gold metal center by a proton and is generally the key step for catalyst regeneration. Surprising as this seems, there is not a great body of experimental or theoretical information in the literature concerning the study of the

Scheme 2.4: An example of an impressive gold catalyzed cascade terminated by the trapping of the vinyl organogold intermediate leading to the formation of a C-C bond\textsuperscript{129a}.

\begin{center}
\includegraphics[width=\textwidth]{Scheme2.4.png}
\end{center}

\textbf{Scheme 2.4:} An example of an impressive gold catalyzed cascade terminated by the trapping of the vinyl organogold intermediate leading to the formation of a C-C bond\textsuperscript{129a}.

\begin{itemize}
\item \textsuperscript{128-132} For the use of Et\textsubscript{3}N.HF as trapping agent, see: (f) Akana, J. A.; Bhattacharyya, K. X.; Müller, P.; Sadighi, J. P.; \textit{J. Am. Chem. Soc.}, 2007, 129, 7736.
\item \textsuperscript{130} Nakamura, I.; Sato, T.; Terada, M.; Yamamoto, Y.; \textit{Org. Lett.}, 2007, 9, 4081.
\end{itemize}
relative basicities of organogold species or concerning the precise mechanism of the proto-deauration step, which is possibly due to the extremely fast nature of this process.

Very recently, Blum et al. reported studies concerning the basicity of organogold compounds to follow the order: \( \text{sp}^3 < \text{sp} < \text{sp}^2 \) as it was observed with complexes \( 2.5 < 2.6 < 2.9 < 2.10 \), but the pK\(_a\) values of the conjugate acids follow the order \( \text{sp} < \text{sp}^2 < \text{sp}^3 \) (i.e., acetylene pK\(_a\) = 25, ethylene pK\(_a\) = 44, methane pK\(_a\) = 48)\(^{133} \). This basicity exhibited by vinylgold, phenylgold, and alkynylgold in comparison to methylgold suggests a stabilizing interaction of the C-C \( \pi \)-bond in the rate-limiting transition state. In addition, it has been reported in the literature a correlation between increased hyperconjugation ability and increased rate of protodeauration for compounds of type \( R_3\text{MPh} \), with \( M = C, Si, Ge, Sn \) and \( Pb \)^{135}. Based on these observations, a transition state involving the overlap of the Au-C \( \sigma \)-bond with the carbon \( \pi \)-bond was proposed\(^{136} \) (Table 2.2, b). This hyperconjugation would further stabilize the developing cationic charge and generate an increased rate of protodeauration in \( 2.9 \) and \( 2.10 \) compared to methyl gold \( 2.5 \), which cannot be stabilized in the same way and to alkynylgold \( 2.6 \), which has reduced potential for hyperconjugative stabilization in the transition state. Furthermore, because the acetylene has a lower HOMO when compared to ethylene, alkynylgold \( 2.6 \) would be expected to have a lower energy \( \pi \)-system than alkenyl gold \( 2.10 \), and thus the \( \pi \)-system of alkynylgold \( 2.6 \) would be less accessible for protonation.

### a) Relative rates of protodemetalation of organogold compounds

<table>
<thead>
<tr>
<th>R(^1)AuPPh(_3) (3 equiv.)</th>
<th>PhAuPPh(_3) (3 equiv.)</th>
<th>HCl.Et(_2)O (1 equiv.)</th>
<th>CD(_2)Cl(_2), 23 °C</th>
<th>R(^1)-H</th>
<th>PhH</th>
<th>k(_{rel}^a)</th>
</tr>
</thead>
<tbody>
<tr>
<td>compound</td>
<td>R(^1)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>2.3</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>0.015(^b)</td>
<td></td>
</tr>
<tr>
<td>2.7</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>0.78</td>
</tr>
</tbody>
</table>

\(^{133}\) For a very recent report on this subject, see: Roth, K.; Blum, S. A.; *Organometallics*, 2010, 29, 1712. The pK\(_a\)'s extracted from this reference differs from the one mentioned in chapter 1 (p.10). This difference can be attributed to different functional groups present in the molecule.

\(^{134}\) Proto-deauration is faster than \( \beta \)-hydrogen elimination: Hashmi, A. S. K.; *Catalysis Today*, 2007, 122, 211.


**b) Proposed transition state for proto-deauration**

![Transition state model for proto deauration, showing partial contribution of the π-system in the protonation event](image)

Table 2.2: **a)** Comparison between reaction rates of selected organogold compounds. **b)** Proposed transition state for protodemetalation step concerned in vinyl and aryl gold species.

One last interesting important feature is the fact when comparing triphenylphosphine and the NHC ligand IPr, the latter showed an increase in the reaction rate by a factor 1.8 (compare compounds \(2.6\) and \(2.8\)), suggesting that NHC ligand can expand the substrate scope of gold catalyzed rearrangements which are terminated by proto-demetalation. Conversely, the slower protodemetalation step for phosphine ligands may be seen as useful, if one envisages submitting these intermediates in further reactions, such as cross-coupling transformations\(^{132}\).
2.1.5.3 Path B: trapping gold intermediates with electrophiles in a “1,2-fashion” and debate on the nature of the intermediate formed

The “push-pull” sequence represented in pathway B of scheme 2.2 plays a central role on the recognized power of gold catalyzed reactions because, once well planned, it allows one to form up to two C-C or C-X bonds, where X is an heteroatom (not considering more elaborated cascades reactions) in just one step and thus rapidly access great molecular complexity. A representative example of this “push-pull” sequence is shown below (Scheme 2.5):

Scheme 2.5: A representative example of the “push-pull” reactivity of gold systems using an acetylenic Schmidt reaction.

One **sine qua non** condition to such a push-pull sequence to happen is that the positive charge in the \( \alpha \)-position to the gold center is stabilized enough. The resulting intermediate can be described either as a cationic form or a carbenoid species. And here, is where an extensive debate about the nature of these intermediates begins.

Among many examples in the literature demonstrating the highly cationic nature of intermediates implicated in cycloisomerization processes, two of the most

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137 For an overview of this rich chemistry, see refs. 102a-g.
representative ones were possibly reported by the Fürstner’s group. In the first one, an enyne of type 2.11 bearing a substituent \( R = H \) was submitted to gold catalysis, there was no regioselectivity on the cyclization and the reaction proceeded with a poor yield (30%, with a 2:1 mixture of regioisomers). When a methyl group was introduced on the double bond (\( R = \text{Me} \)), the reaction efficiency and selectivity increased considerably (80% of just one regioisomer, scheme 2.6), a result that is difficult to explain using a gold carbene intermediate, such as 2.12, but which can be beautifully derived from a charge-delocalized and highly-ordered cationic intermediate (or transition state) such as 2.13.

![Scheme 2.6: Au(I)-catalyzed cycloisomerization cascade. Conditions: PPh\(_3\)PAuCl (5 mol%), AgSbF\(_6\) (5 mol%), CH\(_2\)Cl\(_2\), at room temperature. E = CO\(_2\)Me. The reaction is regioselective for \( R = \text{Me} \), but unselective for \( R = H \). This outcome can be rationalized using a more or less concerted process following the same logic of cationic polyene cyclization reactions (cf. “Stork-Eschenmoser postulate”\(^{141}\), Scheme 2.6 was extracted from ref. 140).](image-url)

A second representative finding, also from Fürstner’s group, was uncovered from a tentative experiment to observe a gold carbene complex, but resulted instead in the formation of a highly cationic form, the other mesomeric form of this intermediate (scheme 141 (a) Stork, G.; Burgstahler, A. W.; J. Am. Chem. Soc., 1955, 77, 5068. (b) Stadler, P. A.; Eschenmoser, A.; Shinz, H.; Stork, G.; Helv. Chim. Acta, 1957, 40, 2191.
This conclusion was based on the following NMR observations: two -OCH₂- groups of the ketal ring in structure 2.16 give rise to just one signal in both the ¹H and the ¹³C NMR spectra, suggesting there is rapid rotation about the C₂-C₃ bond on the NMR timescale, even at -78°C. Whereas this finding is incompatible with the putative gold carbene structure 2.17, it is consistent with the presence of an oxocarbenium cation of type 2.16 (Scheme 2.7). Moreover, the analysis of the pertinent coupling constants reveals the (Z)-olefin character of the C₁-C₂ bond, ²J_H₁-H₂ = 13.9 Hz. If the temperature is carefully raised, the solution gradually changes color from yellow to dark red. NMR spectroscopy reveals that (Z)-2.16 rearranges over the course of several hours to a new compound, featuring the distinctive spectroscopic coupling constant ²J_H₁-H₂ = 19.1 Hz of the corresponding E isomer (E)-2.18. Even in solutions in which both gold species are present, their NMR spectroscopic signals are not significantly broadened by mutual exchange, thus confirming the high barrier of interconversion.

Scheme 2.7: The attempted generation of a gold carbene by rearrangement of cyclopropene 2.15 produced only the corresponding cationic vinyl gold species (Scheme 2.7 was extracted from ref. 142).

As a consequence of these studies, Fürstner argued that even if the intervention of a carbene species cannot be excluded from the gold reactivity scenario, strong evidences have been cumulated to support the view of gold catalyzed processes as falling into the general

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categories of Friedel-Crafts reactions, Prins-type processes and cationic rearrangements, with any claim of the intervention of gold-carbenoid intermediates having to be carefully weighed against the cationic behavior highlighted above.

Conversely, further calculations performed by Toste’s group on cyclopropenes of type 2.15 and other compounds\(^{143}\), supported the idea that even if complex 2.18 (Au\(_0^{PPP}\)), studied by Fürstner was essentially cationic, other analogous vinyl gold complexes, without the cationic stabilizing oxygens, such as Au\(_{Me}^{PMe}\), were better described as carbene-like compounds (Au\(_{X}^{'L}\) refers to a notation where L is the ancillary ligand on gold and X is the substituent on C\(^3\), see Figure 2.7). In order to test this hypothesis experimentally, both complexes were submitted to cyclopropanation protocols. Any productive cyclopropanation of intermediates resulting from the gold-catalyzed reaction using cyclopropene 2.15 as a precursor to Au\(_{O}^{PMe}\) failed (Figure 2.7a). However, the gold-catalyzed reaction of cyclopropene 2.19 (which was assumed to decompose to an intermediate similar to Au\(_{Me}^{PMe}\)) with cis-stilbene (2.20) afforded the product of stereospecific olefin cyclopropanation (Figure 2.7b).

\[ \text{AuMe}^{PMe} \quad \text{AuMe}^{PMe} \quad \text{AuMe}^{PPh} \quad \text{AuMe}^{PPh} \quad \text{AuMe}^{NHC} \quad \text{AuMe}^{Cl} \]

Figure 2.7: **Experimental and theoretical comparison for the carbene reactivity of the substrate with different ancillary ligands.** a, b, Attempted (a) and observed (b) carbene-like reactivity, demonstrating the impact of the ancillary ligand on the yield of cyclopropanation product. c, Bond distances in AuMe complexes (Figure 2.7 was extracted from ref. 143).

The difference in the efficiency observed for the cyclopropanation reaction which strongly depends on the ancillary ligand used on gold can be explained in terms of their $\sigma$-donor and $\pi$-acceptor properties. Because Au(I) complexes have only one valence orbital ($6s$), it results from the Pauli exclusion principle\textsuperscript{143} that the stronger the $\sigma$-donor ligand on gold, the smaller will be the LAu-$C_1$ bond order (which means that this bond, Au-$C_1$, is elongated). This is coherent with the view of figure 2.7c, when comparing the Au-$C_1$ bond lengths between the different phosphine and phosphite ligands (strong $\sigma$-donors and strong $\pi$-acceptors) and AuMe. Additionally, the metal center is able to form two $\pi$-bonds by donation from perpendicular filled $d$-orbitals into empty $\pi$-acceptors on the ancillary ligand and $C_1$. Although these two bonds are not mutually exclusive, they compete for electron density from gold (Figure 2.8). As a consequence, strongly $\pi$-acidic ligands decrease back-donation to the substrate, resulting in even longer Au-$C_1$ bonds ($2.057$ Å for AuMe\textsuperscript{POMe}).

![Diagram](image)

Figure 2.8: The L-Au-$C_1$ bonding interactions can be partitioned into three components. The first one accounts for the $\sigma$-interactions ($6s$ orbital of gold, LUMO), and the other two, implicating two perpendicular and filled $5d$ orbitals (HOMO) of gold are responsible for the $\pi$-interations (Figure 2.8 was extracted from ref. 143).

This bond scenario explains the observed results for the cyclopropanation reaction mentioned earlier. Ancillary ligands which are good $\pi$-acceptors tend to favor carbocationic behavior, because they decrease Au $\rightarrow$ $C_1$ $\pi$-donation. In line with this reasoning, it was
found that strongly \( \pi \)-acidic phosphite ligands provide only traces of the desired product and significant polymerization. Conversely, ligands that increase \( \text{Au} \rightarrow C^1 \) \( \pi \)-donation are expected to reduce carbocationic reactivity, while ligands that decrease \( C^1 \rightarrow \text{Au} \) \( \sigma \)-donation should increase carbene reactivity. This was exactly what was observed for the \( N \)-heterocyclic ligand IPr (1,3-bis(2,6-diisopropylphenyl)imidazol-2-ylidene), which affects both of these parameters (it is strongly \( \sigma \)-donating and only weakly \( \pi \)-acidic). As a result, IPrAu\(^+\) gave the product of stereospecific cyclopropanation with the best yield and diastereoselectivity among all the ligands tried (80% yield, 11:1 cis:trans).

As a consequence of these studies, Toste argued that gold(I)-coordinated carbenes is best accounted for by a continuum ranging from a metal-stabilized singlet carbene to a metal-coordinated carbocation, with the position of a given gold species on this continuum being largely determined by the carbene substituents and the ancillary ligand.

Finally, one may object that these two mesomeric forms are nothing but canonical extremes of the very same intermediate, thus making any discussion pointless. Nevertheless, as it was briefly highlighted here, a growing body of theoretical and experimental information suggests that these mesomeric forms contribute very unevenly to the actual character of \( \text{Au}^+ \) species, actually making it important to distinguish between them (Figure 2.9).

![Figure 2.9: Different reaction profiles are accessed by \( \text{Au}^+ \) catalysis that can be explained by either the carbocationic form (on the left side) or the carbene form (on the right side). The](image-url)
true nature of these intermediates is still source of debate and further studies should be provided to shed light on this subject.

2.1.5.4 The Dewar-Chatt-Duncanson model

A complex of a transition metal with a π-ligand, such as an alkene or an alkyne is generally described using the model of Dewar-Chatt-Duncanson (DCD)\textsuperscript{144}. In general lines, this model states that a σ-bond is made by the overlap of the π-orbital of the ligand (alkyne or alkene) with an empty \(d\) orbital of suitable symmetry from the metal. In another hand, a π-bond derives from the interaction of a filled \(d\) orbital of the metal with the empty antibonding orbital \(π^*\) of the ligand (backdonation).

As detailed by A. Fürstner and P. W. Davies\textsuperscript{102f}, there are 4 orbitals of type \(d\) of suitable symmetry to interact with an alkyne π-ligand (Figure 2.10).

![Diagram of Dewar-Chatt-Duncanson model](Figure 2.10: Considering possible interactions between \(d\) orbitals from the metal and \(π\) orbitals from an alkyne ligand, with the model being well applied to \(d^{10}\)-Au and \(d^8\)-Pt complexes. (Figure 2.10 was extracted from ref. 102f).)

The strongest interactions are the ones coming from the donation \(M (d_{z^2}) \leftrightarrow L (\pi_{||})\) and the back-donation \(M (d_{xz}) \rightarrow L (\pi^*{||})\), which are derived from a better overlap between

parallel orbitals (orbitals drawn in the plane of the paper). Weaker interactions derived from a poorer overlap between orbitals come from the donation \( M (d_{z^2}) \leftarrow L (\pi^\perp) \) and back-donation \( M (d_{xy}) \rightarrow L (\pi^\perp) \) (here, the subscribed sign \( \perp \) represents orbitals drawn perpendicular to the plane of the paper).

The individual contributions of each orbital interaction have been calculated using high level computational methods. Considering the complex derived from the complexation of gold(I) to ethyne, \([Au^+(C_2H_2)]\), it was found that the Au-alkyne bond is described by ca. 65% of the \( \sigma \)-interaction, ca. 27% for the in-plane \( \pi \)-back-donation, ca. 7% for the orthogonal \( \pi^\perp \), and ca. 1% for the \( \delta \)-interaction\(^{125}\). In light of these results, one can conclude that alkynes (but this is also true for alkenes) are strong two-electron \( \sigma \)-donors, but fairly poor \( \pi \)-acceptors toward Au(I) (as well as for Pt(II)). Even if back-donation occur to some extent and should not be overlooked\(^{102f}\), comparison to the analogous Cu-ethyne complex shows that in the case of copper, \( \pi \)-backbonding is much more important for copper than for gold. As a general conclusion, gold cannot be considered to participate significantly in Chatt-Dewar-Duncanson-type bonding\(^{102e}\).

In contrast to this energetic situation of \( \pi \)-ligands, are non-bonding \( p \) orbitals. Because they are lower in energy than the \( \pi^* \) orbitals of alkenes and alkynes, they are generally more suitable to overlap with the 5d filled orbitals of gold. These non-bonding \( p \) orbitals are typically found in cationic intermediates (Figure 2.11).

![Figure 2.11: Non-ligand p-orbitals are lower in energy than \( \pi^* \) orbitals of \( \pi \)-ligands and thus, more suitable for back-donation from gold. In complexes where strong back-donation is present, the carbenoid mesomeric form should better describe the nature of this intermediate.](image-url)
2.1.6 Considerations on the nature of vinyl gold species

After a nucleophilic attack on an alkyne or allene activated by gold, a vinyl gold species is generally expected to be formed and then engaged in either a 1,1-trapping or 1,2-trapping processes, as previously discussed in paragraph 2.1.5.2 and 2.1.5.3, respectively. Despite consistent with one’s intuition and the product structure observation, this was revealed to be a simplistic view of the whole mechanism.

Gagné et al.\textsuperscript{145} reported the existence of a diaurated complex \textbf{2.23} implicated in the intramolecular Friedel-Crafts reaction on an allene catalyzed by PPh\textsubscript{3}AuNTf\textsubscript{2}. This new observed intermediate, which is supposed to engage the vinyl anion in a bridging three-center two-electron mode and being further stabilized by a Au-Au interaction (well established and worth 5-10 kcal.mol\textsuperscript{-1}\textsuperscript{146}) would account for the greater stability of the more electron deficient and crowded diaurated intermediate \textbf{2.23} towards acid when compared to the simpler vinyl gold species \textbf{2.22} (scheme 2.9).

![Scheme 2.8: Friedel-Crafts reaction on an allene activated by PPh\textsubscript{3}Au\textsuperscript{+} complex and the generally accepted simplified mechanism\textsuperscript{147}](image)


\textsuperscript{147} See also: Tarselli, M. A.; Gagné, M. R.; \textit{J. Org. Chem.}, 2008, 73, 2439.
mechanism of scheme 2.8. This means that the mechanism implicated in this hydroarylation process, as well as it possibly suggests for other gold-mediated mechanisms, is more complex than one could initially “naively” imagine.

Scheme 2.9: Study of reactivity of intermediate 2.22 (Scheme 2.9 was extracted from ref. 145).
CHAPTER 3: SYNTHESIS OF FUNCTIONALIZED OXAZOLONES BY A SEQUENCE OF Cu(II) AND Au(I)-CATALYZED TRANSFORMATIONS

This work was developed with contributions from the students Andrea K. Buzas, Florin M. Istrate, and Yann Odabachian. Complementary reading can be found in previous reports on this subject\(^{148}\). With the concern for a clear exposal of the subject, all the experiments performed will be presented herein, nevertheless, the ones done by the author of this thesis will be marked with a red star (*).


CHAPTER 3: SYNTHESIS OF FUNCTIONALIZED OXAZOLONES BY A SEQUENCE OF Cu(II)- AND Au(I)-CATALYZED TRANSFORMATIONS

3.1 INTRODUCTION:

3.1.1 Importance of 4-Oxazol-2-ones

4-oxazol-2-ones are an important class of heterocycles, whose biological importance has been recognized over the past years. For instance, combretoxazolone derivatives\(^{149}\), such as 3.1, are reported to exhibit cytotoxic and antitumor properties; 4-substituted 3-aryl-5-tert-butyl-4-oxazolone-2-ones\(^{150}\), such as 3.2, have been described as powerful herbicides; 3,4-diaryloxazolones\(^{151}\), such as 3.3, as anti-inflammatory agents; and compound 3.4 as possessing antimicrobial properties\(^{152}\) (Figure 3.1).

Figure 3.1: Examples of biological active 4-oxazol-2-ones.

3.1.2 Synthesis of 4-Oxazol-2-ones

4-Oxazol-2-ones can be synthesized following a great number of protocols. Moreover, general strategies to assemble such molecules can be reduced to a few common intermediates. Some of these methods, arguably judged of wide generality, will be briefly presented below.


Oxazolones can be accessed from (N-aryl-N-hydroxy)-acetylamides through a thermally\(^{153}\) or photolytic induced rearrangement\(^{154}\), or upon treatment with triethylamine and \(p\)-nitrobenzenesulphonylchloride\(^{155}\). In addition, mesylated (N-aryl-N-hydroxy)-acetylamide under basic conditions and sonication\(^{156}\) can also be rearranged to the corresponding 4-oxazol-2-one (Scheme 3.1).

Other straightforward methods to assemble the oxazolone core consist of the treatment of \(\alpha\)-hydroxy ketones with isocyanates\(^{157}\), \(\alpha\)-aminoketones with a base and phosgene\(^{158}\), and from the N-/O-exchange from cyclic carbonates\(^{159}\) (scheme 3.2).

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**Scheme 3.1**: Some examples of oxazolones synthesis starting from (N-aryl-N-hydroxy)-acetylamides derivatives.

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\(^{157}\) See refs. 149 and 150.


Since the first preparation of an ynamine reported in the literature in 1892 by J. Bode\textsuperscript{160} and the subsequent developments that culminated with the first general method for ynamine synthesis reported by Viehe in 1963\textsuperscript{161}, reactivity of such alkynes started to be more deeply explored. However, unlike its close relative enamines, their synthetic potential did not consolidate. The reason for that is the fact ynamines are difficult to prepare and to handle, with their sensitivity toward hydrolysis and their reactivity toward electrophiles thus making such molecules generally inaccessible synthetically. A clever way to solve this problem is to attenuate the electron density on the nitrogen atom of ynamines by adding an electron withdrawing group, or through substitutions in the alkyne, thus improving ynamines stability.

Today, the term ynamides is employed to generally refer to electron-deficient ynamines carrying an EWG on the nitrogen atom. Examples of simple ynamines (compounds 3.1-3.5) and ynamides are shown below (compounds 3.6-3.9)\textsuperscript{162} (figure 3.2).

\textsuperscript{160} Bode, J.; Liebigs Ann. Chem., 1892, 267, 268.
\textsuperscript{162} Hsung, R. P.; Tetrahedron, 2006, 62, 3781.
Figure 3.2: Presentation of different classes of ynamines (Figure 3.2 was extracted from ref. 162).

Ultimately, the more stable ynamides\(^{163}\) gave birth to the *renaisance* of ynamines chemistry and have been applied to a large variety of processes, ranging from cycloadditions\(^{164}\), metathesis reactions\(^{165}\), enamide synthesis\(^{166}\), \(\alpha\)-aminoamidines synthesis\(^{167}\), indole synthesis\(^{168}\), radical\(^{169}\), metal\(^{170}\) and Brønsted acid\(^{171}\) promoted


(cyclo)isomerizations to their application in natural product synthesis. Some examples are shown below (scheme 3.3).

a)

\[
\text{I} + \text{Ts} \xrightarrow{\text{Pd(OAc)}_2 (5\% \text{mol}), \text{PPh}_3 (20\% \text{mol}), \text{Bu}_3\text{N} \text{OAc} (3 \text{ equiv.}), \text{DMF, } 60^\circ \text{C}} \text{Bn} \xrightarrow{\text{H}^+} \text{Ts} \]

64%

via:

\[
\begin{align*}
\text{Ph} & \quad \text{N}\text{Me}3\text{Si} & \quad \text{N}\text{Ph} & \quad \text{N}\text{Me}3\text{Si} & \quad \text{N}\text{Ph} \\
\end{align*}
\]

no copper salt used!

b)

\[
\text{MeO}_2\text{C} \quad \text{SiMe}_3 \quad \text{N}\text{Me}3\text{Si} \quad \text{N}\text{Me}3\text{Si} \quad \text{N}\text{Me}3\text{Si} \\
\]

via:

\[
\begin{align*}
\text{Me}_2\text{N} & \quad \text{Ts} & \quad \text{Ph} & \quad \text{Ph} & \quad \text{MeO}_2\text{C} & \quad \text{CO}_2\text{Me} \\
\end{align*}
\]

c)

\[
\text{NH}_2 \quad \text{N}\text{Me}3\text{Si} \quad \text{N}\text{Me}3\text{Si} \quad \text{N}\text{Me}3\text{Si} & \quad \text{N}\text{Me}3\text{Si} \\
\]

via:

\[
\begin{align*}
\text{Me}_2\text{N} & \quad \text{Ts} & \quad \text{Ph} & \quad \text{Ph} & \quad \text{MeO}_2\text{C} & \quad \text{CO}_2\text{Me} \\
\end{align*}
\]

Scheme 3.3: Three examples of ynamines applied in synthesis. a) 2-amino-indole synthesis through a Sonogashira coupling followed by spontaneous amino addition to the triple bond. b) Enantioselective synthesis of axially chiral anilides through rhodium promoted [2+2+2] cycloaddition. c) Synthesis of (-)-Pyridoblastic acid, the pyrimidine metal binding domain of Bleomycin A₂ through a Diels-alder-retro Diels-Alder sequence using 1-(dibenzylamino)propyne.

3.1.3.1 Synthesis of Ynamides

Due to the importance of ynamides as synthetically stable ynamines synthons, growing efforts have been consacrated to the development of new methodologies towards the synthesis of such molecules.

Isomerization of N-propargylated compounds was recognized by Hsung et al.¹⁷³ as a fast and practical route to ynamides. Nevertheless, this process proved to be highly sensitive to the electron-withdrawing group on the nitrogen atom. For instance, carbamates were reported to stop at the allene intermediate. On the other hand, amides are fully isomerized to the corresponding ynamides¹⁷³ (scheme 3.4).

![Scheme 3.4: Isomerization of N-propargylated compounds into ynamides.](image)

Scheme 3.4: Attempts in the isomerization of N-propargyl derivatives. **a)** N-propargyl carbamates fail to completely isomerize to the corresponding ynamide and stop at the allenamide intermediate. **b)** In contrast, N-propargyl amides isomerize under the same conditions to the corresponding ynamides.

Another possible route to ynamides is the elimination reaction from bromoenamides. Although the desired ynamide can be obtained, this process is limited to the elimination of Z-bromo enamides, with E-enamides generally being recovered (scheme 3.5).

![Scheme 3.5](image)

examples:

- O\(\text{N}\)\(\text{C}\)\(\text{N} = \text{Ph}\)\(\text{Bn}\) 75%  
- O\(\text{N}\)\(\text{C} \equiv \text{n-pentyl}\)\(\text{Bn}\) 72%  
- O\(\text{N}\)\(\text{C} \equiv \text{Ph}\)\(\text{Ph}\) 40%  
- O\(\text{N}\)\(\equiv\text{n-hexyl}\) 88%

Scheme 3.5: The procedure reported by Hsung *et al.* based on the bromination of enamides followed by elimination, substantially extends the scope of this method first reported by Viehe *et al.*, but still presents severe limitations as the elimination step is only possible for Z-bromo enamides.

The Corey-Fuchs derived methods from formamides are also a possible route to ynamides. These compounds react with triphenylphosphine and tetrachloromethane to produce \(\beta,\beta\)-dichloroenamides 3.11, which upon treatment with a strong base, such as \(n\)-BuLi, at low temperatures, rearrange to furnish the lithium acetylide anion that can be trapped either by a proton (3.12) or another electrophile (3.13), or can be

---

transmetalated with ZnBr₂ to be directly engaged in a Negishi cross-coupling reaction with aryl iodides (3.5)\(^{177}\). Conversely, dichloro compound 3.11 can be first submitted to a Suzuki cross-coupling reaction and then undergo chlorine elimination upon treatment with a strong base to form the alkyne moiety. Ynamides 3.12 can also be functionalized through a Sonogashira cross-coupling reaction\(^{178}\) (Scheme 3.6).

Scheme 3.6: Synthesis of ynamides from formamides. (Scheme 3.6 was extracted from ref. 163b).

Finally, cross-coupling reactions have been found as one of the most efficient routes to ynamides. Today, six main cross-coupling protocols have been established (Figure 3.3).

\[ \text{Cu source, ligand, base, solvent, temperature} \]

\[ \text{EWG = CO}_2\text{R} \quad \text{SO}_2\text{R} \quad \text{CONR} \]

---


1) Hsung et al.: CuSO₄·5H₂O (5-10 mol%), 1.10-phenantroline (10-20 mol%), K₂PO₄ (2 equiv.) or K₂CO₃ (2 equiv.) toluene, 60-65 °C, X = Br  
ref. 179

2) Danheiser et al.: Cul (1 equiv.), no ligand  
KHMDS (1 equiv.), pyridine, rt, X = Br  
ref. 180

3) Stahl et al.: CuCl₂ (20 mol %), no ligand, O₂ (1 atm)  
N₂H₄ (2 equiv.), pyridine (2 equiv.), toluene, 70 °C, X = H  
ref. 181

4) Jiao et al.: CuCl₂, 2H₂O (10 mol %), no ligand, under air  
N₂H₄ (2 equiv.), toluene, 100 °C, X = CO₂H  
ref. 182

5) Evano et al.: CuCl₂·2H₂O (15 mol %), 1,2-dimethylimidazole (40 mol %), O₂ (1 atm)  
4Å molecular sieves, CH₂Cl₂, rt, X = BF₃K  
ref. 183

6) Evano et al.:  
ref. 184

Figure 3.3: List of cross-coupling reactions employed to assemble ynamides.

All the aforementioned methods afford the desired ynamides with good yields, notwithstanding some advantages of a few protocols over others. For instance, Danheiser’s (Method 2) and Evano’s (Method 5) protocols are the only reports on the coupling reaction at room temperature. On one hand, Danheiser’s method uses 1 equiv. of amine derivative, 2 equiv. of bromoalkyne and a stoichiometric quantity of copper salt. On the other hand, Evano’s method 5 uses catalytic quantity of a copper salt, but 5 equiv. of amine derivatives with 1 equiv. of the alkyne coupling partner (5 equiv. of amine derivative are also needed in Stahl’s protocol, method 3).

Among these previous six methods, two of the most interesting ones are Hsung’s method 1, which employs almost equal amounts of coupling partners (1.1 equiv. of alkyne

---


and 1.0 equiv. of amine derivatives) and Evano’s method 6 where the bromo alkyne is formed in the course of the reaction and its vinyl dibromo precursor being easily obtained from the corresponding aldehydes.

Nitrogen containing compounds successfully employed in the aforementioned cross-coupling reactions are shown below (Figure 3.4).

![Figure 3.4: List of typical frameworks used in Csp-N cross coupling reactions.]

### 3.2 Boc-CYCLIZATION ON ALKYNES: Literature background and previous work in our laboratory

In previous studies from our group and others, the 5-exo dig cyclization of propargyl Boc-carbamates and carbonates was reported to afford alkylidene cyclic carbamates and carbonates, respectively. A brief overview is shown below (Scheme 3.7).

![Scheme 3.7: Boc-cyclization on alkynes.]

---


3.3 RESULTS AND DISCUSSION

3.3.1 The initial idea

Although the alkylidene cyclic carbamates can be isomerized to 4-oxazol-2-ones under acidic conditions\(^ {187}\) (scheme 3.8), our group was interested in a more straightforward route to such molecules.

Furthermore, we were also interested if the 5-\textit{endo dig} cyclization, equally favorable as the previous 5-\textit{exo} process shown above, as predicted by the empirical Baldwin rules\(^ {188}\), would be accessible through the gold mediated rearrangement of the corresponding ynamides homologues (scheme 3.9). Also an aspect of interest to us was the extension of the reaction scope of ynamide cross-coupling reactions for Boc derivatives, first reported by Hsung \textit{et al.}, which at the time our work was performed, was quite restricted (see figure 3.3).


Scheme 3.9: hypothesis concerning the gold-mediated 5-endo dig cyclization process of ynamides tested in this project.

As necessary precursors for the cross-coupling reaction that would afford ynamides \( \mathbf{3.15a-v} \), tert-butyloxy carbamates \( \mathbf{3.18a-h} \) were synthesized using \( \text{Boc}_2\text{O} \) in different solvents\(^\text{189} \) (table 3.1).

\[
\begin{align*}
\text{R}^+\text{NH}_2 + \text{Boc}_2\text{O} & \quad \text{conditions: a) toluene, 110 °C} \\
& \quad \text{b) EtOH, rt} \\
& \quad \text{c) NaHCO}_3 \quad \text{CH}_3\text{CN, rt} \\
& \quad \text{d) Et}_3\text{N} \quad \text{MeOH, rt}
\end{align*}
\]

<table>
<thead>
<tr>
<th>substrate</th>
<th>product</th>
<th>conditions</th>
<th>time</th>
<th>yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>( \text{NH}_2 )</td>
<td>( \text{NHBoc} )</td>
<td>a</td>
<td>2 h</td>
<td>82 %</td>
</tr>
<tr>
<td>( \text{NH}_2 )</td>
<td>( \text{NH}_2 )</td>
<td>a</td>
<td>1 h</td>
<td>85 %*</td>
</tr>
<tr>
<td>( \text{NH}_2 )</td>
<td>( \text{NH}_2 )</td>
<td>a</td>
<td>2 h</td>
<td>43 %</td>
</tr>
<tr>
<td>( \text{NH}_2 )</td>
<td>( \text{NH}_2 )</td>
<td>a</td>
<td>3 h</td>
<td>70 %</td>
</tr>
<tr>
<td>( \text{NH}_2 )</td>
<td>( \text{NH}_2 )</td>
<td>a</td>
<td>1 h</td>
<td>95 %*</td>
</tr>
<tr>
<td>( \text{NH}_2 )</td>
<td>( \text{NH}_2 )</td>
<td>b</td>
<td>5 min</td>
<td>83 %*</td>
</tr>
</tbody>
</table>

The synthesis of ynamides was inspired from Hsung’s protocol, with a slight modification of reagent proportions (our protocol: 1 equiv. of bromoalkyne, 1.2 equiv. of carbamate, 2.4 equiv. of K$_3$PO$_4$, 0.2 equiv. of CuSO$_4$·5H$_2$O and 0.4 equiv. of 1,10-phenanthroline, toluene, 80 °C). The scope of this method was then more deeply studied, with 22 examples of Boc-ynamides being synthesized, possessing neutral, electron rich and electron poor groups on the aromatic ring, esters, silyl ether and a bulky alkyne partner being tolerated (table 3.2).

<table>
<thead>
<tr>
<th>substrate</th>
<th>product</th>
<th>conditions</th>
<th>time</th>
<th>yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>EtO$_2$C—NH$_2$</td>
<td>EtO$_2$C—NHBOc</td>
<td>c</td>
<td>24 h</td>
<td>47 %</td>
</tr>
<tr>
<td>EtO$_2$C—NH$_2$</td>
<td>EtO$_2$C—NHBOc</td>
<td>d</td>
<td>24 h</td>
<td>90 %</td>
</tr>
<tr>
<td>NH$_2$</td>
<td>NHBOc</td>
<td>a</td>
<td>2 h</td>
<td>83 %</td>
</tr>
</tbody>
</table>

Table 3.1: Carbamates synthesized in this project

<table>
<thead>
<tr>
<th>entry</th>
<th>bromoalkyne</th>
<th>carbamate</th>
<th>ynamide</th>
<th>time</th>
<th>yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Br—C≡C—Ph</td>
<td>Boc—NH</td>
<td>Boc—N≡C—Ph</td>
<td>40 h</td>
<td>80 %</td>
</tr>
<tr>
<td>2</td>
<td>3.17a</td>
<td>3.18a</td>
<td>3.15a</td>
<td>16 h</td>
<td>65% *</td>
</tr>
<tr>
<td>entry</td>
<td>bromoalkyne</td>
<td>carbamate</td>
<td>ynamide</td>
<td>time</td>
<td>yield</td>
</tr>
<tr>
<td>-------</td>
<td>-------------</td>
<td>-----------</td>
<td>---------</td>
<td>------</td>
<td>-------</td>
</tr>
<tr>
<td>3</td>
<td>3.17a</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-N=Ph" /></td>
<td>18 h</td>
<td>68%</td>
</tr>
<tr>
<td>4</td>
<td>3.17a</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-N=Ph" /></td>
<td>16 h</td>
<td>48%</td>
</tr>
<tr>
<td>5</td>
<td>3.17a</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-N=Ph" /></td>
<td>48 h</td>
<td>22%*</td>
</tr>
<tr>
<td>6</td>
<td>3.17a</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-N=Ph" /></td>
<td>48 h</td>
<td>62%*</td>
</tr>
<tr>
<td>7</td>
<td>3.17a</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-N=Ph" /></td>
<td>36 h</td>
<td>70%</td>
</tr>
<tr>
<td>8</td>
<td>3.17a</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-N=Ph" /></td>
<td>48 h</td>
<td>23%</td>
</tr>
<tr>
<td>9</td>
<td>3.17b</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-N=Ph" /></td>
<td>38 h</td>
<td>24%</td>
</tr>
<tr>
<td>10</td>
<td>n-C₅H₁₁-N=Br</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-n-C₅H₁₁-N=Ph" /></td>
<td>52 h</td>
<td>75%*</td>
</tr>
<tr>
<td>11</td>
<td>3.17c</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-n-C₅H₁₁-N=Ph" /></td>
<td>48 h</td>
<td>69%</td>
</tr>
<tr>
<td>12</td>
<td>3.17d</td>
<td><img src="image1" alt="Boc-NH" /></td>
<td><img src="image2" alt="Boc-n-C₅H₁₁-N=Ph" /></td>
<td>67 h</td>
<td>72%</td>
</tr>
<tr>
<td>entry</td>
<td>bromoalkyne</td>
<td>carbamate</td>
<td>ynamide</td>
<td>time</td>
<td>yield</td>
</tr>
<tr>
<td>-------</td>
<td>-------------</td>
<td>-----------</td>
<td>---------</td>
<td>-------</td>
<td>-------</td>
</tr>
<tr>
<td>13</td>
<td>3.17d</td>
<td>3.18c</td>
<td><img src="3.15m" alt="Image" /></td>
<td>67 h</td>
<td>80 %</td>
</tr>
<tr>
<td>14</td>
<td>3.17d</td>
<td>3.18g</td>
<td><img src="3.15n" alt="Image" /></td>
<td>67 h</td>
<td>50 %</td>
</tr>
<tr>
<td>15</td>
<td>AcO 3.17e</td>
<td>3.18a</td>
<td><img src="3.15o" alt="Image" /></td>
<td>65 h</td>
<td>55 %*</td>
</tr>
<tr>
<td>16</td>
<td>3.17e</td>
<td>3.18i</td>
<td><img src="3.15p" alt="Image" /></td>
<td>48h</td>
<td>49%</td>
</tr>
<tr>
<td>17</td>
<td>3.17e</td>
<td>3.18f</td>
<td><img src="3.15q" alt="Image" /></td>
<td>48 h</td>
<td>49 %*</td>
</tr>
<tr>
<td>18</td>
<td>3.17e</td>
<td>3.18i</td>
<td><img src="3.15r" alt="Image" /></td>
<td>72 h</td>
<td>49 %*</td>
</tr>
<tr>
<td>19</td>
<td><img src="3.18a" alt="Image" /></td>
<td>3.18a</td>
<td><img src="3.15s" alt="Image" /></td>
<td>45 h</td>
<td>88 %</td>
</tr>
<tr>
<td>20</td>
<td>3.17l</td>
<td>3.18f</td>
<td><img src="3.15t" alt="Image" /></td>
<td>62 h</td>
<td>72 %*</td>
</tr>
<tr>
<td>21</td>
<td><img src="3.17g" alt="Image" /></td>
<td>3.18a</td>
<td><img src="3.15u" alt="Image" /></td>
<td>48 h</td>
<td>74 %</td>
</tr>
<tr>
<td>22</td>
<td>3.17g</td>
<td>3.18i</td>
<td><img src="3.15v" alt="Image" /></td>
<td>48 h</td>
<td>65 %</td>
</tr>
</tbody>
</table>

Table 3.2: Ynamides synthesized in this project.
Having an efficient route to ynamides in hand, we next focused on the cyclization of Boc-ynamides.

The use of the gold complex PPh₃AuNTf₂, which was reported to work nicely in the case studied by Hashmi et al.¹⁹⁰, afforded the desired cyclized product only in poor yields (entry 1, table 3.3). The more electrophilic cationic gold complex (p-CF₃Ph)₃PAuNTf₂ increased the product conversion, but provided access to only modest yields (entries 2 and 3, table 3.3). Conversely, employment of a different counter-ion on the gold complex, by using PPh₃Au(NCCH₃)SbF₆ considerably increased the reaction yield, affording the desired oxazolone in good yields (entries 4 and 5, table 3.3). Only degradation was obtained with the use of HNTf₂ (entry 6, table 3.3). Noteworthy, the reaction also worked for AgNTf₂ but less efficiently (entry 7, table 3.3). Finally, optimal conditions were found to be 1 mol% of PPh₃Au(NCCH₃)SbF₆ in refluxing dichloromethane (entry 5, table 3.3).

![Diagram of cyclization reaction](image)

<table>
<thead>
<tr>
<th>entry</th>
<th>LAu⁺ source</th>
<th>temperature</th>
<th>time</th>
<th>conversion</th>
<th>yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>PPh₃AuNTf₂</td>
<td>20 °C</td>
<td>7 h</td>
<td>63 %</td>
<td>28 %</td>
</tr>
<tr>
<td>2</td>
<td>(p-CF₃Ph)₃PAuNTf₂</td>
<td>20 °C</td>
<td>72 h</td>
<td>85 %</td>
<td>52 %</td>
</tr>
<tr>
<td>3</td>
<td>(p-CF₃Ph)₃PAuNTf₂</td>
<td>40 °C</td>
<td>2 h 30 min</td>
<td>100 %</td>
<td>40 %</td>
</tr>
<tr>
<td>4</td>
<td>PPh₃Au(NCCH₃)SbF₆</td>
<td>20 °C</td>
<td>4 h 30 min</td>
<td>100 %</td>
<td>69 %</td>
</tr>
<tr>
<td>5</td>
<td>PPh₃Au(NCCH₃)SbF₆</td>
<td>40 °C</td>
<td>30 min</td>
<td>100 %</td>
<td>74 %</td>
</tr>
<tr>
<td>6</td>
<td>HNTf₂</td>
<td>20 °C</td>
<td>1 h</td>
<td>100 %</td>
<td>degrad.</td>
</tr>
<tr>
<td>7</td>
<td>AgNTf₂</td>
<td>20 °C</td>
<td>1 h 30 min</td>
<td>100 %</td>
<td>53 %</td>
</tr>
</tbody>
</table>

Table 3.3: Catalyst screening for the Au(I)-mediated cyclization of Boc-ynamides.

Accordingly, substrates 3.15a-3.15v were submitted to the optimal conditions of cyclization. The reaction proved to be compatible with numerous functional groups, aryl,

---

benzyl and acetyl groups on the nitrogen atom, and esters (3.15o-3.15r), silyl ethers (3.15s-3.15t), aryl (3.15a-3.15h), alkyl (3.15j-3.15n, 3.15u-3.15v) and a bulky group (3.15i) on the alkyne moiety afforded the desired oxazolones in general good yields (38-94%, table 3.4).

![Reaction Scheme](image)

<table>
<thead>
<tr>
<th>entry</th>
<th>substrate</th>
<th>R¹</th>
<th>R²</th>
<th>time</th>
<th>product</th>
<th>yield⁰</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>3.15a</td>
<td>Ph</td>
<td></td>
<td>25 min</td>
<td>3.16a</td>
<td>83 %</td>
</tr>
<tr>
<td>2</td>
<td>3.15b</td>
<td>p-FPh</td>
<td></td>
<td>10 min</td>
<td>3.16b</td>
<td>88 %</td>
</tr>
<tr>
<td>3</td>
<td>3.15c</td>
<td>p-ClPh</td>
<td></td>
<td>10 min</td>
<td>3.16c</td>
<td>88 %</td>
</tr>
<tr>
<td>4</td>
<td>3.15d</td>
<td>p-BrPh</td>
<td></td>
<td>10 min</td>
<td>3.16d</td>
<td>83 %</td>
</tr>
<tr>
<td>5</td>
<td>3.15e</td>
<td>Ph</td>
<td>2,4-(OMe)₂Ph</td>
<td>16 h</td>
<td>3.16e</td>
<td>85 %</td>
</tr>
<tr>
<td>6</td>
<td>3.15f</td>
<td>Bn</td>
<td></td>
<td>16 h</td>
<td>3.16f</td>
<td>78 %</td>
</tr>
<tr>
<td>7</td>
<td>3.15g</td>
<td>CH₂CO₂Et</td>
<td></td>
<td>12 h</td>
<td>3.16g</td>
<td>93 %</td>
</tr>
<tr>
<td>8</td>
<td>3.15h</td>
<td>(S)</td>
<td></td>
<td>8 h</td>
<td>3.16h</td>
<td>94 %</td>
</tr>
<tr>
<td>9</td>
<td>3.15i</td>
<td>tBu</td>
<td>Ph</td>
<td>2 h</td>
<td>3.16i</td>
<td>58 %</td>
</tr>
<tr>
<td>10</td>
<td>3.15j</td>
<td>n-C₅H₁₁</td>
<td>Ph</td>
<td>30 min</td>
<td>3.16j</td>
<td>74 %</td>
</tr>
<tr>
<td>11</td>
<td>3.15k</td>
<td>Bn</td>
<td></td>
<td>40 min</td>
<td>3.16k</td>
<td>50 %</td>
</tr>
<tr>
<td>12</td>
<td>3.15l</td>
<td>Ph</td>
<td></td>
<td>30 min</td>
<td>3.16l</td>
<td>78 %</td>
</tr>
<tr>
<td>13</td>
<td>3.15m</td>
<td>p-ClPh</td>
<td></td>
<td>10 min</td>
<td>3.16m</td>
<td>94 %</td>
</tr>
<tr>
<td>14</td>
<td>3.15n</td>
<td>CH₂CO₂Et</td>
<td></td>
<td>5 h</td>
<td>3.16n</td>
<td>70 %</td>
</tr>
<tr>
<td>15</td>
<td>3.15o</td>
<td>Ph</td>
<td></td>
<td>40 min</td>
<td>3.16o</td>
<td>71 %</td>
</tr>
<tr>
<td>16</td>
<td>3.15p</td>
<td>2-Napht</td>
<td></td>
<td>45 min</td>
<td>3.16p</td>
<td>88 %</td>
</tr>
<tr>
<td>17</td>
<td>3.15q</td>
<td>Bn</td>
<td></td>
<td>20 min</td>
<td>3.16q</td>
<td>50 %</td>
</tr>
<tr>
<td>18</td>
<td>3.15r</td>
<td>CH₂CO₂Et</td>
<td></td>
<td>20 min</td>
<td>3.16r</td>
<td>49 %</td>
</tr>
</tbody>
</table>
For the examples 3.15a-3.15d, where the reaction time was short, the Ag(I)-mediated cyclization was attempted. In most cases, the desired oxazolones were isolated in good yields (88-96%, scheme 3.11). Nevertheless, this performance was not general, since substrate 3.15o, for instance, afforded the corresponding oxazolone 3.16o in poor 36% yield (scheme 3.10).

Scheme 3.10: Silver(I)-mediated Boc-cyclization of ynamides proved to work very well in certain cases, although not as general as the gold catalyzed process.

The proposed mechanism is believed to start with the gold activation of the triple bond in the ynamide molecule 3.15 toward cyclization of the nucleophilic oxygen atom of the Boc group, generating the cationic stabilized species 3.19. Cleavage of the C-O bond from the tert-butyloxy group affords isobutene and furnishes the vinyl gold species 3.20, which is protodeurated to give entry to oxazolones 3.16 (Scheme 3.11).
Next, we envisaged a possible domino sequence expecting that the new double bond from the oxazolone ring would be nucleophilic enough to attack a second triple bond in close proximity. Despite all our efforts, this proved difficult, and did not afford in our hands any further cyclized product (scheme 3.12).

Scheme 3.12: First efforts on the direct conversion of ynamides to a tricyclic structure did not afford the desired compounds, being always stopped after the oxazolone formation.
Even submitting the isolated oxazolone (91% from PPh$_3$Au(NCCH$_3$)SbF$_6$, DCM, 40 °C) to different gold catalysts and a stronger electrophile, such as NIS, did not afford any further cyclized products (as observed from the $^1$H NMR of the crude reaction mixture), but only the complete recovery of the starting oxazolone (scheme 3.13).

Scheme 3.13: Conditions tested for the cyclization of oxazone 3.15w under different electrophilic conditions.

Other cyclization attempts using different ynamides also failed. It seems that the double bonds from 4-oxazol-2-ones are simply not nucleophilic enough to attack the alkyne (Scheme 3.14).

Scheme 3.14: Trials of cascade reaction via 4-oxazol-2-ones.
A possible explanation for this inertness is the “push-push” nature exerted both by the oxygen and the nitrogen atoms with competitive antithetic mesomeric modes of action from the enol point of view and the enamine point of view (figure 3.5).

![Enamine Reactivity vs Enol Reactivity](image)

Figure 3.5: The possible reason for the inertness of 4-oxazol-2-ones towards electrophiles can be the low nucleophilicity of the double bond due to the antithetic reactivities accessed by enol against enamine modes of attack (“push-push system”).

### 3.4 CONCLUSION

In summary, an efficient two-steps sequence to access 4-oxazol-2-ones from bromoalkynes and tert-butyloxy carbamates was developed. In one sense we extended the reaction scope of NBoc ynamides synthesis through a Cu(II)-based cross-coupling reaction initially developed by Hsung. In another sense, we have also further extended the feasibility of the Au(I)-promoted cyclization of tert-butyloxy carbamates previously studied in our group from the 5-exo dig to the 5-endo dig cyclization mode.

Preliminary efforts to access a gold(I)-promoted cyclization cascade proved difficult. This can be attributed to the “push-push” nature of 4-oxazol-2-ones, which would account for the low nucleophilicity of the double bond present in the oxazolone ring.

### 3.5 PERSPECTIVES AND FURTHER DEVELOPMENTS

As discussed previously, the “push-push” nature of 4-oxazol-2-ones seems to hamper the envisioned cascade cyclization process, which stops after the first cyclization. Nevertheless, another important aspect worth mentioning is that, because enamines are more nucleophilic than enols (fact that is commonly used in organocatalysis with proline...
derivatives and other amines), all the cyclizations attempted here were sterically favoring the enol nucleophilic attack, and hence, were against the expected double bond polarity. A further way to test if this cascade reaction could be from diyne of type 3.21, which would give a product coming from an exo or/and endo cyclization 3.22 or/ and 3.23, respectively. These cyclized iminium ions could eventually be quenched by a nucleophile to produce 3.24 and/or 3.25 or ring-expand to 3.26 and/or 3.27 (this semi-pinacol-like rearrangement would occur via a stabilized carbocation that would be at the same time allylic and α to an oxygen atom, scheme 3.15).

![Scheme 3.15: Envisaged reaction pathways exploitable in the gold-mediated cascade reaction with diyne 3.21.](image)

Another suggestion of cascade reaction involves dienyne 3.28 and 3.31, which could potentially undergo the Boc rearrangement studied in this chapter to generate compounds 3.29 and 3.32, respectively, followed by a Diels-Alder reaction with the diene in close proximity to furnish tricyclic structures 3.30 and 3.33, respectively\(^1\) (scheme 3.16).

Scheme 3.16: Boc cyclization/ Diels-Alder sequence to construct bicyclic structures 3.30 and 3.33.
CHAPTER 4: GOLD(I)-MEDIATED CYCLOISOMERIZATION OF 1,6-ENYNES – A SHORT DISGRESSION ON CATALYST COUNTER ION EFFECT AND REMOTE SUBSTITUENT CONTROL

The work described in this chapter was performed individually and the results were not yet published
CHAPTER 4: GOLD(I)-MEDIATED CYCLOISOMERIZATION OF 1,6-ENYNES – A SHORT DISGRESSION ON CATALYST COUNTER ION EFFECT AND REMOTE SUBSTITUENT CONTROL

4.1 INTRODUCTION: Gold catalyzed 1,6-ynene cycloisomerizations

The attractiveness of metal-ynene cycloisomerization is pinned on the rapid increase in structural complexity starting from considerably simple acyclic substructures. Among many competent metals, such as Pt, Pd, Ru, In, Rh, etc., gold is particularly efficient, due to its ability to access a wide range of cyclic products under mild conditions and with high chemoselectivity. For this reason, gold-mediated 1,n-ynene (n = 3, 4, 5, 6, etc) cycloisomerization has been extensively studied in the past 10 years.

Although today numerous aspects concerning the mechanistic scenario and product selectivity for the gold-mediated cycloisomerization have been settled, others still remain to be elucidated. In the next sections of this introduction, these well established aspects will be discussed in more detail for 1,6-ynenes (Scheme 4.1). As other enynes are outside the scope of this chapter, the reader looking for further information is invited to visit the excellent reviews listed in ref. 197.

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Scheme 4.1: Reaction scope of cycloisomerization of 1,6-enynes and interception of reaction intermediates by a nucleophile.

4.1.1 Synthesis of 1-alkenyl-1-cyclopentenes (4.2 and 4.3)

In 2004, Echavarren et al. reported the use of gold catalysts for the cycloisomerization of 1,6-enynes 4.11. The generation of highly cationic Ph₃PAu⁺ species from the treatment of PPh₃AuCl with AgSbF₆ was employed to produce dienes 4.12 at room temperature. Previously, only PtCl₂-promoted cycloisomerizations had been described and they required heating at 80°C. Our group next reported the use of stable, highly active and easy to handle Ph₃PAuNTf₂ for the same transformation and many other gold catalysts followed as suitable options for this transformation. A representative reaction scope is exhibited below (scheme 4.2).

200 See reviews in ref 197.
Scheme 4.2: Examples of dienes obtained from gold catalyzed 1,6-enynes cycloisomerizations\textsuperscript{198}.

The reactivity of the gold-alkyne and gold-alkene complexes can be understood from cationic forms 4.14 and 4.17, which are resonance structures of 4.13 and 4.16, respectively. Another important resonance structure of the Au-alkyne complex is the gold carbene 4.15, which is implicated in cyclopropanation reactions discussed below. Complex 4.17 reveals that the [1,2]-hydride shift would produce the corresponding Au carbene 4.18. Applying the principle of microscopic reversibility, one can find a common termination step found in gold catalysis: Au carbene 4.18 serves as a precursor for a [1,2]-hydride shift, which upon elimination of LAu\(^+\) fragment from intermediate 4.17, produces an alkene (formal “\(\beta\)-hydride elimination”).

Scheme 4.3: Resonance structures of gold-alkyne and gold alkene complexes.

The reaction mechanism culminating in products 4.2/4.3 is described by the complexation of the gold metal to the alkyne, followed by a 5-exo dig cyclization. This step
corresponds to the cyclopropanation of the proximate alkene to produce cyclopropyl metal carbene 4.19. This initial step can also be explained by invoking the resonance contribution of carbene 4.15, which results in direct alkyne cyclopropanation. Alternatively, the cyclopropanation can be envisioned to occur step-wise via initial reaction of the alkene with the alkenyl cation 4.14, followed by nucleophilic interception of the resulting carbocation by the alkenyl gold formed (not shown). Echavarren et al. performed DFT calculations\textsuperscript{198}, suggesting that the Au(I)-mediated cyclopropanation to afford intermediate 4.19 occurs directly via a single transition state. In the absence of an external nucleophile, highly electron-deficient gold carbene 4.19 undergo a [1,2]-alkyl shift to produce 4.20.

**Scheme 4.4: Mechanistic rationale for observed cycloisomerized products 4.2 and 4.3 from carbene 4.19.**

Depending on the nature of the substituents $R^2$ and $R^3$ on the alkene, two pathways are possible. Cyclobutane cation 4.20 can either fragment to produce cyclopentene 4.21 (pathway a) or undergo another [1,2]-alkyl shift (dyotropic rearrangement\textsuperscript{201}) to give the spirocycle 4.22, which fragments to produce carbene 4.23 (pathway b). Elimination of the metal fragment from 4.21 and 4.23 affords diene 4.2 (which is originated from the cleavage of one C-C bond), and diene 4.3 (which comes from the cleavage of two C-C bonds), respectively.

\textsuperscript{201} Term defined in 1972 by M. T. Reetz, referring to a new class of pericyclic isomerization reactions, where two $\sigma$-bonds simultaneously migrate intramolecularly. For a review on the subject, see: Fernández, I.; Cossío, F. P.; Sierra, M.; \textit{Chem. Rev.}, 2009, 109, 6687.
Despite the fact that the mechanism in scheme 4.4 is described as proceeding in a step-wise fashion, this is to give the reader a detailed analysis of the bond-breaking and bond-forming steps. DFT calculations showed that the product conversions from 4.19 to either 4.21 or 4.23 are actually direct processes\(^{202}\). In this regard, 4.19 should be seen as a delocalized highly distorted cyclopropylmethyl/cyclobutyl/homoallyl carbocation\(^{203}\) (often called non-classical carbocation\(^{204}\)).

### 4.1.2 Synthesis of alkenylmethylene-cyclopentanes (4.4)

1,6-enynes containing allylsilanes and allylstannanes moieties can be rearranged upon treatment with gold\(^{205}\) (but also a wide range of other metal transition catalysts\(^{206}\)) to afford alkenylmethylene-cyclopentane derivatives (scheme 4.5).

![Scheme 4.5: Examples of gold mediated formation of alkenylmethylene-cyclopentanes from allyltrimethylsilane derivatives.](image)

The mechanism of this reaction can be seen as a concerted process that starts with the gold activation of alkyne 4.25, followed by the formation of a gold stabilized vinyl cation 4.26 and addition of the alkene moiety to generate the carbocationic intermediate 4.28.


Alternatively, this cyclization might proceed in a stepwise manner involving initial generation of cyclopropyl 4.27, which undergoes ring opening by the proximal alkene to generate the same intermediate 4.28. Elimination of either a silyl cation (\(G = \text{TMS}\)) or a tributyltin cation (\(G = \text{Bu}_3\text{Sn}\)) affords the vinyl gold complex 4.29. Protodemetalation takes place to give the observed diene 4.4 and regenerates the active catalyst (scheme 4.6).

![Scheme 4.6: Mechanism of gold-catalyzed cyclization of allylstannanes and allylsylanes derivatives.](image)

Although other metals, such as Pt and Ru, are capable of producing the same products under very similar conditions\(^\text{206}\), these metals can also react through the intermediacy of metalacycles by simultaneous coordination to the alkyne and alkene moieties. Subsequent \(\beta\)-hydrogen elimination from a proximal alkyl substituent affords a formal alder-ene product, distinct from diene 4.4 by the geometry of the double bond carrying the \(R^1\) group (not shown, metalacycle mechanism proved by deuteriation experiments).

### 4.1.3 Hydroxy-, alkoxy-, amine- and (hetero)aryl- addition/5-exo cycloisomerization process with 1,6-enynes (4.5)

1,6-enynes can react with alcolates, hydroxy groups\(^\text{207}\), amines\(^\text{208}\) or electron rich (hetero)aromatic rings\(^\text{209}\) in an inter- or intramolecular fashion, in the presence of Au(I)


catalysts. While the combination of PPh₃AuCl/AgSbF₆ and AuMe(PPh₃)/protic acid affords similar results, gold catalysts carrying bulky biphenyl phosphine ligands have proven to be the catalyst of choice for this transformation. Representative examples of reaction scope are shown below (scheme 4.7). Two of the examples shown, using a phosphite ligand on gold, were performed at -50°C to avoid their fast skeletal rearrangement to dienes of type 4.2/4.3.

Scheme 4.7: Representative scope of nucleophilic addition/ cyclization of 1,6-enynes using different gold catalysts.

Two possible mechanisms for the nucleophilic addition/cyclization can be imagined. The first proceeds in a highly concerted manner implicating simultaneous attack of the gold-alkyne complex by the alkene with a concomitant addition of the nucleophile, as shown in 4.30 (scheme 4.8). The second one proceeds step-wise with formation of the cyclopropane intermediate 4.19, followed by addition of the nucleophile. The final step is the protodemetalation of the vinyl gold complex 4.31 (Scheme 4.8).

4.1.4 Formal intramolecular [4+2] cycloadditions of alkenes with enynes and arylalkynes (4.6)

For 1,6-enynes 4.1 where R¹ is an aromatic ring derivative or a vinyl moiety, a second reaction takes place besides the expected 5-exo cycloisomerization (cf. section 4.1.1). This novel process corresponds to a formal [4+2] cycloaddition of alkenes with enynes or aryl alkynes. Such examples are illustrated below (scheme 4.9).

Scheme 4.9: Two examples of the gold-catalyzed intramolecular [4+2]-cycloaddition of 1,6-enynes.

The proposed reaction mechanism follows an intramolecular olefin cyclopropanation from the gold activated enyne 4.1, thus generating the gold carbene 4.19, which has the proximal alkene or aryl ring nearby, favoring a subsequent nucleophilic attack and opening of the cyclopropane ring in a reminiscent process of the Nazarov cyclization (or analogously to the previous mechanisms, a concerted alkyl addition/Friedel-Crafts alkylation sequence,

---

as depicted in 4.32) to give intermediate 4.33. Subsequent loss of a proton, followed by protodeauration, furnishes tricycle 4.6 and closes the catalytic cycle (scheme 4.10).

Scheme 4.10: Gold(I)-catalized formal [4+2]-cycloaddition of 1,6-enynes.

### 4.1.5 Synthesis of methylenecyclohexenes (4.7)

Further investigation of 1,6-enynes cycloisomerization processes performed by Echavarren et al. revealed that certain compounds, such as 4.34, 4.37 and 4.39, did not afford only cyclopentenes as it could be initially expected (cf. section 4.1.1), but methylenecyclohexenes in different extents (scheme 4.11).

Scheme 4.11: Depending on the substitution pattern of 1,6-enynes, different cycloisomerized products can be obtained. One of the possible outcomes are methylenecyclohexene derivatives.

---

The mechanism for the methylenecyclohexenes synthesis is presented in scheme 4.12. DFT calculations\textsuperscript{211,212} indicate that the pathway lowest in energy follows a 5-exo cyclization to give the gold carbene 4.19 (as in scheme 4.4). Then, a [1,2]-alkyl shift is responsible for the ring expansion to the 6-membered ring, which is accompanied by allylic cation formation to give access to intermediate 4.42. Subsequent elimination of the gold fragment LAu\textsuperscript{+} generates methylenecyclohexene 4.7 (scheme 4.12b).

![Reaction mechanism based on DFT calculations for the formation of methylenecyclohexene 4.7 from the gold-mediated cycloisomerization of 1,6-ene](image)

Scheme 4.12: Reaction mechanism based on DFT calculations for the formation of methylenecyclohexene 4.7 from the gold-mediated cycloisomerization of 1,6-ene 4.1.

From what can be seen until this point, in comparison to section 4.1.1, the substitution of the 1,6 enynes greatly influenciates the outcome of the cycloisomerization process. Temperature also plays an important role in product selectivity between 5- and 6-membered rings. For instance, cyclization of enyne 4.34 at -15 °C in the presence of PPh\textsubscript{3}Au(NCCH\textsubscript{3})SbF\textsubscript{6} produces a mixture of 10:1 of 4.35:4.36, whereas at 0 °C, this ratio drops to 7:1 (scheme 4.11) and at room temperature (23 °C) a 1:1 mixture of 4.35:4.36 is observed. The catalyst is also an important parameter. For instance, PtCl\textsubscript{4}, AuCl, JohnPhosAu(NCCH\textsubscript{3})SbF\textsubscript{6}, [2,4-(iBu\textsubscript{2})\textsubscript{2}PhO\textsubscript{3}PAuCl/ AgSbF\textsubscript{6} afforded virtually only the diene 4.36 at room temperature.

### 4.1.6 Synthesis of bicyclo[4.1.0]heptenes (4.8)

The first two examples of synthesis of bicyclo[4.1.0]heptenes promoted by gold catalysis were reported by Echavarren et al. in 2004\textsuperscript{198} with remarkable efficiency and mild reaction conditions (DCM, 20 °C). Subsequent work from Michelet et al.\textsuperscript{213} reported their

\textsuperscript{212} Curiously, not all authors have considered these calculations and describe an alternative mechanism higher in energy through a 6-endo dig cyclization. See: (a) Gorin, D.; Sherry, B. D.; Toste, D.; Chem. Rev., 2008, 108, 8, 3351 (scheme 22, p 3365). (b) ref. 197b (scheme 34, p. 2283).

enantioselective synthesis, with low to moderate yields, but excellent enantioselectivity (scheme 4.14).

\[
\begin{align*}
X & \quad \text{R1} \\
\text{R2} & \quad \text{R3} \\
\text{R4} & \quad X
\end{align*}
\]

(R)-4-MeO-3,5-(tBu)2MeOBIPHEP(AuCl)₂ (3 mol%)/AgOTf (6 mol%) toluene, 0 °C

\[
\begin{align*}
34 \text{ %}, 98 \text{ %ee} & \quad 59 \text{ %}, 95 \text{ %ee} & \quad 51 \text{ %}, 90 \text{ %ee} & \quad 74 \text{ %}, 98 \text{ %ee (T = 60 °C)}
\end{align*}
\]

Scheme 4.13: Representative scope of the enantiomeric synthesis of bicyclo[4.1.0]heptenes.

The reaction mechanism is believed to proceed through the 6-*endo* dig cyclization, which generates the highly electrophilic gold carbene 4.43. Subsequent [1,2]-hydride shift generates intermediate 4.44, that eliminates the gold fragment LAu⁺ and closes the catalytic cycle with concomitant formation of bicyclo[4.1.0]heptene 4.8.

\[
\begin{align*}
\begin{array}{c}
\begin{array}{c}
X \\
\text{R1} \\
\text{R2} \\
\text{R3} \\
\text{R4}
\end{array}
\end{array} & \quad \text{6-endo dig} & \quad \begin{array}{c}
\begin{array}{c}
\text{H} \\
\text{AuL}
\end{array}
\end{array} & \quad [1,2]-\text{hydride shift} & \quad \begin{array}{c}
\begin{array}{c}
\text{X} \\
\text{R1} \\
\text{R2} \\
\text{R3} \\
\text{R4}
\end{array}
\end{array} & \quad \text{Elimination} & \quad \begin{array}{c}
\begin{array}{c}
\text{X} \\
\text{R1} \\
\text{R2} \\
\text{R3} \\
\text{R4}
\end{array}
\end{array}
\end{array}
\end{align*}
\]

Scheme 4.14: Reaction mechanism for the gold-promoted cycloisomerization of 1,6-enynes generating bicyclo[4.1.0]heptenes 4.8.

In order to rationalize the ease of 1,6-enyne 4.1 to follow either a 5-*exo* or a 6-*endo* cyclization pathway, DFT calculations on the reaction of (E)-6-octen-1-yn with the gold complex [Au(PH₃)]⁺ were performed as a model system. They show the formation of a highly polarized (η¹-alkyne)gold complex 4.45, which reacts readily with the alkene in a 5-*exo* cyclization mode with a very small activation energy of 0.1 kcal.mol⁻¹ to afford intermediate 4.46. The structure of this intermediate can be seen as the canonical form 4.47, corresponding to a gold(I) stabilized homoallylic carbocation. The activation energy for the 6-*endo*-dig process to give the gold carbene 4.48 is significantly higher, equal to 6.1 kcal.mol⁻¹, which suggests that the *exo* cyclization is favored with gold(I) catalysts of type R₃PAu⁺, at
least for substrates related to (E)-6-octen-1-yne. For comparison purposes, the calculated activation energies for the analogous transformation with [Pt(H₂O)Cl₂] are 10.3 and 11.2 kcal.mol⁻¹ for 5-exo and 6-endo cyclizations, respectively.

Scheme 4.15: a) DFT calculations showing the reaction coordinate for the cyclization of (E)-6-octen-1-yne with [AuPH₃]⁺. b)-d) Selected bond lengths in Å for 4.45-4.48. Values in parentheses are for the comparison with the Pt complex trans-Pt(H₂O)Cl₂.

4.1.7 Nucleophilic addition/6-endo cyclization process with 1,6-enynes (4.9)

Following the same pattern of nucleophilic additions to cyclopropyl gold carbenes 4.19, nucleophiles can add to cyclopropyl gold carbenes 4.43 on the carbon of cyclopropyl carrying R³ and R⁴ groups to afford cyclohexene derivatives 4.9²¹⁴. The reaction scope is exhibited below (scheme 4.16).

The mechanism starts by the 6-endo cyclization of the gold-activated triple bond of the 1,6-enzyme 4.1, thus generating cyclopropyl gold carbene 4.43. Subsequent nucleophilic addition to intermediate 4.43 produces vinyl-gold 4.50, which after protodemetalation affords cyclohexene derivative 4.9 (scheme 4.17). In analogy to previous mechanisms, the nucleophilic addition can also be seen as proceeding in a concerted manner through the stabilized vinyl gold cation 4.49.

The regioselectivity between a 5-exo and a 6-endo cycle with the nucleophilic attack, furnishing 4.5 and 4.9, respectively, is mostly dictated by the substitution on the alkyne moiety. Internal alkynes (R<sup>1</sup> ≠ H) generally afford 6-endo cyclization products, although the 5-exo cyclization mode is also followed for internal alkynes with alkenes moieties substituted at C-2 (R<sup>2</sup> ≠ H).
4.1.8 Synthesis of bicyclo[3.2.0]heptenes (4.10)

Although initially postulated as possible intermediates in the formation of dienes 4.2, cyclobutenes were immediately discarded upon examination of the low energies implicated in the supposed conrotatory ring opening necessary to afford dienes 4.2. The energy calculated (DFT methods) to ring-open an hypothetical cyclobutene formed during this process would correspond to an extremely fast process even at low temperatures such as -63°C \( \Delta G_{298}^\circ = 21.7 \text{ kcal/mol} \) for single cleavage mechanism using Cyclohexyl-JohnphosAu(NCCH\(_3\))SbF\(_5\)), thus corresponding to an unrealistic low activation energy when compared to the activation energy for the ring-opening of close-related cyclobutenes \( \Delta G_{298}^\circ = 25.6 \text{ kcal/mol} \) for the conrotatory ring-opening of bicyclo[3.2.0]hept-5-ene to 1-vinyl-1-cyclopentene\(^{202}\).

Despite this fact, cyclobutenes are observed in gold cycloisomerization reactions of 1,6-enynes. Indeed, upon consideration of small changes in the substrate structure, through the interplay with the gold catalyst employed, one is capable of modulating the system reactivity and gain entry to cyclobutenes. Examples of such a modulation is the use of ynamide derivatives 4.51, which generates unstable cyclobutenes 4.52 and have to be immediately hydrolyzed to cyclobutanones 4.53\(^{215}\); and from amide- and ester-tethered 1,6-enynes 4.54 that have the alkyne extremity substituted by an aryl ring (i.e., \( R_1 = \text{Ph or tolyl} \))\(^{216}\) (scheme 4.16).

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Scheme 4.16: Examples of cyclobutenes accessed by gold catalysis, either as a) intermediates or b) products.

The mechanism proposed and supported by DFT calculations\textsuperscript{198,214} on the enyne 4.54 is proposed to start with the gold activation of the alkyne, followed by either the 6-endo or the 5-exo-syn cyclization, which affords intermediates 4.56 (analogous to 4.43) and 4.58 (analogous to 4.19), respectively, depending on the substitution pattern on the 1,6-enyne 4.54. Each intermediate, 4.56 and 4.58, undergoes a [1,2]-alkyl shift to generate, respectively, carbocations 4.57 and 4.60, which upon elimination of the LAu\textsuperscript{+} fragment produce cyclobutene 4.55.

Scheme 4.17: Mechanism of the gold-mediated cyclobutene synthesis from 1,6-enynes. The reaction pathway depends on the enyne substituents.
The reaction mechanism pathway is believed to diverge upon the substituents on the 1,6-enzyme. DFT calculations\textsuperscript{214} on enynes having an ester- or amide- tether and an aryl group at the alkyne (\textit{i.e.} R\textsuperscript{1} = Ph or tolyl) were found to have a 6\textit{-endo} transition state lower in energy than the two possible transition states for the formation of the 5\textit{-exo-syn} and 5\textit{-exo-anti}\textsuperscript{217} products by 14.0 and 2.4 kcal.mol\textsuperscript{-1}, respectively (values calculated for the amide tethered 7-phenyl-1,6-enzyme)\textsuperscript{218}, which suggests that the 6\textit{-endo} cyclization is the preferred pathway. On another hand, DFT calculations\textsuperscript{198} on enynes 4.54 having no carbonyl (Y = H\textsubscript{2}) show that the 6\textit{-endo} transition state is 6.0 kcal.mol\textsuperscript{-1} higher in energy to that arising from the 5\textit{-exo} cyclization product (see scheme 4.15a), which suggests that, in this case, the 5\textit{-exo} pathway is the preferred one. Also noteworthy is the fact 5\textit{-exo-syn} 4.58 and 5\textit{-exo-anti} 4.59 products tend to not interconvert readily, with a high difference of free energy \(\Delta G^\#\) = 24.7 kcal.mol\textsuperscript{-1} (calculated for \((E)\)-6-octen-1-yn with AuPH\textsubscript{3}\textsuperscript{+}) that can be attributed to the loss of conjugation between the gold carbene and the cyclopropane\textsuperscript{202}. Furthermore, the relative orientation of the phosphine gold(I) moiety in the cyclopropyl gold carbene complex largely affects the rearrangement to cyclobutenes. 5\textit{-exo-anti} cyclopropyl gold carbene affords generally dienes 4.2 and/or 4.3 (\textit{cf.} section 4.1.1) with \(\Delta G^\#\) = 9.1 kcal.mol\textsuperscript{-1} to go from 4.19 to 4.21 (single cleavage, value calculated \((E)\)-6-octen-1-yn and H\textsubscript{3}PAu\textsuperscript{+}) and \(\Delta G^\#\) = 14.2 kcal.mol\textsuperscript{-1} to go from 4.19 to 4.23 (double cleavage, value calculated for the same substrate and gold catalyst)\textsuperscript{202}, while 5\textit{-exo-syn} cyclopropyl gold carbene 4.58 has a difference of free energy \(\Delta G^\#\) = 7.8 kcal.mol\textsuperscript{-1} to go to 4.60 (again, value for \((E)\)-6-octen-1-yn and AuPH\textsubscript{3}\textsuperscript{+}). In contrast, formation of 5\textit{-exo-anti} 4.59 (or 4.19) and 5\textit{-exo-syn} 4.58 cyclopropyl gold carbenes from enyne 4.1 have \(\Delta G^\#\) of 0.1 and 9.4 kcal.mol\textsuperscript{-1}, respectively. These differences of values in free energies are consistent with two common experimental observations: one, that 1,6-enynes generally produce dienes 4.2 and/or 4.3 and not cyclobutenes 4.10. Two, that nucleophiles usually add in a \textit{trans} fashion to the gold alkyne complex, which generates 5\textit{-exo-anti} cyclopropyl gold carbene and not in a \textit{cis} fashion, which would afford the 5\textit{-exo-syn} cyclopropyl gold carbene, but which is eventually accessible in some cases (either from an hypothetical nucleophilic \textit{cis}-attack in the gold stabilized vinyl cation formed\textsuperscript{202} or lowering

\textsuperscript{217} Syn- and anti- product are directly related to the positioning of the nucleophilic double bond relative to the gold fragment, \textit{cis} or \textit{trans}, respectively.

\textsuperscript{218} A possible concurrent pathway corresponding to the direct transformation of the ester or amide-tethered enyne 4.54 to cyclobutene 4.55 might also be operating, since the difference in free energy \(\Delta G^\#\) = 15.3 kcal.mol\textsuperscript{-1} is virtually identical to the one of the transformation from enyne 4.54 to the 6\textit{-endo} intermediate 4.56, being \(\Delta G^\#\) = 15.2 kcal.mol\textsuperscript{-1} (values calculated to the amide tethered 7-phenyl-1,6-enzyme).
the energy barrier of *anti*-to-*syn* isomerization). Although formation of the 5-*exo*-cis product is less favorable, it can compete if substitution at the alkene and/or alkyne disfavors the skeletal rearrangement to 4.2 and/or 4.3. For example, DFT calculations on 1-phenyl-6-hexen-1-yne show that only 8.6 kcal.mol⁻¹ is required for the *anti*-to-*syn* isomerization²¹⁰b. Finally, once the 5-*exo*-syn gold carbene 4.58 has been formed, then the most likely outcome is cyclobutene 4.10.

Also noteworthy, the formation of isomerized cyclobutenes 4.63²¹⁰b, corresponding to the proton elimination of intermediate 4.61 (supposed coming from the same pathway as intermediate 4.57) followed by protodeauration has also been observed to internal alkynes substituted with a phenyl ring (scheme 4.18).

\[ \begin{array}{c}
\text{Scheme 4.18: Two examples observed for the corresponding double bond isomerized} \\
cyclobutenes 4.63.
\end{array} \]

### 4.2 RESULTS AND DISCUSSION

#### 4.2.1 The Initial idea

At the beginning of this project, we were interested in the potential cascade reaction shown in scheme 4.19, whereby a substituted 1,6-enzyme of type 4.64 could afford bicyclic product of type 4.65 and/or 4.66 (scheme 4.19).
Scheme 4.19: Hypothetical cascade reaction initially envisaged in this project.

To test our initial hypothesis, substrate 4.64a was synthesized starting from ethyl bromo malonate 4.67 as described below (scheme 4.20)

```
\[ \text{EtO}_2\text{C} \text{CO}_2\text{Et} \xrightarrow{\text{Br, NaH, THF, rt}} \text{EtO}_2\text{C} \text{CO}_2\text{Br} \xrightarrow{\text{Zn, AcOH, rt}} \text{EtO}_2\text{C} \text{CO}_2\text{Et} \]
```

Scheme 4.20: Synthesis of substrate 4.64a.

Enyne 4.64a was submitted to the usual metal catalysts employed in enynes cycloisomerization, PPh₃Au(NCCH₃)SbF₆, PPh₃AuNTf₂ and PtCl₂ (table 4.1). The results obtained were not encouraging. Substrate 4.64a afforded only cyclopentadiene derivative 4.73a, as one could expect from either a single or double cleavage (cf. section 4.1.1), with very slow kinetics.

```
\[ \text{EtO}_2\text{C} \text{CO}_2\text{Et} \xrightarrow{\text{ML}_n, \text{solvent, temp.}} 4.72a + 4.73a \]
```

Table 4.1: First trials of the gold-mediated double cyclization on enyne 4.64a.

<table>
<thead>
<tr>
<th>entry</th>
<th>MLₙ</th>
<th>conditions</th>
<th>time</th>
<th>conversion</th>
<th>ratio 4.72a:4.73a</th>
<th>yield%</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>PPh₃Au(NCCH₃)SbF₆ (4 mol%)</td>
<td>CDCl₃, rt</td>
<td>7 days</td>
<td>100%</td>
<td>0:1</td>
<td>100%</td>
</tr>
<tr>
<td>2</td>
<td>PPh₃AuNTf₂ (4 mol%)</td>
<td>CDCl₃, rt</td>
<td>12 days</td>
<td>50%</td>
<td>1:10</td>
<td>100%</td>
</tr>
<tr>
<td>3</td>
<td>PPh₃AuNTf₂ (4 mol%)</td>
<td>CDCl₃, 60 °C</td>
<td>51 h</td>
<td>81%</td>
<td>1:2</td>
<td>100%</td>
</tr>
<tr>
<td>4</td>
<td>PtCl₂ (10 mol%)</td>
<td>toluene, 80 °C</td>
<td>30 h</td>
<td>50%</td>
<td>0:1</td>
<td>&lt;100%</td>
</tr>
</tbody>
</table>

* determined by ³⁵Cl NMR of the crude reaction mixture. ²estimated by NMR. ³unidentified product present at the reaction mixture.
The reaction required 7 days at room temperature to reach completion and afford the 5-membered cycle \( \text{4.73a} \) as the only product with the more reactive \( \text{PPh}_3\text{Au(NCCH}_3\text{)}\text{SbF}_6 \) (entry 1, table 4.1). The use of \( \text{PPh}_3\text{AuNTf}_2 \) at room temperature gave only 50% conversion after 12 days, in a slightly different 10:1 ratio of \( \text{4.72a: 4.73a} \) (entry 2, table 4.1) and a 1:2 ratio under reflux of \( \text{CDCl}_3 \) (entry 3, table 4.1). The use of catalyst \( \text{PtCl}_2 \) in toluene at 80°C also afforded only 50% conversion for the formation of compound \( \text{4.73a} \) together with unidentified byproducts (entry 4, table 4.1). No trace of a double cyclized product of type \( \text{4.65 or 4.66} \) was observed.

With the unsubstituted enyne \( \text{4.64a} \) proving to be quite reluctant to react under our reaction conditions, we decided to focus our studies on the internal enyne \( \text{4.64b} \), anticipating that this substrate would be more reactive than \( \text{4.64a} \) due to the increased electron density imposed by the heptynyl chain.

Substrate \( \text{4.64b} \) was synthesized from vinyl bromo malonate \( \text{4.69} \) through a Sonogashira cross-coupling reaction, which afforded mono-substituted malonate \( \text{4.74} \). This intermediate was subsequently submitted to standard propargylation conditions to afford \( \text{4.64b} \) (scheme 4.21):
Scheme 4.22: Synthesis of enynes 4.64c-g.

Table 4.2: Second trial on the cascade reaction using 1,6-enynes afforded cycloisomerized products in very different ratios, with a notable preference towards the 6-membered cyclic product in the case of PPh₃AuNTf₂.

<table>
<thead>
<tr>
<th>entry</th>
<th>[Au]</th>
<th>time</th>
<th>conversion</th>
<th>ratio 4.72b : 4.73b</th>
<th>isolated yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>PPh₃Au(NCCH₃)SbF₆</td>
<td>3 h</td>
<td>100%</td>
<td>1.9 : 1</td>
<td>58%</td>
</tr>
<tr>
<td>2</td>
<td>PPh₃AuNTf₂</td>
<td>18 h</td>
<td>100%</td>
<td>14.3 : 1</td>
<td>73%</td>
</tr>
</tbody>
</table>

* ratio determined by ¹H NMR of the crude reaction mixture

From these preliminary studies, we recognized the difficulty of performing our initial idea on the cascade reaction using 1,6-enynes. Nevertheless, from this serependity, we became very interested on the effects governing the selectivity of the cycloisomerization process of 1,6-enynes towards the ring expanded 6-membered cycles.

In order to observe if the same selectivity difference seen for 4.64b was present with other substrates, we synthesized enynes 4.64c-g (scheme 4.22) and submitted them to both catalysts, PPh₃Au(NCCH₃)SbF₆ and PPh₃AuNTf₂.
Despite the small number of molecules tested, the same trend was observed (table 4.3). Accordingly, TMS substituted enyne 4.64c cyclizes with different ratios and in very long reaction times for each gold catalyst (entries 1 and 2, table 4.2), with NTf$_2$ counter-ion favoring the 6-membered ring. A substantial increase in the formation of the 6-membered ring using PPh$_3$AuNTf$_2$ with substrate 4.64c, when compared to enyne 4.64a, suggests that the TMS loss occurs, at least in this case, after cyclization. (entry 2, table 4.3).

Substrate 4.64d afforded a mixture 2.55:1 of compounds 4.72d:4.73d when treated with the gold catalyst PPh$_3$Au(NCCH$_3$)SbF$_6$ (entry 3, table 4.3) and 4.72d as the only product when treated with PPh$_3$AuNTf$_2$ (entry 4, table 4.3). Other catalysts in the same conditions (4mol%, CDCl$_3$, rt) were also tried with this substrate: i) [2,4-(tBu)$_2$PhO]$_3$PAuCl/AgSbF$_6$ afforded a 90% yield of the 6-membered ring 4.72d in less than 30 min. (estimated by $^1$H NMR of the reaction crude, no sign was attributable to the 5-membered ring was detected); ii) XphosAuNTf$_2$ afforded a 2.5:1 mixture of 4.72d:4.73d and iii) AuCl$_3$ did not produce any cyclized product, with starting material being recovered after 19h of reaction.

Substrate 4.64e, possessing a para-methoxy group on the phenyl ring, afforded the 6-membered cyclic adduct as the only product using both catalysts (entries 5 and 6, table 4.3) and substrate 4.64f, possessing a para-chloro group on the phenyl ring was insensitive to both catalysts (entries 7 and 8, table 4.3). On another hand, when both alkyne moieties were substituted by a phenyl ring, the reaction proved to be very sluggish using both catalysts. A 2:1 mixture of 4.72g:4.73g was obtained after 6 days for catalyst PPh$_3$Au(NCCH$_3$)SbF$_6$ (entry 9, table 4.3) and was not complete, even at reflux of chloroform for 7 days, for catalyst PPh$_3$AuNTf$_2$ (entry 10, table 4.3).

Although more examples are necessary to consolidate the trend observed here, the previous results suggest that the interplay of the gold catalyst PPh$_3$AuNTf$_2$ and a vinyl alkyne appendage on the 1,6-enyne are capable of drastically improving the selectivity of the cycloisomerization process for the 6-membered cyclic adducts.

To corroborate our suspicions, similar findings were also recently uncovered in the study of 1,6-diynes$^{219}$, where the 6-endo cyclization was greatly favored in regard to the 5-exo process using Et$_3$PAuNTf$_2$, when compared to Et$_3$PAuSbF$_6$ (interestingly enough, the substrates prepared in this work can be seen as 1,7-diynes and therefore, are structurally close to 1,6-diynes reported in ref. 219).

Scheme 4.23: Synthesis of substrate 4.64h.

Table 4.3: Results of gold-catalyzed cycloisomerization of 1,6- enynes with branched vinyl-alkyne moieties.

<table>
<thead>
<tr>
<th>entry</th>
<th>R1, R2</th>
<th>[Au]</th>
<th>time</th>
<th>conversion</th>
<th>ratio 4.72 : 4.73&lt;sup&gt;a&lt;/sup&gt;</th>
<th>isolated yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>H, TMS (4.64c)</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;Au(NCCH&lt;sub&gt;3&lt;/sub&gt;)SbF&lt;sub&gt;6&lt;/sub&gt;</td>
<td>12 days</td>
<td>100 %</td>
<td>1 : 10 (R&lt;sup&gt;2&lt;/sup&gt; = H, 4.72a : 4.73a)</td>
<td>62 %</td>
</tr>
<tr>
<td>2</td>
<td>4.64c</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;AuNTf&lt;sub&gt;2&lt;/sub&gt;</td>
<td>12 days</td>
<td>68 %</td>
<td>9 : 10 (R&lt;sup&gt;2&lt;/sup&gt; = H, 4.72a : 4.73a)</td>
<td>&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
<tr>
<td>3</td>
<td>H, Ph (4.64d)</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;Au(NCCH&lt;sub&gt;3&lt;/sub&gt;)SbF&lt;sub&gt;6&lt;/sub&gt;</td>
<td>2 h 20 min</td>
<td>100 %</td>
<td>2.55 : 1 (4.72d : 4.73d)</td>
<td>60 %</td>
</tr>
<tr>
<td>4</td>
<td>4.64d</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;AuNTf&lt;sub&gt;2&lt;/sub&gt;</td>
<td>18 h</td>
<td>100 %</td>
<td>1 : 0 (4.72d : 4.73d)</td>
<td>99 %</td>
</tr>
<tr>
<td>5</td>
<td>H, 4-(OMe)-Ph (4.64e)</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;Au(NCCH&lt;sub&gt;3&lt;/sub&gt;)SbF&lt;sub&gt;6&lt;/sub&gt;</td>
<td>2 h</td>
<td>100 %</td>
<td>1.0 (4.72e : 4.73e)</td>
<td>99%</td>
</tr>
<tr>
<td>6</td>
<td>4.64e</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;AuNTf&lt;sub&gt;2&lt;/sub&gt;</td>
<td>7 h</td>
<td>100 %</td>
<td>1.0 (4.72e : 4.73e)</td>
<td>99%</td>
</tr>
<tr>
<td>7</td>
<td>H, 4-(Cl)-Ph (4.64f)</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;Au(NCCH&lt;sub&gt;3&lt;/sub&gt;)SbF&lt;sub&gt;6&lt;/sub&gt;</td>
<td>48 h</td>
<td>0 %</td>
<td>only starting material</td>
<td>&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
<tr>
<td>8</td>
<td>4.64f</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;AuNTf&lt;sub&gt;2&lt;/sub&gt;</td>
<td>48 h</td>
<td>0 %</td>
<td>only starting material</td>
<td>&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
<tr>
<td>9</td>
<td>Ph, Ph (4.64g)</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;Au(NCCH&lt;sub&gt;3&lt;/sub&gt;)SbF&lt;sub&gt;6&lt;/sub&gt;</td>
<td>6 days</td>
<td>100 %</td>
<td>2 : 1 (4.72g : 4.73g)</td>
<td>91 %</td>
</tr>
<tr>
<td>10</td>
<td>4.64g</td>
<td>PPh&lt;sub&gt;3&lt;/sub&gt;AuNTf&lt;sub&gt;2&lt;/sub&gt;</td>
<td>7 days</td>
<td>30 %&lt;sup&gt;c&lt;/sup&gt;</td>
<td>4.3 : 1 (4.72g : 4.73g)</td>
<td>&lt;sup&gt;b&lt;/sup&gt;</td>
</tr>
</tbody>
</table>

<sup>a</sup> ratio determined by 1H NMR of the crude reaction mixture. <sup>b</sup> Not isolated. <sup>c</sup> under reflux (60 °C).

In order to test this hypothesis with some control experiments, we considered compound 4.64h with an extra carbon between the vinyl alkyne branch and the alkene moiety. This substrate was synthesized from 2-(chloromethyl)-3-chloropropene by a cross coupling reaction with heptyne, followed by the exchange of the chlorine atom by a iodine, following typical Finkelstein reaction conditions and a subsequent secondary alkylation of propargyl malonate, to afford 4.64h (scheme 4.23).
The previous large difference in the product distribution observed for 4.64b disappeared with 4.64h, which upon treatment with catalysts PPh₃Au(NCCH₃)SbF₆ and PPh₃AuNTf₂ afforded a 3.3:1 and a 2:1 mixtures of 4.72h:4.73h, respectively (table 4.4).

<table>
<thead>
<tr>
<th>entry</th>
<th>[Au]</th>
<th>time</th>
<th>conversion</th>
<th>ratio 4.72h : 4.73h</th>
<th>isolated yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>PPh₃Au(NCCH₃)SbF₆</td>
<td>5 min</td>
<td>100%</td>
<td>3.3 : 1</td>
<td>99 %</td>
</tr>
<tr>
<td>2</td>
<td>PPh₃AuNTf₂</td>
<td>8 h</td>
<td>100%</td>
<td>2 : 1</td>
<td>99 %</td>
</tr>
</tbody>
</table>

* ratio determined by ¹H NMR of the crude reaction mixture.

Table 4.4: Control experiment on the influence of a vinyl alkyne moiety.

In other efforts to reassure our suspicions on the need of a vinylalkyne moiety for the regioselectivity control of 1,6-enynes cycloisomerization, substrates 4.81-4.84 (synthesized as depicted in scheme 4.24) were submitted to catalysts PPh₃Au(NCCH₃)SbF₆ and PPh₃AuNTf₂. Interestingly, not only the difference in selectivity for the 5- and 6-membered cyclic products was not marked, but also extensive amounts of positional isomers of the double bond for the 6-membered cyclic frameworks were found at both 23 °C (room temperature) and at 0 °C for both catalysts (table 4.5).

Scheme 4.24 : Synthesis of substrates 4.81-4.84.
Table 4.5: Gold(I) mediated cycloisomerization reactions of 1,6-enynes without a stabilizing vinyl alkyne moiety on the alkene.
This confirmed our initial suspicions on the use of substituted vinyl alkyne moieties as directing groups for the regioselectivity of cycloisomerization, but also revealed an extra stabilizing effect from these groups against double bond isomerization (probably due to the conjugation with the alkyne introduced. For instance, substrates 4.64b and 4.83/4.64d and 4.84, are pairs of almost the same product, except for the alkyne moiety present in both 4.64b and 4.64d. They both afford with the use of PPh₃AuNTf₂ selective formation of 6-membered rings, whereas their counter parts, 4.83 and 4.84, respectively, afford mixtures of 5- and 6-membered rings with extensive amounts of double bond isomerization.

Interestingly, similar remote substituent control has also been recently reported on the enyne cross-metathesis/metallotropic [1,3]-shift sequence²²⁰ (scheme 4.25).

Scheme 4.25: Remote control on enyne cross-metathesis and metallotropic [1,3]-shift using similar remote substituents has also been reported in the literature²²⁰.

It is already an accepted concept that a charged transition-metal catalyst can have its activity, lifetime, stability and product selectivity being significantly influenced by the nature of the counterion. In subsequent efforts to understand the effect of counter ion influence on this cycloisomerization process, simple enyne 4.89 was submitted to gold catalysts PPh₃Au⁺ X⁻, with X = OTf, NTf₂, BF₄⁻, and SbF₆⁻. Interestingly, the increase in the 6-membered cyclic product followed the inverse order of counter-ion coordinating ability²²¹: SbF₆⁻ (less-coordinating) > BF₄⁻ > Tf₂N⁻ > TfO⁻ (more coordinating) (table 4.6), which is contrary to the previous results obtained for enynes 4.64b and d. This means that the counter-ion alone cannot explain the difference in the regioselectivity found for enyne 4.64b and d.

Table 4.6: Investigation on the effect of the counter-ion in the cycloisomerization of 1,6-enynes.

<table>
<thead>
<tr>
<th>entry</th>
<th>Counter-ion X</th>
<th>time</th>
<th>conversion</th>
<th>ratio 4.90:4.91⁵</th>
<th>yieldb</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>SbF₆⁻</td>
<td>5 min</td>
<td>100%</td>
<td>5.3 : 1</td>
<td>99 %</td>
</tr>
<tr>
<td>2</td>
<td>BF₄⁻</td>
<td>5 min</td>
<td>100%</td>
<td>4.0 : 1</td>
<td>99 %</td>
</tr>
<tr>
<td>3</td>
<td>NTI₂⁻</td>
<td>53 h</td>
<td>100%</td>
<td>3.6 : 1</td>
<td>99 %</td>
</tr>
<tr>
<td>4</td>
<td>OTI⁻</td>
<td>10 days</td>
<td>32 %</td>
<td>1.3 : 1</td>
<td>99 %</td>
</tr>
</tbody>
</table>

⁵ ratio determined by ¹H NMR of the crude reaction mixture. ⁶ estimated by ¹H NMR of the crude reaction mixture.
⁷ catalyst generated from the PPh₃AuCl/AgX mixture.

It is also noteworthy to mention results reported by Echavarren et al.,¹⁹⁷a, who found for the cycloisomerization reaction of the methyl malonate 4.34 (analogue to enyne 4.89) with catalyst PPh₃Au(NCCH₃)SbF₆ (2 mol%) in DCM at room temperature (23 °C), a 2:1 ratio of 6:-5- membered cyclic products, in stark contrast to entry 1 of table 4.6. The explanation for that, also given in one of Echavarren’s report,²¹¹, was due to degradation of the 6-membered cycle, thus making the relative amount of the 5-membered cycle increase. In our studies, we observed in 5 min, a 5.3:1 ratio for the 6-membered:5-membered cycles; in 15 min, the ratio dropped to 2.8:1 and after 4h, there was no more 6-membered product present in solution, but only the 5-membered cycle and uncharacterizable degradation.

At the same time we were investigating the counter-ion effect on the gold cationic fragment PPh₃Au⁺, we became interested in the possibility of finding a competent catalyst for the selective 6-membered product formation from 1,6-enynes, independent of the appendage on the alkene moiety. Such a catalyst would be of great synthetic value, since 6-membered cycles are rarely obtained alone, but generally with varying amounts of other products.¹⁹⁷a, ¹⁹⁷c, ²¹¹, ²²²

From our initial screening with the most common catalysts (table 4.7), no suitable complex was identified for the task, but some empirical informations can be interpreted: i) simple gold(I) and gold(III) salts are selective towards the 5-membered cycle (entries 1-3,

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table 4.2, the low conversions are probably associated with the fragile catalyst integrity due to high hygroscopicity of these salts). ii) Buchwald biphenyl ligands afford similar results (entries 4-8, table 4.2), but it suggests that product selectivity 4.90/4.91 is insensitive to bulky groups on the phosphorous atom (entries 4 and 6, table 4.2) and it increases with less hindered biphenyl ligands. The best result is obtained for the “extreme case” of just one phenyl ring (entries 4, 5 and 13, table 4.2). Also, no difference in selectivity was marked for biphenyl ligands with different counter-ions (entries 6 and 7, table 4.2). iii) phosphines with electron withdrawing atoms/groups seem to favorise the 6-membered

![Chemical reaction diagram](image)

<table>
<thead>
<tr>
<th>entry</th>
<th>[Au]</th>
<th>ratio 4.90:4.91</th>
<th>time</th>
<th>yield</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>AuCl</td>
<td>0 : 1</td>
<td>7 days</td>
<td>100% (30% conv.)</td>
</tr>
<tr>
<td>2</td>
<td>AuCl₃</td>
<td>0 : 1</td>
<td>56 h</td>
<td>100%</td>
</tr>
<tr>
<td>3</td>
<td>AuBr₃</td>
<td>0 : 1</td>
<td>5 days</td>
<td>100% (25% conv.)</td>
</tr>
<tr>
<td>4</td>
<td>C₁₃H₂₇P₄⁺AuNTI₂⁻</td>
<td>1.1 : 1</td>
<td>20 min</td>
<td>100%</td>
</tr>
<tr>
<td>5</td>
<td>C₁₃H₂₇P₄⁺AuNTI₂⁻</td>
<td>1.6 : 1</td>
<td>1 h 45 min</td>
<td>100%</td>
</tr>
<tr>
<td>6</td>
<td>C₁₃H₂₇P₄⁺AuNTI₂⁻</td>
<td>1.1 : 1</td>
<td>30 min</td>
<td>100%</td>
</tr>
<tr>
<td>7</td>
<td>C₁₃H₂₇P₄⁺Au(NCCH₃)₂SbF₆⁻</td>
<td>1.1 : 1</td>
<td>5 min</td>
<td>100%</td>
</tr>
<tr>
<td>8</td>
<td>C₁₃H₂₇P₄⁺AuNTI₂⁻</td>
<td>1.4 : 1</td>
<td>40 min</td>
<td>100%</td>
</tr>
<tr>
<td>9</td>
<td>(C₆H₅)₃PAuCl/ AgSbF₆</td>
<td>2.3 : 1</td>
<td>1 min</td>
<td>100%</td>
</tr>
<tr>
<td>10</td>
<td>C₂₁H₂₈C₁₂NTI₂⁻</td>
<td>2.1 : 1</td>
<td>11 days</td>
<td>&lt;100%b (75% conv.)</td>
</tr>
<tr>
<td>11</td>
<td>C₂₁H₂₈C₁₂NTI₂⁻</td>
<td>1.1 : 1</td>
<td>16 h</td>
<td>&lt;100%b</td>
</tr>
<tr>
<td>12</td>
<td>C₂₁H₂₈C₁₂NTI₂⁻</td>
<td>1.7 : 1</td>
<td>30 h</td>
<td>&lt;100%b</td>
</tr>
<tr>
<td>13</td>
<td>C₂₁H₂₈C₁₂NTI₂⁻</td>
<td>2 : 1</td>
<td>6 days</td>
<td>&lt;100%b</td>
</tr>
<tr>
<td>14</td>
<td>C₂₁H₂₈C₁₂NTI₂⁻</td>
<td>1.5 : 1</td>
<td>19 h</td>
<td>100%</td>
</tr>
<tr>
<td>15</td>
<td>C₂₁H₂₈C₁₂NTI₂⁻</td>
<td>2.4 : 1</td>
<td>2h 30</td>
<td>100%</td>
</tr>
<tr>
<td>16</td>
<td>C₂₁H₂₈C₁₂NTI₂⁻</td>
<td>1.6 : 1</td>
<td>6 days</td>
<td>&lt;100%b (53% conv.)</td>
</tr>
</tbody>
</table>

*a*estimated by 1HNMR of the crude reaction mixture. *b*subproducts present in a minor extent

**Table 4.2:** Catalyst screening for the regioselective 6-membered cyclic product 4.90.
ring formation (entries 8, 9 and 10, table 4.2). iv) NHC ligands do not give satisfactory results either. But once again, steric demanding groups increase the reaction selectivity (entries 11 and 12, table 4.2).

Three new catalysts were tried in this project (entries 14, 15 and 16, table 4.2). In entry 14, we supposed that a rigid tricyclic structure would be enough hindered to impart selectivity, but was inefficient. In entries 15 and 16 of table 4.2, phosphino N-aryl pyrroles (PAP) ligands reported by Beller et al. in palladium cross coupling reactions were for the first time, to the best of our knowledge, employed in gold chemistry. Beller describes these ligands in terms of reactivity to be close to Buchwald biphenyl phosphine ligands, but easier to prepare. Interestingly, in our case, we found the PAP-gold catalyst in entry 15 of table 4.2 to be more selective than any biphenyl phosphine ligand tried and even the most selective of all catalysts tried before (a similar result was obtained with phosphate ligand, entry 9, table 4.2). Here, the same trend of palladium cross-coupling reactions was observed, the more steric-demanding PAP ligand was more effective in catalysis (entries 15 and 16, table 4.2).

4.3 CONCLUSION

Although not closed, we started here two very interesting projects concerning i) the interplay of the counter-ion NTf2 from triphenyl phosphine gold catalysts with an alkynyl branch on the alkene moiety in 1,6-enynes, which suggests that high regioselectivity towards the 6-membered cycle can be obtained and ii) the use of PAP-ligands in gold catalysis, as efficient and potential selective ligands towards the 6-membered cycle.

Project i) still lacks a larger scope investigation but we believe we have begun a seminal work proving that high differences in regioselectivity for the cyclization of 1,6-enynes can be attained at room temperature, that favorizes the formation of a 6-membered ring. This is an interesting observation, since these 6-membered rings are rarely formed through gold-mediated cycloisomerization of 1,6-enynes (cf. section 4.1.5).

Project ii) reports for the first time the use of PAP-ligands in gold catalysis and reveal these ligands as potential sources for the regioselective 6-membered ring formation. The maxima of catalysis fits well here: “Catalysis is all about fine tuning”. In this regard, new substituents on these ligands must be screened, but these preliminary results look promising.

4.4 PERSPECTIVES AND FURTHER DEVELOPMENTS

In order to enlarge the scope of 1,6-enynes 4.92 that cyclize preferentially to a 6-membered ring, we believe that some substrates would be of particular interest to study, notably by changing the enyne tether to O, NTs, C(SO$_2$Ph)$_2$ and C(CH$_2$OAc)$_2$; and investigating the effect of ester groups on alkyne extremeties. As $R^1 = n$-C$_5$H$_{11}$ and Ph showed the best selectivities (table 4.2 and entries 3 and 4 of table 4.3), we suggest that these two groups are kept in following studies (scheme 4.21).

Scheme 4.21: Substrates 4.92 are possibly interesting substrates to investigate the regioselectivity cycloisomerization processes using triphenyl phosphine gold catalysts.

Having in mind the change in reactivity found for 1,6-enynes 4.92, a further interesting question is about the reaction mechanism. The trapping of reaction intermediates by a nucleophile during the gold mediated cycloisomerization would afford what kind of produt? This is a pertinent question because, despite suggested by DFT calculations that formation of methylene cyclohexenes 4.7 occurs through the ring expansion of intermediate 4.19, as shown in scheme 4.12, slight structure modifications such as the alkyne branch considered in enynes 4.92 can eventually change the reaction pathway. The cyclobutenes synthesis discussed in section 4.1.6 is an example of such a change in reaction mechanism.
Concerning the search for a competent gold catalyst to perform the regioselective 6-membered ring formation from 1,6-enynes, a second catalyst screening can be envisaged. A possible choice regrouping the two main properties observed in table 4.2 regarding selectivity: steric hindrance and the presence of electron-withdrawing groups suggests catalysts following the general structure 4.93, such as 4.94 and 4.95 (Figure 4.1).

Figure 4.1: Possible new catalysts for the gold regioselective synthesis of 6-membered cycles from 1,6-enynes.

Interestingly, recent mass-spectrometry studies report that gold can participate in metathesis reactions\textsuperscript{225}. One attempt to detect this reactivity of gold carbenoids in metathesis reactions can be envisaged from the cycloisomerization of 1,6-enynes in the presence of diazo compounds. Such reactivity is already well established for Ru\textsuperscript{226}, but has also been found in Ni complexes\textsuperscript{227}. One use of gold carbenoids generated from enynes 4.96 can be envisaged. Using enyne 4.96, a typical 5-exo cyclization would generate a methylcyclopropyl gold carbene that would possibly be trapped by a nucleophilic diazonium compound, as shown in 4.97 (Such 1,2 additions to gold carbenes have already been suggested in the literature\textsuperscript{228}). Subsequent elimination of nitrogen and the gold catalyst, would produce alkene 4.98, a product coming from a cycloisomerization/formal cross-metathesis sequence (scheme 4.22). Under the acidic reaction conditions, it is possible that for the use of the trimethylsilyl diazo methane as shown in scheme 4.22, that the TMS group inserted in the substrate is eliminated at some point en route to 4.98.


\textsuperscript{228} Some examples: (a) Davies, P. W.; Albrecht, S. J.-C.; Chem. Comm., 2008, 238. (b) see ref 209b (c) see ref 214.
Scheme 4.22: Envisaged route to a cycloisomerization/ formal cross-metathesis sequence.
CHAPTER 5: HYDROALKYLATION OF ALKYNYL ETHERS VIA A GOLD(I)-CATALYZED 1,5-HYDRIDE SHIFT/ CYCLIZATION SEQUENCE

This work was developed with contribution from the student Yann Odabachian. With the concern for a clear exposal of the subject, all the experiments performed will be presented herein, nevertheless, the ones done by the author of this thesis will be marked with a red star (*).

CHAPTER 5: HYDROALKYLATION OF ALKYNYL ETHERS VIA A GOLD(I)-CATALYZED 1,5-HYDRIDE SHIFT/ CYCLIZATION SEQUENCE

5.1 INTRODUCTION: C-H activation processes and intramolecular redox reactions.

5.1.1 C-H activation processes

Although intensive research on C-H activation contributed to dramatic advances in this area since the 1980’s, the ultimate goal of functionalizing strong C-H bonds still deserves to be named as a Holy Grail in organic chemistry. The importance of this process relies on the ubiquitous presence of these bonds in nature. For instance, considering the energy industry and economical issues, the utilization of methane (the main component of natural gas) from remote locations of many of the world’s known natural gas resources is hampered due to the difficulty of accessing these sites. Today’s solutions to this difficulty are both costly: one is the gas transportation; the other is the current procedure to convert hydrocarbon gas in a more readily transportable liquid, which consists in a two-steps process: the production of synthesis gas (CO and H₂) and its conversion to the desired products\(^\text{229}\). The development of efficient strategies for the direct conversion of methane to methanol or other liquid fuels or chemicals could thus significantly improve methane utilization\(^\text{230}\).

Concerning organic synthesis, a practical, predictable and efficient C-H activation process would represent a revolution\(^\text{231}\) in the logic underpinning the way we build molecules, where functionalization of a certain site is generally based on the proximity of heteroatoms and unsaturations. Viewing C-H bonds as functional groups can notably save functional groups manipulations and allow shortcuts in many steps of tedious procedures\(^\text{232}\).


In the context of C-H functionalization, the preceding C-H “activation” step is commonly understood as defined by Shilov and Shulp’in: “when we refer to the activation of a molecule, we mean that the reactivity of this molecule increases due to some action. (...) The main result of the activation of a C-H bond is the replacement of a strong C-H bond with a weaker, more readily functionalized bond”\textsuperscript{233a}.

In this regard, metal complexes\textsuperscript{233} have been extensively employed in the research of viable catalytic systems capable of efficiently activating C-H bonds. The mechanisms implicated in C-H activations are (i) oxidative addition, which involves late transition metals, such as Re, Fe, Ru, Os, Rh, Ir, Pt. The C-H fragment is added to the metal center to generate two new metal-carbon and metal hydride bonds. (ii) σ-Bond metathesis, which involves early-transition metals with d\textsuperscript{0} electronic configuration, accounts for the interchange of alkyl groups. (iii) 1,2-Insertion involves the addition of an alkane to a metal-nonmetal double bond. (iv) Metalloradical activation implicates two metal centers, which break and share the C-H bond fragments, forming two new metal-carbon and metal-hydrogen fragments. Finally, (v) electrophilic activation refers to the displacement of the hydrogen atom from the C-H bond by another group, generally a halide or a water molecule (scheme 5.1).\textsuperscript{234}

Scheme 5.1: There are five typical metal-promoted C-H activation pathways: i) oxidative addition, ii) σ-bond metathesis, iii) 1,2 insertion, iv) metalloradical activation and v) electrophilic activation.


These processes participate to different extents in the few catalytic systems described to date for the activation and subsequent functionalization of C-H bonds using unreactive hydrocarbons. Three of the most successful transformations dealing with alkanes are borylation reported by Hartwig et al.\textsuperscript{235} and alkane dehydrogenation\textsuperscript{236}, which both proceed through mechanism i), and Shilov’s systems for the activation of methane\textsuperscript{230}, which generally occurs through the electrophilic activation mechanism v) (scheme 5.2).

\textbf{Scheme 5.2:} Examples of a) Hartwig’s borylation and b) alkane dehydrogenation, which both proceeds through an oxidative addition mechanism, whereas c) Pt(II)-based activation of methane to methane bisulphate is believed to occur through an electrophilic activation mechanism.

One significant problem met in these aforementioned C-H activation processes is selectivity. One reason is because the desired functionalized product must not react more


readily with the metal center than the substrate starting material. A second reason is because selectivity among different C-H bonds in the starting substrate is often difficult.

When we consider these selectivity issues in the more general scenario of functionalizing organic molecules, one way to target the activation of a specific C-H bond, is to use the proximity of a heteroatom, such as nitrogen or oxygen, or of an unsaturated multiple bond, to activate the neighboring C-H bond\textsuperscript{237}. In these cases, the mechanisms of activation are completely different from the C-H metalation processes presented in scheme 5.1.

Indeed, typical approaches to functionalize $\alpha$-C-H bonds to a heteroatom or an unsaturated bond generally consist of redox processes promoted by a transition metal catalyst that can be performed either in an inter- or intramolecular manner. In order to activate the targeted $\alpha$-C-H bond through hydrogen abstraction, intermolecular processes use external oxidizing agents (via SET processes), such as 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ), $t$BuOOH (TBHP), $t$BuOO$t$Bu (DTBP), N-bromosuccinimide (NBS) or O$_2$. The process is now termed cross-dehydrogenative coupling (CDC)\textsuperscript{238}. On another hand, intramolecular processes take advantage of an electrophilic moiety present in the substrate, which serves as a hydride acceptor. This electrophilic moiety receives a hydride, thus being reduced and generates an oxidized stabilized carbocation due to the presence of a $\alpha$-heteroatom/allylic (or benzilic) double bond. The hydride addition to the electrophilic moiety is generally accompanied by concomitant formation of an internal nucleophile which cyclizes on the cation previously formed (scheme 5.3).

Scheme 5.3: General schemes for the a) inter- and b) intramolecular redox/cyclization processes, where the directing group is a heteroatom X = O or N.

Since the subject developed in this chapter is essentially an intramolecular redox process (as depicted in scheme 5.3b), literature background concerning this transformation will be more deeply discussed in the next section.

5.1.2 Intramolecular redox reactions: literature background

Literature background concerning 1,5-hydride shift/cyclization sequences is largely dominated by the use of either carbonyl/iminium or Michael acceptors, in a 1,2- or 1,4-hydride additions, respectively, followed by a cyclization step using the nucleophile generated from the 1,5-shift event (scheme 5.4).

Scheme 5.4: Intramolecular approach towards cyclized products following a [1,5]-hydride shift/cyclization sequence, proceeding either by a) 1,2- or b) 1,4-hydride addition.
This 1,5-hydride shift/cyclization sequence was originally described to proceed upon heating in temperatures in the range of 100-230 °C and using tertiary amines as hydride donors. This reactivity was named as tert-amino effect. Later developments extended the generality of this ring closure approach to the use of ethers as linkers. Reactions were described to proceed under milder conditions using catalytic amounts of Brønsted acids, such as TfOH and TFA, and different Lewis acids, such as Gd(OTf)\(_2\), Mg(OTf)\(_2\), SnCl\(_4\), BF\(_3\).OEt\(_2\) and Sc(OTf)\(_3\). Moreover, the first organocatalytic approach using a pyrrolidine derivative/TFA has also been recently documented.

\[
\text{CHO} + N\text{H}_2 \xrightarrow{TfOH (20 mol\%)} \text{EtOH, 78 °C, 3h} \xrightarrow{71 \%} N\text{H}
\]

\[
\text{O} \xrightarrow{\text{BF}_3\text{.OEt}_2 (0.75 \text{ equiv.})} \text{DCM, 40°C, 45h} \xrightarrow{90 \%, > 15:1 \text{ dr}}
\]

\[
\text{OMe} \xrightarrow{\text{Mg(OTf)}_2 (20 \text{ mol\%})} \text{1,2-DCE, 4A MS, 84°C, 18h} \xrightarrow{84 \%, 90:10 \text{ dr} 93/\text{nd \% ee}}
\]

For a review, see: Mátyus, P.; Éliás, O.; Tapolcsányi, P.; Polonka-Bálint, Á.; Halász-Dajka, B.; Synthesis, 2006, 16, 2625.


Scheme 5.5: Selected examples from the literature for the 1,5-hydride shift/cyclization sequence using different Brønsted/Lewis acids and 1,2-/1,4-hydride acceptors.

Important features concerning the substrate structure and the nature of the activating species employed (Brønsted acid, Lewis acid, temperature, etc) are determinants in allowing the 1,5-hydride shift to occur. For instance, in thermal-promoted procedures, which are generally described with double Michael-acceptors, the presence of strong electron-withdrawing groups at the terminal vinylic carbon, such as two cyano groups, are considered as a prerequisite for the 1,5-hydride shift to occur efficiently\(^\text{239}\). Conversely, carbonyl/Lewis acid complexes, which are highly electrophilic\(^\text{248}\), allow a broader use of Michael acceptors, such as \(\alpha,\beta\)-enals, -enones, -amide derivatives, -double esters, etc. (cf. scheme 5.5). Additionally, more demanding substrates, such as 5.1, fail to cyclize thermally, but using Sc(OTf)\(_3\) afford the desired product with excellent yield\(^\text{246}\) (scheme 5.6).

Scheme 5.6: Comparison of a cyclization proceeding via a thermal process and a Sc(OTf)$_3$-catalyzed process using substrate 5.1.

Importantly, because the initiating interaction between the substrate and the activating species (a Brønsted- or Lewis-acid) occurs far from the targeted C-H bond, functionalization can be attained at sterically hindered positions.

Furthermore, the 1,5-relationship between the hydride donor and acceptor moieties seems to be essential, since other substrates possessing a 1,4- or a 1,6-relationships, such as 5.6 and 5.7, respectively, failed to cyclize under typical reaction conditions$^{246}$ (figure 5.1).

![Figure 5.1: Substrates 5.3-5.5 with a 1,5-relationship between hydride donor-acceptor cyclize with excellent yields under treatment with BF$_3$.OEt$_2$ (30 mol%), at room temperature, whereas substrates 5.6 and 5.7, with a 1,4- and 1,6-relationship, respectively, failed to afford any cyclized products.]

In agreement with these observations and deuteration experiments$^{246}$, this process is believed to undergo a cation-induced through space$^{249}$ hydride shift, which should proceed through a six-membered transition state. The ease of the hydride transfer to occur is directly related to the stabilized cation generated, which makes the presence of a heteroatom, such as oxygen or nitrogen, indispensable. Cationic stabilization also explains

---

$^{249}$ The term through space is suggested by D. Sames, one of the main players in this area, to distinguish between this process and other more common 1,5-sigmatropic hydride shifts, which proceeds via a σ- or π-framework. For some examples of such processes, see: (a) Diaz, D.; Martin, V. S.; Org. Lett., 2000, 2, 335. (b) Wölfling, J.; Frank, E.; Schneider, G.; Tietze, L. F.; Angew. Chem. Int. Ed., 1998, 38, 200.
regioselective issues with branched substrates, as for example, substrates 5.8 and 5.10, which both afford only one regioisomer\textsuperscript{243} (scheme 5.6).

Scheme 5.6: The regioselectivity observed in 1,5-hydride shift/ cyclization sequences can be explained from the stabilization of cations generated as intermediates.

Another important structural feature is the effect of neighbor substituents on the reaction rates, as it has been noted for pyridazine\textsuperscript{239} (scheme 5.7a) and phenyl-ether\textsuperscript{244} (scheme 5.7b) derivatives. Accordingly, the bulkier the neighboring group, the faster is the reaction rate and lower is the catalyst loading required. This enhancement is attributed to two factors, one is the conformational behavior of the chain bearing the targeted C-H bond, which tends to spend more time in a productive conformation, due to an $A_{1,3}$-allylic strain-like repulsion, and another is the “buttressing effect” between the two ortho-substituents (scheme 5.7).
The reaction scope of this 1,5-hydride shift/cyclization sequence has also been extended to include alkynes as hydride acceptors. Urabe et al.\textsuperscript{250} reported the use of alkynylsulfones using Rh(tfa)\textsubscript{4} as catalyst in refluxing toluene. Interestingly, the use of sulfones was necessary to activate the alkyne moiety, whereas other substituents, such as hydrogen (terminal alkyne), alkyl, phenyl, trimethylsilyl or ester, all failed to produce the corresponding cyclized products (scheme 5.8).

Scheme 5.8: Selected examples to demonstrate the reaction scope of alkynyl sulfones in the Rh-catalyzed 1,5-hydride shift/cyclization sequence.

While the work developed in this chapter was under evaluation for publication in the Journal of American Chemical Society in late 2009, a very close report focusing on the use of PtI₄ for the activation of alkynes toward the 1,5-hydride shift/cyclization sequence performed by Vadola and Sames²⁵¹ appeared on the same journal’s website as an ASAP publication. In their study, Vadola and Sames showed that unactivated alkynes could undergo a through-space hydride shift, followed by a cyclization event using cyclic ethers and carboxamides as hydride sources (scheme 5.9).

![Scheme 5.9: Selected examples to demonstrate the reaction scope on the platinum-catalyzed 1,5-shift/cyclization sequence reported by Vadola and Sames (In red the C-C bond formed and the transferred hydride are highlighted).](image)

Because C(2)-linked substrates have the targeted C-H bond on a tertiary carbon, which ultimately would mean a more stabilized carbocation, they were expected to be more reactive than C(3)-substituted substrates. Indeed, these substrates afforded only degradation upon treatment with the highly electrophilic PtI₄. In such cases, the less active platinum chloride salt, K₂PtCl₄, was used (scheme 5.9).

Recognizing the difference in reactivity between different platinum salts depending on the halide on the platinum center, an interesting study with common platinum salts was also carried out. The authors hypothesized that by increasing both the bond length and by

following the trend of spectrochemical series\textsuperscript{252} when moving from a chloride to an iodide as ligand, platinum iodides would be more electrophilic than other platinum halides, and the platinum metal would be more accessible to the alkyne, thus resulting in a more active catalyst\textsuperscript{253}. Indeed, this has shown to be the case. In order to confirm if the observed halide effect was due to increased catalyst activity and not catalyst stability, the formation of fused bicycle ring 5.13a from substrate 5.12a was investigated using different platinum halides. Interestingly, PtI\textsubscript{2} was shown to be the most active, but reactions catalyzed by this salt failed to proceed to completion due to the instability of this catalyst under the reaction conditions. Overall, the best result was obtained with PtI\textsubscript{4} (scheme 5.10).

![Scheme 5.10](image)

Scheme 5.10: Effect of the halide ligand on the catalyst activity (scheme 5.10 was reproduced from ref. 251).

\textsuperscript{252} Ligands are capable of modifying the difference energy between the \textit{d} orbitals of the metal center, called the \textit{ligand-field splitting parameter} (\(\Delta\)). For halides, the order from small \(\Delta\) to large \(\Delta\) (spectrochemical series) is: I < Br < Cl < F. For a broader discussion, see: (a) Les orbitales moléculaires dans les complexes. Jean, Y.; Éditions de l'École Polytechnique, Palaiseau: 2003. (b) Chimie Moléculaire des Éléments de Transition, Mathey, F.; Sevin, A.; Éditions de l'École Polytechnique, Palaiseau: 2000.

\textsuperscript{253} For examples in the literature where platinum iodides and bromides performed better than platinum chlorides, see: (a) Fürstner, A.; Szillat, H.; Gabor, B.; Mynott, R. J.; \textit{J. Am. Chem. Soc.}, 1998, 120, 8305. (b) Hardin, A. R.; Sarpong, R.; \textit{Org. Lett.}, 2007, 9, 4547.
5.2 RESULTS AND DISCUSSION

5.2.1 The initial idea

The literature background we had at the time was the 1,5-hydride shift/cyclization sequences reported on Michael acceptors and aldehyde/iminium groups and Urabe’s work on alkynyl sulfones (cf. section 5.1.2). Recognizing the limiting aspect of Urabe’s report of the necessary presence of a sulfone substituent at the alkynyl moiety and knowing the high \(\pi\)-Lewis acidity of gold complexes (cf. chapter 2), we were interested in the possibility of using alkynes as 1,5-hydride acceptors\(^{254}\), speculating that they could subsequently follow the same corresponding cyclization process typically observed in tert-amino effects (scheme 5.11).

\[
\begin{align*}
\text{H} & \quad \text{H} \\
\text{X} & \quad \text{X} \\
\text{Au} & \quad \text{Au} \\
\end{align*}
\]

\(X = \text{O, N}\)

Scheme 5.11: Initial hypothesis concerning the 1,5-hydride shift/cyclization sequence catalyzed by gold.

Investigation towards the best reaction conditions was performed under the following experimental conditions: treatment of 0.1 mmol of model substrate 5.17a, with 4 mol% of catalyst and heating to reflux in 500\(\mu\)L of solvent. Substrate 5.17a was synthesized by two straightforward alkylation steps of dimethyl malonate 5.15 (scheme 5.12).

\[
\begin{array}{ccc}
\text{CO}_2\text{Me} & \text{CO}_2\text{Me} & \text{CO}_2\text{Me} \\
\text{Br} & \text{Br} & \text{Br} \\
\end{array}
\]

Scheme 5.12: Synthesis of substrate 5.17a.

The first trial using PtCl$_2$, which had already been documented to efficiently catalyze hydride shifts\(^{255}\), failed to provide any expected cyclization product (entry 1, table 5.1). The use of AuBr$_3$ alone or in combination with the silver salt AgN Tf$_2$ did not afford any productive cyclization either, but only marginal amounts of the hydrated-derived ketone 5.21a\(^{256}\) (entries 2 and 3, table 5.1). While the use of ligands L1-L4 on gold metal centers did not produce any efficient catalyst for the desired transformation, affording only trace amounts of the 6-membered cycle 5.18a and degradation under the reaction conditions (entries 4-7, table 5.1), L5-L7 performed substantially better. Initial experiments in refluxing CDCl$_3$ using L5 and L6, with SbF$_6$ as counter-ion, afforded 45% and 57%, respectively, of the 6-membered cycle 5.18a. Surprisingly, tricyclic structure 5.20a which was attributed to a double C-H activation and represents a formal [3+2]-cycloaddition product was also produced in the reaction mixture to minor extents (entries 8 and 9, table 5.1). Changing the counter-ion on the gold center from SbF$_6$ to N Tf$_2$ decreased the reaction yield of 5.18a to 41% (entry 10, table 5.1). Ligand L7 afforded a slightly better yield of 5.18a than L6 (entry 11, table 5.1). As undesired hydration products were consistently being formed due to trace amounts of water present in the solvent, we hypothesized that changing the reaction solvent could increase the reaction rate of 1,5-hydride-shift/cyclization sequence over the hydration pathway\(^{257}\). Solvent screening using toluene (entry 12, table 5.1), dioxane (entry 13, table 5.1) and nitromethane (entry 13, table 5.1) revealed the latter as the optimal choice in reducing hydration. Furthermore, by stopping reactions after 2h, allowed us to compare reaction rates using ligands L7-L10 (entries 15-18, table 5.1), which also pointed to the higher activity of ligand L7 over L8, L9 and L10.

---


Table 5.1: Catalytic system optimization for the 1,5-hydride shift/cyclization sequence using unactivated alkynes.
With the optimal conditions in hand, we had set the stage to start the investigation on reaction scope and functional group tolerance of this process. Accordingly, alkynes derived from propargyl malonate 5.17a were synthesized from four different methods: a Sonogashira cross-coupling reaction leading to substrate 5.17b, a bromination step leading to substrate 5.17c, reduction with LiAlH₄/acetylation sequence leading to substrate 5.17d and treatment with LDA/ClCO₂Et leading to substrate 5.17e (scheme 5.13).

Scheme 5.13: Alkynes synthesized from propargyl malonate 5.17a.

Other terminal alkynes 5.17f-t were also synthesized via two straightforward alkylation steps. The first alkylation inserted either a THF or a sulfonamide ring from a bromo (5.14f-j, 5.14l, 5.14p) or a tosylate (5.14k)/ mesylate (5.14q-r) derivative, respectively. Bromo derivatives 5.14f-j, 5.14l, 5.14p were constructed from a NBS-promoted cyclization process from the corresponding homoallyl alcohol 5.26f-j, 5.26l, 5.26p (ester) or from either a chloro- or an iodo-ether derivative, 5.29 and 5.31, respectively.

The second alkylation step consisted of the propargylation of the monosubstituted malonate previously prepared. From these double alkylated substrates, compounds 5.17f, 5.17k-l were treated with LDA and ethyl chloroformate to afford the corresponding conjugate esters 5.17m-o (scheme 5.14).
Scheme 5.14: Detailed synthesis of the main C(2)-linked THF/sulfonamides rings in this project.
Other C(2)-linked heterocycles synthesized in the context of this work by analogous alkylation procedures are shown below (figure 5.1).

Figure 5.1: Other substrates synthesized in the context of this project.

These C(2)-linked heterocycles (5.17a-5.17aa) were submitted to our optimized reaction conditions. In this context, only oxygenated 5-membered rings with either a malonate or a diacetoxyethyl groups as tethers and bearing terminal alkynes or alkynes substituted by an ester group proceeded through the gold-catalyzed 1,5-hydride shift/cyclization sequence.

For instance, internal alkynes 5.17b,c afforded only starting material after prolonged reaction times (entries 1 and 2, table 5.2). On another hand, the use of a diacetate in the tether of 5.17d allowed the reaction to proceed smoothly, affording a 3.7:1 mixture of cyclized products 5.18d:5.20d (entry 3, table 5.2). Conjugate esters 5.17e and 5.17m-o were highly selective towards 6-membered cycles 5.18e and 5.18m-o (entries 4, 12 and 14, table 5.2), with exception of 5.17n, where the presence of the aromatic ring destabilizes the carbocation at the α-position to the oxygen due its inductive electron-withdrawing effect (entry 12, table 5.2). Double substituted THF rings at positions 2 and 5, 5.17f-h, and even triple substituted 5.17i-j, afforded mixtures of diastereoisomers for spiro compound 5.18 and the formal [3+2]-addition product 5.20 (entries 5-9, table 5.2) in ratios that are difficult to rationalize only on the grounds of carbocation stabilization, but where steric hindrance should also play an important role. For instance, the THF ring bearing an ester branch 5.17h afforded a richer mixture in formal [3+2]-product 5.20h (1.5:1 of 5.18h:5.20h, entry 7, table 5.2) than 5.17g (2.3:1 of 5.18g:5.20g, entry 6, table 5.2) that was approximately obtained in the same ratio for compound 5.17i (2.1:1 of 5.18i:5.20i, entry 8, table 5.2).
Interestingly, compound 5.17j, which has a trans ring-junction, did not afford any formal [3+2]-addition product (entry 9, table 5.2), whereas 5.17i with a cis ring-junction

<table>
<thead>
<tr>
<th>entry</th>
<th>substrate</th>
<th>trans:cis&lt;sup&gt;a&lt;/sup&gt;</th>
<th>products</th>
<th>isolated yield 5.17-5.20&lt;sup&gt;b&lt;/sup&gt;</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Ph</td>
<td>-</td>
<td>5.17b</td>
<td>-</td>
</tr>
<tr>
<td>2</td>
<td>Br</td>
<td>-</td>
<td>5.17c + hydration</td>
<td>-</td>
</tr>
<tr>
<td>3</td>
<td>OAc</td>
<td>-</td>
<td>5.17d:5.18d:5.20d</td>
<td>70%:3.7:1</td>
</tr>
<tr>
<td>4</td>
<td>EtO&lt;sub&gt;2&lt;/sub&gt;C</td>
<td>-</td>
<td>5.17e:5.18e</td>
<td>91%*</td>
</tr>
<tr>
<td>5</td>
<td>2:1</td>
<td>5.18f:1.4:5.20f</td>
<td>5.18f:5.20f</td>
<td>82%:&lt;sup&gt;c&lt;/sup&gt; 5.1:1</td>
</tr>
<tr>
<td>6</td>
<td>Ph</td>
<td>2:1</td>
<td>5.18g:1.5:5.20g</td>
<td>76%:&lt;sup&gt;c&lt;/sup&gt; 2.3:1</td>
</tr>
<tr>
<td>7</td>
<td>2.5:1</td>
<td>5.18h:1.3:5.20h</td>
<td>5.18h:5.20h</td>
<td>82%:&lt;sup&gt;c&lt;/sup&gt; 1.5:1</td>
</tr>
<tr>
<td>8</td>
<td>1.6:1</td>
<td>5.18i:1.7:5.20i</td>
<td>5.18i:5.20i</td>
<td>74%:&lt;sup&gt;c&lt;/sup&gt; 2.1:1</td>
</tr>
<tr>
<td>9</td>
<td>1:0</td>
<td>5.18j:1.8:5.18j</td>
<td>5.18j</td>
<td>50%&lt;sup&gt;c&lt;/sup&gt;</td>
</tr>
<tr>
<td>10</td>
<td></td>
<td>(5.17k + hydration)</td>
<td>5.17k</td>
<td>-</td>
</tr>
</tbody>
</table>
Table 5.2: Main results of the gold-catalyzed 1,5-hydride shift/cyclization sequence using C(2)-substituted heterocycles.

afforded a 2.1:1 mixture of 5.18i:5.20i (entry 8, table 5.2). This difference can only be rationalized in steric grounds. The formation of a formal [3+2]-addition product in the cyclization of 5.17j is not observed probably because the conformation necessary to the second hydrogen transfer to occur is hindered by the presence of the fused 6-membered cycle with the trans junction (scheme 5.15, see also the reaction mechanism proposition in scheme 5.19).

Scheme 5.15: Geometric constraints suffered by the trans-fused ring 5.17j in comparison to the cis-fused ring 5.17i (in red, the hydrogens that will be transferred).
Also noteworthy, the synthesis of substrate 5.17j in only one diastereoisomer is at first sight, of no consequence, as the tertiary center in the α-position to the oxygen proximal to the tether is always destroyed during the reaction process (even if an eventual "chiral memory" can nevertheless be present, see mechanism discussion in scheme 5.19). By the same reason evoked for the conjugate alkyne 5.17n, terminal alkyne 5.17k does not undergo cyclization, but rather affords only starting material and the hydration product 5.21k (entry 10, table 5.2). Finally, dioxolane 5.18l can also be obtained from 5.17l. Nevertheless, it should be noted that a poor conversion was obtained for this substrate under reflux of nitromethane (20% after 6h). This reaction was then conducted at 130 °C in nitropropane. Furthermore, even if the dioxolane ring survived our reaction conditions (as indicated by 1H NMR analysis of the crude reaction mixture) it proved to be rather labile and did not resist the acidic conditions of column chromatography, being isolated as the corresponding deprotected ketone (entry 11, table 5.2).

Other C(2)-linked heterocycles 5.17p-z and 5.17aa were also submitted to our optimized reaction conditions, but they afforded only starting material and hydration products of the general structure type 5.21 in different extents. Indeed, sulfonamides and 6-membered heterocycles 5.17q, 5.17r and 5.17u, which were expected to be less reactive\textsuperscript{251,258}, gave only starting material. Lactone 5.17p and carbonate 5.17v and acetal 5.17w were not able to sufficiently stabilize the cation in the α-position of the oxygen and were unreactive. Greater rotation freedom also seems a restricting factor for the occurrence of the 1,5-hydride shift, as no productive cyclization was observed for acetal 5.17x and for ethers 5.17s and 5.17t. Nor are other linkers allowed, as only starting material and hydration products (of type 5.21) in different extents were obtained for these reactions with substrates 5.17y, 5.17z and 5.17aa.

After having learned about the reactivity pattern demonstrated by C(2)-substituted heterocycles, we were interested on the reactivity of C(3)-substituted THF rings, which were expected to be less prone to undergo the 1,5-hydride shift, because of their secondary C-H bonds. In this regard, substrates 5.36a-l were synthesized using classic straightforward
routes. Homoallyl alcohols 5.32b-e, which are readily obtained from the Barbier reaction\textsuperscript{259} were submitted to a bromination step\textsuperscript{260}, affording dibromo intermediates 5.33b-e, which were directly engaged in a \( \text{S}_2\text{2} \)-intramolecular substitution without purification. The C(3)-bromo THF rings thus obtained are submitted to the same propargylation, esterification, bromination and cross-coupling reactions previously employed for C(2)-substituted THFs (scheme 5.15).

Other related structure synthesized in this project:

\[
\text{5.36a}^{11}, \text{R} = \text{H}, 71\%
\]
\[
\text{5.36b}^{11}, \text{R} = \text{Pr}, 97\%
\]

Scheme 5.15: Synthesis of malonates bearing C(3)-substituted THF rings.

Subjection of alkynes 5.36a-l to our optimized conditions furnished similar results as C(2)-linked substrates. Internal alkynes did not react. Only starting material was recovered when substrates 5.36a, 5.36h and 5.36i were submitted to gold-catalysis (entries 1, 8 and 9,


<table>
<thead>
<tr>
<th>entry</th>
<th>substrate</th>
<th>trans/cis&lt;sup&gt;a&lt;/sup&gt;</th>
<th>products</th>
<th>isolated yield 5.37/5.38&lt;sup&gt;b&lt;/sup&gt;</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>5.36a</td>
<td>-</td>
<td>5.36a&lt;sup&gt;h&lt;/sup&gt;</td>
<td>-</td>
</tr>
<tr>
<td>2</td>
<td>5.36b</td>
<td>-</td>
<td>5.37b, 5.38b</td>
<td>95%&lt;sup&gt;c&lt;/sup&gt;, 6:1</td>
</tr>
<tr>
<td>3</td>
<td>5.36c</td>
<td>1:3</td>
<td>5.37c&lt;sup&gt;(1:3)&lt;sup&gt;f&lt;/sup&gt;</td>
<td>93%&lt;sup&gt;c&lt;/sup&gt;, -</td>
</tr>
<tr>
<td>4</td>
<td>5.36d</td>
<td>1:3</td>
<td>5.37d&lt;sup&gt;(1:3)&lt;sup&gt;f&lt;/sup&gt;</td>
<td>88%&lt;sup&gt;c&lt;/sup&gt;, -</td>
</tr>
<tr>
<td>5</td>
<td>5.36e</td>
<td>1:3</td>
<td>5.37e&lt;sup&gt;(1:3)&lt;sup&gt;f&lt;/sup&gt;</td>
<td>29%&lt;sup&gt;c&lt;/sup&gt;, -</td>
</tr>
<tr>
<td>6</td>
<td>5.36f</td>
<td>CO₂Et</td>
<td>5.37f, 5.38f</td>
<td>81%&lt;sup&gt;c&lt;/sup&gt;, 1:1</td>
</tr>
<tr>
<td>7</td>
<td>5.36g</td>
<td>CO₂Et</td>
<td>5.37g&lt;sup&gt;(6:7:1)&lt;sup&gt;f&lt;/sup&gt;</td>
<td>90%&lt;sup&gt;c&lt;/sup&gt;, 1.5:1</td>
</tr>
<tr>
<td>8</td>
<td>5.36h</td>
<td>Ph</td>
<td>5.36h&lt;sup&gt;h&lt;/sup&gt;</td>
<td>-</td>
</tr>
<tr>
<td>9</td>
<td>5.36i</td>
<td>-</td>
<td>5.36i&lt;sup&gt;h&lt;/sup&gt;</td>
<td>-</td>
</tr>
<tr>
<td>10&lt;sup&gt;f&lt;/sup&gt;</td>
<td>5.36j</td>
<td>Br</td>
<td>Br&lt;sup&gt;5.37j&lt;sup&gt;(4.3:1)&lt;sup&gt;f&lt;/sup&gt;</td>
<td>75%&lt;sup&gt;d&lt;/sup&gt;, -</td>
</tr>
<tr>
<td>11</td>
<td>5.36k</td>
<td>1:3</td>
<td>5.36k&lt;sup&gt;b&lt;/sup&gt;</td>
<td>-</td>
</tr>
<tr>
<td>12</td>
<td>5.36l</td>
<td>(5.36l + hydration)</td>
<td>-</td>
<td>-</td>
</tr>
</tbody>
</table>

<sup>a</sup> Determined by <sup>1</sup>H NMR spectroscopy. <sup>b</sup> SM fully recovered. <sup>c</sup> dr determined by <sup>1</sup>H NMR spectroscopy. <sup>f</sup> The conversion of 5.36j was 85%.
Table 5.3: Results of the gold catalyzed 1,5-hydride shift/cyclization sequence using C(3)-linked THF rings.

On another hand, conjugate alkynes 5.36f and 5.36g afforded cyclized products, but without any regioselectivity, with 1:1 and 1.5:1 mixtures of 5:6-membered rings, respectively (entries 6 and 7, table 5.3). Some selectivity was observed using terminal alkyne 5.36b, which afforded a 6:1 mixture of 5:6-membered cycles (entry 2, table 5.3). Interestingly, complete regioselectivity was observed with 3,5-substituted THF rings bearing terminal alkynes 5.36c-e, which afforded only 5-membered ring products (entries 3-5, table 5.3). Curiously, this time, brominated alkyne 5.36j furnished the corresponding 5-membered product regioselectively, in good 75% yield (entry 10, table 5.3), but this reactivity was not general, as 5.36k failed to undergo the same reaction process (entry 11, table 5.3). Furthermore, an ethereal linker proved to be prejudicial, as compound 5.36l also failed to afford any desired cyclization (entry 12, table 5.3). We next attempted this 1,5-hydride shift/cyclization sequence with the oxygen being part of the tether. Results obtained with terminal alkynes 5.40a-d were highly disappointing, as no trace of cyclized products 5.42 or 5.43 were observed (scheme 5.16).

We next turned our attention to conjugate esters 5.41a-e. These esters were synthesized from propargyl alcohols 5.39 a-d that were alkylated on the oxygen, to furnish terminal alkynes 5.40a-d, and then treated with n-BuLi and quenched with ethyl chloroformate to give access to the desired conjugate esters 5.41a-d. Alkynyl ester 5.41e was synthesized from epoxide 5.42, which was opened with an acetylide anion generated from the treatment of trimethylsilyl ethyne with n-BuLi, then methanolysis of the TMS group (K₂CO₃ (cat)/MeOH) and benzylation of the alcohol generated benzyl ether 5.43. This alkynyl
ether was deprotonated with \( n\text{-BuLi} \) and quenched with ethyl chloroformate to afford alkynyl ester \( 5.41e \) (scheme 5.16).

![Scheme 5.17: Synthesis of conjugate esters 5.41a-e.](image)

Subjection of benzyl derivatives \( 5.41a \) and \( 5.41d-e \) to our optimized conditions afforded cyclic ethers \( 5.44a \) and \( 5.44d-e \) with excellent diastereoselectivities (entries 1, 4, 5, table 5.4). In the cases where an allyl or a PMB group were used, only degradation was obtained in both cases (entries 2 and 3, table 5.4).
In order to investigate the reaction mechanism of this 1,5-hydride shift/cyclization process, deuterated compounds **5.49** and **5.50** were prepared. The deuteration in **5.49** was obtained from the deprotonation at the \( \alpha \)-position of the ester **5.45** and quench with MeOD, which afforded derivative 63\%D-**5.46**. Subsequent reduction with lithium aluminium hydride slightly eroded its deuterium content to 50\%D. Then, tosylation and the same two alkylation steps used for the synthesis of previous substrates afforded compound **5.49** with 50\%D (scheme 5.18a). The deuteration in **5.50** was performed in a more straightforward manner. Substrate **5.17** was deprotonated with LDA and then quenched with D\(_2\)O (scheme 5.18b).

Table 5.4: Alkynyl ethers with the oxygen being part of the tether employed in the 1,5-hydride shift/cyclization sequence.

<table>
<thead>
<tr>
<th>Entry</th>
<th>Substrate</th>
<th>Products</th>
<th>Isolated Yield</th>
<th>cis:trans(^a)(R(^1), R(^2))</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td><strong>5.41a</strong></td>
<td><strong>5.44a</strong></td>
<td>65%*</td>
<td>16:1</td>
</tr>
<tr>
<td>2</td>
<td><strong>5.41b</strong></td>
<td>degradation</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>3</td>
<td><strong>5.41c</strong></td>
<td>degradation</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>4</td>
<td><strong>5.41d</strong></td>
<td><strong>5.44d</strong></td>
<td>74%*</td>
<td>&gt; 25:1</td>
</tr>
<tr>
<td>5</td>
<td><strong>5.41e</strong></td>
<td><strong>5.44e</strong></td>
<td>88%</td>
<td>&gt; 25:1</td>
</tr>
</tbody>
</table>

\(^a\) Determined by \(^1\)H NMR spectroscopy
Scheme 5.18: a) Synthesis of deuterated substrate 5.49. b) Synthesis of deuterated substrate 5.50.

The cyclization of 5.49 and 5.50 afforded 5.51 and 5.52, respectively (scheme 5.19a). The observed deuteration pattern suggests that an intramolecular 1,5-hydride shift takes place and generates an stabilized cation in an oxonium form, as the one drawn in 5.53. This cationic form can be seen as being delocalized on the vinylic gold moiety, as shown in 5.54. At this stage, two reaction pathways are possible. Either this delocalized carbocation rearranges to a 5-membered cycle 5.55 or to the 6-membered cycle 5.56, which through a 1,2-alkyl shift ring-expands to give the same 6-membered cycle 5.56. Finally, the 6-membered cycle 5.56 regenerates the gold fragment and affords the major product 5.18a. Formation of the by-products 5.19a and 5.20a can be explained from the 5-membered ring intermediate 5.55, which can either undergo a 1,2-hydride shift through the intermediacy of 5.57 and subsequent gold loss to afford compound 5.19a, or can follow another 1,5-hydride shift to generate oxonium 5.58 that evolves by ring-closure to produce the tricycle 5.20a (scheme 5.19b).

At this point, important remarks should be made. Although in the previously discussed mechanism for the synthesis of 5.18a, the direct route from 5.54 to the 6-membered cycle 5.56 cannot be excluded, the passage through the 5-membered cycle 5.55
is supported by the formation of by-products \textbf{5.19a} and \textbf{5.20a}, and presumably due to the nucleophilic nature of the \(\beta\)-carbon of vinylgold \textbf{5.53}. In contrast to the previous work reported by Vadola and Sames\textsuperscript{251}, where 5-membered cycles were reported as major products, we obtained mainly 6-membered rings. A plausible explanation for this difference in regioselectivity is that the ring expansion from \textbf{5.55} to \textbf{5.56} might be under thermodynamic control using XphosAu(NCCH\textsubscript{3})SbF\textsubscript{6}, whereas the use of the stronger Lewis acid Ptl\textsubscript{4} would implicate essentially a kinetic control.

\[ \text{a) } \begin{align*}
\text{MeO}_2\text{C} & \text{ CO}_2\text{Me} \\
\text{MeO}_2\text{C} & \text{ CO}_2\text{Me} \\
\text{MeO}_2\text{C} & \text{ CO}_2\text{Me} \\
\text{MeO}_2\text{C} & \text{ CO}_2\text{Me} \\
\end{align*} \]

\[ \text{b) } \begin{align*}
\text{MeO}_2\text{C} & \text{ CO}_2\text{Me} \\
\text{MeO}_2\text{C} & \text{ CO}_2\text{Me} \\
\text{MeO}_2\text{C} & \text{ CO}_2\text{Me} \\
\text{MeO}_2\text{C} & \text{ CO}_2\text{Me} \\
\end{align*} \]
Scheme 5.19: **a)** Results of deuterium experiments using \textit{5.49} and \textit{5.50} suggests that **b)** the mechanism for the formation of main product \textit{5.18a} as starting with a 1,5-hydride shift, followed by ring-closure to produce the 5-membered cycle, which would ultimately ring-expands to a 6-membered ring. Formation of by-products \textit{5.19a} and \textit{5.20a} are also exhibited.

Upon careful examination of the \textsuperscript{1}H NMR of the crude reaction mixtures of each of the deuterium-labeled experiments, it is interesting to note that for both of the by-products, \textit{5.19a} and \textit{5.20a}, the deuterium atom was found to be exclusively incorporated at a single vinylic position (scheme 5.20a). This stereoselectivity is believed to derive from the existence of only one productive conformer \textit{5.59}, as the presence of the THF ring hinders the formation of the other conformer \textit{5.60} (scheme 5.20b).

\begin{itemize}
\item a)
\begin{center}
\begin{tikzpicture}
\node (a) at (0,0) {CO$_2$Me};
\node (b) at (1,0) {CO$_2$Me};
\node (c) at (0.5,0.5) {AuL};
\node (d) at (0.5,-0.5) {H$_b$};
\node (e) at (1.5,0) {H$_a$};
\node (f) at (0.5,0) {H$_a$};
\node (g) at (1.5,0.5) {H$_b$};
\end{tikzpicture}
\end{center}
\end{itemize}
Scheme 5.20: a) Comparison between two possible conformers that culminate with the stereoselective hydride shift in all three observed products 5.18a, 5.19a and 5.20a. b) $^1$H NMR of representative signals using 5.17a. c) $^1$H NMR of representative signals using 5.49. d) $^1$H NMR of representative signals using 5.50.

Concerning the 1,5-shift/cyclization sequence using C(3)-linked THFs, 5-membered rings with an exo methylene were obtained as the major outcome, instead of the 6-membered cycles that were mainly observed when using C(2)-linked THFs. Although both substrates proceed essentially by the same mechanism discussed above in scheme 5.19, the difference in selectivity can be accounted for by the different stabilities of intermediates 5.62 and 5.63, which lead to products 5.37b and 5.38b, respectively. For the cyclization of substrate 5.17a, the steric contraints in the spiro intermediate 5.55 should favor its rearrangement into cation 5.56 and then the formation of cyclohexene 5.18a as the major product (scheme 5.19). The steric contraints should be weaker for the fused bycyclic intermediate 5.62, thus allowing for a rapid 1,2-hydride shift leading to 5.37b, rather than a 1,2-alkyl shift leading to 5.38b (scheme 5.21).

Scheme 5.21: The divergent reaction pathway observed for C(3)-linked THFs, which lead to 5-membered cycles as major products, contrasts with the reaction pathway observed for C(2)-linked THFs, which lead usually to 6-membered cycles. This difference can be explained in terms of different steric hindrance present in intermediates 5.62 and 5.63.
An alternative reaction mechanism for alkynyl esters, such as 5.41a-e, could also be imagined to proceed through the cationic gold(I) coordination to the carbonyl of the ester group, followed by a hydride conjugate addition. This would generate an enolate form that would then attack the cation formed and produce the ring closure (not shown). This hypothesis is considered less likely, because gold(I) complexes are known to preferentially activate alkynes rather than keto functionalities. Furthermore, it would not explain the formation of isomers 5.38f and 5.38g from compounds 5.36f and 5.36g, respectively (see entries 6 and 7, table 5.3).

The high selectivity towards 6- when compared to 5-membered rings observed in the reaction of conjugate alkynyl esters 5.41a and 5.41d-e, might be accounted by the destabilization of gold carbene 5.65 caused by the ester group that is on the same carbon that carries the gold fragment. This destabilization would accelerate the subsequent 1,2-alkyl shift leading to intermediate 5.66 thus explaining why 5-membered cycles are not observed in this case (but also why 6-membered cycles are observed to some extent in the case of substrates 5.36f and 5.36g, scheme 5.22). The high diastereoselectivities observed should be considered by the fact that the diastereoselective hydride addition/cyclization sequence, which proceeds via a Zimmerman-Traxler transition state that is responsible to set the relative stereochemistry between (R1, R3) and R2 (Scheme 5.22).

---

Scheme 5.22 Reaction mechanism for the 1,5-hydride shift/cyclization sequence using alkynyl esters 5.41.

---

261 Au(I) catalysts were recognized as more carbophilic Lewis acids, while Au(III) more oxophilic. For a more detailed discussion, see: Yamamoto, Y.; J. Org. Chem., 2007, 72, 7817.
5.3 CONCLUSION

Various fused and spiro tetrahydrofurans and dihydropyrans were obtained via a 1,5-hydride shift/cyclization sequence performed on alkynyl ethers by using the gold complex XphosAu(CH$_3$CN)SbF$_6$ as the catalyst in refluxing nitromethane. This hydroalkylation process showed good substrate scope and functional group compatibility, being applicable to aliphatic and cyclic ethers possessing either terminal or substituted alkynes. This route, that formally transforms a C(sp$^3$)-H bond into a new C-C bond, represents not only a practical entry to ethers possessing five- or six-membered rings, but is also one of the first reports where an hydrogen atom is intramolecularly transferred as an hydride to an unactivated alkyne.
CHAPTER 6: FORMATION OF CINNOLINE DERIVATIVES BY A GOLD(I)-CATALYZED HYDROARYLATION OF N-PROPARGYL-N’-ARYLHYDRAZINES

The work described in this chapter was performed individually and was reported in the Journal of Organometallic Chemistry: Jurberg, I. D.; Gagosz, F.; J. Organom. Chem., 2011, 696, 37.
CHAPTER 6: FORMATION OF CINNOLINE DERIVATIVES BY A GOLD(I)-CATALYZED HYDROARYLATION OF N-PROPARGYL-N'-ARYLHYDRAZINES

6.1 INTRODUCTION: Intramolecular hydroarylation processes and Importance of Cinnoline derivatives

6.1.1 Intramolecular Hydroarylation Processes

Among the large-quantity chemicals manufactured by the chemical industry, arene derivatives, such as benzenes, naphtalenes, phenols and anilines have a great importance as they serve as raw material for a plethora of organic transformations. In this context, conception of catalytic and straightforward processes derived from these compounds, which functionalize aromatic C-H bonds leading to C-C bond formation in more complex structures are very practical and therefore of great interest for the chemical and pharmaceutical community.

In this regard, an intramolecular hydroarylation procedure, which proceeds by a formal addition of an arene C-H bond across a multiple bond represents a flexible and conceptually simple approach allowing rapid access to valuable annulated compounds. Other advantages using this approach in comparison to other common annulation strategies, such as the Heck reaction for instance, is that the hydroarylation process not only eliminates the need for a halogen or a triflate group, but allows also multiple mechanistic possibilities, ultimately allowing access to different products (scheme 6.1).

Figure 6.1: Comparison between the hydroarylation versus the Heck approach for the ring-closure of propargyl/allyl arene derivatives.

The three main alternative mechanistic routes available from the hydroarylation approach are i) arene metalation-Heck type addition\textsuperscript{265}, ii) multiple bond activation-Friedel-Crafts type substitution\textsuperscript{266} and iii) metal-promoted Claisen rearrangement-heteroatom addition sequences\textsuperscript{267,268} (scheme 6.1).


Scheme 6.1: Hydroarylation of alkynes and alkenes can follow three main pathways: i) arene metalation-Heck type addition, ii) multiple bond activation-Friedel-Crafts substitution and iii) metal-promoted Claisen rearrangement-heteroatom addition.

Common metals employed in intramolecular hydroarylation processes with alkenes and alkynes are Pt\textsuperscript{269}, Pd\textsuperscript{270}, Ru\textsuperscript{271}, Ga\textsuperscript{272}, Hg\textsuperscript{273}, Cu\textsuperscript{274} and Au\textsuperscript{275}. For instance, the synthesis of important biological agents, such as chromenes \textsuperscript{6.1\textsuperscript{276}}, coumarins \textsuperscript{6.2\textsuperscript{277}}, dihydroquinolines

\textsuperscript{277} Coumarins are used in the pharmaceutical industry as precursors of a number of synthetic anticoagulant drugs, where a notable one is warfarin. For references, see: (a) Lowenthal, J.; Birnbaum, H.; \textit{Science}, 1969, 164, 181. (b) O'Reilly, R. A.; Ohms, J. I.; Motley, C. H.; \textit{J. Biol. Chem.}, 1969, 244, 1303.
6.3 and quinolinones 6.4 have been reported using PtCl₄, Pd(OAc)₂/ TFA, Hg(OTf)₂·(TMU)₃ (TMU = tetramethylurea), CuCl₂, AuCl₃/AgOTf and JohnPhosAu(NCCH₃)SbF₆ (scheme 6.1).

Scheme 6.1: Selected substrate scope of chromenes, coumarins, dihydroquinolines and quinolinones obtained from different transition metal-catalyzed transformations.

6.1.2 Importance of cinnolines

Another important class of benzo-fused compounds is derived from cinnolines, the focus of this project. Because cinnolines exhibit an exceptional spectrum of pharmaceutical activities, their use in drug design have been investigated and numerous tetrahydrocinnolines and 1,2-dihydrocinnolines, such as 6.5 or the commercially

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278 Diydroquinolines have been used in a variety of ways as antioxidants. For references, see: (a) Johnson, J. V.; Rauckman, B. S.; Baccanari, D. P.; Roth, B.; J. Med. Chem., 1989, 32, 1942. (b) Prohaszka, L.; Rozsnya, T.; Avian Pathology, 1990, 19.
available drugs *cinnopentazole 6.6* and *cinnofuradione 6.7* have demonstrated important biological activity (scheme 6.2).

Scheme 6.2: Examples of bioactive 1,2-dihydro- and tetrahydrocinnolines.

### 6.2 RESULTS AND DISCUSSION

#### 6.2.1 The initial idea

Previous work in our group describes the 6-exo dig cyclization of propargyl aniline derivatives 6.7 bearing a malonate as tether upon treatment with XphosAu(NCCH₃)SbF₆ (1 mol%) in refluxing nitromethane to afford mixtures of tetrahydroquinolines 6.8 and dihydroquinolines 6.9. These mixtures were isomerized by treatment with a catalytic amount of PTSA to obtain dihydroquinoline 6.9 as the sole product. Subsequent exposure of these dihydroquinolines 6.9 to the sun light promoted a 6π-electrocyclic ring opening-ring closure sequence that afforded indoles 6.10 (scheme 6.3).

Scheme 6.3: Previous work developed in our group towards the synthesis of dihydroquinolines 6.9 and indoles 6.10.

---

Based on these results, we envisaged a similar reaction sequence from N-propargyl-N'-arylhydrazines 6.11, which would ultimately afford cinnoline derivatives 6.13 and aminoindoles 6.14 (scheme 6.4).

Scheme 6.4: Reaction sequence envisaged to access cinnoline derivatives 6.13 and aminoindoles 6.14.

In order to start our investigation, N-propargyl-N'-phenyl hydrazine 6.15 was synthesized from diethyl azodicarboxylate (DEAD) in a two-steps sequence consisting of the conjugate addition of phenyl magnesium bromide\textsuperscript{284} and standard propargylation. Subsequent gold catalysis using the previously optimized protocol for N-aminophenyl propargyl malonates 6.7 (\textit{i.e.} 1 mol\% XphosAu(NCCH\textsubscript{3})SbF\textsubscript{6} in refluxing CH\textsubscript{3}NO\textsubscript{2}) did not afford any products, except for the starting substrate. Higher catalyst loadings and different solvents did not change this result. We speculate that the origin of this lack of reactivity comes from the steric constraint between the two ester groups present in 6.15\textsuperscript{285}. Other possible reasons are the dipole repulsion between both ester groups and/or the low nucleophilicity of the phenyl ring due to the electron-withdrawing carboxymethyl group on the proximal nitrogen atom (scheme 6.5).

\textsuperscript{284} Demers, J. P.; Klaubert, D. H.; \textit{Tetrahedron Lett.}, 1987, 28, 42, 4933

\textsuperscript{285} In contrast to amide bonds, whose energy barriers of rotation depend on the solvent used, carbamates rotation barriers were reported to be independent of solvent variation. See: Cox, C.; Lectka, T.; \textit{J. Org. Chem.}, 1998, 63, 2426.
Scheme 6.15: First trials towards the cyclization of N-propargyl-N’-aryl hydrazines using 6.15.

As a second attempt, as well as a proof-of-principle experiment, we speculated that hydrazine derivative 6.11a carrying two phenyl rings on the same nitrogen atom would exhibit a substantially faster cyclization rate, as both aromatic rings are available to attack the alkyne moiety. Substrate 6.11a was prepared from phenyl hydrazine 6.16 following protection of the primary nitrogen as a carbamate, oxidation, Grignard conjugate addition and propargylation. Only one purification step by column chromatography was necessary at the end of this sequence (scheme 6.16).

Scheme 6.16: Synthesis of substrate 6.11a.

Treatment of compound 6.11a with 1 mol% of XphosAu(NCCH₃)SbF₆ in refluxing nitromethane did not afford any product, but only starting substrate. It was by increasing the amount of gold catalyst to 4 mol% that the desired cyclization was observed, and with an excellent 99% yield for 6.12a as the only isomer in 1h. No trace of the internal double-bond isomer 6.13a was observed under these reaction conditions, but the exocyclic double bond...
isomer 6.12a could be isomerized to the internal position isomer 6.13a upon treatment with a catalytic amount of PTSA in refluxing chloroform (Scheme 6.17).

Scheme 6.17: Hydroarylation attempt with 6.11a.

After being reinvigorated by this positive result, we turned our attention to less-biased substrates. Hydrazine derivatives 6.11b-e with different substituents on the nitrogen proximal to the aromatic ring were prepared. The synthetic route chosen for substrates 6.11b-c was identical to that used to prepare 6.11a (scheme 6.18a). In the case where a benzyl group is used, a double alkylation strategy using n-BuLi was employed for substrate 6.11d (scheme 6.18b). Although our attempt to add tert-butyl magnesium chloride to the phenyl intermediate 6.17 failed, with only the starting substrate being recovered, the reverse addition, i.e., the addition of phenyl magnesium bromide to the previously prepared tert-butyl intermediate 6.20 afforded the desired hydrazine derivative in a good 66% overall yield (scheme 6.18c).

---

The subsequent application of our hydroarylation-isomerization protocol to substrates 6.11b-d furnished the desired cyclic products in 75-99% yields. However not surprisingly, the tert-butyl group in substrate 6.11e did not resist the acidic conditions and only a poor 25% yield was obtained for 6.12e according to $^1$H NMR analysis of the crude reaction mixture, with concomitant formation of an unidentified complex mixture of products (scheme 6.19).

\[
\begin{array}{cccc|ccc|c}
R & \text{product} & \text{time} & \text{yield} & \text{product} & \text{time} & \text{yield} \\
\hline
n-C_3H_7 & 6.12b & 2h 30min & 99\% & 6.13b & 1h & 91\% \\
Bn & 6.12c & 1h 30min & 87\% & 6.13c & 30min & 80\% \\
^3\text{Pr} & 6.12d & 1h & 96\% & 6.13d & 30min & 75\% \\
^3\text{Bu} & 6.12e & 3h & 25\%^a & & & \\
\end{array}
\]

\(^a\) estimated by $^1$H NMR spectroscopy. Not isolated.


We next turned our attention to aromatic rings possessing different functional groups and the regioselectivity coming from the hydroarylation process when asymmetrically substituted aromatic nuclei (such as those with meta-substitutents) were employed. Substrates 6.11f-p were prepared by the same synthetic route as substrates 6.11a-c, with this time, MeMgBr being systematically added. Gold catalysis with molecules 6.11f-k and subsequent treatment with a catalytic amount of p-TSA afforded products in good 63-85% isolated yields. Substrate 6.11l bearing the strong electron withdrawing NO\_2 group did not react (scheme 6.20a).

Aromatic rings with a methyl and chloro group as meta-substituents, 6.11m and 6.11n, respectively, afforded 1:1 mixtures of both possible regioisomers 6.12ma and...
6.12mb, and 6.12na and 6.12nb, respectively, without any selectivity (scheme 6.20b). Aromatic rings with a methyl and a chloro group as ortho-substituents produced very sluggish reactions, affording also 7-endo cyclization products, 6.12ob and 6.12pb, respectively. We speculate that the steric constraint caused by the necessary alignment of the methyl on the nitrogen atom and the substituent in the ortho-position of the aromatic rings renders this step so slow that the 7-membered ring enters as a competing pathway in the process (scheme 6.20c).

![Chemical reaction image]

<table>
<thead>
<tr>
<th>FG</th>
<th>product</th>
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<th>yield</th>
<th>product</th>
<th>time</th>
<th>yield</th>
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<td>OMe</td>
<td>6.11f</td>
<td>60%</td>
<td>6.12f</td>
<td>9h</td>
<td>83%</td>
<td>6.13f</td>
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<td>6.12g</td>
<td>18h</td>
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<td>6.13g</td>
<td>1h</td>
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<td>71%</td>
<td>6.12i</td>
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<td>63%</td>
<td>6.13j</td>
<td>4h 30min</td>
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<td>77%</td>
<td>6.12k</td>
<td>17h</td>
<td>76%</td>
<td>6.13k</td>
<td>1h 30min</td>
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<td>24h</td>
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</table>

*a*: Global yield from the corresponding hydrazine. *b*: Isolated yields

b) ![Chemical reaction image]

6.11m, R = Me, 73%
6.11n, R = Cl, 70%

6.12ma-na 3h 30min (6.12ma + 6.12mb) 96% (ratio 6.12ma:6.12mb 1:1)²
6.12na + 6.12nb 14h (6.12na + 6.12nb) 96% (ratio 6.12na:6.12nb 1:1)²

*²* Yield estimated from °H NMR spectroscopy. Although a 1:1 mixture is observed after 1h by °H NMR when the experiment is carried out in CDCl₃, prolonged reaction time promotes the degradation of regiosomer 6.12ma, while regiosomer 6.12mb can be isolated alone in 48%.

c) ![Chemical reaction image]

6.11a, R = Me, 72%
6.11b, R = Cl, 70%

6.12oa-7ap 32h (6.12oa + 6.12ob) 81% (ratio 6.12oa:6.12ob 6:1)

Scheme 6.20: Results obtained for the cyclization of aromatic hydrazines bearing different functional groups a) in the para-position 6.11f-l, and b) chloro- and methyl-substituted aryl hydrazines in the meta-position, 6.11m-n, and c) in the ortho-position 6.11o-p.
Internal alkynes \textit{6.11q-s} with a vinyl, a phenyl or an allyl group, respectively, and other substrates as \textit{6.11t}, -u did not afford any cyclized product, rather only the starting substrate was recovered (scheme 6.21a). Trials toward the rearrangement of dihydroquinolines \textit{6.13} to indoles \textit{6.14} with compounds \textit{6.13a-b} or \textit{6.13f-g} upon heating, acid catalyzed conditions or microwave irradiation gave only the corresponding starting materials. Although the use of a \( h \) lamp (wave length of sun light) afforded a complex mixture of products using compound \textit{6.13b}, the more electron-rich derivative \textit{6.13f} rearranged in a substantially cleaner transformation to indole \textit{6.14f}\textsuperscript{287} (scheme 6.21b). This represents our first successful lead towards the expected indole derivative of type \textit{6.14} but further experiments are still necessary to confirm the structure obtained and to study the reaction scope (\textit{cf.} previous experiments with malonate derivatives \textit{6.9} and \textsuperscript{1}H NMR of the crude reaction mixture. The small scale employed of \textit{ca.} 10 mg precluded product isolation).

\textsuperscript{287} Although we could observe the substantially clean formation of a major product from the \textsuperscript{1}H NMR of the crude reaction mixture, TLC analysis revealed a series of minor products to be also present in the reaction mixture.
Scheme 6.21: Trials concerning a) internal alkynes 6.11q-s and other substrates 6.11t,u also submitted to gold catalysis and b) rearrangement of dihydroquinolines 6.13a-b and 6.13f-g toward the corresponding indole derivatives.

Concerning the reaction mechanism, a detailed study was not performed and an unambiguous mechanism cannot be determined. Nevertheless, in recent related studies reported in the literature, gold(I)-catalysts have been documented to activate aryl C-H bonds\(^{288}\) and have also been proposed to react through a dual activation mode in the NXS-promoted halogenation of aromatic compounds by activating both the C-H bond and the substrate electrophilic moiety\(^{289}\). Although in our case, we cannot exclude the gold-metallation of the aromatic C-H bond to participate in the process, it seems hard to explain the reactivity pattern previously observed without passing by the activation of the alkyne triple bond in a Friedel-Crafts type mechanism. The main reason for this is that electron poor aromatic rings afforded reaction times considerably longer, presumably due to their lower nucleophilicity.

Furthermore, it is worth-mentioning that surprising results were obtained when different aryl groups on the nitrogen atom were employed in competition experiments performed with substrates 6.11v-x. Little discrimination in product regioselectivities was observed for the cyclization of 6.11v-x, which gave a 1:2 mixture of 6.12va:6.12vb and a 1:1.5 mixture of 6.12wa:6.12wb, respectively. Moreover, the phenyl cyclized product 6.12vb obtained as the major product was not anticipated, as we were expecting that the more electron-rich 4-(OMe)Ph ring would have a higher cyclization rate (scheme 6.21).


Scheme 6.21: Competition experiments using different substituted aromatic rings.

Although, in light of these results, rationalization of the observed selectivity is a difficult matter and further investigation on the reaction mechanism is still necessary, we speculate that the observed results can be possibly accounted by two effects: i) the electrophilic gold activation of the alkyne triple bond towards nucleophilic addition in a Friedel-Crafts-type mechanism and ii) the reversible complexation of the gold complex on the Lewis-basic nitrogen atom carrying both aromatic rings. This complexation would force either one aromatic ring or the other to be in a productive close-proximity position of the alkyne triple bond. This dual action promoted by the gold complex can possibly explain the little discrimination found in the cyclization observed with the methoxy and the ester groups in 6.11v and 6.11w, respectively, and also why the more electron poor cycle in 6.11v furnished the major cyclization product.

6.3 CONCLUSION

In summary, we developed in this chapter a practical and efficient synthetic route for the synthesis of cinnoline derivatives by a gold(I)-catalyzed hydroarylation process. Starting substrates 6.11 can be synthesized in a 4 steps/1 purification process from commercially available arylhydrazines. This procedure was also found to be compatible with the presence of different aryl or alkyl groups on the nitrogen atom proximal to the functionalized aromatic rings.
Based on cross-dehydrogenative coupling oxidations (cf. introduction chapter 5) and the already reported conjugate addition on indole derivatives\(^{290}\), a further functionalization of tetrahydrocinnoline derivatives can be proposed (scheme 6.22)

![Scheme 6.22: Potential functionalization route from tetrahydro cinnoline derivatives 6.12.](image)

A possible route to cinnoline rings can also be envisaged from the use of 4-Phenylpyrazole-3,5-dione 6.22, which upon conjugate addition of an ArMgX derivative and a propargylation step should afford intermediate 6.23 that should cyclize to tetrahydrocinnoline 6.24 upon treatment with a catalytic amount of XphosAu(NCCH\(_3\))SbF\(_6\). Intermediate 6.24 can either be submitted to a functionalization step as the one discussed in scheme 6.22 and further cleavage of the amide bonds by action of a strong base, such as KOH, to give access to cinnolines 6.26 or can be isomerized by a catalytic amount of \(\text{pTSA}\) and then have their amides bond cleaved to produce cinnolines of type 6.25.

![Scheme 6.23: Potential route to functionalized cinnolines 6.25 and 6.26.](image)

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APPENDICES A & B: ACRONYMS AND ABREVIATIONS & EXPERIMENTAL SECTION
APPENDIX A: Acronyms and abbreviations

A.1 Units

°C  Celsius degree
atm  Standard atmosphere
g  Gram
h  Hour
Hz  Hertz
kg  Kilogram
L  Liter
M  Molar concentration
mg  Milligram
MHz  Megahertz
min  Minute
mL  Milliliter
mmol  Millimole
mol  Mole
m/z  Mass unity on elementary charge
N  Normal concentration
ppm  parts per million

A.2 Chemical groups and compounds

Ac  Acetyl
Ad  Adamantyl

Acetyl

Ar  Aryl
Bn  Benzyl
Boc  tert-Butoxycarbonyl
Bu  Butyl
Bz  Benzoyl
Cy  Cyclohexyl
1,2-DCE  1,2-dichloroethane
DCM  Dichloromethane
DMAP  N,N-Dimethyl-4-aminopyridine
DMF  |  Dimethylformamide
DMSO |  Dimethylsulfoxide
dppm |  1,1-Bis(diphenylphosphino)methane
Et   |  Ethyl
IMes |  \(N,N'\)-bis(2,4,6-trimethylphenyl)imidazol-2-ylidene

\(\text{^{'Pr}}\) |  Isopropyl
IPr  |  \(N,N'\)-bis(2,6-diisopropylphenyl)imidazol-2-ylidene

JohnPhos |  2-(Di-\(\text{^{'Bu}}\)-tert-butylphosphino)biphenyl

KHMDMS |  Potassium bis(trimethylsilyl)amide
LDA   |  Lithiumdiisopropylamide
Me    |  Methyl
Mes   |  Mesityl, \textit{i.e.}, 2,4,6-trimethylphenyl
\(m\)-CPBA |  \textit{meta}-chloroperbenzoic acid
Ms    |  Mesyl, \textit{i.e.}, Methanesulfonyl
2-Napht |  2-Naphtyl
\(n\)-Bu |  \(n\)-Butyl
NBS   |  \(N\)-bromosuccinimide
NHC   |  \(N\)-heterocyclic carbene
NIS   |  \(N\)-iodosuccinimide
\(n\)-Pr |  \(n\)-Propyl
NXS   |  \(N\)-halosuccinimide
PE    |  Petroleum ether
Ph    |  Phenyl
phen  |  1,10-phenanthroline
<table>
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<tr>
<th>Abbreviation</th>
<th>Definition</th>
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<td>Piv</td>
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<tr>
<td>PMB</td>
<td>para-methoxy benzyl</td>
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<td>p-TSA</td>
<td>para-toluene sulfonic acid</td>
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<tr>
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</tr>
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<td>TFA</td>
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<tr>
<td>Tf</td>
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</tr>
<tr>
<td>Tol</td>
<td>Toly, i.e., 4-methyl-phenyl</td>
</tr>
<tr>
<td>Ts</td>
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</tr>
<tr>
<td>Xphos</td>
<td>2-Dicyclohexylphosphino-2′,4′,6′-triisopropylbiphenyl</td>
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### A.3 Other acronyms and abbreviations

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<tr>
<th>Abbreviation</th>
<th>Definition</th>
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<td>Aqueous</td>
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<td>Axial</td>
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<td>bs</td>
<td>Broad singulet (in NMR)</td>
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<tr>
<td>cat.</td>
<td>Catalytic</td>
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<tr>
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<td>de</td>
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<td>Degradation</td>
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</tr>
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<td>ee</td>
<td>Enantiomeric excess</td>
</tr>
<tr>
<td>Abbreviation</td>
<td>Full Form</td>
</tr>
<tr>
<td>--------------</td>
<td>-----------</td>
</tr>
<tr>
<td>EI⁺</td>
<td>Positive electron ionization (in MS)</td>
</tr>
<tr>
<td>eq</td>
<td>Equatorial</td>
</tr>
<tr>
<td>equiv.</td>
<td>Equivalent</td>
</tr>
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<tr>
<td>EWG</td>
<td>Electron withdrawing group</td>
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<td>Heptuplet (in NMR)</td>
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<td>High resolution mass spectroscopy</td>
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<td>meta-</td>
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<td>Molecular formula</td>
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<td>Mass spectroscopy</td>
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<tr>
<td>Nu</td>
<td>Nucleophile</td>
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<tr>
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<td>ortho-</td>
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<tr>
<td>p-</td>
<td>para-</td>
</tr>
<tr>
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</tr>
<tr>
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<td>Quantitative</td>
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<tr>
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</tr>
<tr>
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</tr>
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<td>Temperature</td>
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<td>TLC</td>
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</tr>
<tr>
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<td>Turnover frequency</td>
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<tr>
<td>TON</td>
<td>Turnover number</td>
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<td>UV</td>
<td>Ultraviolet</td>
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<td>Wavenumber (in IR)</td>
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APPENDIX B: Experimental Section

Part of the work presented in this manuscript is the result of the collaboration with some members of our research group. In subjects where other students were involved, a red star (*) was previously used to mark the experiments performed by the author of this manuscript. Only these experiments will be described in this section. The contribution of each member was acknowledged at the beginning of each chapter in the Main Sections and will be restated here, together with the reference of the journal in which the results were published.

**Chapter 1:** Individual work under the guidance of Fabien Gagosz and Samir Zard.

**Chapter 3:** Collaboration with Andrea K. Buzas, Florin Istrate and Yann Odabachian under the guidance of Fabien Gagosz.

**Chapter 4:** Individual work under the guidance of Fabien Gagosz.

*Unpublished results*

**Chapter 5:** Collaboration with Yann Odabachian under the guidance of Fabien Gagosz.

**Chapter 6:** Individual work under the guidance of Fabien Gagosz
Compounds characterized in Chapter 1:

1.17f

1.17g

1.17m

1.19d

1.19e

1.19f

1.19g

1.19h

1.19i

1.19j

1.22a

1.22b

1.22c

1.22d

1.22e

1.22f

1.22g

1.22h

1.22i

1.22j

1.22k

1.22l

1.22m

1.22n
Compounds characterized in Chapter 3:

- Boc\(\text{N} = \text{Ph}\) 3.15b
- Boc\(\text{N} = \text{n-C}_8\text{H}_{11}\) 3.15j
- Boc\(\text{N} = \text{OAc}\) 3.15p
- Boc\(\text{N} = \text{OAc}\) 3.15q
- Boc\(\text{N} = \text{Ph}\) precursor 3.15w/x
- Boc\(\text{N} = \text{Ph}\) 3.15y
- Boc\(\text{N} = \text{OAc}\) 3.15o
- Boc\(\text{N} = \text{Ph}\) 3.15f

3.15k
3.15l
3.15m
3.15n
3.15o
3.15p
3.15q
3.15w
3.15x
3.16a
3.16b
3.16c
3.16d
3.16e
3.16f
3.16g
3.16h
3.16i
3.16j
3.16k
3.16l
3.16m
3.16n
3.16o
3.16p
3.16q
3.16r
3.16s
3.16t
3.16u
3.16v
3.16w
3.16x
3.16y
3.16z
Compounds characterized in Chapter 4:

- EtO₂C CO₂Et
- MeO
- Cl
- Br
- TMS
- Ph
- n-C₃H₇
Compounds characterized in Chapter 5:

5.17a

5.17b

5.17e
Compounds characterized in Chapter 6:
B.1 General Methods

B.1.1 Reagents and Solvents

Commercially available reagents were used as received without further purification.

All reaction solvents correspond to SDS “pure for synthesis” solvents and were used as received, with the exception of dry Et\textsubscript{2}O and THF, which were obtained by distillation from Na/benzophenone and dry DCM and toluene, which were obtained by distillation from CaH\textsubscript{2}.

Solvents for flash column chromatographies correspond to SDS “pure for synthesis” solvents and were used as received.

The same holds for HPLC solvents, which correspond to the SDS “HPLC grade”.

B.1.2 Experimental Procedures

All non-aqueous reactions were performed under dry Ar or N\textsubscript{2} atmosphere using standard syringe/cannula/septum techniques, unless specified otherwise.

Concentration under reduced pressure was performed by rotator evaporation at room temperature using a water jet pump.

Purified compounds were further dried on a high vacuum pump.

B.1.3 Chromatography

Thin layer chromatography (TLC) were done on Merck Silica Gel 60 F\textsubscript{254} plates (aluminium plates coated with silica gel 60). They were visualized under UV light at 254 or 365 nm, then stained using KMnO\textsubscript{4} or anisaldehyde solution. KMnO\textsubscript{4} solution was prepared from 600 mL of water, 6g of KMnO\textsubscript{4} 40 g of K\textsubscript{2}CO\textsubscript{3} and 0.5mL concentrated acetic acid. Anisaldehyde solution was prepared from 26 mL \textit{p}-anisaldehyde, 950 mL ethanol 95%, 35 mL concentrated sulfuric acid and 10.5 mL concentrated acetic acid.

Flash column chromatography were performed on SDS 60 CC 40-63 silica gel (pore-size 60 Å, particle size 40-63 µm) using a forced flow of eluent at 0.1-0.5 bar pressure.

B.1.4 Analytical Methods

NMR spectra were recorded on a Bruker Avance 400 operating at a 400 MHz for \textsuperscript{1}H and 100MHz for \textsuperscript{13}C nuclei.
In $^1$H NMR chemical shifts ($\delta$) are expressed in ppm using tetramethylsilane ($\delta = 0$ ppm) or the residual peak of chloroform ($\delta = 7.26$ ppm) as an internal reference. Coupling constants $J$ are given in Hertz (Hz).

In $^{13}$C NMR, chemical shifts ($\delta$) are expressed in ppm taking the chloroform residual peak ($\delta = 77.0$ ppm) as an internal reference.

**IR spectra** were recorded with a Perkin Elmer FT-1600 spectrometer. Oil or solid samples were dissolved in carbon tetrachloride and then placed in a sodium chloride cell. The spectra values are reported as absorption maxima and expressed in cm$^{-1}$.

**Mass spectra (MS)** were recorded on a Hewlett Packard HP-5890 B using positive electron ionization (EI$^+$) or positive chemical ionization with ammonia (Cl$^+$, NH$_3$) methods. Fragment signals are given as a mass-to-charge ratio (m/z).

All analyses were performed at the DCSO laboratory (Laboratoire de Chimie de Synthèse Organique) of Ecole Polytechnique.

**B.1.5 Softwares**

The 1D and 2D NMR FID spectral data was processed and visualized using MesctReC 4.7.0.0.

This manuscript was written using Microsoft Word, versions 2002 and 2007. Chemical formulas were edited in Cambridge Soft’s Chemdraw Standard 6.0 and 8.0.

The name of the synthesized compounds was determined according to the CAS nomenclature using Beilstein AutoNom 2000, version 4.01.305.
B.2 The catalysts synthesis

B.2.1 Synthesis of phospine gold(I) bis(trifluoromethanesulfonylimidate complexes

The phospine gold(I) bis(trifluoromethanesulfonylimidate complexes used during our work were obtained according to the method developed in our laboratory from the corresponding phospine gold(I) chloride derivatives upon treatment with AgNTf₂ in DCM at rt.

\[
\text{R}_3\text{P} = \text{PPh}_3, (p\text{-CF}_3\text{C}_6\text{H}_4)_3\text{P}, \text{Ad}_2(n\text{-Bu})\text{P}, \text{Johnphos}, \text{Xphos}, \text{etc.}
\]

**General Procedure B.2.1:** The phospine gold(I) chloride complex (1 equiv.) was dissolved in DCM (0.04M) at rt and AgNTf₂ (1 equiv.) was added resulting in the instantaneous formation of the expected silver chloride precipitate. The mixture was stirred for an additional 15 min and the formation of a single complex was observed in \(^{31}\text{P} \) NMR of the crude mixture. Filtration over celite to remove the silver chloride salts resulted in a pale colored solution. The expected complex was obtained quantitavely by evaporation to dryness under reduced pressure and drying under vacuum.

B.2.2 Synthesis of phospine gold chloride complexes

Phosphine gold chloride complexes, such as XPhosAuCl, tert-butylixphosAuCl and Ad₂(n-Bu)PAuCl can be prepared from Me₂S.AuCl upon treatment with the corresponding phospine by simple ligand exchange.

**General Procedure B.2.2:** Me₂S.AuCl (1 equiv.) and the phospine (1 equiv.) were weighed in air then placed under nitrogen. DCM (0.04 M) was added, which resulted in the dissolution of the starting material and the formation of a pale yellow solution. The mixture was stirred at rt for 15 min and the complete formation of the desired gold chloride complex was verified by \(^{31}\text{P} \) NMR. The volume of the solution was reduced to 1 mL and then 5 mL of hexane was added, resulting in the precipitation of the complex. The solid was filtered,
washed with hexane and dried under vacuum, resulting in the quantitative isolation of the title compound.

**B.2.3 Synthesis of \( \text{PR}_3\text{Au(NCCH}_3\text{)}\text{SbF}_6 \)**

The synthesis of the gold complex \( \text{PR}_3\text{Au(NCCH}_3\text{)}\text{SbF}_6 \) is performed by treating the phosphine gold complex \( \text{PR}_3\text{AuCl} \) with the silver salt \( \text{AgSbF}_6 \) in acetonitrile at rt.

\[
\text{R}_3\text{P—Au—Cl} \xrightarrow{\text{AgSbF}_6, \text{CH}_3\text{CN, rt, 24 h}} \left[ \text{R}_3\text{P—Au—NCCMe} \right]^+ \text{SbF}_6^- + \text{AgCl}
\]

**General procedure B.2.3:** To a mixture of \( \text{PR}_3\text{AuCl} \) (1 equiv.) and \( \text{AgSbF}_6 \) (1 equiv.), acetonitrile (0.125 M) was added. The reaction mixture was allowed to stir at rt for 24h and then it was filtered through a pad of celite. The filtrate was evaporated under reduced pressure to give the expected cationic complex.
B.3 Synthesized Molecules

B.3.1. Chapter 1: Unusual approach to branched 3-alkynylamides and to 1,5-dihydropyrrrol-2-ones

All molecules synthesized for the publication of this work\textsuperscript{291} are described below. This project was developed individually.

B.3.1.1 Synthesis of isoxazolin-5-ones

Isoxazolin-5-ones were synthesized as depicted below:

![Diagram of synthetic route]

Isoxazolin-5-ones synthesized:

- 1.22a, $R_1^1 = \text{Ph}$, $R_2^1 = \text{H}$, $R_3^1 = \text{H}$, $R_4^1 = \text{Bu}$
- 1.22b, $R_1^1 = \text{Ph}$, $R_2^1 = 4$-Me-Ph, $R_3^1 = \text{H}$, $R_4^1 = \text{Bu}$
- 1.22c, $R_1^1 = \text{Ph}$, $R_2^1 = 4$-Me-Ph, $R_3^1 = \text{H}$, $R_4^1 = \text{CH}_2\text{CO}_2\text{Et}$
- 1.22d, $R_1^1 = \text{Ph}$, $R_2^1 = 4$-Me-Ph, $R_3^1 = \text{H}$, $R_4^1 = \text{CH}_2\text{Ph}$
- 1.22e, $R_1^1 = \text{Ph}$, $R_2^1 = 2$-thiophenyl, $R_3^1 = \text{H}$, $R_4^1 = \text{Bu}$
- 1.22f, $R_1^1 = \text{Ph}$, $R_2^1 = \text{Me}$, $R_3^1 = \text{Me}$, $R_4^1 = \text{Bu}$
- 1.22g, $R_1^1 = \text{Ph}$, $R_2^1 = \text{H}$, $R_3^1 = (\text{CH}_2)_4$, $R_4^1 = \text{Bu}$
- 1.22h, $R_1^1 = \text{CH}_2\text{CO}_2\text{Me}$, $R_2^1 = \text{H}$, $R_3^1 = \text{H}$, $R_4^1 = \text{Bu}$
- 1.22i, $R_1^1 = \text{CH}_2\text{CO}_2\text{Me}$, $R_2^1 = (\text{CH}_2)_4$, $R_3^1 = \text{Bu}$
- 1.22j, $R_1^1 = \text{CH}_2\text{OCH}_2\text{Ph}$, $R_2^1 = 4$-Me-Ph, $R_3^1 = \text{H}$, $R_4^1 = \text{Bu}$
- 1.22k, $R_1^1 = \text{CH}_2\text{CO}_2\text{Me}$, $R_2^1 = \text{H}$, $R_3^1 = \text{H}$, $R_4^1 = \text{Bu}$
- 1.22l, $R_1^1 = \text{CH}_2\text{OCH}_2\text{Ph}$, $R_2^1 = \text{H}$, $R_3^1 = \text{H}$, $R_4^1 = \text{CH}_2\text{Ph}$
- 1.22m, $R_1^1 = \text{CH}_2\text{OCH}_2\text{Ph}$, $R_2^1 = (\text{CH}_2)_4$, $R_3^1 = \text{Bu}$
- 1.22n, $R_1^1 = \text{CH}_2\text{OCH}_2\text{Ph}$, $R_2^1 = (\text{CH}_2)_4$, $R_3^1 = \text{CH}_2\text{CO}_2\text{Et}$

Scheme B.3.1.1: Synthetic route employed for the synthesis of substrates 1.22a-n.

4-Benzyllox-3-oxo-butryric acid ethyl ester and 3-oxo-hept-6-enoic acid ethyl ester were synthetized as described in the literature\textsuperscript{292}. All other β-ketoesters were commercially afforded.

General procedure 1.1A, *Synthesis of the 5-isoxazolone ring:* A mixture of NH$_2$OH.HCl (1 equiv.) and NaOAc (1 equiv.) in acetic acid (0.5 M) was allowed to stir for 15 min. at room temperature. The β-ketoester (1 equiv.) was slowly added to the reaction mixture and the reaction was allowed to stir for 1 h at room temperature until complete consumption of the starting β-ketoester (a mixture of the corresponding oxime and the final product is formed). The mixture was then heated up to 80-90 °C for 30 min, when no more oxime was detected on TLC. The solution was then filtered and concentrated under reduced pressure. H$_2$O and DCM were added. The organic layers were separated and extracted with DCM (2x). The combined organic extracts were dried (MgSO$_4$) and concentrated under reduced pressure. In most of the cases, the desired compound can be obtained pure and no purification step is necessary.

General procedure 1.1B, *Synthesis of the isoxazolin-5-ones:* A mixture of the β-ketoester (1 equiv.), EtOH (0.5 M), NH$_2$OH.HCl (1.5 equiv.) and NaOAc (1.5 equiv.) was heated up to reflux (78 °C) until complete consumption of the starting material (TLC). The temperature was allowed to cool to room temperature. At this point, a catalytic amount of HCl 37% (ca. 20 µL per mmol of β-ketoester) was added and the reaction was heated to reflux (78 °C) until the complete disparition of the oxime (TLC). The mixture was filtered, extracted with DCM (3x), dried (MgSO$_4$) and concentrated under reduced pressure. The product was generally obtained pure and no purification step was necessary.

General procedure 1.2, *Knoevenagel condensation of isoxazolin-5-ones with aldehydes/ketones:* To a round bottom flask charged with isoxazolin-5-one (1 equiv.) in isopropanol (0.5 M) was added the aldehyde (1.2 equiv.) or ketone (2.0 equiv.) and a catalytic amount of piperidine (ca. 0.1 mL/mm mol isoxazolin-5-one). The resulting solution was then stirred to 50 °C in the case of aldehydes or to reflux (82 °C) in the case of ketones. Upon completion (TLC), most of the solvent was removed under partial vacuum. In the case of aldehydes, the products generally precipitated and were isolated by filtration followed by washes with petroleum ether. In the other cases, the solution was completely evaporated and purification was accomplished by flash column chromatography (silica gel, PE:AcOEt).


Isocyanides EtO$_2$CCH$_2$NC 1.20b and PhCH$_2$NC 1.20c were synthesized from the corresponding amines as described in the literature. $^1$BuNC was commercially afforded.

**General Procedure 1.3, 1,4-addition of isocyanides to alkylidene isoxazolin-5-ones:** To a round bottom flask charged with isocyanide (1 equiv.) and THF (0.33 M) was added the alkylidene isoxazolin-5-one (1 equiv.), followed by a small amount of water (50 µL/mmol isoxazolin-5-one). The temperature was heated up to reflux (67°C) and the reaction mixture was allowed to stir at this temperature. Upon reaction completion (TLC), the solvent was removed under reduced pressure. Purification by flash column chromatography afforded the pure products in the stated yields.

<table>
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<tr>
<th>(5-Oxo-4,5-dihydro-isoxazol-3-yl)-acetic acid methyl ester</th>
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<tbody>
<tr>
<td><img src="" alt="Chemical Structure" /></td>
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</tr>
<tr>
<td>MF: C$_6$H$_7$O$_4$N</td>
<td></td>
</tr>
<tr>
<td>MW = 157 g.mol$^{-1}$</td>
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</table>

**Method:** See general procedure 1.1.B using (1 equiv., 20 mmol, 3.48 g) of 3-oxo-pentanedioic acid dimethyl ester.

**Purification:** none/ $R_f$ (1:1 PE:AcOEt): 0.36.

**Product:** Yellow solid.

**Yield:** 98%.

$^1$H NMR (δ, ppm) 3.78 (s, 3H, OCH$_3$), 3.61 (s, 2H, CH$_2$), 3.57 (s, 2H, CH$_2$).

(CDCl$_3$, 400 MHz)

$^{13}$C NMR (δ, ppm) 174.7 (Cq, C=O), 167.7 (Cq, C=O), 160.7 (Cq, C=N), 52.8 (OCH$_3$), 35.7 (CH$_2$), 34.4 (CH$_2$).

(CDCl$_3$, 100 MHz)

---

3-Benzylxoxymethyl-4H-isoxazol-5-one 1.24c

\[
\text{MF: } C_{11}H_{11}O_3N \\
\text{MW = 205 g.mol}^{-1}
\]

**Method:** See general procedure 1.1.A using (1 equiv., 5 mmol, 1.18 g) of 4-benzylxoxo-3-oxo-butyric acid ethyl ester.

**Purification:** Flash column chromatography (silica gel, 8:2 PE: AcOEt)/ \text{Rf} (7:3 PE: AcOEt): 0.42.

**Product:** Dark orange oil.

**Yield:** 68%.

**\(^1\)H NMR (\(\delta\), ppm)**

(CDCl\(_3\), 400 MHz) 7.40-7.32 (m, 5H, CH-Ph), 4.58 (s, 2H, OCH\(_2\)), 4.32 (s, 2H, OCH\(_2\)), 3.44 (s, 2H, CH\(_2\)).

**\(^13\)C NMR (\(\delta\), ppm)**

(CDCl\(_3\), 100 MHz) 174.7 (Cq, C=O), 164.5 (Cq, C=N), 136.4 (Cq, Ph), 128.6 (CH-Ph), 128.3 (CH-Ph), 128.0 (CH-Ph), 73.4 (OCH\(_2\)), 64.8 (OCH\(_2\)), 34.4 (CH\(_2\)).

**IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))**

3003 (w), 2955 (w), 2848 (w), 1813 (s), 1748 (s), 1611 (w), 1509 (w), 1437 (w), 1407 (w), 1374 (w), 1293 (w), 1205 (m), 1156 (m), 1003 (w).

**HRMS (EI+, m/z):**

3-But-3-enyl-4H-isoxazol-5-one

\[
\begin{align*}
\text{Method:} & \quad \text{See general procedure 1.1.A using (1 equiv., 5 mmol, 850 mg) of 3-oxo-hept-6-enoic acid ethyl ester.} \\
\text{Purification:} & \quad \text{none/} \quad R_f (9:1 \ PE: \text{AcOEt}): 0.38. \\
\text{Product:} & \quad \text{Dark orange oil.} \\
\text{Isolated yield:} & \quad 95\%. \\
\end{align*}
\]

\[\begin{align*}
^1\text{H NMR (}\delta, \text{ ppm}) & \quad 5.81 (\text{ddt, } 1\text{H, } J = 6.6\text{Hz, } J = 10.2\text{Hz, } J = 15.3\text{Hz, } 1\text{H, CH}=\text{CH}_2), 5.13 (\text{ddt, } J = 1.5\text{ Hz, } J = 1.6\text{Hz, } J = 15.3\text{Hz, } 1\text{H, CH}=\text{CH}_2), 5.09 (\text{ddt, } J = 1.5\text{Hz, } J = 1.6\text{Hz, } J = 6.6\text{Hz, } 1\text{H, CH}_2-\text{homo allyl chain}), 3.38 (\text{s, } 2\text{H, CH}_2\text{CO}_2), 2.59 (\text{t, } J = 7.4\text{ Hz, } 2\text{H, CH}_2-\text{homo allyl chain}), 2.42-2.37 (\text{m, } 2\text{H, CH}_2-\text{homo allyl chain}). \\
^{13}\text{C NMR (}\delta, \text{ ppm}) & \quad 175.1 (\text{Cq, OC}=\text{O}), 166.2 (\text{Cq, C}=\text{N}), 135.6 (\text{CH}=\text{CH}_2), 116.9 (\text{CH}_2=\text{CH}_2), 35.9 (\text{CH}_2\text{CO}_2), 29.7 (\text{CH}_2-\text{homo allyl chain}), 28.4 (\text{CH}_2-\text{homo allyl chain}). \\
\text{IR (}\nu, \text{ cm}^{-1}\text{)} (\text{CCl}_4) & \quad 3082 (\text{w}), 2980 (\text{w}), 2924 (\text{w}), 1807 (\text{s}), 1692 (\text{w}), 1641 (\text{w}), 1611 (\text{w}), 1441 (\text{w}), 1380 (\text{w}), 1320 (\text{w}), 1260 (\text{w}), 1160 (\text{m}). \\
\text{HRMS (EI+, m/z):} & \quad \text{Calculated: 139.0633 } \text{found: 139.0629.}
\end{align*}\]

\[
\begin{align*}
\{4-[1-\text{Cyclopropyl-methylidene]-5-oxo-4,5-dihydro-isoxazol-3-yl]-acetic acid methyl ester (E and Z)\}
\end{align*}
\]

\[
\begin{align*}
\text{MF:} & \quad C_{10}H_{11}O_4N \\
\text{MW:} & \quad 209 \text{ g.mol}^{-1}
\end{align*}
\]
Method: See general procedure 1.2 using (1 equiv., 10 mmol, 1.57 g) of (5-oxo-4,5-dihydro-isoxazol-3-yl)-acetic acid methyl ester.

Purification: Flash column chromatography (silica gel, 6:4 PE:AcOEt)/ $R_f$ (1:1 PE: AcOEt): 0.40.

Product: Pale yellow solid.

Yield: 48% (inseparable mixture, ratio isomers 5:1).

$^1$H NMR (δ, ppm) [CDCl$_3$, 400 MHz] Major isomer: 6.49 (d, $J = 11.6$ Hz, 1H, =CH), 3.59 (s, 3H, OCH$_3$), 3.06-2.97 (m, 1H, CH-cyclopropane), 1.36-1.31 (m, 2H, CH$_2$-cyclopropane). Minor isomer: 6.43 (d, $J = 12.4$ Hz, 1H, =CH), 3.72 (s, 2H, COCH$_3$), 3.63 (s, 3H, OCH$_3$), 1.91-1.82 (m, 1H, CH-cyclopropane), 1.02-0.97 (m, 2H, CH$_2$-cyclopropane).

$^{13}$C NMR (δ, ppm) [CDCl$_3$, 100 MHz] Major isomer: 169.1 (Cq, C=O), 167.5 (Cq, C=O), 166.1 (=CH), 156.2 (Cq, C=N), 117.7 (Cq, =CCO$_2$N), 52.3 (OCH$_3$), 31.6 (CH$_2$CO$_2$Me), 14.6 (CH-cyclopropane), 13.4 (CH$_2$-cyclopropane). Minor isomer: 169.1 (Cq, C=O), 167.5 (Cq, C=O), 163.5 (=CH), 155.7 (Cq, C=N), 117.7 (Cq, =CCO$_2$N), 52.4 (OCH$_3$), 33.9 (CH$_2$CO$_2$Me), 15.4 (CH-cyclopropane), 13.3 (CH$_2$-cyclopropane).

IR (ν, cm$^{-1}$) (CCl$_4$) 3075 (w), 3010 (w), 2955 (w), 2848 (w), 1770 (s), 1750 (s), 1646 (s), 1436 (w), 1330 (w), 1297 (w), 1256 (w), 1203 (w), 1171 (w), 1109 (w), 1057 (w), 1019 (w)

HRMS (EI+, m/z): Calculated: 209.0688 found: 209.0689.

(4-Cyclopentylidene-5-oxo-4,5-dihydro-isoxazol-3-yl)-acetic acid methyl ester 1.19g

MF: C$_{11}$H$_{23}$O$_4$N

MW = 223 g.mol$^{-1}$

Method: See general procedure 1.2 using (1 equiv., 10 mmol, 1.57 g) of (5-oxo-4,5-dihydro-isoxazol-3-yl)-acetic acid methyl ester.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ $R_f$ (7:3 PE: AcOEt): 0.58.

Product: Brown oil.
Isolated yield: 74%.

$^1$H NMR (δ, ppm) 3.77 (s, 3H, OCH$_3$), 3.73 (s, 2H, CH$_2$CO$_2$), 3.12 (t, $J = 7.4$ Hz, 2H, CH$_2$-cyclopentane), 2.75 (t, $J = 6.2$Hz, 2H, CH$_2$-cyclopentane), 1.90-1.85 (m, 4H, CH$_2$-cyclopentane).

$^{13}$C NMR (δ, ppm) 182.8 (Cq, C=O), 169.0 (Cq, C=O), 168.3 (Cq, C=N), 155.7 (Cq), 114.2 (Cq, =COC$_2$N), 52.8 (OCH$_3$), 35.6 (CH$_2$-cyclopentane), 35.0 (CH$_2$-cyclopentane), 34.4 (CH$_2$CO$_2$), 25.9 (CH$_2$-cyclopentane), 25.2 (CH$_2$-cyclopentane).

IR (ν, cm$^{-1}$) (CCl$_4$) 3509 (w), 2965 (s), 2881 (s), 1750 (s), 1649 (s), 1555 (w), 1438 (m), 1409 (s), 1330 (s), 1295 (m), 1260 (s), 1202 (s), 1175 (s), 1134 (s), 1017(s), 993 (m).

HRMS (EI+, m/z) Calculated: 223.0845 found: 223.0845.

3-Benzylxymethyl-4-[1-p-tolyl-methylidene]-4H-isoxazol-5-one 1.19h

MF: C$_{19}$H$_{17}$O$_3$N

MW = 307 g.mol$^{-1}$

Method: See general procedure 1.2 using (1 equiv., 5 mmol, 1.03 g) of 3-benzylxymethyl-4H-isoxazol-5-one.

Purification: Flash column chromatography (silica gel, 9:1 PE: AcOEt)/ $R_f$ (7:3 PE: AcOEt): 0.61.

Product: Yellow solid.

Isolated yield: 52% (for two steps).

$^1$H NMR (δ, ppm) 8.26 (d, $J = 8.4$ Hz, 2H, CH-tolyl), 7.83 (s, 1H, =CH), 7.38-7.32 (m, 7H, CH-Ph + CH-tolyl), 4.60 (s, 2H, OCH$_3$), 4.60 (s, 2H, OCH$_2$), 2.46 (s, 3H, CH$_3$).

$^{13}$C NMR (δ, ppm) 161.1 (Cq, C=O), 152.1 (=CH), 146.1 (Cq, C=N), 136.7 (Cq, Ar), 134.4 (CH-Ar), 130.2 (Cq, Ar), 129.9 (CH-Ar), 128.7 (CH-Ar), 128.5 (Cq, Ar), 128.3 (CH-Ar), 116.3 (Cq, =COC$_2$N), 72.8 (OCH$_3$), 63.7 (OCH$_2$), 22.1 (CH$_3$).

IR (ν, cm$^{-1}$) (CCl$_4$) 3067 (w), 3032 (w), 2923 (w), 2865 (w), 1761 (s), 1696 (s), 1622 (s), 1597 (s), 1559 (s), 1499 (w), 1452 (w), 1419 (w), 1372 (w), 1349 (w), 1318 (w), 1287 (w),
3-But-3-enyl-4-[1-cyclopropyl-meth-(Z)-ylidene]-4H-isoxazol-5-one (E and Z)  \[ \text{1.19i} \]

**Method:** See **general procedure 1.2** using (1 equiv., 10 mmol, 1.39 g) of 3-but-3-enyl-4H-isoxazol-5-one.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt) / \( \text{Rf} \) (9:1 PE:AcOEt): 0.45.

**Product:** Pale green solid.

**Isolated yield:** 70% (inseparable mixture, ratio isomers 10: 1).

**\[ ^1\text{H} \text{NMR} \] (\( \delta \), ppm)**

**Major isomer:** 6.30 (d, \( J = 11.2 \) Hz, 1H, =CH), 5.89-5.79 (m, 1H, CH=CH\(_2\)), 5.12-5.03 (m, 2H, CH=CH\(_2\)), 3.25-3.16 (m, 1H, CH-cyclopropane), 2.58 (dd, \( J = 6.6 \) Hz, \( J = 8.6 \) Hz, 2H, CH\(_2\)-homo allyl chain), 1.45-1.40 (m, 2H, CH\(_2\)-cyclopropane), 1.01-0.97 (m, 2H, CH\(_2\)-cyclopropane).

**Minor isomer:** 6.26 (d, \( J = 12.0 \) Hz, 1H, =CH.), 5.89-5.79 (m, 1H, CH=CH\(_2\)), 5.12-5.03 (m, 2H, CH=CH\(_2\)), 2.85 (ddd, \( J = 6.6 \) Hz, \( J = 8.6 \) Hz, 2H, CH\(_2\)-homo allyl chain), 2.54-2.49 (m, 2H, CH\(_2\)-homo allyl chain), 2.12-2.03 (m, 1H, CH-cyclopropane), 1.45-1.40 (m, 2H, CH\(_2\)-cyclopropane), 1.10-1.07 (m, 2H, CH\(_2\)-cyclopropane).

**\[ ^{13}\text{C} \text{NMR} \] (\( \delta \), ppm)**

**Major isomer:** 169.8 (C\(_q\), C=O), 162.9 (=CH), 161.1 (C\(_q\), C=N), 136.2 (CH=CH\(_2\)), 119.5 (C\(_q\), \( \text{CCO}_2\)), 116.3 (CH=CH\(_2\)), 30.0 (CH\(_2\)-homo allyl chain), 25.2 (CH\(_2\)-homo allyl chain), 14.1 (CH-cyclopropane), 13.2 (CH\(_2\)-cyclopropane).

**Minor isomer:** 170.0 (C\(_q\), C=O), 161.7 (=CH), 161.1 (C\(_q\), C=N), 136.1 (CH=CH\(_2\)), 119.2 (C\(_q\), \( \text{CCO}_2\)), 116.3 (CH=CH\(_2\)), 29.8 (CH\(_2\)-homo allyl chain), 28.3 (CH\(_2\)-homo allyl chain), 15.0 (CH-cyclopropane), 13.1 (CH\(_2\)-cyclopropane).

**IR (\( v \), cm\(^{-1}\))** (CCl\(_4\))

3512 (w), 3080 (w), 3009 (w), 2985 (w), 2926 (w), 2858 (w), 1767 (s), 1647 (s), 1564 (w), 1446 (w), 1288 (w), 1180 (w), 1120 (w)

\[ \text{MF}: \text{C}_{11}\text{H}_{13}\text{O}_2\text{N} \]

\[ \text{MW} = 191 \text{ g.mol}^{-1} \]
**HRMS** (EI+, m/z) : calculated: 191.0946 found: 191.0950.

<table>
<thead>
<tr>
<th><strong>3-But-3-enyl-4-cyclopentyldiene-4H-isoxazol-5-one</strong></th>
<th>1.19j</th>
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<tbody>
<tr>
<td><img src="image" alt="Structure" /></td>
<td>MF: C_{12}H_{15}O_{2}N</td>
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<tr>
<td>MW = 205 g. mol^{-1}</td>
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</table>

**Method**: See *general procedure 1.2* using (1 equiv., 10 mmol, 1.39 g) of 3-but-3-enyl-4H-isoxazol-5-one.

**Purification**: Flash column chromatography (silica gel, 9:1 PE:AcOEt)/ R_f (9:1 PE:AcOEt): 0.30.

**Product**: Orange pale solid.

**Isolated yield**: 61%.

<table>
<thead>
<tr>
<th><strong>^1H NMR</strong> (δ, ppm) (CDCl₃, 400 MHz)</th>
<th></th>
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<tbody>
<tr>
<td>5.90 (ddt, J = 6.5Hz, J = 10.2Hz, J = 17.0Hz, 1H, CH=CH₂), 5.12 (ddd, J = 1.5Hz, J = 3.2Hz, J = 17.0Hz, 1H, CH=CH₂), 5.07 (ddd, J = 1.5Hz, J = 2.7Hz, J = 10.2Hz, 1H, CH=CH₂), 3.11 (t, J = 6.8 Hz, 2H, CH₂-cyclopentane), 2.85 (t, J = 6.4 Hz, 2H, CH₂-cyclopentane), 2.73 (m, 2H, CH₂-cyclopentane), 2.51-2.45 (m, 2H, CH₂-cyclopentane), 1.93-1.83 (m, 4H, CH₂-homo allyl chain).</td>
<td></td>
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</tbody>
</table>

<table>
<thead>
<tr>
<th><strong>^13C NMR</strong> (δ, ppm) (CDCl₃, 100 MHz)</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>181.2 (Cq, C=O), 169.6 (Cq, C=N), 161.0 (Cq), 136.3 (CH=CH₂), 116.1 (CH=CH₂), 114.9 (Cq, =CCO₂N), 25.4 (CH₂-cyclopentane), 35.2 (CH₂-cyclopentane), 29.7 (CH₂-cyclopentane), 28.5 (CH₂-cyclopentane), 25.9 (CH₂-homo allyl chain), 25.2 (CH₂-homo allyl chain).</td>
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<table>
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<tr>
<th><strong>IR</strong> (ν, cm^{-1}) (CCl₄)</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>3080 (w), 2969 (m), 2880 (m), 1761 (s), 1648 (s), 1449 (w), 1409 (w), 1316 (w), 1295 (w), 1179 (m), 1138 (m), 1025 (w), 993.3 (w).</td>
<td></td>
</tr>
</tbody>
</table>

**HRMS** (EI+, m/z) : Calculated: 205.1103 found: 205.1100.
*N*-tert-Butyl-2-cyclopropyl-2-(5-oxo-3-phenyl-4,5-dihydro-isoxazol-4-yl)-acetamide and *N*-tert-Butyl-2-cyclopropyl-2-(5-oxo-3-phenyl-2,5-dihydro-isoxazol-4-yl)-acetamide

Method: See general procedure 1.3 using (1 equiv. 2 mmol, 426 mg) of 4-[1-cyclopropyl-methylidene]-3-phenyl-4H-isoxazol-5-one and (1 equiv, 2 mmol, 166 mg) of tert butyl isocyanide.


Product: Pale pink solid.

Isolated Yield: ~100% (inseparable mixture, ratio maj. dia.: min. dia.: NH form 1: 0.76 : 0.40).

1H NMR (δ, ppm) (CDCl3, 400 MHz)

**Major diastereoisomer:** 7.69-7.67 (m, 2H, CH-Ph), 7.52-7.40 (m, 3H, CH-Ph), 6.10 (br s, 1H, NHCO), 4.85 (d, J = 1.9 Hz, 1H, CHCO2), 2.15 (dd, J = 1.9 Hz, J = 10.6 Hz, 1H, CHCON), 1.39-1.37 (m, 1H, CH2-cyclopropane), 1.39 (s, 9H, CH3-tBu), 0.68-0.63 (m, 2H, CH2-cyclopropane), 0.43-0.39 (m, 1H, CH2-cyclopropane), 0.34-0.30 (m, 1H, CH2-cyclopropane).

**Minor diastereoisomer:** 7.69-7.67 (m, 2H, CH-Ph), 7.52-7.40 (m, 3H, CH-Ph), 6.32 (br s, 1H, NHCO), 4.83 (d, J = 2.4 Hz, 1H, CHCO2), 2.01 (dd, J = 2.4 Hz, J = 10.8 Hz, 1H, CHCON) 1.28 (s, 9H, CH3-tBu), 0.89-0.83 (m, 1H, CH2-cyclopropane), 0.74-0.70 (m, 1H, CH2-cyclopropane), 0.68-0.63 (m, 1H, CH2-cyclopropane), 0.60-0.54 (m, 1H, CH2-cyclopropane), -0.07- -0.16 (m, 1H, CH2-cyclopropane).

**NH form:** 7.61-7.60 (m, 2H, CH-Ph), 7.52-7.40 (m, 3H, CH-Ph), 6.43 (br s, 1H, NHCO), 2.55 (d, J = 10.4 Hz, 1H, CHCON), 1.39-1.37 (m, 1H, CH2-cyclopropane), 1.37 (s, 9H, CH3-tBu), 0.68-0.63 (m, 1H, CH2-cyclopropane), 0.28-0.23 (m, 1H, CH2-cyclopropane), 0.18-0.13 (m, 1H, CH2-cyclopropane), 0.04- -0.06 (m, 1H CH2-cyclopropane).

13C NMR (δ, ppm) (CDCl3, 100 MHz)

**Major diastereoisomer:** 179.05 (Cq, C=O), 168.9 (Cq, C=O), 166.3 (Cq, C=N), 131.5 (CH-Ph), 129.1 (CH-Ph), 128.1 (Cq, Ph), 127.5 (CH-Ph), 52.5 (CHCON), 51.8 (Cq, tBu), 46.5 (CHCO2), 28.5 (CH3-tBu), 9.7 (CH2-cyclopropane), 5.1 (CH2-cyclopropane), 4.0 (CH2-cyclopropane).

**Minor diastereoisomer:** 176.6 (Cq, C=O), 168.7 (Cq, C=O), 166.7 (Cq, C=N), 131.8 (CH-Ph), 129.4 (CH-Ph), 128.4 (Cq, Ph), 127.1 (CH-Ph), 51.8 (Cq, tBu), 51.6 (CHCON), 47.3 (CHCO2), 28.6 (CH3-tBu), 8.9 (CH2-cyclopropane), 4.4 (CH2-cyclopropane), 4.3 (CH2-cyclopropane).

**NH form:** no visible pics.
**IR** (ν, cm$^{-1}$) (CCl$_4$)  
3314 (w), 3047 (w), 3003 (w), 2966 (w), 2927 (w), 2856 (w), 2802 (w), 1716 (s), 1619 (s), 1545 (m), 1453 (m), 1366 (w), 1299 (w), 1255 (w), 1220 (w), 1141 (w), 1039 (w).

**HRMS** (EI+, m/z): Calculated: 314.1631 found: 314.1630.

**N-tert-Butyl-2-(5-oxo-3-phenyl-4,5-dihydro-isoxazol-4-yl)-2-p-tolyl-acetamide** and **N-tert-Butyl-2-(5-oxo-3-phenyl-2,5-dihydro-isoxazol-4-yl)-2-p-tolyl-acetamide**

1.22b

**MF**: C$_{22}$H$_{24}$O$_3$N$_2$

**MW**: 364 g.mol$^{-1}$

**Method**: See general procedure 1.3 using (1 equiv. 2 mmol, 526 mg) of 3-phenyl-4-[1-p-tolyl-methylidene]-4H-isoxazol-5-one and (1 equiv, 2 mmol, 166 mg) of tert butyl isocyanide.

**Purification**: Filtration/ R$_f$ (7:3 PE: AcOEt): 0.76.

**Product**: Yellow solid.

**Isolated Yield**: 58% (inseparable mixture, ratio maj. dia.: min. dia: NHform 0.60 : 0.44 : 1).

**$^1$H NMR** (δ, ppm) (CDCl$_3$, 400 MHz)

**Major diastereoisomer**: 7.54-7.37 (m, 3H, CH-Ph), 7.18-7.13 (m, 2H, CH-Ph), 7.10 (d, J = 8.0 Hz, 2H, CH-tol), 6.99 (d, J = 8.0 Hz, 2H, CH-tol), 5.20 (br s, 1H, NHCO), 4.75 (d, J = 2.4 Hz, 1H, CH$_2$CO$_2$), 4.15 (d, J = 2.4 Hz, 1H, NCOCH), 2.33 (s, 3H, CH$_3$), 1.19 (s, 9H, CH$_3$-tBu).

**Minor diastereoisomer**: 7.54-7.37 (m, 3H, CH-Ph), 7.18-7.13 (m, 2H, CH-Ph), 7.10 (d, J = 8.0 Hz, 2H, CH-tol), 6.79 (d, J = 8.0 Hz, 2H, CH-tol), 5.60 (br s, 1H, NHCO), 5.00 (d, J = 3.6 Hz, 1H, CH$_2$CO$_2$), 4.04 (d, J = 3.6 Hz, 1H, NCOCH), 2.30 (s, 3H, CH$_3$), 1.32 (s, 9H, CH$_3$-tBu),

**NH form**: 7.68 (d, J = 7.8 Hz, 2H, CH-tol), 7.54-7.37 (m, 3H, CH-Ph) 7.47 (d, J = 7.8 Hz, 2H, CH-tol), 7.18-7.13 (m, 2H, CH-Ph), 6.21 (br s, 1H, NHCO), 4.41 (s, 1H, NCOCH), 2.33 (s, 3H, CH$_3$), 1.30 (s, 9H, CH$_3$-tBu).

**$^{13}$C NMR** (δ, ppm) (CDCl$_3$, 100 MHz)

**Major diastereoisomer, minor diastereoisomer and NH form** (three Cq carbons can not be unambiguously assigned): 177.9 (Cq, C=O), 176.3 (Cq, C=O), 172.3 (Cq, C=O), 168.4 (Cq, C=O), 167.8 (Cq, C=O), 166.3 (Cq, C=O), 166.0 (Cq, C=N), 164.2 (Cq, C=N), 139.0 (Cq), 138.9 (Cq), 137.5 (Cq), 135.2 (Cq), 131.6 (CH-Ar), 131.4 (CH-Ar), 131.2 (CH-Ar), 131.1 (Cq), 130.1 (CH-Ar),
IR (ν, cm⁻¹) (CCl₄): 3425 (w), 3275 (w), 3064 (w), 3028(w), 2971 (m), 2925 (w), 1796 (s), 1738 (w), 1676 (s), 1644 (s), 1557 (w), 1515 (s), 1475 (s), 1450 (s), 1394 (s), 1366 (s), 1311 (w), 1268 (w), 1219 (m), 1150 (w), 1118 (w), 1029 (w).

HRMS (EI+, m/z): Calculated: 364.1787 found: 364.1791.

[2-(5-Oxo-3-phenyl-4,5-dihydro-isoxazol-4-yl)-2-p-tolyl-acetylamino]-acetic acid ethyl ester and [2-(5-Oxo-3-phenyl-2,5-dihydro-isoxazol-4-yl)-2-p-tolyl-acetylamino]-acetic acid ethyl ester

\[
\begin{align*}
\text{MF: } C_{22}H_{22}O_5N_2 \quad \text{MW: } 394 \text{ g.mol}^{-1}
\end{align*}
\]

Method: See general procedure 1.3 using (1 equiv., 2 mmol, 526 mg) of 3-phenyl-4-[1-p-tolyl-methylidene]-4H-isoxazol-5-one and (1 equiv., 2 mmol, 226 mg) of methyleneamino-acetic acid methyl ester.

Purification: Flash column chromatography (silica gel, 1:1 PE: AcOEt) RF (7:3 PE: AcOEt): 0.20.

Product: Pale green solid.

Isolated yield: 38% (inseparable mixture, ratio dia. 1 : dia. 2 : NHform 1 : 1 : 2).

¹H NMR (δ, ppm) (CDCl₃, 400 MHz)

Diasteroisomer 1: 7.64 (d, J = 7.2 Hz, 2H, CH-tolyl), 7.54-7.39 (m, 7H, CH-Ph + CH-tolyl), 6.90 (br t, J = 5.2 Hz, 1H, NHCO), 4.66-4.65 (m, 1H, CHCON) 4.18-4.10 (m, 3H, OCH₂CH₃ + CHCO₂N), 3.90 (t, J = 5.2 Hz, 2H, CH₂CO₂Et), 2.33 (s, 3H, CH₃), 1.28-1.19 (m, 3H, OCH₂CH₃).

Diasteroisomer 2: 7.54-7.39 (m, 5H, CH-Ph), 7.05 (d, J = 7.8Hz, 2H, CH-tolyl) 6.83 (d, J = 7.8Hz, 2H, CH-tolyl), 6.29 (br t, J = 4.8Hz, 1H, NHCO), 5.04 (d, J = 3.6Hz, 1H, CHCON), 4.27 (d, J = 2.0 Hz, 1H, CHCO₂N ), 4.18-4.10 (m, 2H, OCH₂CH₃ dia. 2) 4.06 (dd, J = 3.0 Hz, J = 5.0 Hz, 2H, CH₂CO₂Et), 2.30 (s, 3H, CH₃), 1.28-1.19 (m, 3H, OCH₂CH₃).

NH form: 7.54-7.39 (m, 3H, CH-Ph), 7.24 (d, J = 7.6Hz, 2H, CH-tolyl), 7.15-7.11 (m, 5H, CH-Ph + CH-tolyl + NHCO), 4.66-4.65 (m, 1H, CHCON), 4.18-
\[
\begin{align*}
\text{IR (\nu, \text{cm}^{-1}) (CCl_4)} & \quad 3419 (w), 3311 (w), 3060 (w), 2984 (w), 2927 (w), 2868 (w), 1797 (m), 1747 (s), 1679 (s), 1620 (m), 1516 (m), 1477 (m), 1444 (m), 1378 (m), 1205 (s), 1115 (w), 1025 (w). \\
\text{HRMS (El+, m/z)} & \quad \text{Calculated: 394.1529  found: 394.1531.}
\end{align*}
\]
= 2.4 Hz, 1H, CHCO₂), 2.30 (s, 3H, CH₃).

**Diasteroisomer 2:** 7.55-7.00 (m, 12H, CH-Ph + CH-tolyl), 6.79 (d, J = 8.0 Hz, 2H, CH-tolyl), 6.07 (br s, 1H, CONH), 5.02 (d, J = 3.2 Hz, 1H, CHCON), 4.42-4.35 (m, 2H, CH₂NH), 4.13 (d, J = 3.2 Hz, 1H, CHCO₂), 2.27 (s, 3H, CH₃).

**NH form:** 7.55-7.00 (m, 15H, CH-Ph + 2x CH-tolyl + CONH), 4.62 (s, 1H, CHCON), 4.42-4.35 (m, 2H, CH₂NH), 2.32 (s, 3H, CH₃).

**13C NMR (δ, ppm)** Diasteroisomer 1, diasteroisomer 2 and NH form (seven carbons CH-Ar and one Cq carbon can not be unambiguously assigned): 177.7 (Cq, C=O), 176.3 (Cq, C=O), 172.4 (Cq, C=O), 169.5 (Cq, C=O), 168.8 (Cq, C=O), 166.0 (Cq, C=O), 166.0 (Cq, C=O), 164.2 (Cq, C=O), 139.2 (Cq, Ar), 139.0 (Cq, Ar), 137.6 (Cq, Ar), 137.4 (Cq, Ar), 137.3 (Cq, Ar), 137.1 (Cq, Ar), 134.8 (Cq, Ar), 131.7 (CH-Ar), 131.5 (CH-Ar), 130.7 (Cq, Ar), 130.5 (Cq, Ar), 130.4 (Cq, Ar), 130.1 (Cq, Ar), 129.8 (CH-Ar), 129.4 (CH-Ar), 129.2 (CH-Ar), 129.1 (CH-Ar), 129.0 (CH-Ar), 128.9 (CH-Ar), 128.7 (CH-Ar), 128.6 (CH-Ar), 128.5 (CH-Ar), 128.2 (CH-Ar), 128.1 (Cq, Ar), 127.7 (CH-Ar), 127.6 (CH-Ar), 127.5 (CH-Ar), 127.3 (CH-Ar), 127.1 (CH-Ar), 53.1 (CH), 51.9 (CH), 48.0 (CH), 47.9 (CH), 47.6 (CH), 44.2 (CH₂), 44.0 (CH₂), 43.9 (CH₂), 21.0 (CH₃), 21.0 (CH₃x2).

**IR (ν, cm⁻¹) (CCl₄)** 3434 (w), 3276 (w), 3064 (w), 3032 (w), 2925 (w), 1797 (s), 1676 (s), 1644 (s), 1566 (m), 1515 (s), 1475 (s), 1449 (s), 1396 (s), 1363 (s), 1310 (w), 1254 (w), 1181 (w), 1074 (w), 1026 (w).

**HRMS (EI+, m/z):** Calculated: 398.1631 found: 398.1621.

**N-tert-Butyl-2-(5-oxo-3-phenyl-4,5-dihydro-isoxazol-4-yl)-2-thiophen-2-yl-acetamide and N-tert-Butyl-2-(5-oxo-3-phenyl-2,5-dihydro-isoxazol-4-yl)-2-thiophen-2-yl-acetamide**

**MF:** C₁₉H₂₀O₃N₂S

**MW** = 356g.mol⁻¹

**Method:** See general procedure 1.3 using (1 equiv., 2 mmol, 510 mg) of 3-phenyl-4-[1-thiophen-2-yl-methylidene]-4H-isoxazol-5-one and (1 equiv, 2 mmol, 166 mg) of tert butyl isocyanide.

**Purification:** Flash column chromatography (silica gel, 7:3 PE: AcOEt)/ Rᵢ (1:1 PE: EtOAc): 0.42.

**Product:** Brown solid.

**Isolated Yield:** 24% (inseparable mixture, ratio dia. 1: dia. 2: NH form 0.42 : 0.42 : 1).
**1H NMR (δ, ppm)**

(CDCl₃, 400 MHz)

**Diasteroisomer 1:** 7.72-7.70 (m, 2H, CH-Ph), 7.53-7.38 (m, 3H, CH-Ph), 7.28 (dd, J = 0.8 Hz, J = 5.2 Hz, 1H, CH-tiophene), 6.96-6.93 (m, 1H, CH-tiophene), 6.84 (d, J = 2.8 Hz, 1H, CH-tiophene), 5.50 (br s, 1H, NHCO), 4.88 (d, J = 2.4 Hz, 1H, CHCO₂), 4.49 (d, J = 2.4 Hz, 1H, CHCON), 1.20 (s, 9H, CH₃-tBu).

**Diasteroisomer 2:** 7.53-7.38 (m, 5H, CH-Ph), 7.24-7.21 (m, 1H, CH-tiophene), 6.96-6.93 (m, 1H, CH-tiophene), 6.78 (d, J = 3.6 Hz, 1H, CH-tiophene), 5.95 (br s, 1H, NHCO), 5.03 (d, J = 3.6 Hz, 1H, CHCO₂), 4.33 (d, J = 3.6 Hz, 1H, CHCON), 1.34 (s, 9H, CH₃-tBu).

**NH form:** 7.53-7.38 (m, 5H, CH-Ph), 7.24-7.21 (m, 1H, CH-tiophene), 6.96-6.93 (m, 2H, CH-tiophene), 6.58 (br s, 1H, NHCO), 4.69 (s, 1H, CHCON), 1.32 (s, 9H, CH₃-tBu).

**13C NMR (δ, ppm)**

(CDCl₃, 100 MHz)

Diasteroisomer 1, diasteroisomer 2 and NH form (one Cq carbon can not be unambigously assigned): 177.4 (Cq, C=O), 176.1 (Cq, C=O), 172.2 (Cq, C=O), 167.2 (Cq, C=O), 166.6 (Cq, C=O), 166.0 (Cq, C=O), 165.7 (Cq, C-N), 163.7 (Cq, C-N), 141.2 (Cq), 134.9 (Cq), 134.2 (Cq), 131.8 (CH-Ar), 131.6 (CH-Ar), 130.5 (CH-Ar), 129.3 (CH-Ar), 129.2 (Cq), 129.1 (CH-Ar), 129.1 (CH-Ar), 129.0 (CH-Ar), 128.2 (CH-Ar), 128.0 (CH-Ar), 127.9 (CH-Ar), 127.9 (Cq), 127.8 (CH-Ar), 127.7 (CH-Ar), 127.7 (Cq), 127.4 (CH-Ar), 127.3 (CH-Ar), 127.1 (CH-Ar), 127.0 (Cq), 126.9 (CH-Ar), 125.8 (CH-Ar), 125.3 (CH-Ar), 52.4 (Cq, 'Bu), 52.2 (Cq, 'Bu), 52.0 (Cq, 'Bu), 48.8 (CH), 48.5 (CH), 48.2 (CH), 47.7 (CH), 44.2 (CH), 29.6 (CH₃-tBu), 28.4 (CH₃-tBu), 28.3 (CH₃-tBu).

**IR (v, cm⁻¹) (CCl₄)**

3417 (w), 3282.9 (w), 3069 (w), 2969 (m), 2928 (m), 2870 (w), 1797 (m), 1680 (s), 1647 (s), 1563 (m), 1517 (s), 1475 (s), 1452 (s), 1394 (s), 1366 (s), 1280 (s), 1222 (s), 1072 (w), 1043 (w), 1027 (w).

**HRMS (EI+, m/z)**: Calculated: 356.1195 found: 356.1183.

**N-tert-Butyl-2-(5-oxo-3-phenyl-4,5-dihydro-isoxazol-4-yl)-isobutyramide** 1.22f

![N-tert-Butyl-2-(5-oxo-3-phenyl-4,5-dihydro-isoxazol-4-yl)-isobutyramide](image)

MF: C₁₇H₂₂O₃N₂

MW = 302 g·mol⁻¹

**Method**: See general procedure 1.3 using (1 equiv., 2 mmol, 402 mg) of 4-isopropylidene-3-phenyl-4H-isoxazol-5-one and (1 equiv., 2 mmol, 166 mg) of tert butyl isocyanide.

**Purification**: filtration/ Rᵣ (7:3 PE: AcOEt): 0.31.

**Product**: White solid.
Isolated Yield : 58% (CH form only).

$^1$H NMR (δ, ppm) 
(CDCls, 400 MHz)  
7.59-7.58 (m, 2H, CH-Ph), 7.50-7.41 (m, 3H, CH-Ph), 5.25 (br s, 1H, NH), 4.55 (s, 1H, CHCO₂), 1.59 (s, 3H, CH₃), 1.17 (s, 9H, CH₃-tBu), 1.12 (s, 3H, CH₃).

$^{13}$C NMR (δ, ppm)  
(CDCls, 100 MHz)  
176.49 (Cq, C=O), 172.4 (Cq, C=O), 167.3 (Cq, C=N), 131.2 (CH-Ph), 128.9 (CH-Ph), 128.7 (Cq, Ph), 128.1 (CH-Ph), 51.4 (Cq), 50.9 (CHCO₂), 45.9 (Cq), 28.4 (CH₃-tBu), 23.7 (CH₃), 22.6 (CH₃).

IR (ν, cm⁻¹) (CCl₄)  
3313 (w), 2973 (s), 2863 (s), 1811 (w), 1706 (s), 1626 (m), 1535 (m), 1451 (m), 1384 (w), 1363 (m), 1341 (m), 1303 (w), 1258 (w), 1213 (w), 1156 (w), 1069 (s), 1001 (w).

HRMS (EI+, m/z) :  
Calculated: 302.1631       found: 302.1633.

**1-(5-Oxo-3-phenyl-4,5-dihydro-isoxazol-4-yl)-cyclopentanecarboxylic acid tert-butylamide and 1-(5-Oxo-3-phenyl-2,5-dihydro-isoxazol-4-yl)-cyclopentanecarboxylic acid tert-butylamide**  

| MF: C₁₉H₂₄O₃N₂ | MW = 328 g.mol⁻¹ |

**Method :**  
See general procedure 1.3 using (1 equiv., 2 mmol, 454 mg) of 4-cyclopentylidene-3-phenyl-4H-isoxazol-5-one and (1 equiv, 2 mmol, 166 mg) of tertbutyl isocyanide.

**Purification :**  
Flash column chromatography (silica gel, 8:2 PE: AcOEt) / Rf(7:3 PE: AcOEt): 0.30.

**Product :**  
White solid.

**Isolated Yield :**  
51% (inseparable mixture, ratio CHform: NHform 1:1).

$^1$H NMR (δ, ppm)  
(CDCls, 400 MHz)  
**CH form and NH form:**  
7.52-7.41 (m, 10H, CH-Ph), 6.76 (br s, 1H, NH), 5.29 (br s, 1H, NH), 4.16 (s, 1H, CHCO₂ CHform), 2.38-2.35 (m, 2H, CH₂-cyclopentane), 2.00-1.96 (m, 1H, CH₂-cyclopentane), 1.89-1.40 (m, 13H, CH₂-cyclopentane), 1.24 (s, 9H, CH₃-tBu), 1.20 (s, 9H, CH₃-tBu).
CH form and NH form: 176.8 (Cq, C=O), 174.0 (Cq, C=O), 173.6 (Cq, C=O), 171.5 (Cq, C=O), 167.4 (Cq, C=N), 163.2 (Cq, C-N), 131.1 (CH-Ph), 131.0 (CH-Ph), 128.9 (CH-Ph), 128.9 (Cq, Ph), 128.7 (CH-Ph), 128.6 (CH-Ph), 128.0 (CH-Ph), 127.9 (Cq, Ph), 104.1 (Cq, C=CO₂N NH form), 57.0 (Cq, CH₂CO₂CH₃), 52.1 (Cq), 51.5 (Cq), 50.7 (Cq), 35.5 (CH₂-cyclopentane x2), 34.6 (CH₂-cyclopentane), 33.7 (CH₂-cyclopentane), 28.4 (CH₃, t-Bu), 28.3 (CH₃-t-Bu), 23.7 (CH₂-cyclopentane), 23.6 (CH₂-cyclopentane), 23.3 (CH₂-cyclopentane x2).

IR (v, cm⁻¹) (CCl₄) 3434 (w), 3339 (w), 3061 (m), 2962 (s), 2869 (m), 2788 (w), 1797 (m), 1710 (s), 1675 (s), 1643 (s), 1606 (s), 1535 (s), 1451 (s), 1362 (s), 1292 (m), 1268 (m), 1216 (m), 1176 (m), 1109 (w), 1020 (m).

HRMS (EI+, m/z) : Calculated: 328.1787  found: 328.1777.

[4-(tert-Butylcarbamoyl-cyclopropyl-methyl)-5-oxo-4,5-dihydro-isoxazol-3-yl]-acetic acid methyl ester and [4-(tert-Butylcarbamoyl-cyclopropyl-methyl)-5-oxo-2,5-dihydro-isoxazol-3-yl]-acetic acid methyl ester

Method : See general procedure 1.3 using (1 equiv., 2 mmol, 418 mg) of {4-[1-cyclopropyl-methylidene]-5-oxo-4,5-dihydro-isoxazol-3-yl}-acetic acid methyl ester and (1 equiv., 2 mmol, 166 mg) of tert butyl isocyanide.

Purification : None/ Rf (1:1 PE:AcOEt): 0.43.

Product : Orange solid.

Isolated Yield : ~100% (inseparable mixture, ratio maj.: dia.:min. dia.:NH form  2:1:4).

¹H NMR (δppm) (CDCl₃, 400 MHz)

Major diastereoisomer: 6.14 (br s, 1H, NH), 4.15 (d, J = 2.4 Hz, 1H, CHCO₂N), 3.73 (s, 3H, OCH₃), 3.61 (s, 2H, CH₂CO₂Me), 2.08-2.04 (m, 1H, CHCON), 1.32 (s, 9H, CH₃-Bu), 1.20-1.44 (m, 1H, CH-cyclopropane), 0.80-0.72 (m, 1H, CH₂-cyclopropane), 0.71-0.63 (m, 1H, CH₂-cyclopropane), 0.44-0.32 (m, 2H, CH₂-cyclopropane).

Minor diastereoisomer: 5.87 (br s, 1H, NH), 3.97 (d, J = 1.6 Hz, 1H, CHCO₂N) 3.70 (s, 3H, OCH₃) 3.57 (s, 2H, CH₂CO₂Me), 2.08-2.04 (m, 1H, CHCON), 1.30 (s, 9H, °Bu), 1.20-1.44 (m, 1H, CH-cyclopropane), 0.80-0.72 (m, 1H, CH₂-cyclopropane), 0.71-0.63 (m, 1H, CH₂-cyclopropane), 0.44-0.32 (m, 2H,
CH₂-cyclopropane).

**NH form:** 6.31 (br s, 1H, NH), 3.73-3.65 (m, 5H, OCH₃ + CH₂CO₂Me) 2.55 (d, J = 9.2 Hz, CHCON), 1.36-1.30 (m, 1H, CH₂-cyclopropane), 1.33 (s, 9H, tBu), 0.60-0.54 (m, 1H, CH₂-cyclopropane), 0.53-0.46 (m, 1H, CH₂-cyclopropane), 0.30-0.22 (m, 1H, CH₂-cyclopropane), 0.21-0.12 (m, 1H, CH₂-cyclopropane).

**¹³C NMR (δ, ppm)**

<table>
<thead>
<tr>
<th>Major diastereoisomer</th>
<th>Minor diastereoisomer</th>
<th>NH form</th>
</tr>
</thead>
<tbody>
<tr>
<td>177.7 (Cq, C=O)</td>
<td>177.3 (Cq, C=O)</td>
<td>177.0 (Cq, C=O)</td>
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<tr>
<td>173.7 (Cq, C=O)</td>
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<td>172.3 (Cq, C=O)</td>
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<td>169.3 (Cq, C=O)</td>
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<td>169.2 (Cq, C=O)</td>
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<td>168.1 (Cq, C=O)</td>
</tr>
<tr>
<td>168.0 (Cq, C=O)</td>
<td>168.0 (Cq, C=O)</td>
<td>168.0 (Cq, C=O)</td>
</tr>
<tr>
<td>163.3 (Cq, C=N)</td>
<td>157.6 (Cq, C=N)</td>
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</tr>
<tr>
<td>157.6 (Cq, C=N)</td>
<td>157.1 (Cq, C=N)</td>
<td>157.1 (Cq, C=N)</td>
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<td>92.3 (Cq, C=O₂, NH form)</td>
<td>92.3 (Cq, C=O₂, NH form)</td>
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<td>52.6 (OCH₃ x₂)</td>
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<tr>
<td>3.8 (CH₂-cyclopropane)</td>
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</tr>
</tbody>
</table>

**IR (ν, cm⁻¹) (CCl₄)**

3433 (w), 3081 (w), 2967 (m), 1793 (s), 1741 (s), 1678 (s), 1647 (s), 1514 (s), 1457 (m), 1400 (m), 1366 (m), 1331 (m), 1212 (s), 1168 (m), 1016 (w).

**HRMS (EI⁺, m/z)**

Calculated: 310.1529  found: 310.1524.

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**[4-(1-tert-Butylcarbamoyl-cyclopentyl)-5-oxo-4,5-dihydro-isoxazol-3-yl]-acetic acid methyl ester**

**MF:** C₁₆H₂₄O₅N₂  
**MW:** 324 g.mol⁻¹

**Method:**

See **general procedure 1.3** using (1 equiv., 2 mmol, 446 mg) of (4-cyclopentylidene-5-oxo-4,5-dihydro-isoxazol-3-yl)-acetic acid methyl ester and (1 equiv., 2 mmol, 166 mg) of tert butyl isocyanide.

**Purification:**

Flash column chromatography (silica gel, 7:3 PE:AcOEt)/ Rₚ (1:1 PE: AcOEt): 0.50.

**Product:**

Brown oil.

**Isolated yield:**

56% (CHform only).
1H NMR (δ, ppm) 5.49 (br, 1H, NH), 3.81-3.62 (m, 6H, CH₂CO₂ + OCH₃ + CHCO₂N), 2.34-2.28 (m, 1H, CH₂-cyclopentane), 2.21-1.42 (m, 1H, CH₂-cyclopentane), 1.88-1.77 (m, 2H, CH₂-cyclopentane), 1.75-1.59 (m, 3H, CH₃-cyclopentane), 1.28 (s, 9H, CH₃-Bu)

13C NMR (δ, ppm) 176.9 (Cq, C=O), 171.6 (Cq, C=O), 168.8 (Cq, C=O), 163.4 (Cq, C=N), 52.3 (Cq), 52.7 (CHCO₂N), 52.1 (OCH₃), 51.8 (Cq), 34.9 (CH₂-cyclopentane), 33.6 (CH₂CO₂), 28.5 (CH₃-Bu), 24.0 (CH₂-cyclopentane), 23.9 (CH₂-cyclopentane)

IR (ν, cm⁻¹) (CCl₄) 3438 (w), 2962 (s), 2877 (m), 1795 (s), 1743 (s), 1673 (s), 1513 (s), 1453 (s), 1406 (m), 1365 (s), 1324 (m), 1208 (s), 1176 (s), 1006 (w)

HRMS (EI+, m/z) : Calculated: 324.1685 found: 324.1681.

2-(3-Benzoyloxymethyl-5-oxo-2,5-dihydro-isoxazol-4-yl)-N-tert-butyl-2-p-tolyl-acetamide 1.22j

MF: C₂₄H₂₈O₄N₂

MW = 408 g.mol⁻¹

Method : See general procedure 1.3 using (1 equiv., 1.5 mmol, 464 mg) of 3-benzoyloxymethyl-4-[1-p-tolyl-methylidene]-4H-isoxazol-5-one and (1 equiv., 1.5 mmol, 126 mg) of tert butyl isocyanide.

Purification : Flash column chromatography (silica gel, 7:3 PE: AcOEt) / Rf (7:3 PE: AcOEt): 0.55.

Product : Brown solid.

Yield : 72% (ratio CH forms: NHform traces : 1).

1H NMR (δ, ppm) NH form: 7.35-7.30 (m, 3H, CH-Ph), 7.21-7.19 (m, 2H, CH-Ph), 7.12 (d, J = 8.2 Hz, 2H, CH-tolyl), 7.08 (d, J = 8.2Hz, 2H, CH-tolyl), 6.02 (br s, 1H, NH), 4.58 (s, 1H, CHCON), 4.43 (d, J = 13.1Hz, 1H, CH₂), 4.42 (d, J = 11.5Hz, 1H, CH₂), 4.34 (d, J = 11.5Hz, 1H, CH₂), 4.30 (d, J = 13.1Hz, 1H, CH₂), 2.31 (s, 3H, CH₃), 1.29 (s, 9H, CH₃-Bu).

13C NMR (δ, ppm) NH form: 174.0 (Cq, C=O), 171.8 (Cq, C=O), 161.3 (Cq, =CNH), 137.4 (Cq, Ar), 137.2 (Cq, Ar), 135.6 (Cq, Ar), 129.6 (CH-Ar), 128.4 (CH-Ar), 128.0 (CH-Ar), 128.0 (CH-Ar), 126.9 (CH-Ar), 95.2 (Cq, =CO₂N), 72.2 (OCH₃), 63.5
2-(3-But-3-ethyl-5-oxo-4,5-dihydro-isoxazol-4-yl)-N-tert-butyl-2-cyclopropyl-acetamide

**MF:** C_{16}H_{24}O_{3}N_{2}

**MW:** 292 g.mol^{-1}

**Method:** See general procedure 1.3 using (1 equiv., 2 mmol, 382 mg) of 3-but-3-ethyl-4-[(1-cyclopropyl-methylidene)-4H-isoxazol-5-one and (1 equiv., 2 mmol, 166 mg) of tert butyl isocyanide.

**Purification:** Flash column chromatography (silica gel, 8:2 PE: AcOEt)/ R\textsubscript{f} (8:2 PE: AcOEt): 0.23.

**Product:** White solid.

**Isolated yield:** 89% (ratio maj.: min. dia.: NHform 2: 1: traces).

**\textsuperscript{1}H NMR (δ, ppm)**

<table>
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<th>Compound</th>
<th>Value</th>
</tr>
</thead>
<tbody>
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<td>Major diasteroisomer</td>
<td>6.29 (br s, 1H, CONH), 5.89-5.76 (m, 1H, CH=CH\textsubscript{2}), 5.13-5.03 (m, 2H, CH=CH\textsubscript{2}), 4.19 (d, J = 3.2 Hz, 1H, CHCO\textsubscript{2}), 2.72-2.34 (m, 4H, CH\textsubscript{2}-homo allyl chain)</td>
</tr>
<tr>
<td>Minor diasteroisomer</td>
<td>6.04 (br s, 1H, CONH), 5.89-5.76 (m, 1H, CH=CH\textsubscript{2}), 5.13-5.03 (m, 2H, CH=CH\textsubscript{2}), 3.79 (d, J = 3.2 Hz, 1H, CHCO\textsubscript{2}), 2.72-2.34 (m, 4H, CH\textsubscript{2}-homo allyl chain)</td>
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**\textsuperscript{13}C NMR (δ, ppm)**

<table>
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<tr>
<th>Compound</th>
<th>Value</th>
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<tbody>
<tr>
<td>Major diasteroisomer</td>
<td>178.5 (Cq, C=O), 168.9 (Cq, C=O), 168.1 (Cq, C=N), 136.0 (CH=CH\textsubscript{2}), 116.3 (CH=CH\textsubscript{2}), 51.6 (Cq, 'Bu)</td>
</tr>
<tr>
<td>Minor diasteroisomer</td>
<td>121.7 (CH\textsubscript{2}-cyclo propane), 50.9 (CHCON), 48.1 (CHCO\textsubscript{2}), 29.2 (CH\textsubscript{2}-homo allyl chain), 28.6 (CH\textsubscript{2}-'Bu)</td>
</tr>
</tbody>
</table>
propane).

**Minor diastereoisomer:** 177.4 (Cq, C=O), 168.7 (Cq, C=O), 168.3 (Cq, C=N), 135.9 (CH=CH₂), 116.5 (CH=CH₃), 51.8 (CHCON), 51.7 (Cq, tBu), 49.1 (CH₂CON), 51.7 (Cq, tBu), 49.1 (CH₂CON), 51.7 (Cq, tBu), 49.1 (CH₂CON).

**IR (ν, cm⁻¹) (CCl₄)**

3434 (w), 3296 (w), 3074 (w), 2969 (w), 2922 (w), 1791 (m), 1681 (s), 1643 (s), 1605 (s), 1515 (s), 1453 (m), 1392 (m), 1364 (m), 1298 (m), 1220 (m), 1104 (m), 1028 (m), 995 (m), 965 (m).

**HRMS (EI+, m/z):**

Calculated: 292.1787

found: 292.1780.

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**N-Benzyl-2-(3-but-3-enyl-5-oxo-4,5-dihydro-isoxazol-4-yl)-2-cyclopropylacetamide and N-Benzyl-2-(3-but-3-enyl-5-oxo-2,5-dihydro-isoxazol-4-yl)-2-cyclopropylacetamide**

**Method:** See general procedure 1.3 using (1 equiv., 2 mmol, 382 mg) of 3-but-3-enyl-4-[1-cyclopropyl-methyldiene]-4H-isoxazol-5-one and (1 equiv., 2 mmol, 234 mg) of benzyl isocyanide.

**Purification:** Flash column chromatography 7:3 PE: AcOEt/ Rf (7:3 PE: AcOEt): 0.24.

**Product:** Brown oil.

**Isolated yield:** 30% (ratio maj. dia.: min. dia. NHform 2:1:2).

**¹H NMR (δ, ppm)**

**Major diastereoisomer:** 7.36-7.23 (m, 5H, CH-Ph), 6.74 (br s, 1H, CONH), 5.86-5.72 (m, 1H, CH=CH₂), 5.13-5.00 (m, 2H, CH=CH₂), 4.58-4.41 (m, 2H, CH₂NH), 4.12 (d, J = 2.4Hz, 1H, CHCO₂), 2.71-2.12 (m, 4H, CH₂-homo allyl chain), 2.20 (dd, J = 2.8Hz, J = 10.8Hz, 1H, CHCON), 1.01-0.93 (m, 1H, CH-cyclo propane), 0.83-0.61 (m, 2H, CH₂-cyclo propane), 0.48-0.36 (m, 2H, CH₂-cyclo propane).

**Minor diastereoisomer:** 7.36-7.23 (m, 5H, CH-Ph), 6.67 (br s, 1H, CONH), 5.86-5.72 (m, 1H, CH=CH₂), 5.13-5.00 (m, 2H, CH=CH₂), 4.58-4.41 (m, 2H, CH₂NH) 3.86 (d, J = 2.8Hz, 1H, CHCO₂), 2.71-2.12 (m, 4H, CH₂-homo allyl chain), 2.13 (dd, J = 2.8Hz, J = 11.2 Hz, 1H, CHCON), 1.14-1.06 (m, 1H, CH-cyclo propane), 0.83-0.61 (m, 2H, CH₂-cyclo propane), 0.48-0.36 (m, 1H, CH₂
cyclo propane), 0.33-0.27 (m, 1H, CH₂-cyclo propane).

**NHform**: 7.36-7.23 (m, 6H, CH-Ph + CONH), 5.86-5.72 (m, 1H, CH=CH₂), 5.13-5.00 (m, 2H, CH=CH₂), 4.58-4.41 (m, 2H, CH₂-CH₂NH), 2.71-2.12 (m, 5H, CH₂- homo allyl chain + CHCON), 1.55-1.47 (m, 1H, CH-cyclo propane), 0.58-0.51 (m, 1H, CH₂-cyclo propane), 0.48-0.36 (m, 2H, CH₂-cyclo propane), 0.22-0.15 (m, 1H, CH₂-cyclo propane).

**13C NMR** (δ, ppm) Major diastereoisomer, minor diastereoisomer and NHform: 178.3 (Cq, C=O), 177.2 (Cq, C=O), 173.4 (Cq, C=O), 172.7 (Cq, C=O), 170.0 (Cq, C=O), 169.7 (Cq, C=O), 168.1 (Cq, C=O), 167.9 (Cq, C=O), 166.4 (Cq, C=O), 137.8 (Cq, Ar), 137.6 (Cq, Ar), 137.6 (Cq, Ar), 136.7 (CH=CH₂), 136.0 (CH=CH₂), 135.8 (CH=CH₂), 128.9 (CH=CH₂), 128.8 (CH=CH₂), 128.6 (CH=CH₂), 127.7 (CH=CH₂), 127.6 (CH=CH₂), 127.5 (CH=CH₂), 127.4 (CH=CH₂), 116.6 (CH=CH₂), 116.5 (CH=CH₂), 116.4 (CH=CH₂), 99.5 (Cq), 50.8 (CH), 50.7 (CH), 50.3 (CH), 49.0 (CH), 48.3 (CH), 44.0 (CH₂N), 43.9 (CH₂N), 43.7 (CH₂N), 31.5 (CH₂-homo allyl chain), 29.3 (CH₂-homo allyl chain), 29.1 (CH₂-homo allyl chain), 28.3 (CH₂-homo allyl chain), 28.2 (CH₂-homo allyl chain), 24.9 (CH₂-homo allyl chain), 13.1 (CH₂-cyclopropane), 10.6 (CH₂-cyclopropane), 9.7 (CH₂-cyclopropane), 5.9 (CH₂-cyclopropane), 5.2 (CH₂-cyclopropane), 5.1 (CH₂-cyclopropane), 4.7 (CH₂-cyclopropane), 4.6 (CH₂-cyclopropane), 4.3 (CH₂-cyclopropane).

**IR** (ν, cm⁻¹) (CCl₄) 3443 (w), 3079 (w), 3032 (w), 3007 (w), 2926 (w), 1794 (m), 1712 (s), 1680 (m), 1648 (m), 1550 (w), 1516 (m), 1430 (w), 1356 (w), 1307 (m), 1288 (w), 1211 (m), 1175 (w), 1099 (w), 1024 (w).

**HRMS (El+,** m/z): Calculated: 326.1631 found: 326.1622.

**1-(3-But-3-enyl-5-oxo-4,5-dihydro-isoxazol-4-yl)-cyclopentanecarboxylic acid tert-butylamide and 1-(3-But-3-enyl-5-oxo-2,5-dihydro-isoxazol-4-yl)-cyclopentanecarboxylic acid tert-butylamide**

[Chemical structure images]

**MF**: C₁₇H₂₆O₃N₂

**MW** = 306 g mol⁻¹

**Method**: See general procedure 1.3 using (1 equiv., 2 mmol, 410 mg) of 3-but-3-enyl-4-cyclo pentylidene-4H-isoxazol-5-one and (1 equiv., 2 mmol, 166 mg) of tert butyl isocyanide.

**Purification**: Flash column chromatography (silica gel, 8:2 PE: AcOEt)/ Rf (8:2 PE: AcOEt): 0.32.

**Product**: Orange solid.
Isolated yield: 60% (ratio CHform: NHform 1:0.6).

\( ^{1}H \text{ NMR} \) (\( \delta, \text{ ppm} \)) (CDCl\(_{3}, 400 \text{ MHz} \))

CHform: 8.35-5.71 (m, 1H, CH=CH\(_{2}\)), 5.47 (br s, 1H, CONH), 5.06-4.98 (m, 2H, CH=CH\(_{2}\)), 3.60 (s, 1H, CHCO\(_{2}\)), 2.61-2.53 (m, 2H, CH\(_{2}\)-homo allyl chain), 2.47-2.33 (m, 2H, CH\(_{2}\)-homo allyl chain), 2.22-2.01 (m, 2H, CH\(_{2}\)-cyclopentane), 1.83-1.58 (m, 6H, CH\(_{2}\)-cyclopentane), 1.29 (s, 9H, CH\(_{3}\)-tBu).

NH form: 6.50 (br s, 1H, CONH), 5.83-5.71 (m, 1H, CH=CH\(_{2}\)), 5.06-4.98 (m, 2H, CH=CH\(_{2}\)), 2.70 (t, \( J = 7.7 \text{Hz} \), 2H, CH\(_{2}\)-homo allyl chain), 2.47-2.33 (m, 2H, CH\(_{2}\)-homo allyl chain), 2.22-2.01 (m, 2H, CH\(_{2}\)-cyclopentane), 1.83-1.58 (m, 6H, CH\(_{2}\)-cyclopentane), 1.24 (s, 9H, CH\(_{3}\)-tBu).

\( ^{13}C \text{ NMR} \) (\( \delta, \text{ ppm} \)) (CDCl\(_{3}, 100 \text{ MHz} \))

CHform: 177.2 (Cq, C=O), 172.1 (Cq, C=O), 168.1 (Cq, C=N), 136.1 (CH=CH\(_{2}\)), 116.1 (CH=CH\(_{2}\)), 55.8 (Cq), 52.2 (Cq), 52.0 (CHCO\(_{2}\)N), 35.3 (CH\(_{2}\)), 35.1 (CH\(_{2}\)), 29.2 (CH\(_{2}\)), 28.4 (CH\(_{2}\)), 28.3 (CH\(_{2}\)), 24.6 (CH\(_{2}\)), 24.3 (CH\(_{3}\)-tBu).

NHform: 174.2 (Cq, C=O), 173.2 (Cq, C=O), 163.8 (Cq, C=N), 135.8 (CH=CH\(_{2}\)), 116.6 (CH=CH\(_{2}\)), 101.4 (Cq, C=CO\(_{2}\)N), 51.6 (Cq), 50.9 (Cq), 33.2 (CH\(_{2}\)), 31.4 (CH\(_{2}\)), 28.8 (CH\(_{2}\)), 28.5 (CH\(_{2}\)), 28.4 (CH\(_{2}\)), 25.8 (CH\(_{2}\)), 23.1 (CH\(_{3}\)-tBu).

IR (\( \nu, \text{ cm}^{-1} \)) (CCl\(_{4}\))

3438 (w), 3080 (w), 2965 (m), 2876 (w), 1787 (s), 1733 (w), 1675 (s), 1608 (w), 1512 (s), 1453 (m), 1392 (w), 1366 (w), 1327 (w), 1217 (w), 1161 (w), 1094 (w).

HRMS (EI+, m/z):


\{[1-(3-But-3-enyl-5-oxo-4,5-dihydro-isoxazol-4-yl)-cyclopentanecarbonyl]-amino}-acetic acid ethyl ester and \{[1-(3-But-3-enyl-5-oxo-2,5-dihydro-isoxazol-4-yl)-cyclopentanecarbonyl]-amino}-acetic acid ethyl ester

\[
\text{MF: C}_{17}\text{H}_{24}\text{O}_{5}\text{N}_{2}
\]

MW = 336 g.mol\(^{-1}\)

Method: See general procedure 1.3 using (1 equiv., 2 mmol, 410 mg) of 3-but-3-enyl-4-cyclopentylidene-4\( H \)-isoxazol-5-one and (1 equiv., 2 mmol, 226 mg) of methyleneamino-acetic acid methyl ester.

Purification: Flash column chromatography (silica gel, 7:3 PE: AcOEt)/ \( R_{f} \) (7:3 PE: AcOEt):
Product: Pale brown solid.

Isolated yield: 49% (ratio CH form: NH form 1: 0.88).

\[ \text{H NMR (δ, ppm)} \]

CH form: 6.35 (br s, 1H, CONH), 5.90-5.78 (m, 1H, H\(_2\)C=CH), 5.14-5.04 (m, 2H, H\(_2\)C=CH), 4.24 (q, J = 7.2 Hz, 2H, OCH\(_3\)), 4.03 (dd, J = 5.2 Hz, J = 12.4 Hz, 2H, NH), 3.70 (s, 1H, CHCO\(_2\)), 2.89-2.18 (m, 7H, CH\(_2\))-cyclopentane + homoallyl chain), 1.98-1.69 (m, 5H, CH\(_2\)-cyclopentane), 1.32 (t, J = 7.2 Hz, 3H, OCH\(_3\)).

NH form: 7.10 (t, J = 5.5 Hz, 1H, CONH), 5.90-5.78 (m, 1H, H\(_2\)C=CH), 5.14-5.04 (m, 2H, H\(_2\)C=CH), 4.18 (q, J = 7.2 Hz, 2H, OCH\(_3\)), 3.97 (d, J = 5.5 Hz, 2H, NH), 2.89-2.18 (m, 7H, CH\(_2\)) cyclopentane + homoallyl chain), 1.98-1.69 (m, 5H, CH\(_2\)-cyclopentane), 1.28 (t, J = 7.2 Hz, 3H, OCH\(_3\)).

\[ \text{C NMR (δ, ppm)} \]

CH form and NH form: 176.8 (Cq, C=O), 175.5 (Cq, C=O), 174.1 (Cq, C=O), 173.0 (Cq, C=O), 169.8 (Cq, C=O), 169.6 (Cq, C=O), 167.8 (Cq, C-N), 164.6 (Cq, C-N), 136.2 (CH=CH\(_2\)), 136.0 (CH=CH\(_2\)), 116.7 (CH=CH\(_2\)), 116.1 (CH=CH\(_2\)), 102.5 (Cq, NH), 61.7 (OCH\(_3\)), 61.2 (OCH\(_3\)), 55.4 (Cq), 55.2 (Cq), 51.6 (CHCO\(_2\), CH form), 41.6 (CH\(_2\)), 37.8 (CH\(_2\)), 35.3 (CH\(_2\)), 35.3 (CH\(_2\)), 34.6 (CH\(_2\)), 33.2 (CH\(_2\)), 31.2 (CH\(_2\)), 29.1 (CH\(_2\)), 28.8 (CH\(_2\)), 25.9 (CH\(_2\)), 24.9 (CH\(_2\)), 24.8 (CH\(_2\)), 23.7 (CH\(_2\)), 23.2 (CH\(_2\)), 14.1 (OCH\(_3\)), 14.1 (OCH\(_3\)).

IR (ν, cm\(^{-1}\)) (CCl\(_4\)) 3253 (w), 3081 (w), 2943 (m), 2864 (w), 2750 (w), 1748 (s), 1703 (s), 1630 (s), 1550 (m), 1516 (m), 1447 (w), 1398 (w), 1374 (w), 1349 (w), 1285 (w), 1256 (w), 1204 (s), 1114 (w), 1019 (m).

HRMS (EI+, m/z): Calculated: 336.1685 found: 336.1696.
B.3.1.2 Synthesis of β-branched alkynes and dihydropyrrolones

Alkynes and dihydro pyrrolones were synthesized as described below:

![Chemical structure diagram]

**Procedure 1.4**

FeSO₄•7H₂O, NaNO₂
AcOH, H₂O, rt

**Procedure 1.5**

NIS, NaH
acetone, rt

**Alkynes synthesized:**

1.23a, R¹ = Ph, R² = \( \text{Ph} \), R³ = H, R⁴ = \( \text{Ph} \)
1.23b, R¹ = Ph, R² = 4-Me-Ph, R³ = H, R⁴ = \( \text{Ph} \)
1.23c, R¹ = Ph, R² = 4-Me-Ph, R³ = H, R⁴ = \( \text{CH}_3\text{CO}_2\text{Et} \)
1.23d, R¹ = Ph, R² = 4-Me-Ph, R³ = H, R⁴ = \( \text{CH}_3\text{Ph} \)
1.23e, R¹ = Ph, R² = 2-tiophenyl, R³ = H, R⁴ = \( \text{Ph} \)
1.23f, R¹ = Ph, R² = Me, R³ = Me, R⁴ = \( \text{Ph} \)
1.23g, R¹ = Ph, R² = \( \text{Ph} \), R³ = \( \text{CH}_3\)₂H, R⁴ = \( \text{Ph} \)
1.23h, R¹ = \( \text{CH}_3\text{CO}_2\text{Me} \), R² = \( \text{Ph} \), R³ = H, R⁴ = \( \text{Bu} \)
1.23i, R¹ = \( \text{CH}_3\text{CO}_2\text{Me} \), R² = \( \text{Ph} \), R³ = \( \text{CH}_3\)₂H, R⁴ = \( \text{Bu} \)
1.23j, R¹ = \( \text{CH}_3\text{OCH}_2\text{Ph} \), R² = 4-Me-Ph, R³ = H, R⁴ = \( \text{Bu} \)
1.23k, R¹ = \( \text{Ph} \), R² = \( \text{Ph} \), R³ = \( \text{Ph} \), R⁴ = \( \text{Bu} \)
1.23l, R¹ = \( \text{Ph} \), R² = \( \text{Ph} \), R³ = \( \text{Ph} \), R⁴ = \( \text{Bu} \)
1.23m, R¹ = \( \text{Ph} \), R² = \( \text{Ph} \), R³ = \( \text{CH}_3\)₂H, R⁴ = \( \text{Bu} \)
1.23n, R¹ = \( \text{Ph} \), R² = \( \text{Ph} \), R³ = \( \text{CH}_3\)₂H, R⁴ = \( \text{CH}_3\text{CO}_2\text{Et} \)

**Dihydro pyrrolones synthesized:**

1.25a, R¹ = Ph, R³ = \( \text{Ph} \), R⁴ = \( \text{Bu} \), Nu = \( \text{OH} \)
1.25b, R¹ = Ph, R³ = \( \text{Ph} \), R⁴ = \( \text{Bu} \), Nu = \( \text{OCH}_3 \)
1.25c, R¹ = Ph, R³ = \( \text{Ph} \), R⁴ = \( \text{Bu} \), Nu = \( \text{OCH}_3 \)
1.25d, R¹ = Ph, R³ = 4-Me-Ph, R⁴ = \( \text{CH}_3\text{Ph} \), Nu = \( \text{OH} \)
1.25e, R¹ = \( \text{Ph} \), R³ = \( \text{Ph} \), R⁴ = \( \text{CH}_3\text{Ph} \), Nu = \( \text{OCH}_3 \)

**Scheme B.3.1.2:** Synthesis of alkynes 1.23a-n and dihydropyrrolones 1.25a-e.

**General Procedure 1.4,** *Nitrosative cleavage of the 5-isoxazolino ring:* All the solutions must be thoroughly degassed beforehand. To a three neck flask under an argon atmosphere charged with FeSO₄•7H₂O (5.5 equiv.) and AcOH (0.25 M to isoxazolinone), equipped with two addition funnels, one charged with a solution of NaNO₂ (10 equiv.) in H₂O (0.25 M to isoxazolinone) and the other with isoxazolinone (1 equiv.) in AcOH (0.25 M to isoxazolinone) was added half of a solution of sodium nitrite in water. The remainder was added slowly simultaneously with a solution of isoxazolinone in acetic acid at room temperature. The reaction was allowed to stir at rt until completion (TLC). The set-up was then flushed with nitrogen for 30 minutes, water was added and the reaction extracted with DCM (3x). The organic extracts were treated with NaHCO₃ (with stirring) for 40 min and dried (MgSO₄). Concentration under reduced pressure and purification by flash column chromatography of the residue provided alkynes in the stated yields.

**General Procedure 1.5:** To a round bottom flask charged with alkyne (1 equiv.), wet acetone (0.2 M) and the corresponding alcohol used as nucleophile (10 equiv.) was added N-iodo-succinimide (2.1 equiv.) dissolved in wet acetone (0.2 M). In the case where water is inserted, no nuceleophile is added (water comes from acetone). The reaction was allowed to stir at room temperature. After complete consumption of the starting material (TLC), a
solution of Na$_2$S$_2$O$_3$ in water was added. The mixture was extracted with Et$_2$O (3x), dried (MgSO$_4$) and concentrated under reduced pressure. Purification by flash column chromatography (silica, PE:AcOEt) afforded the pure product in the stated yields.

<table>
<thead>
<tr>
<th>2-Cyclopropyl-4-phenyl-but-3-ynoic acid tert-butylamide</th>
<th>1.23a</th>
</tr>
</thead>
<tbody>
<tr>
<td><img src="image" alt="Chemical Structure" /></td>
<td></td>
</tr>
<tr>
<td>MF: C$<em>{17}$H$</em>{21}$ON</td>
<td></td>
</tr>
<tr>
<td>MW = 255 g.mol$^{-1}$</td>
<td></td>
</tr>
</tbody>
</table>

**Method:**  See general procedure 1.4 using (1 equiv., 1.6 mmol, 510 mg) of the corresponding isoxazolin-5-one.

**Purification:**  Flash column chromatography (silica gel, 6:4 PE:Et$_2$O)/ $R_f$ (6:4 PE:Et$_2$O): 0.88.

**Product:**  Pale pink solid.

**Yield:**  92%.

**$^1$H NMR** ($\delta$, ppm) (CDCl$_3$, 400 MHz)

- 7.42-7.39 (m, 2H, CH$_2$-Ph), 7.33-7.30 (m, 3H, CH$_2$-Ph), 6.32 (br s, 1H, NH), 3.39 (d, $J = 5.6$ Hz, CHCON), 1.41-1.36 (m, 1H, CH$_2$-cyclopropane), 1.39 (s, 9H, CH$_3$-Bu), 0.63-0.54 (m, 2H, CH$_2$-cyclopropane), 0.53-0.43 (m, 2H, CH$_2$-cyclopropane).

**$^{13}$C NMR** ($\delta$, ppm) (CDCl$_3$, 100 MHz)

- 168.9 (Cq, C=O), 131.6 (CH$_2$-Ph), 128.5 (CH$_2$-Ph), 128.4 (CH$_2$-Ph), 122.4 (Cq, Ph), 86.4 (Cq, C=CC), 84.8 (Cq, C=CC), 51.3 (Cq, $^3$Bu), 43.5 (CHCON), 28.6 (CH$_3$-Bu), 12.5 (CH-cyclopropane), 1.1 (CH$_2$-cyclopropane), 1.6 (CH$_3$-cyclopropane).

**IR** ($\nu$, cm$^{-1}$) (CCl$_4$)

- 3410 (m), 3082 (w), 3001 (w), 2968 (m), 2928 (w), 2873 (w), 1685 (s), 1513 (s), 1453 (m), 1391 (w), 1364 (m), 1274 (m), 1222 (m), 1021 (w).

**HRMS** (EI+, m/z)

- Calculated: 255.1623  found: 255.1622.
4-Phenyl-2-p-tolyl-but-3-ynoic acid tert-butylamide  

\[
\text{MF: } C_{21}H_{23}ON \\
\text{MW } = 305 \text{ g.mol}^{-1}
\]

**Method:** See general procedure 1.4 using (1 equiv., 1.15 mmol, 420 mg) of the corresponding isoxazolin-5-one.

**Purification:** Flash column chromatography (silica gel, 1:1 PE: Et\(_2\)O) / \(R_f\) (1:1 PE: Et\(_2\)O): 0.50.

**Product:** Pale yellow solid.

**Isolated Yield:** 87%.

**\(^1\)H NMR** (δ, ppm) (CDCl\(_3\), 400 MHz) 7.52-7.49 (m, 2H, CH-Ph), 7.48 (d, J = 8.0 Hz, 2H, CH-tolyl), 7.42-7.40 (m, 3H, CH-Ph), 7.20 (d, J = 8.0 Hz, 2H, CH-tolyl), 6.50 (br s, 1H, NH), 4.61 (s, 1H, CHCON), 2.37 (s, 3H, CH\(_3\)), 1.39 (s, 9H, CH\(_3\)-tBu).

**\(^{13}\)C NMR** (δ, ppm) (CDCl\(_3\), 100 MHz) 167.7 (Cq, C=O), 137.1 (Cq, Ar), 133.6 (Cq, Ar), 131.4 (CH-Ph), 129.3 (CH-tolyl), 128.4 (CH-Ph), 128.3 (CH-Ph), 127.5 (CH-tolyl), 122.3 (Cq, Ar), 87.0 (Cq, C=CC), 86.3 (Cq, C=CC), 51.3 (Cq, 'Bu), 46.5 (CHCON), 28.4 (CH\(_3\)-tBu), 21.0 (CH\(_3\)).

**IR** (ν, cm\(^{-1}\)) (CCl\(_4\)) 3410 (w), 3056 (w), 3028 (w), 2969 (m), 2926 (w), 2871, 1689 (s), 1512 (s), 1452 (w), 1391 (w), 1366 (w), 1271 (w), 1224 (w), 1045 (w), 1026 (w).

**HRMS** (EI+, m/z): Calculated: 305.1780 found: 305.1777.

---

(4-Phenyl-2-p-tolyl-but-3-ynoylamino)-acetic acid ethyl ester  

\[
\text{MF: } C_{21}H_{21}O_3N \\
\text{MW } = 335 \text{ g.mol}^{-1}
\]

---
Method: See general procedure 1.4 using (1 equiv., 0.20 mmol, 65 mg) of the corresponding isoxazolin-5-one.

Purification: Flash column chromatography (silica gel, 6:4 Et$_2$O: PE) / R$_f$ (7:3 PE:EtOAc): 0.50.

Product: Brown solid.

Isolated yield: 81%.

$^1$H NMR (δ, ppm) 7.54-7.52 (m, 2H, CH$_2$-Ph), 7.46 (d, J = 8.2 Hz, 2H, CH-tolyl), 7.35-7.34 (m, 3H, CH-Ph), 7.19 (d, J = 8.2 Hz, 2H, CH-tolyl), 7.01 (br t, J = 5.2 Hz, 1H, NH), 4.73 (s, 1H, CHCON), 4.21 (q, J = 7.1 Hz, 2H, OCH$_2$CH$_3$), 4.08 (dd, J = 5.2 Hz, J = 18.5 Hz, 1H, CH$_3$NH), 4.00 (dd, J = 5.2, J = 18.5 Hz, 1H, CH$_2$NH), 2.35 (s, 3H, CH$_3$), 1.26 (t, J = 7.1 Hz, 3H, OCH$_2$CH$_3$)

$^{13}$C NMR (δ, ppm) 169.5 (Cq, C=O), 168.8 (Cq, C=O), 137.7 (Cq, Ar), 133.1 (Cq, Ar), 131.8 (CH-Ph), 129.5 (CH-tolyl), 128.6 (CH-Ph), 128.4 (CH-Ph), 127.8 (CH-tolyl), 122.3 (Cq, Ar), 87.5 (Cq, C==C), 85.3 (Cq, C==C), 61.6 (OCH$_2$CH$_3$), 45.7 (CHCON), 41.8 (CH$_3$NH), 21.1 (CH$_3$), 14.1 (OCH$_2$CH$_3$)

IR (ν, cm$^{-1}$) (CCl$_4$) 3406 (w), 3028 (w), 2984 (w), 2926 (w), 2864 (w), 1745 (s), 1690 (s), 1512 (s), 1443 (w), 1376 (w), 1351 (w), 1205 (s), 1023 (w).

HRMS (EI+, m/z): Calculated: 335.1521 found: 335.1529.

4-Phenyl-2-p-tolyl-but-3-ynoic acid benzylamide 1.23d

MF: C$_{24}$H$_{21}$ON

MW = 339g.mol$^{-1}$

Method: See general procedure 1.4 using (1 equiv., 0.25 mmol, 100 mg) of the corresponding isoxazolin-5-one.

Purification: Flash column chromatography (silica gel, 6:4 Et$_2$O: PE) / R$_f$ (8:2 PE:EtOAc): 0.39.

Product: Yellow solid.

230
Isolated yield : 90%.

$^1$H NMR ($\delta$, ppm) 7.52-7.48 (m, 4H, CH-Ar), 7.39-7.24 (m, 10H, CH-Ar), 6.81 (br t, $J = 5.8$ Hz, 1H, NH), 4.83 (s, 1H, CHCON), 4.56 (dd, $J = 5.8$ Hz, $J = 15.0$ Hz, 1H, CH$_2$NH), 4.48 (dd, $J = 5.8$ Hz, $J = 15.0$ Hz, 1H, CH$_2$NH), 2.42 (s, 3H, CH$_3$)

$^{13}$C NMR ($\delta$, ppm) 168.8 (Cq, C=O), 137.9 (Cq, Ar), 137.6 (Cq, Ar), 133.2 (Cq, Ar), 131.7 (CH-Ar), 129.5 (CH-Ar), 128.7 (CH-Ar), 128.6 (CH-Ar), 128.3 (CH-Ar), 127.7 (CH-Ar), 127.5 (CH-Ar), 127.4 (CH-Ar), 122.3 (Cq, Ar), 87.3 (Cq, C=CC), 85.7 (Cq, C=CC), 46.0 (CHCON), 43.8 (CH$_3$NH), 21.1 (CH$_3$).

IR ($\nu$, cm$^{-1}$) (CCl$_4$) 3424 (w), 3062 (w), 3031 (w), 2925 (w), 1690 (s), 1512 (s), 1451 (w), 1253 (w), 1196 (w), 1196 (w).

HRMS (EI+, m/z) : Calculated: 339.1623  found: 339.1620.

---

4-Phenyl-2-thiophen-2-yl-but-3-ynoic acid tert-butyramide

![Chemical Structure]

MF: C$_{18}$H$_{19}$ONS

MW = 297 g.mol$^{-1}$

Method : See general procedure 1.4 using (1 equiv., 0.36 mmol, 130 mg) of the corresponding isoxazolin-5-one.

Purification : Flash column chromatography (silica gel, 6:4 Et$_2$O: PE) / $R_f$ (6:4 Et$_2$O: PE): 0.66.

Product : Brown solid.

Isolated Yield : 92%.

$^1$H NMR ($\delta$, ppm) 7.51-7.48 (m, 2H, CH-Ph), 7.38-7.35 (m, 3H, CH-Ph), 7.25 (dd, $J = 1.2$ Hz, $J = 4.8$ Hz, 1H, CH thiophene), 7.19 (dt, $J = 1.2$ Hz, $J = 3.6$ Hz, 1H, CH-thiophene), 6.98 (dd, $J = 3.6$ Hz, $J = 4.8$ Hz, 1H, CH-thiophene), 6.45 (br s, 1H, NH), 4.85 (s, 1H, CHCON), 1.37 (s, 9H, CH$_3$-Bu).

$^{13}$C NMR ($\delta$, ppm) 166.5 (Cq, C=O), 138.9 (Cq, Ar), 131.6 (CH-Ph), 128.8 (CH-Ph), 128.5 (CH-Ph), 126.6 (CH-thiophene), 125.9 (CH-thiophene), 125.4 (CH-thiophene), 122.1 (Cq, Ar), 87.3 (Cq, C=CC), 85.3 (Cq, C=CC), 51.6 (Cq, 'Bu), 42.5 (CHCON), 28.4 (CH$_3$-Bu).
**2,2-Dimethyl-4-phenyl-but-3-ynoic acid tert-butylamide**

**MF:** C_{16}H_{21}ON

**MW:** 243 g. mol\(^{-1}\)

**Method:** See general procedure 1.4 using (1 equiv., 1.14 mmol, 344 mg) of the corresponding isoxazolin-5-one.

**Purification:** Flash column chromatography (silica gel, 9:1 PE: AcOEt)/ \(R_f\) (7:3 PE: AcOEt): 0.92.

**Product:** White solid.

**Yield:** 76%.

**\(^1\)H NMR** (\(\delta\), ppm) (CDCl\(_3\), 400 MHz)

7.42-7.40 (m, 2H, CH-Ph), 7.33-7.32 (m, 3H, CH-Ph), 6.67 (br s, 1H, NH), 1.50 (s, 6H, CH\(_3\)), 1.37 (s, 9H, CH\(_3\)-t-Bu).

**\(^13\)C NMR** (\(\delta\), ppm) (CDCl\(_3\), 100 MHz)

172.7 (Cq, C=O), 131.4 (CH-Ph), 128.4 (CH-Ph), 128.4 (CH-Ph), 122.6 (Cq, Ph), 92.8 (Cq, C \(\equiv\) C ), 84.8 (Cq, C \(\equiv\) C ), 51.0 (Cq), 39.8 (Cq), 28.5 (CH\(_3\)-t-Bu), 27.5 (CH\(_3\)).

**IR** (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))

3410 (w), 3062 (w), 2973 (w), 2973 (w), 2933 (w), 2871 (w), 1812 (w), 1683 (s), 1511 (s), 1455 (w), 1391 (w), 1365 (w), 1271 (w), 1223 (w), 1178 (w), 1178(w).

**HRMS** (EI+, m/z):

Calculated: 243.1623  found: 243.1617.
**N-tert-Butyl-2-(4-nitro-5-oxo-3-phenyl-4,5-dihydro-isoxazol-4-yl)-isobutyramide**

\[ \text{MF: C}_{17}\text{H}_{21}\text{O}_{5}\text{N}_{3} \]

\[ \text{MW} = 347\text{.mol}^{-1} \]

**Method:**
See general procedure 1.4 using (1 equiv., 1.14 mmol, 344 mg) of the corresponding isoxazolin-5-one.

**Purification:**
Flash column chromatography (silica gel, 9:1 PE: AcOEt) / \( R_f \) (9:1 PE: AcOEt): 0.31.

**Product:**
White solid.

**Isolated yield:**
20%.

**\(^1\text{H NMR}\) (δ, ppm)**

(CDCl₃, 400 MHz)

7.57-7.51 (m, 3H, CH-Ph), 7.47-7.43 (m, 2H, CH-Ph), 5.31 (br s, 1H, NH) 1.71 (s, 3H, CH₃), 1.26 (s, 9H, CH₃-tBu), 1.21 (s, 3H, CH₃).

**\(^{13}\text{C NMR}\) (δ, ppm)**

(CDCl₃, 100 MHz)

171.3 (Cq, C=O), 167.3 (Cq, C=O), 161.8 (Cq, C=N), 131.7 (CH-Ph), 129.3 (CH-Ph), 128.9 (CH-Ph), 128.1 (Cq, Ph), 94.3 (Cq), 52.1 (Cq), 51.8 (Cq), 28.3 (CH₃-tBu), 22.1 (CH₃), 21.7 (CH₃).

**IR (ν, cm\(^{-1}\)) (CCl₄)**

3583 (w), 3456 (w), 3064 (w), 2971 (w), 2932 (w), 2873 (w), 1809 (s), 1706 (s), 1676 (s), 1562 (s), 1513 (s), 1454 (m), 1392 (w), 1366 (w), 1340 (m), 1272 (w), 1219 (m), 1179 (w), 1145 (w), 1096 (w).

**HRMS (El+, m/z):**

Calculated: 347.1481 found: 347.1496.

---

**1-Phenylethynyl-cyclopentanecarboxylic acid tert-butylamide**

\[ \text{MF: C}_{18}\text{H}_{23}\text{ON} \]

\[ \text{MW} = 269\text{.g.mol}^{-1} \]
Method: See general procedure 1.4 using (1 equiv., 0.40 mmol, 130 mg) of the corresponding isoxazolin-5-one.

Purification: Flash column chromatography (silica gel, 9:1 PE: AcOEt)/ Rf (7:3 PE: AcOEt): 0.69.

Product: White solid.

Isolated Yield: 71%.

1H NMR (δ, ppm) (CDCl3, 400 MHz) 7.41-7.39 (m, 2H, CH-Ph), 7.33-7.31 (m, 3H, CH-Ph), 6.70 (br s, 1H, NH), 2.28-2.21 (m, 2H, CH2-cyclopentane), 2.03-1.97 (m, 2H, CH2-cyclopentane), 1.87-1.83 (m, 4H, CH2-cyclopentane), 1.38 (s, 9H, CH3-tBu).

13C NMR (δ, ppm) (CDCl3, 100 MHz) 172.4 (Cq, C=O), 131.4 (CH-Ph), 128.4 (CH-Ph), 128.2 (CH-Ph), 122.9 (Cq, Ph), 93.0 (Cq, C≡C), 85.0 (Cq, C≡C), 51.1 (Cq), 50.0 (Cq), 39.8 (CH2-cyclopentane), 28.6 (CH3-tBu), 25.7 (CH2-cyclopentane).

IR (v, cm⁻¹) (CCl4) 3409 (w), 3060 (w), 2966 (s), 2871 (w), 1680 (s), 151 (s), 1452 (w), 1391 (w), 1365 (w), 1322 (w), 1267 (w), 1222 (w).

HRMS (EI+, m/z) Calculated: 269.1780, found: 269.1778.

1-(4-Nitro-5-oxo-3-phenyl-4,5-dihydro-isoxazol-4-yl)-cyclopentanecarboxylic acid tert-butylamide 1.17g

MF: C19H23O5N3

MW = 373 g.mol⁻¹

Method: See general procedure 1.4 using (1 equiv., 0.40 mmol, 130 mg) of the corresponding isoxazolin-5-one.

Purification: Flash column chromatography (silica gel, 9:1 PE: AcOEt)/ Rf (7:3 PE: AcOEt): 0.53.

Product: dark red solid.
Isolated yield : 18%.

$^1$H NMR ($\delta$, ppm) ($\text{CDCl}_3$, 400 MHz) 7.57-7.41 (m, 5H, CH-Ph), 5.19 (br s, 1H, NH), 2.98-2.91 (m, 1H, CH$_2$-cyclopentane), 2.15-2.08 (m, 1H, CH$_2$-cyclopentane), 2.04-1.95 (m, 1H, CH$_2$-cyclopentane), 1.87-1.75 (m, 2H, CH$_2$-cyclopentane), 1.71-1.61 (m, 1H, CH$_2$-cyclopentane), 1.53-1.43 (m, 2H, CH$_2$-cyclopentane), 1.25 (s, 9H, CH$_3$-t-Bu).

$^{13}$C NMR ($\delta$, ppm) ($\text{CDCl}_3$, 100 MHz) 171.3 (Cq, C=O), 167.6 (Cq, C=O), 162.4 (Cq, C=N) 131.8 (CH-Ph), 129.4 (CH-Ph), 128.7 (CH-Ph), 127.3 (Cq, Ph), 94.4 (Cq), 61.5 (Cq), 52.1 (Cq), 33.1 (CH$_2$-cyclopentane), 32.7 (CH$_2$-cyclopentane), 28.3 (CH$_3$-Bu), 25.9 (CH$_2$-cyclopentane), 24.7 (CH$_2$-cyclopentane).

IR ($\nu$, cm$^{-1}$) ($\text{CCl}_4$) 3455 (w), 3063 (w), 2966 (s), 2874 (w), 1804 (s), 1720 (s), 1676 (s), 1562 (s), 1513 (s), 1453 (s), 1392 (w), 1365 (m), 1338 (s), 1263 (m), 1219 (m), 1127 (w), 1072 (w).

HRMS (EI+, m/z) : Calculated: 373.1648 found: 373.1638.

X-ray structure for $^{1.17}$g

![X-ray structure](image)

Table 1. Crystal data for $^{1.17}$g

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<th>Compound</th>
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<td>Molecular formula</td>
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<tr>
<td>Molecular weight</td>
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<tr>
<td>Crystal habit</td>
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<tr>
<td>Crystal dimensions(mm)</td>
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Crystal system: monoclinic
Space group: P2_1/c

a(Å)  7.507(1)
b(Å)  18.276(1)
c(Å)  14.379(1)
α(°)  90.00
β(°)  112.491(1)
γ(°)  90.00
V(Å³)  1822.7(3)
Z  4

d(g·cm⁻³)  1.361
F(000)  792
λ(Å)  0.100

Absorption corrections: multi-scan, 0.9669 min, 0.9960 max

Diffractometer: KappaCCD
X-ray source: MoKα
μ(Å)  0.71069
Monochromator: graphite
T (K)  150.0(1)
Scan mode: phi and omega scans
Maximum θ  27.48

HKL ranges  -9 9 , -21 23 , -14 18

Reflections measured  10476
Unique data  4171

Rint  0.0307
Reflections used  2721
Criterion  I > 2σ(I)

Refinement type: Fsqd
Hydrogen atoms: mixed
Parameters refined  247

Reflections / parameter  11

wR2  0.1069
R1  0.0412

Weights a, b  0.0506 , 0.0000
GoF  1.021

difference peak / hole (e Å⁻³)  0.230(0.042) / -0.239(0.042)

Table 2. Atomic Coordinates (Å x 10⁴) and equivalent isotropic displacement parameters (Å² x 10³) for 1.17g

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**U(eq) is defined as 1/3 the trace of the Uij tensor.**

**Table 3. Bond lengths (Å) and angles (deg) for 1.17g**

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Table 4. Anisotropic displacement parameters (Å² x 10³) for 1.17g

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The anisotropic displacement factor exponent takes the form
\[2 \pi^2 [h^2 a^* U(11) + ... + 2hka*b*U(12)]\]

Table 5. Hydrogen Coordinates (Å x 10⁴) and equivalent isotropic displacement parameters (Å² x 10³) for 1.17g

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H(19B) 7558  4315  6117  43
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H(20B) 9263  3617  5446  38
H(25A) 6276  1028  4359  54
H(25B) 4470  495  4139  54
H(25C) 4944  1107  4996  54
H(26A) 2967  1779  2060  57
H(26B) 3145  917  2278  57
H(26C) 5048  1409  2570  57
H(27A) 1571  1579  4157  70
H(27B) 1126  978  3287  70
H(27C) 800  1828

5-tert-Butylcarbamoyl-5-cyclopropyl-pent-3-ynoic acid methyl ester 1.23h

MF: C_{14}H_{21}O_{3}N

MW = 251 g.mol^{-1}

Method : See general procedure 1.4 using (1 equiv., 1.64 mmol, 508 mg) of the corresponding isoxazolin-5-one.

Purification : Flash column chromatography (silica gel, 7:2:1 tol: DCM: AcOEt)/ R_{f} (1:1 PE: AcOEt): 0.67.

Product : Yellow oil.

Isolated yield : 9% (difficult to isolate).

^1H NMR (δ, ppm) (CDCl_{3}, 400 MHz) 6.66 (br s, 1H, NH), 3.74 (s, 3H, OCH_{3}), 3.30 (d, J = 2.4Hz, CH_{2}CO_{2}), 3.18-3.16 (m, 1H, CHCON), 1.37 (s, 9H, CH_{3}-tBu), 1.30-1.24 (m, 1H, CH-cyclopropane), 0.54-0.33 (m, 4H, CH_{2}-cyclopropane).

^13C NMR (δ, ppm) (CDCl_{3}, 100 MHz) 169.5 (Cq, OC=O), 168.9 (Cq, NC=O), 79.5 (Cq, C $\equiv$ C), 78.5 (Cq, C $\equiv$ C), 52.6 (OCH_{3}), 51.4 (Cq, tBu), 42.8 (CHCON), 28.6 (CH_{3}-tBu), 25.7 (CH_{2}CO_{2}), 12.3 (CH-cyclopropane), 3.1 (CH_{2}-cyclopropane), 1.4 (CH_{2}-cyclopropane).

IR (v, cm^{-1}) (CCl_{4}) 3382 (w), 2968 (m), 1748 (s), 1721 (s), 1680 (s), 1609 (w), 1562 (m), 1519 (s), 1454 (m), 1398 (w), 1365 (m), 1338 (m), 1267 (m), 1209 (s), 1177 (s), 1021
4-(1-tert-Butylcarbamoyl-cyclopentyl)-but-3-ynoic acid methyl ester

MF: C_{15}H_{23}O_{3}N

MW = 265 g mol\(^{-1}\)

Method:
See general procedure 1.4 using (1 equiv., 1.13 mmol, 366 mg) of the corresponding isoxazolin-5-one.

Purification:
Flash column chromatography (silica gel, 2x, 95:5 toluene:AcOEt, then 9:1 PE:AcOEt)/ \(R_f\) (9:1 PE:AcOEt): 0.38.

Product:
Transparent oil.

Isolated yield:
15%.

\(^1\)H NMR (\(\delta\), ppm) (CDCl\(_3\), 400 MHz)
6.96 (br s, 1H, NH), 3.73 (s, 3H, OCH\(_3\)), 3.30 (s, 2H, CH\(_2\)CO\(_2\)), 2.15-2.10 (m, 2H, CH\(_2\)-cyclopentane), 1.88-1.83 (m, 2H, CH\(_2\)-cyclopentane), 1.79-1.73 (m, 4H, CH\(_2\)-cyclopentane), 1.36 (s, 9H, CH\(_3\)-tBu).

\(^{13}\)C NMR (\(\delta\), ppm) (CDCl\(_3\), 100 MHz)
172.5 (Cq, C=O), 169.0 (Cq, C=O), 87.7 (Cq, C\(\equiv\)C), 76.5 (Cq, C\(\equiv\)C), 52.5 (OCH\(_3\)), 51.1 (Cq), 49.3 (Cq), 39.6 (CH\(_2\)), 28.6 (CH\(_3\)-Bu), 25.8 (CH\(_3\)), 25.5 (CH\(_2\)).

IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))
3384 (m), 2963 (s), 2871 (m), 1750 (s), 1677 (s), 1515 (s), 1451 (s), 1396 (w), 1364 (w), 1341 (w), 1319 (w), 1265 (s), 1204 (s), 1173 (s).

HRMS (EI+, m/z): MS (HRMS EI): calculated: 265.1678 found: 265.1665.
**5-Benzzyloxy-2-p-tolyl-pent-3-ynoic acid tert-butylamide** 1.23j

![Chemical structure of 5-Benzzyloxy-2-p-tolyl-pent-3-ynoic acid tert-butylamide]

**MF:** C\textsubscript{23}H\textsubscript{27}O\textsubscript{2}N  
**MW:** 349 g.mol\textsuperscript{-1}

**Method:** See *general procedure 1.4* using (1 equiv., 0.25 mmol, 100 mg) of the corresponding isoxazolin-5-one.

**Purification:** Flash column chromatography (silica gel, 95:5 toluene: AcOEt)/ R\textsubscript{f} (95:5 toluene: AcOEt): 0.29.

**Product:** White solid.

**Yield:** 47%.

**\textsuperscript{1}H NMR (\textdelta, ppm)**  
(CDCl\textsubscript{3}, 400 MHz)  
7.36-7.32 (m, 7H, CH-Ph + CH-tolyl), 7.16 (d, \textit{J} = 8.0 Hz, 2H, CH-tolyl), 6.24 (br s, 1H, NH), 4.62 (s, 2H, OCH\textsubscript{2}Ph), 4.42 (t, \textit{J} = 2.0 Hz, CHCON), 4.29 (d, \textit{J} = 2.0 Hz, 2H, CH\textsubscript{2}O), 2.34 (s, 3H, CH\textsubscript{3}), 1.32 (s, 9H, CH\textsubscript{3}-tBu)

**\textsuperscript{13}C NMR (\textdelta, ppm)**  
(CDCl\textsubscript{3}, 100 MHz)  
167.5 (Cq, C=O), 137.4 (Cq, Ar), 137.2 (Cq, Ar), 133.6 (Cq, Ar), 129.4 (CH-Ar), 128.5 (CH-Ar), 128.0 (CH-Ar), 127.6 (CH-Ar), 83.9 (Cq, C=C=C), 83.0 (Cq, C=C=C), 71.7 (OCH\textsubscript{2}Ph), 57.5 (OCH\textsubscript{2}Ph), 51.5 (Cq, tBu), 46.3 (CHCON), 28.5 (CH\textsubscript{3}-tBu), 21.1 (CH\textsubscript{3})

**IR (\nu, \textit{cm}^{-1})** (CCl\textsubscript{4})  
3412 (w), 3031 (w), 2968 (w), 2926 (w), 2862 (w), 2862 (w), 1689 (s), 1512 (s), 1454 (w), 1389 (w), 1363 (w), 1270 (w), 1222 (w), 1077 (w), 1023 (w).

**HRMS (EI*, m/z):**  
Calculated: 349.2042  
found: 349.2049

---

**2-Cyclopropyl-oct-7-en-3-ynoic acid tert-butylamide** 1.23k

![Chemical structure of 2-Cyclopropyl-oct-7-en-3-ynoic acid tert-butylamide]

**MF:** C\textsubscript{15}H\textsubscript{23}ON  
**MW:** 233 g.mol\textsuperscript{-1}

**Method:** See *general procedure 1.4* using (1 equiv., 0.30 mmol, 100mg) of the corresponding isoxazolin-5-one.
Purification: Flash column chromatography (silica gel, 8:2 PE: AcOEt) / $R_f$ (8:2 PE:AcOEt): 0.68.

Product: Pale yellow oil.

Isolated yield: 94%.

$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz) 6.31 (br, 1H, NH), 5.87-5.77 (m, 1H, CH=CH$_2$), 5.10-5.02 (m, 2H, CH=CH$_2$), 3.16-3.14 (m, 1H, CHCON), 2.31-2.21 (m, 4H, CH$_2$-homo allyl chain), 1.34 (s, 9H, CH$_3$-tBu), 1.26-1.22 (m, 1H, CH-cyclo propane), 0.52-0.31 (m, 4H, CH$_2$-cyclo propane).

$^{13}$C NMR (δ, ppm) (CDCl$_3$, 100 MHz) 169.7 (Cq, C=O), 136.6 (CH=CH$_2$), 115.9 (CH=CH$_2$), 86.1 (Cq, C≡C C), 76.0 (Cq, C≡C C), 51.1 (Cq, Bu), 42.9 (CHCON), 32.8 (CH$_2$-homo allyl chain), 28.6 (CH$_2$-homo allyl chain), 18.4 (CH$_3$-Bu), 12.3 (CH-cyclo propane), 2.8 (CH$_2$-cyclo propane), 1.2 (CH$_2$-cyclo propane).

IR (ν, cm$^{-1}$) (CCl$_4$) 3406 (w), 3082 (w), 2970 (w), 2926 (m), 2873 (w), 1807 (w), 1683 (s), 1565 (s), 1514 (s), 1454 (w), 1391 (w), 1365 (w), 1332 (w), 1274 (w).

HRMS (EI+, m/z): Calculated: 233.1780; found: 233.1777.

2-Cyclopropyl-oct-7-en-3-ynoic acid benzylamide

Method: See general procedure 1.4 using (1 equiv., 0.55 mmol, 190 mg) of the corresponding isoxazolin-5-one.

Purification: Flash column chromatography (silica gel, 95:5 toluene: AcOEt) / $R_f$ (7:3 PE:AcOEt): 0.55.

Product: Pale yellow solid.

Isolated yield: 60%.

$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz) 7.36-7.33 (m, 2H, CH-Ph), 7.29-7.27 (m, 3H, CH-Ph), 6.82 (br t, J = 4.8 Hz, 1H, NH), 5.74 (ddt, 1H, J = 6.4Hz, J = 10.2Hz, J = 16.9Hz, 1H, CH=CH$_2$), 4.95 (ddt, J = 1.5Hz, J = 1.6 Hz, J = 16.9 Hz, 1H, CH=CH$_2$), 4.91 (ddt, J = 1.0Hz, J = 1.6Hz, J = 10.2 Hz, 1H, CH=CH$_2$), 4.50 (dd, J = 4.8Hz, J = 14.0Hz, 1H, CH$_2$NH), 4.45 (dd, J = 4.8Hz, J = 14.0Hz, 1H, CH$_2$NH), 3.37-3.35 (m, 1H, CH$_2$NH), 3.37-3.35 (m, 1H, CH$_2$NH).
CHCON), 2.30-2.26 (m, 2H, CH$_2$-homo allyl chain), 2.22-2.17 (m, 2H, CH$_2$-homo allyl chain), 1.39-1.31 (m, 1H, CH-cyclo propane), 0.55-0.37 (m, 4H, CH$_2$-cyclo propane).

$^{13}$C NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz) 170.5 (Cq, C=O), 138.2 (Cq, Ph), 136.8 (CH=CH$_2$), 128.6 (CH-Ph), 127.4 (CH-Ph), 115.9 (CH=CH$_2$), 86.4 (Cq, C=C=C), 75.4 (Cq, C=C=C), 43.6 (CH$_2$-cyclo propane), 42.2 (CHCON), 32.8 (CH$_2$-homo allyl chain), 12.4 (CH-cyclo propane), 2.8 (CH$_2$-cyclo propane), 1.4 (CH$_2$-cyclo propane).

IR ($\nu$, cm$^{-1}$) (CCl$_4$) 3418 (w), 3081 (w), 3030 (w), 3007 (w), 2925 (w), 2851 (w), 1682 (s), 1514 (s), 1454 (w), 1398 (w), 1356 (w), 1333 (w), 1252 (w), 1022 (w).

HRMS (EI+, m/z) : Calculated: 267.1623 found: 267.1622.

1-Hex-5-en-1-ynyl-cyclopentanecarboxylic acid tert-butylamide 1.23m

![Chemical Structure](image)

MF: C$_{16}$H$_{25}$ON

MW = 247g.mol$^{-1}$

Method : See general procedure 1.4 using (1 equiv., 0.33 mmol, 100 mg) of the corresponding isoxazolin-5-one.

Purification : Flash column chromatography (silica gel, 95:5 PE:AcOEt)/ $R_{f}$ (8:2 PE:AcOEt): 0.72.

Product : Transparent oil.

Isolated yield : 68%.

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz) 6.69 (br s, 1H, NH), 5.84 (ddt, J = 6.3Hz, J = 10.2Hz, J = 16.9Hz, 1H, CH=CH$_2$), 5.10 (ddt, J = 1.4Hz, J = 1.5Hz, J = 16.9Hz, 1H, CH=CH$_2$), 5.04 (ddm, J = 1.4, J = 10.2, 1H, CH=CH$_2$), 2.34-2.30 (m, 2H, CH$_2$-cyclo pentane), 2.28-2.23 (m, 2H, CH$_2$-homo allyl chain), 2.15-2.08 (m, 2H, CH$_2$-cyclo pentane) 1.85-1.75 (m, 6H, CH$_2$-cyclo pentane + homo allyl chain), 1.34 (s, 9H, CH$_3$-Bu).

$^{13}$C NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz) 173.0 (Cq, C=O), 136.8 (CH=CH$_2$), 115.9 (CH=CH$_2$), 84.6 (Cq, C=C=C), 84.4 (Cq, C=C=C), 50.9 (Cq), 50.0 (Cq), 39.8 (CH$_2$-homo allyl chain), 32.8 (CH$_2$-cyclo pentane), 28.6(CH$_3$-Bu), 25.4 (CH$_2$-homo allyl chain), 18.5 (CH$_2$-cyclo pentane).

IR ($\nu$, cm$^{-1}$) (CCl$_4$) 3405 (w), 3080 (w), 2965 (w), 2871 (w), 1678 (s), 1512 (s), 1452 (w), 1391 (w), 1364 (w), 1334 (w), 1267 (w), 1223 (w).
HRMS (El+, m/z) : Calculated: 247.1936  found: 247.1946.

1-{(3-But-3-enyl-4-nitro-5-oxo-4,5-dihydro-isoxazol-4-yl)-cyclopentanecarboxylic acid tert-butylamide}  

\[
\begin{align*}
\text{MF: } & C_{17}H_{25}O_5N_3 \\
\text{MW: } & 351 \text{ g.mol}^{-1}
\end{align*}
\]

Method: See general procedure 1.4 using (1 equiv., 0.33 mmol, 100 mg) of the corresponding isoxazolin-5-one.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ Rf (8:2 PE:AcOEt): 0.25.

Product: White solid.

Isolated yield: 26%.

\[\text{^1H NMR (δ, ppm)}\]

(\text{CDCl}_3, 400 MHz)

5.86 (ddt, \(J = 6.4\text{Hz}, J = 10.2\text{Hz}, J = 17.0\text{Hz}\), 1H, CH=CH_2), 5.48 (br s, 1H, NH), 5.11 (ddt, \(J = 1.4\text{Hz}, J = 1.6\text{Hz}, J = 17.0\text{Hz}\), 1H, CH=CH_2), 5.05 (ddt, \(J = 1.3\text{Hz}, J = 1.4\text{Hz}, J = 10.2\text{Hz}\), 1H, CH=CH_2), 2.79-2.60 (m, 2H, CH_2-homo allyl chain), 2.54-2.45 (m, 3H, CH_2-homo allyl chain+ CH_2-cyclo pentane), 2.33-2.24 (m, 2H, CH_2-cyclo pentane), 2.19-2.13 (m, 1H, CH_2-cyclo pentane), 1.86-1.74 (m, 2H, CH_2-cyclo pentane), 1.73-1.62 (m, 2H, CH_2-cyclo pentane), 1.30 (s, 9H, CH_3-Bu).

\[\text{^13C NMR (δ, ppm)}\]

(\text{CDCl}_3, 100 MHz)

170.3 (Cq, C=O), 167.4 (Cq, C=O), 163.8 (Cq, C=N), 136.0 (CH=CH_2), 116.2 (CH=CH_2), 93.6 (Cq), 60.4 (Cq), 52.3 (Cq), 32.6 (CH_2), 32.2 (CH_2), 28.6 (CH_2), 28.3 (CH_3-Bu), 27.6 (CH_2), 25.4 (CH_2), 24.7 (CH_2).

IR (\(\nu, \text{cm}^{-1}\)) (CCl_4)  3580 (w), 3436 (w), 3389 (w), 3079 (w), 2963 (s), 2930 (s), 2874 (s), 1802 (s), 1720 (s), 1679 (s), 1648 (s), 1562 (s), 1513 (s), 1454 (s), 1392 (w), 1365 (s), 1334 (s), 1263 (s), 1221 (s), 1163 (s), 1096 (s), 1020 (s).

HRMS (El+, m/z) : Calculated: 351.1802  found: 351.1794.
[(1-Hex-5-en-1-ynyl-cyclopentanecarbonyl)-amino]-acetic acid ethyl ester

\[
\begin{align*}
\text{MF: } & C_{16}H_{23}O_3N \\
\text{MW: } & 277 \text{ g.mol}^{-1}
\end{align*}
\]

**Method:** See **general procedure 1.4** using (1 equiv., 0.30 mmol, 100 mg) of the corresponding isoxazolin-5-one.

**Purification:** Flash column chromatography (silica gel, 6:4 Et₂O: PE)/ \( R_f \) (7:3 PE:AcOEt): 0.59.

**Product:** Transparent oil.

**Isolated yield:** 70%.

**\(^1\)H NMR (δ, ppm)**

(CDCl\(_3\), 400 MHz)

- 7.35 (br s, 1H, NH), 5.85 (ddt, \( J = 6.3 \text{Hz}, J = 10.3 \text{Hz}, J = 16.4 \text{Hz} \)), 1H, CH=CH\(_2\)), 5.08 (ddt, 1H, \( J = 1.5 \text{Hz}, J = 1.6 \text{Hz}, J = 17.2 \text{Hz}, 1H, CH=CH\(_2\) \)), 5.03 (ddm, \( J = 1.0 \text{Hz}, J = 10.2 \text{Hz}, 1H, CH=CH\(_2\) \)), 4.21 (q, \( J = 7.2 \text{Hz}, 1H, OCH\(_2\)CH\(_3\) \)), 4.00 (d, \( J = 5.3 \text{Hz}, 2H, CH\(_2\)NH \)), 2.35-2.25 (m, 4H, CH\(_2\)-homo allyl chain), 2.17-2.10 (m, 2H, CH\(_2\)-cyclo pentane), 1.92-1.86 (m, 2H, CH\(_2\)-cyclo pentane), 1.82-1.76 (m, 4H, CH\(_2\)-cyclo pentane), 1.28 (t, \( J = 7.2 \text{Hz}, 3H, OCH\(_2\)CH\(_3\) \)).

**\(^{12}\)C NMR (δ, ppm)**

(CDCl\(_3\), 100 MHz)

- 174.2 (Cq, C=O), 169.8 (Cq, C=O), 137.1 (CH=CH\(_2\) ), 115.7 (CH=CH\(_2\) ), 84.9 (Cq, C \( \equiv \) C), 83.5 (Cq, C \( \equiv \) C), 61.4 (OCH\(_2\)CH\(_3\) ) , 49.0 (Cq-cyclo pentane), 41.8 (CH\(_2\)NH), 39.9 (CH\(_2\)-cyclo pentane), 33.0 (CH\(_2\)-homo allyl chain), 25.2 (CH\(_2\)-cyclo pentane), 18.5 (CH\(_2\)-homo allyl chain), 14.1 (OCH\(_2\)CH\(_3\) ).

**IR (ν, cm\(^{-1}\))** (CCl\(_4\))

- 3402 (w), 3079 (w), 2954 (w), 2870 (w), 1747 (s), 1679 (s), 1512 (m), 1444 (w), 1375 (w), 1350 (w), 1206 (s), 1023 (w).

**HRMS (El+, m/z)**: Calculated: 277.1678 found: 277.1677.
1-tert-Butyl-3-cyclopropyl-5-hydroxy-4-iodo-5-phenyl-1,5-dihydro-pyrrol-2-one

MF: C_{17}H_{20}O_{2}NI

MW = 397 g.mol\(^{-1}\)

**Method:**
See general procedure 1.5 using (1 equiv., 0.10 mmol, 25 mg) of the corresponding alkyne. No nucleophile added.

**Purification:**
Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ \(R_f\) (8:2 PE:AcOEt): 0.55.

**Product:**
White solid.

**Isolated yield:**
72%.

**\(^1\)H NMR** (\(\delta\), ppm) (CDCl\(_3\), 400 MHz)

- 7.77 (br s, 1H, CH-Ph)
- 7.40 (br s, 1H, CH-Ph)
- 7.32 (t, \(J = 5.1\)Hz, 2H, CH-Ph)
- 7.02 (br s, 1H, CH-Ph)
- 2.63 (s, 1H, OH)
- 1.64-1.59 (m, 1H, CH\(_2\)-cyclo propane)
- 1.50-1.45 (m, 1H, CH\(_2\)-cyclo propane)
- 1.42-1.37 (m, 1H, CH\(_2\)-cyclo propane)
- 1.32 (s, 9H, CH\(_3\)-tBu)
- 0.90-0.81 (m, 2H, CH\(_2\)-cyclo propane)

**\(^13\)C NMR** (\(\delta\), ppm) (CDCl\(_3\), 100 MHz)

- 167.2 (Cq, C=O)
- 141.9 (Cq)
- 139.1 (Cq)
- 128.2 (CH-Ph)
- 127.4 (CH-Ph)
- 125.4 (CH-Ph)
- 118.9 (Cq)
- 93.4 (Cq)
- 57.2 (Cq-tBu)
- 28.8 (CH\(_3\)-tBu)
- 12.3 (CH-cyclo propane)
- 6.7 (CH\(_2\)-cyclo propane)
- 6.2 (CH\(_2\)-cyclo propane)

**IR** (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))

- 3.584 (w)
- 3.065 (w)
- 3.009 (w)
- 2.965 (w)
- 2.928 (w)
- 1.695 (s)
- 1.642 (w)
- 1.451 (w)
- 1.346 (m)
- 1.270 (w)
- 1.209 (w)
- 1.170 (w)
- 1.130 (w)
- 1.014 (w)

**HRMS** (EI+, m/z):
Calculated: 397.0539  found: 397.0540.
X-ray structure for 1.25a

![X-ray structure image]

Table 1. Crystal data for 1.25a

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<td>Criterion</td>
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Fsqd
Hydrogen atoms
mixed
Parameters refined
193
Reflections / parameter
18
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0.0932
R1
0.0373
Weights a, b
0.0421, 0.0000
GoF
0.983
difference peak / hole (e Å⁻³)
0.699(0.093) / -0.664

Table 2. Atomic Coordinates (Å x 10⁴) and equivalent isotropic displacement parameters (Å² x 10³) for 1.25a

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U(eq) is defined as 1/3 the trace of the Uij tensor.

Table 3. Bond lengths (Å) and angles (deg) for 1.25a

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Table 4. Anisotropic displacement parameters (Å² x 10³) for 1.25a

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The anisotropic displacement factor exponent takes the form 2π² [h²a*²U(11) + ... + 2hka*b*U(12)]

Table 5. Hydrogen Coordinates (Å x 10⁴) and equivalent isotropic displacement parameters (Å² x 10³) for 1.25a

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**1-tert-Butyl-3-cyclopropyl-4-iodo-5-methoxy-5-phenyl-1,5-dihydro-pyrrol-2-one**

![Structural formula](image)

**MF:** C_{18}H_{22}O_{2}NI

**MW:** 411 g.mol^{-1}

**Method:** See **general procedure 1.5** using (1 equiv., 0.10 mmol, 26 mg) of the corresponding alkyne and (10 equiv., 1 mmol, 41 µL) of MeOH.

**Purification:** Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ R_f (8:2 PE:AcOEt): 0.86.

**Product:** Yellow solid.

**Isolated yield:** 76%.

**^1H NMR** (δ, ppm) (CDCl₃, 400 MHz)
- 7.78 (d, J = 7.4Hz, 1H, CH-Ph), 7.39 (t, J = 7.2Hz, 1H, CH-Ph), 7.30 (dt, J = 7.4Hz, J = 7.2Hz, 2H, CH-Ph), 6.96 (d, J = 7.2Hz, 1H, CH-Ph), 3.25 (s, 3H, OCH₃), 1.67 (tt, J = 5.3Hz, J = 8.7Hz, 1H, CH-cyclo propane), 1.50-1.40 (m, 2H, CH₂-cyclo propane), 1.28 (s, 9H, CH₃-tBu), 0.89-0.84 (m, 2H, CH₂-cyclo propane).

**^13C NMR** (δ, ppm) (CDCl₃, 100 MHz)
- 167.8 (Cq, C=O), 143.6 (Cq), 139.0 (Cq), 128.3 (CH-Ph), 128.2 (CH-Ph), 127.8 (CH-Ph), 127.7 (CH-Ph), 125.7 (CH-Ph), 115.8 (Cq), 97.5 (Cq), 56.9 (Cq, tBu), 49.8 (OCH₃), 28.1 (CH₃-tBu), 12.2 (CH-cyclo propane), 6.7 (CH₂-cyclo propane), 6.3(CH₂-cyclo propane)

**IR** (v, cm⁻¹) (CCl₄)
- 3064 (w), 3004 (w), 2965 (w), 2935 (w), 2833 (w), 1693 (s), 1643 (w), 1488 (w), 1394 (w), 1362 (w), 1337 (m), 1272 (w), 1247 (w), 1210 (w), 1117 (w), 1075 (w).

**HRMS** (EI+, m/z)
- calculated: 411.0696 found: 411.0702.
**5-Allyloxy-1-tert-butyl-3-cyclopropyl-4-iodo-5-phenyl-1,5-dihydro-pyrrol-2-one**  

![Chemical Structure](image)

**MF:** C_{20}H_{24}O_{2}NI

**MW:** 437 g.mol⁻¹

**Method:** See **general procedure 1.5** using (1 equiv., 0.10 mmol, 26 mg) of the corresponding alkyne and (10 equiv., 1 mmol, 70 µL) of allyl alcohol.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt)/R_f (9:1 PE:AcOEt): 0.74.

**Product:** Yellow solid.

**Isolated yield:** 72%.

**¹H NMR** (δ, ppm) (CDCl₃, 400 MHz)

7.86 (d, J = 7.8Hz, 1H, CH-Ph), 7.40 (t, J = 7.4Hz, 1H, CH-Ph), 7.34-7.28 (m, 2H, CH-Ph), 6.96 (d, J = 7.4Hz, 1H, CH-Ph), 6.04 (ddt, J = 5.1Hz, J = 10.4Hz, J = 17.2Hz, 1H, CH=CH₂), 5.46 (ddt, J = 1.5Hz, J = 1.7Hz, J = 17.2Hz, 1H, CH=CH₂), 5.26 (ddt, J = 1.5Hz, J = 1.7Hz, J = 10.4Hz, 1H, CH=CH₂), 3.93 (ddt, J = 1.7 Hz, J = 5.2 Hz, J = 12.4 Hz, 1H, OCH₂), 3.88 (ddt, J = 1.7 Hz, J = 5.2 Hz, J = 12.4 Hz, 1H, OCH₂), 1.67 (tt, J = 5.3Hz, J = 8.6Hz, 1H, CH-cyclopropane), 1.50-1.41 (m, 2H, CH₂-cyclopropane), 1.28 (s, 9H, CH₃-tBu), 0.91-0.82 (m, 2H, CH₂-cyclopropane).

**¹³C NMR** (δ, ppm) (CDCl₃, 100 MHz)

167.7 (Cq, C=O), 143.4 (Cq), 139.0 (Cq), 133.6 (CH=CH₂), 128.4 (CH-Ph), 128.2 (CH-Ph), 127.8 (CH-Ph), 127.8 (CH-Ph), 125.7 (CH-Ph), 116.8 (CH=CH₂), 115.9 (Cq), 96.9 (Cq), 63.1 (OCH₂), 57.0 (Cq, tBu), 28.1 (CH₃-tBu), 12.2 (CH₂-cyclopropane), 6.7 (CH₂-cyclopropane), 6.3 (CH₂-cyclopropane).

**IR** (ν, cm⁻¹) (CCl₄)

3081 (w), 3008 (w), 2967 (w), 2927 (w), 1693 (s), 1643 (w), 1551 (w), 1451 (w), 1395 (w), 1363 (w), 1337 (m), 1272 (w), 1248 (w), 1212 (w), 1125 (w), 1062 (w), 1023 (w).

**HRMS** (EI⁺, m/z): Calculated: 437.0852 found: 437.0837.
**1-Benzyl-5-hydroxy-4-iodo-5-phenyl-3-p-tolyl-1,5-dihydro-pyrrol-2-one**

**MF:** C_{24}H_{20}O_{2}NI

**MW:** 481 g·mol^{-1}

**Method:** See general procedure 1.5 using (1 equiv., 0.10 mmol, 36 mg) of the corresponding alkyne. No nucleophile added.

**Purification:** Flash column chromatography (silica gel, 95:5 PE:AcOEt)/ R_{f} (8:2 PE: AcOEt): 0.54.

**Product:** Yellow solid.

**Isolated yield:** 68%.

**{^1}H NMR (δ, ppm) (CDCl₃, 400 MHz)**

- 7.68 (d, J = 8.2Hz, 2H, CH-tolyl), 7.39-7.37 (m, 5H, CH-Ph), 7.27-7.26 (m, 5H, CH-Ph), 7.22 (d, J = 8.2Hz, 2H, CH-tolyl), 4.82 (d, J = 15.0Hz, 1H, CH₂N), 4.01 (d, J = 15.0Hz, 1H, CH₂N), 2.42 (s, 1H, OH), 2.39 (s, 3H, CH₃).

**{^{13}}C NMR (δ, ppm) (CDCl₃, 100 MHz)**

- 166.8 (Cq, C=O), 140.6 (Cq), 139.5 (Cq), 137.9 (Cq), 136.4 (Cq), 129.0 (CH-Ar), 128.9 (CH-Ar), 128.9 (CH-Ar), 128.7 (CH-Ar), 128.4 (CH-Ar), 127.5 (Cq), 127.3 (CH-Ar), 126.6 (CH-Ar), 120.0 (Cq), 93.2 (Cq), 44.6 (CH₂N), 21.5 (CH₃).

**IR (ν, cm⁻¹) (CCl₄)**

- 3577 (w), 3065 (w), 3033 (w), 2926 (w), 2859 (w), 1708 (s), 1622 (w), 1551 (w), 1498 (w), 1450 (w), 1385 (w), 1335 (w), 1280 (w), 1166 (w), 1073 (w), 1047 (w).

**HRMS (EI+, m/z):** calculated: 481.0539 found: 481.0527.
1-Benzy1-5-but-3-enyl-3-cyclopropyl-4-iodo-5-methoxy-1,5-dihydro-pyrrol-2-one

MF: C_{19}H_{22}O_2NI

MW = 423 g.mol⁻¹

Method : See general procedure 1.5 using (1 equiv., 0.10 mmol, 27 mg) of the corresponding alkyne and (10 equiv., 1 mmol, 41µL) of MeOH.

Purification : Flash column chromatography (silica gel, 95:5 PE:AcOEt) / R_f (95:5 PE:AcOEt): 0.54.

Product : Yellow oil.

Isolated yield : 70%.

¹H NMR (δ, ppm) (CDCl₃, 400 MHz) 7.36-7.30 (m, 4H, CH-Ph), 7.24-7.20 (m, 1H, CH-Ph), 5.77 (ddt, J = 6.2Hz, J = 10.2Hz, J = 16.9Hz, 1H, CH=CH₂), 4.99 (ddm, J = 1.2Hz, J = 16.9Hz, 1H, CH=CH₂), 4.94 (ddm, J = 1.2Hz, J = 10.2 Hz, 1H, CH=CH₂), 4.58 (d, J = 16.2Hz, 1H, CH₂N), 4.53 (d, J = 16.2Hz, 1H, CH₂N), 3.09 (s, 3H, OCH₃), 2.06-1.90 (m, 4H, CH₂-homo allyl chain), 1.78-1.71 (m, 1H, CH-cyclo propane), 1.60-1.54 (m, 2H, CH₂-cyclo propane), 0.90-0.83 (m, 2H, CH₂-cyclo propane).

¹³C NMR (δ, ppm) (CDCl₃, 100 MHz) 156.5 (Cq, C=O), 144.4 (Cq), 140.5 (Cq), 137.4 (Cq=CH₂), 128.2 (CH-Ph), 127.4 (CH-Ph), 126.3 (CH-Ph), 114.9(CH=CH₂), 111.2 (Cq), 106.2 (Cq), 50.5 (CH₂N), 49.5 (OCH₃), 36.1 (CH₂-homo allyl chain), 27.0 (CH₂-homo allyl chain), 12.3 (CH₂-cyclo propane), 7.1 (CH₂-cyclo propane), 6.5 (CH₂-cyclo propane).

IR (ν, cm⁻¹) (CCl₄) 3069 (w), 3012 (w), 2958 (w), 2933 (w), 2360 (w), 2337 (w), 1688 (s), 1642 (w), 1618 (w), 1496 (w), 1451 (w), 1354 (w), 1327 (w), 1327 (w), 1299 (w), 1268 (w), 1170 (w), 1130 (w), 1037 (w), 987 (w).

HRMS (EI+, m/z) : calculated: 423.696    found: 423.0677.
**6-tert-Butyl-4-cyclopropyl-3-methylene-6a-phenyl-2,3,6,6a-tetrahydrofuro[2,3-b]pyrrol-5-one**

MF: C_{20}H_{23}O_{2}N

MW = 309 g.mol^{-1}

**Method:**

To a round bottom flask at room temperature under an argon atmosphere charged with 5-allyloxy-1-tert-butyl-3-cyclopropyl-4-iodo-5-phenyl-1,5-dihydropyrrol-2-one (1 equiv., 0.07 mmol, 31 mg), CH_{3}CN (850 µL) and THF (280 µL) was added Et_{3}N (1.5 equiv., 0.11 mmol, 15 µL) and then Pd(PPh_{3})_{4} (0.10 equiv., 0.007 mmol, 8 mg). The reaction was allowed to heat up to reflux (T < 82 °C) and to stir at this temperature overnight. Upon completion (TLC), the volatiles were evaporated. Purification by flash column chromatography afforded the title compound in the stated yield.


**Purification:** Flash column chromatography (silica gel, 95:5 PE AcOEt)/R_{f} (9:1 PE:AcOEt): 0.38.

**Product:** White solid.

**Isolated yield:** 51%.

**{H NMR (δ, ppm)} (CDCl_{3}, 400 MHz)**

7.35-7.27 (m, 5H, CH-Ph), 5.36 (t, J = 2.2Hz, 1H, C=CH_{2}), 5.10 (t, J = 2.2Hz, 1H, C=CH_{2}), 4.92 (dt, J = 2.2Hz, J = 13.1Hz, 1H, OCH_{2}), 4.78 (dt, J = 2.2Hz, J = 13.1Hz, 1H, OCH_{2}), 1.81-1.74 (m, 1H, CH-cyclopropane), 1.53-1.48 (m, 1H, CH-cyclopropane), 1.26-1.22 (m, 1H, CH-cyclopropane), 1.24 (s, 9H, CH_{2}-^1Bu), 0.93-0.86 (m, 2H, CH_{2}-cyclopropane).

**{C NMR (δ, ppm)} (CDCl_{3}, 100 MHz)**

171.9 (Cq, C=O), 153.3 (Cq), 137.5 (Cq), 136.1 (Cq), 132.0 (Cq), 128.3 (CH-Ph), 127.9 (CH-Ph), 126.8 (CH-Ph), 109.2 (C=CH_{2}), 100.3 (Cq), 75.8 (OCH_{2}), 56.0 (Cq, ^1Bu), 28.2 (CH_{2}-^1Bu), 8.4 (CH-cyclopropane), 7.6 (CH_{2}-cyclopropane), 5.3 (CH_{2}-cyclopropane).

**IR (ν, cm^{-1}) (CCl_{4})**

3064 (w), 2964 (w), 2928 (w), 2869 (w), 1694 (s), 1597 (w), 1485 (w), 1451 (w), 1392 (w), 1363 (w), 1333 (w), 1293 (w), 1263 (w), 1215 (w), 1142 (w), 1025 (w).

**HRMS (El+, m/z):** Calculated: 309.1729 found: 309.1742.

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B.3.2 Chapter 3: Synthesis of functionalized oxazolones by a sequence of copper (II) and gold (I)-catalyzed transformations

Only experiments performed by this author are presented in this section. The substrates reported in this section are marked with a red star (*).

B.3.2.1 Synthesis of ynamides via a Cu-mediated coupling reaction:

The synthetic route employed for the synthesis of ynamides is depicted in the scheme below:

Ynamides synthesized:

3.15a, $R^1 = \text{Ph}, R^2 = \text{Ph}$

3.15b, $R^1 = \text{Ph}, R^2 = 4$-F-Ph

3.15c, $R^1 = \text{Ph}, R^2 = 4$-Cl-Ph

3.15d, $R^1 = \text{Ph}, R^2 = 4$-Br-Ph

3.15e, $R^1 = \text{Ph}, R^2 = 2$-OMe-4$\text{-}$OMe-Ph

3.15f, $R^1 = \text{Ph}, R^2 = \text{Bn}$

3.15g, $R^1 = \text{Ph}, R^2 = \text{CH}_2\text{CO}_2\text{Et}$

3.15h, $R^1 = \text{Ph}, R^2 = \text{CH}$(-$\text{S}$)-Me-\text{CO}_2\text{Me}$

3.15i, $R^1 = \text{tBu}, R^2 = \text{Ph}$

3.15j, $R^1 = n$-C$_5$H$_{11}$, $R^2 = \text{Ph}$

3.15k, $R^1 = n$-C$_5$H$_{11}$, $R^2 = \text{Bn}$

3.15l, $R^1 = \text{Ph}$, $R^2 = \text{Ph}$

3.15m, $R^1 = 4$-Cl-Ph, $R^2 = \text{Ph}$

3.15n, $R^1 = \text{Ph}$, $R^2 = \text{CH}_2\text{CO}_2\text{Et}$

3.15o, $R^1 = \text{CH}_2\text{OAc}, R^2 = \text{Ph}$

3.15p, $R^1 = \text{CH}_2\text{OAc}, R^2 = \text{2-Napht}$

3.15q, $R^1 = \text{CH}_2\text{OAc}, R^2 = \text{Bn}$

3.15r, $R^1 = \text{CH}_2\text{OAc}, R^2 = \text{CH}_2\text{CO}_2\text{Et}$

3.15s, $R^1 = \text{CH}_2\text{CH}_2\text{OHOTIPS}, R^2 = \text{Ph}$

3.15t, $R^1 = \text{CH}_2\text{CH}_2\text{OHOTIPS}, R^2 = \text{Bn}$

3.15u, $R^1 = \text{Ph}$, $R^2 = \text{Ph}$

3.15v, $R^1 = \text{Bn}$, $R^2 = \text{Bn}$

3.15w, $R^1 = \text{Ph}$, $R^2 = \text{Ph}$

3.15x, $R^1 = \text{Bn}$, $R^2 = \text{Bn}$

3.15y, $R^1 = \text{Ph}$, $R^2 = \text{Ph}$

Scheme B.3.2.1: Synthesis of ynamides.
General Procedure 3.1, *NBS promoted bromination of terminal alkynes*: To a round bottom flask charged with the alkyne (1 equiv.) and acetone (0.15 M) at 0 °C. was added *N*-bromo succinimide (1.2 equiv) and AgNO$_3$ (0.1 equiv.). The reaction mixture was allowed to warm up to room temperature, the flask was covered with an aluminium foil to avoid light exposure and the reaction was allowed to stir at room temperature until completion (TLC). The reaction mixture was then diluted in H$_2$O, extracted with Et$_2$O (3x), dried (MgSO$_4$) and concentrated under reduced pressure. Purification of the crude mixture by flash column chromatography afforded the bromoalkynes in the stated yields.

_Synthesis of tBu-carbamates (Boc protection of amines):_

**General Procedure 3.2.A** 296: To a round bottom flask charged with aniline (1 equiv.) and toluene (1M) at room temperature was added Boc$_2$O (1.5 equiv.). The resulting mixture was heated up to reflux (110 °C) and allowed to stir at this temperature until completion (TLC). The reaction usually takes one to two hours. The mixture is then concentrated under reduced pressure and the solid obtained recrystallized from an adequate solvent mixture.

**General Procedure 3.2.B** 297: To a round bottom flask charged with amine (1 equiv.) and EtOH (1M) at room temperature was added Boc$_2$O (1.1 equiv) and allowed to stir at this temperature. Upon completion (TLC), the reaction mixture is concentrated under reduced pressure and the resulting solid recrystallized from an adequate solvent mixture. The reaction usually takes about 10 min.

**General Procedure 3.2.C** 298: To a round bottom flask charged with amine (1 equiv.) and MeCN (0.2 M) at 0 °C was added NaHCO$_3$ (3 equiv.) and Boc$_2$O (1 equiv.). The reaction was allowed to warm up and to stir at room temperature until completion (TLC). The reaction was generally left for 24h. The solution is then concentrated under reduced pressure. The residue is then diluted in water, extracted with DCM (3x), washed with brine (1x), dried (MgSO$_4$) and concentrated under reduced pressure. Purification by flash column chromatography afforded the expected pure product.

General Procedure 3.2.D: To a round bottom flask charged with amine (1 equiv.) and MeOH (0.75 M) were added Et$_3$N (1.5 equiv.) and Boc$_2$O (1.05 equiv.). The reaction mixture was allowed to stir at room temperature until completion (TLC). The reaction was generally left for 24h. The solution is then concentrated under reduced pressure. The residue is then diluted in water, extracted with DCM (3x), washed with brine (1x), dried (MgSO$_4$) and concentrated under reduced pressure. The expected product is generally pure and used directly in the following step.

General Procedure 3.3, Cu-mediated coupling reaction between carbamates and bromoalkynes: To a round bottom flask at room temperature, under an argon flow and closed with a rubber septum were sequentially added: bromoalkyne (1 equiv.), anhydrous toluene (0.33 M), carbamate (1.2 equiv.) K$_3$PO$_4$ (2.4 equiv.), CuSO$_4$.5H$_2$O (0.2 equiv.) and 1,10-phenanthroline (0.4 equiv.). The argon flow was removed and the flask was heated up to 80 °C for 16 to 72 h, while being periodically monitored by TLC. Upon completion, the reaction was left to cool down to room temperature, diluted in AcOEt, filtered through a short pad of celite and concentrated under reduced pressure. Purification of the crude mixture by flash column chromatography afforded the title compounds in the stated yields.

(4-Fluoro-phenyl)-phenylethynyl-carbamic acid tert-butyl ester 3.15b

\[
\begin{align*}
\text{MF: } & \text{C}_{11}\text{H}_{14}\text{O}_2\text{NF} \\
\text{MW = } & 311 \text{ g.mol}^{-1}
\end{align*}
\]

Method: See general procedure 3.3 using (1 equiv., 2 mmol, 362 mg) of bromoethynyl-benzene and (1.2 equiv., 2.4 mmol, 506 mg) of (4-fluoro-phenyl)-carbamic acid tert-butyl ester.

**Purification:** Flash column chromatography (silica gel, 95:5 PE:EtO).

**Product:** Yellow oil that crystallizes in the fridge to give a transparent solid.

**Conversion:** 100%.

**Isolated yield:** 65%.

**1H NMR** ($\delta$, ppm) (CDCl$_3$, 400 MHz)
- 7.49 (dd, $J_{HH} = 8.9$Hz, $J_{HF} = 4.8$Hz, 2H, CH-Ar), 7.39 (dd, $J_{HH} = 1.6$Hz, $J_{HF} = 8.0$Hz, 2H, CH-Ph), 7.32-7.26 (m, 3H, CH-Ph), 7.08 (t, $J_{HH} = 8.9$Hz, 2H, CH-Ar), 1.57 (s, 9H, CH$_3$-Bu).

**13C NMR** ($\delta$, ppm) (CDCl$_3$, 100 MHz)
- 161.0 (d, $J_{CF} = 244.9$Hz, Cq, Ar), 152.9 (Cq, NC=O), 135.6 (d, $J_{CF} = 3.1$Hz, Cq, Ar), 130.9 (CH-Ph), 128.3 (CH-Ph), 127.5 (CH-Ph), 126.6 (d, $J_{CF} = 8.4$Hz, CH-Ar), 123.2 (Cq, Ph), 115.6 (d, $J_{CF} = 22.8$Hz, CH-Ar), 83.7 (Cq, Bu), 83.5 (Cq, C≡C), 70.1 (Cq, C≡C), 28.0 (CH$_3$-Bu).

**IR** ($\nu$, cm$^{-1}$) (CCl$_4$)
- 3058 (w), 2980 (m), 2933 (w), 2252 (s), 1875 (w), 1737 (s), 1599 (m), 1508 (s), 1453 (w), 1391 (w), 1364 (s), 1290 (s), 1238 (s), 1158 (s), 1096 (w), 1068 (w), 1009 (m).

**HRMS** (El+, m/z):
- Calculated: 311.1322  found: 311.1312.

<table>
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<tr>
<th>(2,4-Dimethoxy-phenyl)-phenylethynyl-carbamic acid tert-butyl ester</th>
<th>3.15e</th>
</tr>
</thead>
</table>

![Chemical Structure]

- **MF:** C$_{21}$H$_{23}$O$_4$N
- **MW:** 353 g.mol$^{-1}$

**Method:** See general procedure 3.3 using (1 equiv., 2 mmol, 362 mg) of bromoethynylbenzene and (1.2 equiv., 2.4 mmol, 607 mg) of (2,4-dimethoxy-phenyl)-carbamic acid tert-butyl ester.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:** Brown solid in foam form.

**Conversion:** 33%.
Isolated Yield: 22%.

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz) 7.37-7.35 (m, 2H, CH-Ph), 7.29-7.21 (m, 4H, CH-Ph + CH-Ar), 6.52-6.48 (m, 2H, CH-Ar), 3.86 (s, 3H, OCH$_3$), 3.82 (s, 3H, OCH$_3$), 1.51 (s, 9H, CH$_3$-Bu).

$^{13}$C NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz) 160.5 (Cq, NC=O), 155.4 (Cq, Ar), 153.4 (Cq, Ar), 130.7 (CH-Ph), 128.6 (CH-Ar), 128.0 (CH-Ph), 126.9 (CH-Ph), 123.7 (Cq, Ph), 121.8 (Cq, Ar), 104.2 (CH-Ar), 99.5 (CH-Ar), 84.7 (Cq, 'Bu), 82.5 (Cq, C=CC), 67.9 (Cq, C=CC), 55.7 (OCH$_3$), 55.4 (OCH$_3$), 27.8 (CH$_3$-Bu).

IR ($\nu$, cm$^{-1}$) (CCl$_4$) 3079 (w), 2977 (s), 2936 (s), 2838 (s), 2251 (s), 1734 (s), 1607 (s), 1462 (s), 1420 (w), 1367 (s), 1321 (s), 1295 (s), 1254 (s), 1213 (s), 1162 (s), 1130 (s), 1041 (s), 1005 (m), 972 (w).

HRMS (El+, m/z): calculated: 353.1627 found: 353.1626.

Benzyl-phenylethynyl-carbamic acid tert-butyl ester 3.15f

MF: C$_{20}$H$_{21}$O$_2$N

MW = 307 g.mol$^{-1}$

Method: See general procedure 3.3, using (1 equiv., 2 mmol, 362 mg) of bromoethynyl-benzene and (1.2 equiv., 2.4 mmol, 497 mg) of benzyl-carbamic acid tert-butyl ester.

Purification: Flash column chromatography (silica gel, 95:5 PE:Et$_2$O)/ R$_f$ (95:5 PE: Et$_2$O): 0.56.

Product: Yellow oil that crystallizes in the fridge to give a white solid.

Conversion: < 100% (difficult to estimate).

Isolated Yield: 62%.

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz) 7.40-7.24 (m, 10H, CH-Ph), 4.67 (s, 2H, CH$_2$N), 1.53 (s, 9H, CH$_3$-Bu).
\[ ^{13}C \text{NMR} (\delta, \text{ppm}) \]

\[
\begin{align*}
(CDCl_3, 100 \text{ MHz}) & \quad 153.7 \text{ (Cq, NC=O)}, 136.3 \text{ (Cq, Ph)}, 130.5 \text{ (CH-Ph)}, 128.4 \text{ (CH-Ph)}, 128.2 \text{ (CH-Ph)}, 128.0 \text{ (CH-Ph)}, 127.8 \text{ (CH-Ph)}, 127.0 \text{ (CH-Ph)}, 123.5 \text{ (Cq, Ph)}, 84.0 \text{ (Cq, C=)} \text{C, 82.5 (Cq, } \text{^3} \text{Bu)}, 70.9 \text{ (Cq, C=)} \text{C, 53.0 (CH_2N), 27.9 (CH_3^3\text{Bu).}}
\end{align*}
\]

\[ \text{IR} (\nu, \text{cm}^{-1}) (CCl_4) \]

\[
3063 \text{ (w), 3033 (w), 2979 (w), 2934 (w), 2243 (m), 1722 (s), 1496 (w), 1447 (w), 1393 (m), 1366 (m), 1302 (s), 1244 (m), 1159 (s), 1072 (w), 1026 (w), 995 (w).}
\]

\[ \text{HRMS (El+, m/z) : calculated: 307.1572 found: 307.1579.} \]

---

**Hept-1-ynyl-phenyl-carbamic acid tert-butyl ester**

\[
\begin{align*}
\text{MF: C}_{16}\text{H}_{25}\text{O}_2\text{N} \\
\text{MW} = 287 \text{ g.mol}^{-1}
\end{align*}
\]

**Method:**

See general procedure 3.3, using (1 equiv., 2 mmol, 350 mg) of 1-bromo-hept-1-yne and (1.2 equiv., 2.4 mmol, 464 mg) of phenyl-carbamic acid tert-butyl ester.

**Purification:**

Flash column chromatography (silica gel, 95:5 PE: EtO).

**Product:**

Yellow oil.

**Conversion:**

76%.

**Isolated Yield:**

69%.

\[ ^1\text{H NMR} (\delta, \text{ppm}) \]

\[
(CDCl_3, 400 \text{ MHz}) \quad 7.48 \text{ (d, } J = 7.7\text{Hz, 2H, CH-Ph), 7.35 \text{ (t, } J = 7.7\text{Hz, 2H, CH-Ph), 7.20 \text{ (t, } J = 7.7\text{Hz, 1H, CH-Ph), 2.33 \text{ (t, } J = 7.0\text{Hz, 2H, CH}_2\text{-pentyl chain), 1.59-1.54 \text{ (m, 2H, CH}_2\text{-pentyl chain), 1.63 \text{ (s, 9H, CH}_3^3\text{-Bu), 1.45-1.28 \text{ (m, 4H, CH}_2\text{-pentyl chain), 0.91 \text{ (t, } J = 7.2\text{Hz, 3H, CH}_2\text{-pentyl chain),}}}
\]

\[ ^{13}\text{C NMR} (\delta, \text{ppm}) \]

\[
(CDCl_3, 100 \text{ MHz}) \quad 153.5 \text{ (Cq, NC=O), 140.2 \text{ (Cq, Ph), 128.5 \text{ (CH-Ph), 126.0 \text{ (CH-Ph), 124.4 \text{ (CH-Ph), 82.7 \text{ (Cq, } \text{^3} \text{Bu), 74.2 \text{ (Cq, C=) C, 69.2 \text{ (Cq, C=) C, 30.9 \text{ (CH}_2\text{-pentyl chain), 28.5 \text{ (CH}_2\text{-pentyl chain), 27.9 \text{ (CH}_3^3\text{-Bu), 22.1 \text{ (CH}_2\text{-pentyl chain), 18.4 \text{ (CH}_2\text{-pentyl chain), 13.9 \text{ (CH}_2\text{-pentyl chain).}}}
\]

\[ \text{IR} (\nu, \text{cm}^{-1}) (CCl_4) \]

\[
3068 \text{ (w), 3039 (w), 2959 (s), 2932 (s), 2862 (m), 2267 (m), 1731 (s), 1594 (m), 1495 (m), 1460 (m), 1389 (w), 1366 (m), 1296 (s), 1256 (s), 1215 (m), 1161 (s), 1073 (w), 1049 (w), 1019 (w).}
\]
HRMS (El+, m/z) : calculated: 287.1885 found: 287.1889.

Benzyl-hept-1-ynyl-carbamic acid tert-butyl ester  

\[
\begin{array}{c}
\text{O} \\
\text{N} \\
\text{O} \\
\end{array}
\]

MF: C\(_{19}\)H\(_{27}\)O\(_2\)N  
MW = 301 g.mol\(^{-1}\)

Method : See general procedure 3.3 using (1 equiv, 2 mmol, 350 mg) of 1-bromo-hept-1-yne and (1.2 equiv., 2.4 mmol, 497 mg) of benzyl-carbamic acid tert-butyl ester.

Purification : Flash column chromatography (silica gel, PE:Et\(_2\)O)/ \(R_f\) (95:5 PE:Et\(_2\)O): 0.63.

Product : Transparent oil.

Conversion < 100% (difficult to estimate).

Isolated Yield : 69%.

\(^1\)H NMR (\(\delta\), ppm) (CDCl\(_3\), 400 MHz)  
7.39-7.31 (m, 5H, CH-Ph), 4.60 (s, 2H, CH\(_2\)N), 2.29 (t, \(J = 6.4\)Hz, 2H, CH\(_2\)-pentyl chain), 1.54 (s, 9H, CH\(_3\)-tBu), 1.57-1.46 (m, 2H, CH\(_2\)-pentyl chain), 1.41-1.32 (m, 4H, CH\(_2\)-pentyl chain), 0.94 (t, \(J = 7.1\)Hz, 3H, CH\(_3\)-pentyl chain).

\(^{13}\)C NMR (\(\delta\), ppm) (CDCl\(_3\), 100 MHz)  
154.5 (Cq, NC=O), 136.8 (Cq, Ph), 128.3 (CH-Ph), 128.0 (CH-Ph), 127.5 (CH-Ph), 81.9 (Cq, \(^t\)Bu), 74.5 (Cq, C\(\equiv\)C), 69.6 (Cq, C\(\equiv\)C), 52.9 (CH\(_2\)N), 30.8 (CH\(_2\)-pentyl chain), 28.6 (CH\(_2\)-pentyl chain), 28.0 (CH\(_3\)^tBu), 22.1 (CH\(_2\)-pentyl chain), 18.3 (CH\(_2\)-pentyl chain), 13.9 (CH\(_3\)-pentyl chain).

IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))  
3066 (w), 3032 (w), 2958 (s), 2932 (s), 2862 (m), 2260 (m), 1718 (s), 1455 (m), 1389 (s), 1367 (s), 1295 (s), 1256 (s), 1230 (s), 1164 (s), 1077 (w), 1028 (w), 1009 (w), 996 (w).

HRMS (El+, m/z) : calculated: 301.2042 found: 301.2033.
Acetic acid 3-(tert-butoxycarbonyl-phenyl-amino)-prop-2-ynyl ester

Method: See general procedure 3.3 using (1 equiv., 2 mmol, 354 mg) of 4-bromo-but-3-ynoic acid methyl ester and (1.2 equiv., 2.4 mmol, 464 mg) of phenyl-carbamic acid tert-butyl ester.

Purification: Flash column chromatography (silica gel, 9:1 PE:Et₂O)/ Rf (9:1 PE:Et₂O): 0.34.

Product: Transparent oil that crystallizes in the fridge to give a white solid.

Conversion: 91%.

Isolated Yield: 52%.

¹H NMR (δ, ppm) (CDCl₃, 400 MHz) 7.45 (d, J = 7.7Hz, 2H, CH-Ph), 7.39 (t, J = 7.7Hz, 2H, CH-Ph), 7.26 (t, J = 7.7Hz, CH-Ph), 4.88 (s, 2H, OCH₂), 2.10 (s, 3H, COCH₃), 1.55 (s, 9H, CH₃-tBu).

¹³C NMR (δ, ppm) (CDCl₃, 100 MHz) 170.1 (Cq, C=O), 152.7 (Cq, C=O), 139.2 (Cq, Ph), 128.6 (CH-Ph), 126.7 (CH-Ph), 124.6 (CH-Ph), 83.5 (Cq, tBu), 80.8 (Cq, C≡C), 64.4 (Cq, C≡C), 52.6 (OCH₂), 27.7 (CH₃-tBu), 20.6 (CH₃CO₂).

IR (ν, cm⁻¹) (CCl₄) 3068 (w), 3039 (w), 2980 (m), 2936 (m), 2262 (m), 1742 (s), 1595 (w), 1496 (m), 1454 (m), 1369 (m), 1287 (s), 1224 (s), 1158 (s), 1075 (w), 1048 (m), 1020 (s).

HRMS (El+, m/z): calculated: 289.1314 found: 289.1322.

MF: C₁₆H₁₉O₄N

MW = 289 g.mol⁻¹
Acetic acid 3-(benzyl-tert-butoxycarbonyl-amino)-prop-2-ynyl ester

MF: C₁₇H₂₁O₄N

MW = 303 g.mol⁻¹

Method: See general procedure 3.3 using (1 equiv., 2 mmol, 354 mg) of 4-bromo-but-3-ynoic acid methyl ester and (1.2 equiv., 2.4 mmol, 497 mg) of benzyl-carbamic acid tert-butyl ester.

Purification: Flash column chromatography (silica gel, 95:5 toluene: Et₂O)/ Rf (95:5 toluene: Et₂O): 0.40.

Product: Transparent oil.

Conversion: 95%.

Isolated Yield: 14% (difficult to isolate).

¹H NMR (δ, ppm) (CDCl₃, 400 MHz) 7.39-7.31 (m, 5H, CH-Ph), 4.83 (s, 2H, CH₂O), 4.61 (s, 2H, CH₂N), 2.11 (s, 3H, CH₃CO), 1.53 (s, 9H, CH₃-tBu).

¹³C NMR (δ, ppm) (CDCl₃, 100 MHz) 170.3 (Cq, C=O), 153.8 (Cq, C=O), 136.2 (Cq, Ph), 128.4 (CH-Ph), 128.1 (CH-Ph), 127.8 (CH-Ph), 82.9 (Cq, tBu), 81.4 (Cq, C≡C), 65.2 (Cq, C≡C), 53.1 (CH₂N), 52.8 (OCH₂), 27.9 (CH₃-tBu), 20.8 (CH₃C=O).

IR (ν, cm⁻¹) (CCl₄) 3459 (w), 3066 (w), 3033 (m), 2980 (s), 2938 (s), 2259 (s), 1845 (w), 1726 (s), 1608 (w), 1497 (m), 1477 (m), 1436 (s), 1400 (s), 1365 (s), 1294 (s), 1224 (s), 1161 (s), 1076 (w), 1020 (s), 957 (s), 903 (w).

HRMS (EI+, m/z): Calculated: 303.1421  found: 303.1469.
[(3-Acetoxy-prop-1-ynyl)-tert-butoxycarbonyl-amino]-acetic acid ethyl ester

\[
\begin{align*}
\text{MF: } & C_{14}H_{21}O_6N \\
\text{MW: } & 299 \text{ g.mol}^{-1}
\end{align*}
\]

**Method:** See general procedure 3.3 using (1 equiv., 2 mmol, 354 mg) of 4-bromo-but-3-ynoic acid methyl ester and (1.2 equiv., 2.4 mmol, 488 mg) of tert-butoxycarbonylamino-acetic acid ethyl ester.

**Purification:** Flash column chromatography (silica gel, 9:1 toluene:Et\textsubscript{2}O)/ \textit{R} \textsubscript{f} (9:1 toluene:Et\textsubscript{2}O): 0.30.

**Product:** Yellow oil that crystallizes in the fridge to give a transparent solid.

**Conversion:** 95%.

**Isolated Yield:** 48%.

**\textsuperscript{1}H NMR** (\(\delta, \text{ ppm}\)) (CDCl\textsubscript{3}, 400 MHz) 4.73 (s, 2H, CH\textsubscript{2}O), 4.16 (q, \(J = 7.0\text{Hz}\), 2H, OCH\textsubscript{2}CH\textsubscript{3}), 4.00 (s, 2H, CH\textsubscript{2}N), 2.00 (s, 3H, CH\textsubscript{3}C=O), 1.43 (s, 9H, CH\textsubscript{3}t-Bu), 1.21 (t, \(J = 7.0\text{Hz}\), 3H, OCH\textsubscript{2}C\textsubscript{H}\textsubscript{3}).

**\textsuperscript{13}C NMR** (\(\delta, \text{ ppm}\)) (CDCl\textsubscript{3}, 100 MHz) 170.1 (Cq, C=O), 167.6 (Cq, C=O), 153.5 (Cq, C=O), 152.9 (Cq, C=O \textit{rotamer}), 83.2 (Cq, t-Bu), 81.4 (Cq, C \textit{\equiv}\text{C rotamer}), 80.7 (Cq, C \textit{\equiv}\text{C}), 64.3 (Cq, C \textit{\equiv}\text{C}), 63.7 (Cq, C \textit{\equiv}\text{C rotamer}), 61.3 (OCH\textsubscript{2}CH\textsubscript{3}), 52.7 (OCH\textsubscript{2}), 50.6 (CH\textsubscript{2}N), 27.7 (CH\textsubscript{3}t-Bu), 20.6 (C=OCH\textsubscript{2}), 14.0 (OCH\textsubscript{2}CH\textsubscript{3}).

**IR** (\(\nu, \text{ cm}^{-1}\)) (CCl\textsubscript{4}) 2982 (m), 2939 (m), 2263 (m), 1738 (s), 1473 (w), 1404.8 (m), 1370 (m), 1349 (s), 1318 (s), 1220 (s), 1158 (s), 1099 (w), 1026.8 (s), 960 (m).

**HRMS** (EI+, m/z): Calculated: 299.1369   found: 299.1374.
**Method:** See general procedure 3.3 using (1 equiv., 1.95mmol, 624 mg) of (5-bromo-pent-4-ynyl)-triisopropyl-silane and (1.2 equiv., 2.34 mmol, 485 mg) of benzyl-carbamic acid tert-butyl ester.

**Purification:** Flash column chromatography (silica gel, 95:5 toluene:Et₂O)/ Rₚ (toluene:Et₂O): 0.55.

**Product:** Pale yellow oil.

**Conversion:** 77%.

**Isolated Yield:** 72%.

**¹H NMR** (δ, ppm) (CDCl₃, 400 MHz)

<table>
<thead>
<tr>
<th>Chemical Shift (ppm)</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>7.38-7.31</td>
<td>(m, 5H, CH-Ph)</td>
</tr>
<tr>
<td>4.59</td>
<td>(s, 2H, CH₂N)</td>
</tr>
<tr>
<td>3.77</td>
<td>(t, J = 6.0Hz, 2H, CH₂O)</td>
</tr>
<tr>
<td>2.42</td>
<td>(t, J = 6.0Hz, 2H, CH₂)</td>
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<tr>
<td>1.75</td>
<td>(t, J = 6.0Hz, 2H, CH₂)</td>
</tr>
<tr>
<td>1.54</td>
<td>(s, 9H, CH₃-tBu)</td>
</tr>
<tr>
<td>1.12-1.10</td>
<td>(d, J = 3.7Hz, 3H, CH-Pr)</td>
</tr>
<tr>
<td>1.11</td>
<td>(d, J = 3.7Hz, 18H, CH₃-iPr)</td>
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**¹³C NMR** (δ, ppm) (CDCl₃, 100 MHz)

<table>
<thead>
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<th>Chemical Shift (ppm)</th>
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<td>154.5</td>
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</tr>
<tr>
<td>136.8</td>
<td>(Cq, Ph)</td>
</tr>
<tr>
<td>128.3</td>
<td>(CH-Ph)</td>
</tr>
<tr>
<td>127.9</td>
<td>(CH-Ph)</td>
</tr>
<tr>
<td>127.5</td>
<td>(CH-Ph)</td>
</tr>
<tr>
<td>81.9</td>
<td>(Cq, tBu)</td>
</tr>
<tr>
<td>74.6</td>
<td>(Cq, C≡C)</td>
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<tr>
<td>69.2</td>
<td>(Cq, C≡C)</td>
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<tr>
<td>61.8</td>
<td>(CH₂O)</td>
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<tr>
<td>53.0</td>
<td>(CH₂N)</td>
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<tr>
<td>32.3</td>
<td>(CH₂)</td>
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<tr>
<td>27.9</td>
<td>(CH₃-tBu)</td>
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<td>17.9</td>
<td>(CH₃-iPr)</td>
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<tr>
<td>14.8</td>
<td>(CH₂)</td>
</tr>
<tr>
<td>11.9</td>
<td>(CH-iPr)</td>
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**IR** (ν, cm⁻¹) (CCl₄)

<table>
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<th>Frequency (cm⁻¹)</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>3066</td>
<td>(w)</td>
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<tr>
<td>3032</td>
<td>(w)</td>
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<tr>
<td>2943</td>
<td>(s)</td>
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<tr>
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</tr>
<tr>
<td>2751</td>
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<td>(w)</td>
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<tr>
<td>918</td>
<td>(w)</td>
</tr>
<tr>
<td>901</td>
<td>(w)</td>
</tr>
</tbody>
</table>

**HRMS** (EI+, m/z):

- Calculated: 445.3012
- Found: 445.3023
**1-Bromoethyl-2-hept-1-ynyl-benzene**

**MF:** C_{15}H_{15}Br

**MW = 275 g·mol^{-1}

**Method:**

To a round bottom flask under an argon atmosphere at room temperature charged with 2-bromo-benzaldehyde (1 equiv., 5 mmol, 583 µL), 1-heptyne (1.2 equiv. 6.0 mmol, 786 µL) and Et$_3$N (0.25 M, 20 mL) was added PdCl$_2$(PPh$_3$)$_2$ (0.02 equiv, 0.1 mmol, 70 mg). The mixture was allowed to stir at this temperature for 5 min, then CuI (0.01 equiv., 0.05 equiv., 10 mg) was added and the resulting mixture was heated up to 50 °C and allowed to stir at this temperature overnight. Upon completion (TLC), the reaction was filtered through a short pad of celite and the volatiles evaporated under reduced pressure. Purification of the crude mixture by flash column chromatography (silica gel, 95:5 PE:EtO) afforded 2-heptynyl benzaldehyde (959 mg, 96%).

Next, a solution of 2-heptynylbenzaldehyde (1 equiv., 4.4 mmol, 880 mg) dissolved in anhydrous DCM (3.5 mL) was slowly added to a previously prepared solution of PPh$_3$ (4 equiv, 17.6 mmol, 4.61 g) and CBr$_4$ (2 equiv., 8.8 mmol, 2.92 g) in anhydrous DCM (12.5 mL) at 0°C. The reaction was allowed to stir at 0°C. Upon completion (TLC), the reaction was quenched with a saturated solution of NaHCO$_3$, extracted with DCM (3x), washed with brine (1x), dried (MgSO$_4$) and concentrated under reduced pressure. To the white solid obtained was added petrol ether and the solid was carefully triturated and left to stir for some minutes to ensure the best extraction of the dibromo vinyl compound from the insoluble triphenyl phosphine. The solution was then filtered and concentrated under reduced pressure again. Purification by flash column chromatography (silica gel, PE) of the crude, afforded the corresponding vinyldibromide (1.22 g, 78%) as a yellow oil.

Next, to a solution of 1-(2,2-dibromo-vinyl)-2-hept-1-ynyl-benzene (1 equiv., 3.34 mmol, 1.19 g) in distilled THF (0.1M, 35mL) under an argon atmosphere at -78°C was added LiHMDS (2 equiv., 1M solution in hexanes, 6.7 mmol, 6.7 mL) dropwise. The reaction was allowed to stir at -78°C. After the complete consumption of the starting material, the reaction was quenched with a saturated solution of NH$_4$Cl at -78°C and allowed to warm up to room temperature. The reation was extracted with EtO (3x), washed with brine (1x), dried (MgSO$_4$) and concentrated under reduced pressure. Filtration through a pad of silica gel (PE) afforded the title compound (903 mg, 98%).
Purification: See above/ Rf (PE): 0.48.

Product: Yellow oil.

Isolated yield: 73 % for three steps.

\[ \text{1H NMR} (\delta, \text{ppm}) \]
\[ (\text{CDCl}_3, 400 \text{ MHz}) \]
\[ 7.42 (d, J = 6.8 \text{ Hz}, 1 \text{H}, \text{CH}-\text{Ar}), 7.38 (d, J = 7.6 \text{ Hz}, 1 \text{H}, \text{CH}-\text{Ar}), 7.26-7.18 (m, 2\text{H}, \text{CH}-\text{Ar}), 2.47 (t, J = 6.8\text{Hz}, 2\text{H}, \text{CH}_2\text{-pentyl chain}), 1.68-1.61 (m, 2\text{H}, \text{CH}_2\text{-pentyl chain}), 1.54-1.46 (m, 2\text{H}, \text{CH}_2\text{-pentyl chain}), 1.43-1.34 (m, 2\text{H}, \text{CH}_2\text{-pentyl chain}). 0.941 (t, J = 7.2\text{Hz}, 3\text{H}, \text{CH}_3\text{-pentyl chain}). \]

\[ \text{13C NMR} (\delta, \text{ppm}) \]
\[ (\text{CDCl}_3, 100 \text{ MHz}) \]
\[ 132.2 (\text{CH}-\text{Ar}), 131.7 (\text{CH}-\text{Ar}), 128.3 (\text{CH}-\text{Ar}), 127.3 (\text{Cq}, \text{Ar}), 127.1 (\text{CH}-\text{Ar}), 125.1 (\text{Cq}, \text{Ar}), 95.2 (\text{Cq}, \text{C} \equiv \text{C}), 79.1 (\text{Cq}, \text{C} \equiv \text{C}), 79.0 (\text{Cq}, \text{C} \equiv \text{C}), 53.0 (\text{Cq}, \text{C} \equiv \text{Br}), 31.0 (\text{CH}_2\text{-pentyl chain}), 28.4 (\text{CH}_2\text{-pentyl chain}), 22.3 (\text{CH}_2\text{-pentyl chain}), 19.6 (\text{CH}_2\text{-pentyl chain}), 14.0 (\text{CH}_2\text{-pentyl chain}). \]

IR (\( \nu, \text{cm}^{-1} \)) (CCl\(_4\))
\[ 3063 (w), 2955 (s), 2931 (s), 2862 (s), 2232 (m), 2200 (m), 1476 (s), 1444 (s), 1376 (w), 1332 (w), 1298 (w), 1105 (w), 1036 (w). \]

HRMS (El+, m/z) : Calculated: 274.0357  found: 274.0348.

\[
\text{(2-Hept-1-ynyl-phenylethynyl)-phenyl-carbamic acid} \ \text{tert}-\text{butyl ester} \quad 3.15w
\]

\[
\text{MF: C}_{26}\text{H}_{29}\text{O}_2\text{N} \\
\text{MW = 387 g.mol}^{-1}
\]

Method: See general procedure 3.3 using (1 equiv., 2 mmol, 550 mg) of 1-bromoethynyl-2-hept-1-ynyl-benzene and (1.2 equiv., 2.4 mmol, 464 mg) of phenyl-carbamic acid \text{tert}-\text{butyl ester}.

Purification: Flash column chromatography (silica gel, 98:2 PE: AcOEt)/ Rf (95:5 PE:AcOEt): 0.31.

Product: Yellow oil.

Isolated yield: 78%. 

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**Benzyl-(2-hept-1-ynyl-phenylethynyl)-carbamic acid tert-butyl ester**

![Chemical Structure]

**MF:** $C_{27}H_{31}O_2N$

**MW:** 401 g mol$^{-1}$

**Method:** See **general procedure 3.3** using (1 equiv., 1.2 mmol, 330 mg) of 1-bromomethynyl-2-hept-1-ynyl-benzene and (1.2 equiv., 1.44 mmol, 298 mg) of benzyl-carbamic acid tert-butyl ester.

**Purification:** Flash column chromatography (silica gel, 98:2 PE:AcOEt) / $R_f$ (95:5 PE:AcOEt): 0.56.

**Product:** Orange oil.

**Isolated yield:** 81%.

**$^1$H NMR** ($\delta$, ppm)

- 7.51 (d, $J$ = 7.2 Hz, 2H, CH-Ar), 7.41-7.30 (m, 5H, CH-Ar), 7.20-7.13 (m, 2H, CH-Ar), 4.73 (s, 2H, CH$_2$N), 2.40 (t, $J$ = 7.2 Hz, 2H, CH$_2$-pentyl chain), 1.64-1.54 (m, 2H, CH$_2$-pentyl chain), 1.57 (s, 9H, CH$_3$-Bu), 1.46-1.38 (m, 2H, CH$_2$-pentyl chain), 0.89 (t, $J$ = 7.2 Hz, 3H, CH$_3$-pentyl chain).

**$^{13}$C NMR** ($\delta$, ppm)

- 152.8 (Cq, NC=O), 139.7 (Cq, Ar), 131.9 (CH-Ar), 130.7 (CH-Ar), 128.6 (CH-Ar), 127.1 (CH-Ar), 126.9 (CH-Ar), 126.4 (CH-Ar), 125.9 (Cq, Ar), 125.4 (Cq, Ar), 124.4 (CH-Ar), 94.3 (Cq, C≡C), 86.9 (Cq, C≡C), 83.5 (Cq, 'Bu), 79.4 (Cq, C≡C), 69.9 (Cq, C≡C), 31.2 (CH$_2$-pentyl chain), 28.4 (CH$_2$-pentyl chain), 28.0 (CH$_3$-Bu), 22.2 (CH$_2$-pentyl chain), 19.6 (CH$_2$-pentyl chain), 14.0 (CH$_3$-pentyl chain).

**IR** ($\nu$, cm$^{-1}$) (CCl$_4$)

- 3064 (w), 2959 (m), 2932 (m), 2864 (w), 2251 (m), 1737 (s), 1490 (w), 1458 (w), 1365 (s), 1285 (s), 1254 (m), 1157 (s), 1008 (w).

**HRMS** (EI+, m/z)

- Calculated: 387.2198, found: 387.2198.
(CDCl₃, 400 MHz) chain), 1.39-1.29 (m, 2H, CH₂-pentyl chain), 0.92 (t, J = 7.2Hz, 3H, CH₃-pentyl chain).

¹³C NMR (δ, ppm) 153.5 (Cq, NC=O), 136.6 (Cq, Ar), 131.9 (CH-Ar), 130.6 (CH-Ar), 128.4 (CH-Ar), 128.1 (Cq, Ar), 127.7 (CH-Ar), 127.0 (CH-Ar), 126.5 (CH-Ar), 125.9 (CH-Ar), 125.0 (Cq, Ar), 93.8 (Cq, C ≡≡ C), 87.9 (Cq, C ≡≡ C), 82.6 (Cq, C ≡≡ C), 79.6 (Cq, Bu), 70.1 (Cq, C ≡≡ C), 53.5 (CH₃N), 31.1 (CH₂-pentyl chain), 28.5 (CH₂-pentyl chain), 27.9 (CH₃-Bu), 22.1 (CH₂-pentyl chain), 19.6 (CH₂-pentyl chain), 13.9 (CH₃-pentyl chain).

IR (ν, cm⁻¹) (CCl₄) 3064 (w), 2958 (w), 2933 (m), 2863 (w), 2242 (m), 1723 (s), 1390 (m), 1366 (m), 1303 (m), 1246 (m), 1159 (m).

HRMS (EI+, m/z) : Calculated: 401.2355 found: 401.2346.

(2-Hept-1-ynyl-phenyl)-phenylethynyl-carbamic acid tert-butyl ester 3.15y

MF: C₂₆H₂₉O₂N

MW = 387 g.mol⁻¹

Method:

To a round bottom flask charged with 2-iodo aniline (1 equiv., 2.28 mmol, 500 mg) and triethylamine (9 mL) under an argon atmosphere was added 1-heptyne (1.2 equiv., 2.74 mmol, 360 µL) and PdCl₂(PPh₃)₂ (0.02 equiv., 0.045 mmol, 32 mg). The reaction mixture was allowed to stir for 5 minutes at room temperature. Cul (0.01 equiv., 0.023 mmol, 4.4 mg) was then added and the reaction mixture heated up to 30°C and allowed to stir at this temperature overnight. Upon completion (TLC), the mixture was then cooled to room temperature, filtered through a short pad of celite and concentrated under reduced pressure. Purification by flash column chromatography (silica gel, 95:5 PE:AcOEt) afforded the coupled product (414 mg, 97 %) as a light red/brown oil.

Next, to a round bottom flask charged with this coupled aniline (1 equiv., 412 mg, 2.20 mmol) and anhydrous THF (1M, 2.20 mL) at room temperature was added Boc₂O (1.10 equiv., 2.43 mmol, 545 mg) and the reaction mixture was then heated to reflux. After complete consumption of the starting amine (TLC), the mixture was concentrated under reduced pressure. Purification by flash column chromatography (silica gel, 98:2 PE:AcOEt) afforded the Boc protected...
aniline (500 mg, 79 %) as a yellow oil.

Next, **general procedure 3.3** using this previously prepared carbamate (1.2 equiv., 1.67 mmol, 480 mg) and bromoethylbenzene (1.0 equiv. 1.39 mmol, 252 mg) afforded after purification by flash column chromatography (silica gel, 95:5 PE:AcOEt) the title compound (110 mg, 20 %).

**Purification:** See above/ \( R_f \) (95:5 PE:AcOEt): 0.38.

**Product:** Orange/brown oil.

**Isolated yield:** 16% for three steps.

**\(^1\)H NMR** (\( \delta \), ppm) (CDCl\(_3\), 400 MHz) 7.53 (d, \( J = 6.4\) Hz, 1H, CH-Ar), 7.46-7.27 (m, 8H, CH-Ar), 2.47 (t, \( J = 7.2\)Hz, 2H, CH\(_2\)-pentyl chain), 1.69-1.62 (m, 2H, CH\(_2\)-pentyl chain), 1.57 (br s, 9H, CH\(_3\)-tBu), 1.53-1.45 (m, 2H, CH\(_2\)-pentyl chain), 1.39-1.32 (m, 2H, CH\(_2\)-pentyl chain), 0.93 (t, \( J = 7.2\)Hz, 3H, CH\(_3\)-pentyl chain).

**\(^{13}\)C NMR** (\( \delta \), ppm) (CDCl\(_3\), 100 MHz) 153.0 (Cq, NC=O), 132.9 (CH-Ar), 130.9 (CH-Ar), 128.3 (CH-Ar), 128.0 (CH-Ar), 127.9 (CH-Ar), 127.8 (Cq, Ar), 127.3 (CH-Ar), 127.1 (CH-Ar), 123.7 (Cq, Ar), 122.9 (Cq, Ar), 96.3 (Cq, C \( \equiv \) C), 82.9 (Cq, C \( \equiv \) C), 83.9 (Cq, C \( \equiv \) C), 82.9 (Cq, C \( \equiv \) C), 31.0 (CH\(_2\)-pentyl chain), 28.3 (CH\(_2\)-pentyl chain), 27.9 (CH\(_2\)-tBu), 22.2 (CH\(_2\)-pentyl chain), 19.7 (CH\(_2\)-pentyl chain), 13.9 (CH\(_3\)-pentyl chain).

**IR** (\( \nu \), cm\(^{-1}\)) (CCl\(_4\)) 2959 (w), 2932 (m), 2863 (w), 2253 (m), 1737 (s), 1489 (w), 1452 (w), 1367 (m), 1298 (s), 1251 (s), 1160 (s).

**HRMS** (El+, m/z) : calculated: 387.2198  found: 387.2194.
B.3.2.2 Synthesis of 4-oxazolin-2-ones via a Au/Ag-mediated rearrangement of 1Bu-ynamides

N-alkynyl tert-butyloxycarbamates in the presence of the gold catalyst PPh₃Au(NCCH₃)SbF₆ or the silver catalyst AgNTf₂ rearrange to afford 4-oxazolin-2-ones (Scheme B.3.2).

Ynamides synthesized:
3.16a, R¹ = Ph, R² = Ph
3.16b, R¹ = Ph, R² = 4-F-Ph
3.16c, R¹ = Ph, R² = 4-Cl-Ph
3.16d, R¹ = Ph, R² = 4-Br-Ph
3.16e, R¹ = Ph, R² = 2-O-Me-4-O-Me-Ph
3.16f, R¹ = Ph, R² = Bn
3.16g, R¹ = Ph, R² = CH₂CO₂Et
3.16h, R¹ = Ph, R² = CH(Me)-CO₂Me
3.16i, R¹ = t-Bu, R² = Ph
3.16j, R¹ = n-C₅H₁₁, R² = Ph
3.16k, R¹ = n-C₅H₁₁, R² = Bn
3.16l, R¹ = Ph, R² = Ph
3.16m, R¹ = Ph, R² = 4-Cl-Ph
3.16n, R¹ = , R² = CH₂CO₂Et
3.16o, R¹ = CH₂OAc, R² = Ph
3.16p, R¹ = CH₂OAc, R² = 2-Napht
3.16q, R¹ = CH₂OAc, R² = Bn
3.16r, R¹ = CH₂OAc, R² = CH₂CO₂Et
3.16s, R¹ = CH₂CH₂CH₂OTIPS, R² = Ph
3.16t, R¹ = CH₂CH₂CH₂OTIPS, R² = Bn
3.16u, R¹ = , R² = Ph
3.16v, R¹ = , R² = Bn
3.16w, R¹ = , R² = Ph
n-C₅H₁₁
3.16x, R¹ = , R² = Bn
n-C₅H₁₁
3.16y, R¹ = Ph, R² = , n-C₅H₁₁

Scheme B.3.2.2: 4-Oxal-2-ones synthesized in this project

General Procedure 3.4.A, Au-mediated rearrangement of 1Bu-ynamides to oxazolinones: To a round bottom flask charged with ynamide (1 equiv.) and anhydrous DCM (0.5 M) at room temperature and under an argon atmosphere was added PPh₃Au(CH₂CN)SbF₆ (0.01 equiv). The reaction mixture was heated to reflux (40 °C) and allowed to stir at this temperature. Upon completion (TLC), the solution was concentrated
under reduced pressure and purified either by flash column chromatography or by recrystallization from a suitable solvent mixture to afford the title compound.

In the case of oxazolones remarked instable, which made isolation impossible, the reaction was performed again, changing the previous solvent for CDCl$_2$. Upon completion (NMR), 1,3,5-trimethoxybenzene (1 equiv.) was added as an internal standard and the yield was determined by $^1$H NMR analysis of the crude reaction mixture.

**General Procedure 3.4.B, Ag-mediated rearrangement of $^1$Bu-ynamides:** To a round bottom flask charged with ynamide (1 equiv.) and anhydrous DCM (0.5 M) at room temperature and under an argon atmosphere was added AgNTf$_2$ (0.05 equiv). The reaction mixture was allowed to stir at this temperature (rt). Upon completion (TLC), the solution was concentrated under reduced pressure and the yield of the rearranged oxazolone was determined through $^1$H NMR analysis of the crude reaction mixture using 1,3,5-trimethoxybenzene (1 equiv.) as an internal standard.

<table>
<thead>
<tr>
<th>3,5-Diphenyl-3H-oxazol-2-one</th>
<th>3.16a</th>
</tr>
</thead>
<tbody>
<tr>
<td><img src="image.png" alt="Chemical Structure" /></td>
<td></td>
</tr>
<tr>
<td>MF: C$<em>{15}$H$</em>{11}$NO$_2$</td>
<td></td>
</tr>
<tr>
<td>MW = 237 g.mol$^{-1}$</td>
<td></td>
</tr>
</tbody>
</table>

**Method:** See general procedure 3.4.B using (1 equiv., 0.25 mmol, 74 mg) of phenyl-phenylethynyl-carbamic acid tert-butyl ester.

**Purification:** Recrystallization from petrol ether.

**Product:** White solid.

**Isolated Yield:** 83%.

$^1$H NMR ($\delta$, ppm) 7.63 (d, $J = 7.6$ Hz, 2H, CH-Ph), 7.56 (d, $J = 7.3$ Hz, 2H, CH-Ph), 7.48 (t, $J = 7.6$ Hz, 2H, CH-Ph), 7.42 (t, $J = 7.3$ Hz, 2H, CH-Ph), 7.34 (t, $J = 7.3$ Hz, 1H, CH-
(CDCl$_3$, 400 MHz) Ph), 7.31 (t, $J = 7.6$ Hz, 1H, CH-Ph), 7.18 (s, 1H, NCH=).

$^{13}$C NMR (δ, ppm) 152.5 (Cq, NC=O), 139.8 (Cq, O=C=CHN), 135.4 (Cq, Ph), 129.4 (CH-Ph), 128.8 (CH-Ph), 128.5 (CH-Ph), 126.8 (Cq, Ph), 126.5 (CH-Ph), 123.1 (CH-Ph), 120.9 (CH-Ph), 108.4 (O=C=CHN).

IR (ν, cm$^{-1}$) (CCl$_4$) 3153 (w), 3055 (w), 2958 (m), 2927 (w), 2860 (w), 1773 (s), 1595 (m), 1503 (s), 1450 (m), 1393 (s), 1212 (w), 1114 (m).

HRMS (EI+, m/z) : Calculated: 237.0790  found: 237.0793.

3-(4-Chloro-phenyl)-5-phenyl-3H-oxazol-2-one 3.16c

![Chemical Structure]

MF: C$_{15}$H$_{10}$NO$_2$Cl

MW = 271 g.mol$^{-1}$

Method : See general procedure 3.4.B using (1 equiv., 0.25 mmol, 82 mg) of (4-chlorophenyl)-phenylethynyl-carbamic acid tert-butyl ester.

Purification : Recrystallization from petrol ether.

Product : White solid.

Isolated Yield : 88%.

$^1$H NMR (δ, ppm) 7.62-7.55 (m, 4H, CH-Ar), 7.47-7.41 (m, 4H, CH-Ar), 7.36 (t, $J = 7.4$Hz, 1H, CH-Ar), 7.16 (s, 1H, NCH=).

$^{13}$C NMR (δ, ppm) 152.3 (Cq, NC=O), 140.1 (O=C=CHN), 134.0 (Cq, Ar), 132.0 (Cq, Ar), 129.6 (CH-Ar), 128.9 (CH-Ar), 128.7 (CH-Ar), 126.6 (Cq, Ar), 123.2 (CH-Ar), 122.0 (CH-Ar), 107.9 (O=C=CHN).

IR (ν, cm$^{-1}$) (CCl$_4$) 3177 (w), 3062 (w), 2975 (m), 2928 (w), 1731 (s), 1654 (s), 1498 (m), 1449 (m), 1393 (s), 1279 (s), 1203 (m), 1095 (m), 1052 (w), 1016 (m).

HRMS (EI+, m/z) : Calculated: 271.0400  found: 271.0403.
3-(2,4-Dimethoxy-phenyl)-5-phenyl-3H-oxazol-2-one

MF: C\textsubscript{17}H\textsubscript{15}NO\textsubscript{4} \\
MW = 297 g.mol\textsuperscript{-1}

Method: See general procedure 3.4.A using (1 equiv., 0.10 mmol, 32 mg) of (2,4-

dimethoxyphenyl)-phenylethynyl-carbamic acid tert-butyl ester.

Purification: Recrystallization from petrol ether.

Product: Orange solid.

Isolated Yield: 78%.

\textsuperscript{1}H NMR (δ, ppm) \\
(CDCl\textsubscript{3}, 400 MHz) \\
7.53 (d, J = 7.2 Hz, 2H, CH-Ar), 7.41-7.36 (m, 3H, CH-Ar), 7.30 (t, J = 7.4 Hz, \\
1H, CH-Ar), 6.96 (s, 1H, NCH=), 6.57 (s, 1H, CH-Ar), 6.56 (dd, J = 2.6Hz, J = \\
10.8 Hz, 1H, CH-Ar), 3.84 (s, 3H, OCH\textsubscript{3}), 3.83 (s, 3H, OCH\textsubscript{3}).

\textsuperscript{13}C NMR (δ, ppm) \\
(CDCl\textsubscript{3}, 100 MHz) \\
160.7 (Cq, Ar), 154.9 (Cq, Ar), 153.9 (Cq, NC=O), 138.5 (OC=CHN), 128.7 \\
(CH-Ar), 128.0 (CH-Ar), 127.9 (CH-Ar), 127.4 (Cq, Ar), 122.8 (CH-Ar), 116.6 \\
(Cq, Ar), 112.0 (OC=CHN), 104.5 (CH-Ar), 99.6 (CH-Ar), 55.7 (OCH\textsubscript{3}), 55.5 \\
(OCH\textsubscript{3}).

IR (ν, cm\textsuperscript{-1}) (CCl\textsubscript{4}) \\
3061 (w), 2927 (w), 2854 (m), 2360 (w), 1768 (s), 1611 (s), 1518 (m), 1460 (s), \\
1287 (s), 1212 (m), 1162 (m), 1109 (s), 1039 (s).

HRMS (EI+, m/z): Calculated: 397.1001 found: 297.0998.
**5-(2-Hept-1-ynyl-phenyl)-3-phenyl-3H-oxazol-2-one**

MF: C_{22}H_{21}O_{2}N

MW = 331 g. mol^{-1}

**Method:** See general procedure 3.4.A using (1 equiv., 0.52 mmol, 200 mg) of (2-hept-1-ynyl-phenylethynyl)-phenyl-carbamic acid tert-butyl ester.

**Purification:** Flash column chromatography (silica gel, 98:2 PE: AcOEt)/ R_{f} (9:1 PE:AcOEt): 0.42.

**Product:** Yellow solid.

**Isolated yield:** 91%.

**{H NMR} (δ, ppm) (CDCl₃, 400 MHz)**

- 7.94 (s, 1H, NCH=)
- 7.73 (d, J = 8.0Hz, 1H, CH-Ar)
- 7.63 (dd , J = 0.9Hz, J = 8.0Hz, 2H, CH-Ar)
- 7.48 (t, J = 8.0Hz, 3H, CH-Ar)
- 7.38-7.30 (m, 2H, CH-Ar)
- 7.26-7.22 (m, 1H, CH-Ar)
- 2.53 (t, J = 7.2Hz, 2H, CH₂-pentyl chain)
- 1.71-1.64 (m, 2H, CH₂-pentyl chain)
- 1.49-1.41 (m, 2H, CH₂-pentyl chain)
- 1.38-1.29 (m, 2H, CH₂-pentyl chain)
- 0.88 (t, J = 7.4Hz, 3H, CH₃-pentyl chain).

**{C NMR} (δ, ppm) (CDCl₃, 100 MHz)**

- 152.1 (Cq, NC=O)
- 138.2 (Cq)
- 135.6 (Cq)
- 133.8 (CH-Ar)
- 129.5 (CH-Ar)
- 128.1 (CH-Ar)
- 127.5 (CH-Ar)
- 127.4 (CH-Ar)
- 126.6 (Cq)
- 124.5 (CH-Ar)
- 121.0 (CH-Ar)
- 118.4 (Cq)
- 112.6 (NCH=)
- 97.2 (Cq, C ≡ C)
- 80.2 (Cq, C ≡ C)
- 31.3 (CH₂-pentyl chain)
- 28.3 (CH₂-pentyl chain)
- 22.2 (CH₂-pentyl chain)
- 19.8 (CH₂-pentyl chain)
- 13.9 (CH₃-pentyl chain).

**IR (ν, cm⁻¹) (CCl₄)**

- 3066 (w)
- 2958 (s)
- 2932 (s)
- 2863 (s)
- 1772 (s)
- 1719 (s)
- 1639 (w)
- 1596 (s)
- 1503 (s)
- 1460 (s)
- 1389 (s)
- 1308 (s)
- 1272 (m)
- 1223 (m)
- 1158 (s)
- 1114 (s)
- 1027 (s)
- 978 (s).

**HRMS (EI+, m/z):** Calculated: 331.1572  found: 331.1581.
3-BenzyI-5-(2-hept-1-ynyl-phenyl)-3H-oxazol-2-one

MF: C_{23}H_{23}O_{2}N

MW = 345 g/mol

Method: See general procedure 3.4.A using (1 equiv. 0.10 mmol, 40 mg) of benzyl-(2-hept-1-ynyl-phenylethynyl)-carbamic acid tert-butyl ester.

Purification: Flash column chromatography (silica gel, 9:1 PE:AcOEt)/ R_{f} (9:1 PE:AcOEt): 0.29.

Product: White solid.

Isolated yield: 90%.

^{1}H NMR (δ, ppm) (CDCl_{3}, 400 MHz)
7.65 (d, J = 8.0Hz, 1H, CH-Ar), 7.41-7.29 (m, 8H, NCH= + CH-Ar), 7.17 (dt, J = 1.3Hz, J = 7.6Hz, 1H, CH, CH-Ar), 4.80 (s, 2H, CH_{2}N), 2.28 (t, J = 7.2Hz, CH_{2}-pentyl chain), 1.49-1.42 (m, 2H, CH_{2}-pentyl chain), 1.38-1.30 (m, 4H, CH_{2}-pentyl chain), 0.90 (t, J = 7.2Hz, 3H, CH_{3}-pentyl chain).

^{13}C NMR (δ, ppm) (CDCl_{3}, 100 MHz)
154.4 (Cq, NC=O), 137.6 (Cq), 135.1 (Cq), 133.5 (CH-Ar), 129.0 (CH-Ar), 128.4 (CH-Ar), 128.0 (CH-Ar), 127.9 (CH-Ar), 127.0 (Cq), 124.2 (CH-Ar), 124.2 (CH-Ar), 118.1 (Cq), 113.0 (NCH=), 96.8 (Cq, C =C), 80.0 (Cq, C =C), 47.9 (CH_{3}N), 31.1 (CH_{2}-pentyl chain), 28.1 (CH_{2}-pentyl chain), 22.1 (CH_{2}-pentyl chain), 19.5 (CH_{3}-pentyl chain), 13.9 (CH_{3}-pentyl chain).

IR (ν, cm\(^{-1}\)) (CCl_{4})
3066 (w), 3033 (w), 2930 (s), 2862 (s), 1757 (s), 1701 (s), 1630 (s), 1601 (s), 1562 (s), 1490 (s), 1453 (s), 1392 (s), 1364 (s), 1312 (s), 1265 (s), 1173 (s), 1113 (s), 1085 (s), 1022 (s).

HRMS (EI+, m/z): Calculated: 345.1729  found: 345.1727.
3-(2-Hept-1yny1-phenyl)-5-phenyl-3H-oxazol-2-one

\[
\text{MF: C}_{22}\text{H}_{21}\text{O}_{2}\text{N}
\]

\[
\text{MW} = 331\text{g. mol}^{-1}
\]

**Method:** See general procedure 3.4.A using (1 equiv. 0.10 mmol, 39 mg) of (2-hept-1yny1-phenyl)-phenylethynyl-carbamic acid tert-butyl ester.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt)/R\(_f\)(9:1 PE:AcOEt): 0.27.

**Product:** Yellow oil.

**Isolated yield:** 57%.

\[^1\text{H NMR}\] (\(\delta\), ppm) \(\text{(CDCl}_3\text{, 400 MHz)}\)

7.57-7.51 (m, 4H, CH-Ar), 7.42-7.37 (m, 3H, CH-Ar), 7.34-7.30 (m, 2H, CH-Ar), 7.24 (s, 1H, NCH=), 2.37 (t, \(J = 7.0\text{Hz}\), 2H, CH\(_2\)-pentyl chain), 1.54-1.47 (m, 2H, CH\(_2\)-pentyl chain), 1.34-1.26 (m, 2H, CH\(_2\)-pentyl chain), 1.24-1.15 (m, 2H, CH\(_2\)-pentyl chain), 0.79 (t, \(J = 7.3\text{Hz}\), 3H, CH\(_3\)-pentyl chain).

\[^{13}\text{C NMR}\] (\(\delta\), ppm) \(\text{(CDCl}_3\text{, 100 MHz)}\)

153.3 (Cq, C=O), 138.6 (Cq), 136.0 (Cq), 133.5 (CH-Ar), 128.9 (CH-Ar), 128.6 (CH-Ar), 128.2 (CH-Ar), 128.1 (CH-Ar), 127.4 (Cq), 126.2 (CH-Ar), 122.9 (CH-Ar), 120.5 (Cq), 111.2 (NCH=), 97.7 (Cq, C\(\equiv\)C), 76.4 (Cq, C\(\equiv\)C), 31.0 (CH\(_2\)-pentyl chain), 28.1 (CH\(_2\)-pentyl chain), 22.1 (CH\(_2\)-pentyl chain), 19.6 (CH\(_2\)-pentyl chain), 13.8 (CH\(_3\)-pentyl chain).

**IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))**

2957 (w), 2931(w), 2861(w), 1776 (s), 1497 (w), 1453 (w), 1391 (w), 1293 (w), 1214 (w), 1125 (w), 1101 (w), 1051 (w).

**HRMS (El+, m/z):** calculated: 331.1572 found: 331.1581.
B.3.3 Chapter 4: Gold(I) mediated cycloisomerization of 1,6-enynes – a short disgression on catalyst counter ion effect and remote substituent control

All molecules synthesized for this work are described here. This project was developed individually.

B.3.3.1 Synthesis of 1,6-enynes

1,6-enynes were synthesized as depicted in the scheme below:

**Scheme B.3.3.1 Synthesis of 1,6-enynes.**
General Procedure 4.1\textsuperscript{301}, Sonogashira coupling reaction with vinyl bromides: To a round bottom flask charged with vinyl bromide (1 equiv.) and Pd(PPh\textsubscript{3})\textsubscript{4} (0.01 equiv.) under an argon atmosphere was sequentially added CuI (0.01 equiv.), piperidine (1 equiv.), THF (0.10 M) and alkyne (2 equiv.). The reaction mixture was heated up to 60 °C and allowed to stir at this temperature upon completion (TLC). The reaction was generally left overnight. The mixture was then quenched with a saturated solution of NH\textsubscript{4}Cl, extracted with Et\textsubscript{2}O (3x), dried (MgSO\textsubscript{4}) and concentrated under reduced pressure. Purification by flash column chromatography afforded the title compounds in the stated yields.

General Procedure 4.2\textsuperscript{302}, Sonogashira coupling reaction with aryl iodides: To a round bottom flask charged with alkyne (1 equiv.), aryl iodide (1.1 equiv.) and Et\textsubscript{3}N (0.10 M) was added PdCl\textsubscript{2}(PPh\textsubscript{3})\textsubscript{4} (0.02 equiv.) and CuI (0.02 equiv.). The mixture was heated up to 50 °C and allowed to stir at this temperature upon completion (TLC). The reaction was usually left overnight. Then a saturated solution of NH\textsubscript{4}Cl was added, the reaction was extracted with AcOEt (3x), dried (MgSO\textsubscript{4}) and concentrated under reduced pressure. Purification by flash column chromatography afforded the title alkynes in the stated yields.

General Procedure 4.3\textsuperscript{303}, Iron-mediated coupling reaction: To a round bottom flask charged with vinyl bromide (1 equiv.) and Fe(acac)\textsubscript{3} (0.02 equiv.) flushed with argon for 15 min was added THF (0.70 M to vinyl bromide) and NMP (0.80 M to vinyl bromide). The temperature was cooled to 0 °C and the Grignard reagent (3 equiv.) was added. The reaction was allowed to stir at 0 °C upon completion (TLC). The reaction usually takes not more than 1h. The reaction was then quenched with a saturated solution of NH\textsubscript{4}Cl, extracted with Et\textsubscript{2}O (3x), dried (MgSO\textsubscript{4}) and concentrated under reduced pressure. Purification by flash column chromatography afforded the title compounds in the stated yields.

General procedure 4.4, Propargylation of monosubstituted malonates: To a round bottom flask charged with monosubstitued malonate (1equiv.) and THF (0.125 M) was sequentially added NaH (60% in mineral oil, 1.5 equiv.) and propargyl bromide (1.5 equiv., 80% w/w solution in toluene). The reaction was allowed to stir at room temperature upon completion (TLC). The reaction was usually left overnight. Then, the reaction was quenched with a saturated solution of NH\textsubscript{4}Cl, extracted with AcOEt (3x), dried (MgSO\textsubscript{4}) and concentrated under reduced pressure.


under reduced pressure. Purification by flash column chromatography afforded the title compounds in the stated yields.

**General Procedure 4.5**<sup>304</sup>, *Deprotection of TMS-alkynes*: To a round bottom flask charged with TMS-protected alkyne (1 equiv.) and EtOH (0.50 M) at room temperature was added K$_2$CO$_3$ (0.20 equiv.). The reaction was allowed to stir at this temperature upon completion (TLC). The mixture was then diluted in H$_2$O, extracted with DCM (3x), washed with brine (1x), dried (MgSO$_4$) and concentrated under reduced pressure. Purification by flash column chromatography afforded the terminal alkyne.

**2-(2-Bromo-allyl)-malonic acid diethyl ester**

![Structural formula of 2-(2-Bromo-allyl)-malonic acid diethyl ester]

**Method**: To a round bottom flask charged with 2-bromo-malonic acid diethyl ester (1 equiv., 20 mmol, 3.42 mL), 2,3-dibromo-propene (1.2 equiv., 24 mmol, 3.10 mL) and DMF (1M, 20 mL) at 0 °C was added K$_2$CO$_3$ (2 equiv., 40 mmol, 5.52 g). The temperature was allowed to warm up and to stir at room temperature. Upon completion (TLC), the reaction was diluted in water, extracted with Et$_2$O (3x), washed with brine (1x), dried (MgSO$_4$) and concentrated under reduced pressure. Purification by flash column chromatography afforded 2-bromo-2-(2-bromo-allyl)-malonic acid diethyl ester (6.36 g, 89%).

This compound (1 equiv., 16.8 mmol, 6.0 g) was then diluted in AcOH (1M, 17 mL). To this solution at room temperature was added Zn dust (2 equiv. 33.6 mmol, 2.20 g) and the reaction was allowed to stir at room temperature upon completion (TLC). The reaction was then filtered through a small pad of celite, washed with a saturated solution of Na$_2$CO$_3$ with stirring (in a erlenmeyer), then the solution was extracted with Et$_2$O (3x), washed with brine (1x), dried (MgSO$_4$) and concentrated under reduced pressure. Filtration through silica gel

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<sup>304</sup>Allevi, P.; Ciuffreda P.; Anastasia M.; *Tetrahedron: Asymmetry*; 1997, 8, 1, 93.
(8:2 PE:AcOEt) afforded the title compound (4.4 g, 94%).

**Purification:** See above.

**Product:** Transparent oil.

**Isolated yield:** 84% (for two steps)

**$^1$H NMR** ($\delta$, ppm) ($\text{CDCl}_3$, 400 MHz)
- 5.69 (s, 1H, C=CH$_2$),
- 5.47 (s, 1H, C=CH$_2$),
- 4.21 (q, $J = 7.1$ Hz, 4H, OCH$_2$CH$_3$),
- 3.78 (t, $J = 7.6$ Hz, 1H, CH),
- 3.01 (d, $J = 7.6$ Hz, 2H, CH$_2$),
- 1.27 (t, $J = 7.1$ Hz, 6H, OCH$_2$C$_3$H$_3$).

**$^{13}$C NMR** ($\delta$, ppm) ($\text{CDCl}_3$, 100 MHz)
- 168.1 (Cq x2, C=O),
- 129.4 (Cq, C=CH$_2$),
- 119.7 (C=CH$_2$),
- 61.7 (OCH$_2$CH$_3$ x2),
- 50.6 (CH),
- 40.4 (CH$_2$),
- 14.0 (OCH$_2$C$_3$H$_3$ x2).

**IR** ($\nu$, cm$^{-1}$) ($\text{CCl}_4$)
- 3467 (w),
- 2984 (s),
- 2939 (s),
- 2908 (s),
- 2874 (m),
- 1737 (s),
- 1632 (s),
- 1465 (s),
- 1445 (s),
- 1428 (s),
- 1392 (s),
- 1369 (s),
- 1176 (s),
- 1038 (s).

**MS** (Cl, NH$_3$, m/z)
- M + H$: 280,
- M + NH$_4^+$: 297.

### 2-(2-Methylene-but-3-ynyl)-2-prop-2-ynyl-malonic acid diethyl ester 4.64a

![Chemical Structure](image)

**MF:** C$_{15}$H$_{18}$O$_4$

**MW:** 262 g mol$^{-1}$

**Method:** See **general procedure 3.1** with (1 equiv., 1.0 mmol, 280 mg) of 2-(2-bromoallyl)-malonic acid diethyl ester and ethynyl-trimethyl-silane (2 equiv., 2.0 mmol, 277 µL), followed by **general procedure 3.5** with (1 equiv., 0.95 mmol, 282 mg) of the previously prepared coupled product, followed by **general procedure 3.4** using (1 equiv., 0.54 mmol, 120 mg) of the previously prepared terminal alkyne.

**Purification:** Flash column chromatography (silica gel 9:1 PE AcOEt)/ $R_f$ (95:5 PE:AcOEt): 0.30.

**Product:** Transparent oil.
Isolated yield: 38\% for three steps.

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz)

5.59 (s, 1H, C=CH$_2$), 5.51 (s, 1H, C=CH$_2$), 4.23-4.13 (m, 4H, OCH$_2$CH$_3$), 2.97 (s, 2H, CH$_2$), 2.89 (d, $J = 2.8$Hz, 2H, CH$_2$-propargyl chain), 2.86 (s, 1H, CH), 2.02 (t, $J = 2.7$Hz, 1H, CH), 1.24 (t, $J = 7.2$ Hz, 6H, OCH$_2$CH$_3$).

$^{13}$C NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz)

169.3 (Cq x2, C=O), 128.5 (C=CH$_2$), 124.5 (Cq, C=CH$_2$), 79.0 (Cq, C=C), 72.8 (Cq, C=CH), 61.7 (OCH$_2$CH$_3$), 56.2 (Cq), 38.2 (CH$_2$), 22.2 (CH$_2$), 13.9 (OCH$_2$CH$_3$).

IR ($\nu$, cm$^{-1}$) (CCl$_4$)

3313 (s), 2954 (w), 2842 (w), 1744 (s), 1610 (w), 1437 (m), 1324 (w), 1296 (m), 1274 (m), 1237 (s), 1199 (s), 1182 (s), 1075 (w), 1053 (w), 982 (w).

HRMS (EI+, m/z):  Calculated: 262.1205 found: 262.1210.

2-(2-Methylene-non-3-ynyl)-2-prop-2-ynyl-malic acid diethyl ester 4.64b

Method:

See general procedure 4.1 using (1 equiv., 0.9 mmol, 250 mg) of 2-(2-bromoallyl)-malonic acid diethyl ester and hept-1-yn-1-ene (3 equiv., 2.7 mmol, 360 µL), followed by general procedure 4.4 using (1 equiv., 0.41 mmol, 120 mg) of the previously prepared compound.

Purification:

Flash column chromatography (silica gel 95:5 PE: AcOEt)/ $R_f$ (9:1 PE:AcOEt): 0.38.

Product:

Orange oil.

Isolated yield:

47\% for two steps.

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz)

5.40 (s, 1H, C=CH$_2$), 5.39 (s, 1H, C=CH$_2$), 4.27-4.12 (m, 4H, OCH$_2$CH$_3$), 2.94 (m, 4H, CH$_2$), 2.24 (t, $J = 7.2$Hz, 2H, CH$_2$-pentylchain), 2.00 (t, $J = 2.6$Hz, 1H, CH =CH), 1.53-1.48 (m, 2H, CH$_2$-pentylchain), 1.37-1.30 (m, 4H, CH$_2$-pentylchain), 1.26 (t, $J = 7.2$Hz, 6H, OCH$_2$CH$_3$), 0.90 (t, $J = 7.0$ Hz, 3H, CH$_3$-
pentylchain).

**13C NMR (δ, ppm)**

169.5 (Cq x2, C=O), 126.0 (Cq, C=CH₂), 125.2 (C=CH₂), 91.3 (Cq, C==C), 80.3 (Cq, C==C), 79.3 (Cq, C==C), 71.5 (C==CH), 61.6 (OCH₂CH₃ x2), 56.5 (Cq), 38.2 (CH₂), 22.3 (CH₂), 14.0 (CH₂), 13.9 (CH₃).

**IR (ν, cm⁻¹) (CCl₄)**

3314 (m), 3295 (w), 2982 (m), 2961 (m), 2935 (m), 2910 (m), 2874 (w), 2862 (w), 2838 (w), 2224 (w), 1740 (s), 1606 (w), 1466 (w), 1458 (w), 1446 (w), 1390 (m), 1367 (w), 1323 (w), 1280 (m), 1241 (m), 1206 (s), 1197 (m), 1097 (s), 1054 (m), 1018 (m), 1011 (m).

**HRMS (EI+, m/z):**


2-(2-Methylene-4-trimethylsilany1-but-3-ynyl)-2-prop-2-ynyl-malonic acid diethyl ester

\[
\begin{array}{c}
\text{EtO} \\
\text{TMS} \\
\text{CH} \\
\text{O} \\
\text{Et}
\end{array}
\]

MF: C₁₈H₂₆O₄Si

MW = 334 g.mol⁻¹

**Method:**

See general procedure 4.1 with (1 equiv., 3.6 mmol, 1.0 g) of 2-(2-bromo-allyl)-malonic acid diethyl ester followed by general procedure 4.4 using (1 equiv., 0.14 mmol, 40 mg) of the previously prepared monosubstituted malonate.

**Purification:**

Flash column chromatography (silica gel 95:5 PE:AcOEt)/ Rₖ (95:5 PE:AcOEt): 0.23.

**Product:**

Yellow pale oil.

**Isolated yield:**

37% for two steps.

**1H NMR (δ, ppm)**

5.55 (s, 1H, C=CH₂), 5.43 (s, 1H, C=CH₂), 4.27-4.12 (m, 4H, OC₃H₂CH₃), 2.96 (d, J = 2.6Hz, 2H, CH₂-propargyl chain), 2.95 (s, 2H, CH₂), 2.02 (t, J = 2.6Hz, 1H, CH), 1.26 (t, J = 7.2Hz, 6H, OCH₂C₃H₃), 0.19 (s, 9H, CH₃-TMS).

**13C NMR (δ, ppm)**

169.3 (Cq x2, C=O), 127.6 (C=CH₂), 125.8 (Cq, C=CH₂), 104.9 (Cq, C==C), 94.9 (Cq, C==C), 79.3 (Cq, C==C), 71.6 (C==CH), 61.6 (OCH₂CH₃ x2), 56.7 (Cq), 38.2 (CH₂), 22.3 (CH₂), 14.0 (OCH₂CH₃ x2), -0.20 (CH₃-TMS).

**IR (ν, cm⁻¹) (CCl₄)**

3314 (m), 2982 (m), 2963 (m), 2938 (w), 2902 (w), 2873 (w), 2145 (w), 1740 (s), 1604 (w), 1464 (w), 1445 (w), 1391 (w), 1367 (w), 1323 (w), 1297 (m), 1275 (m), 1251 (s), 1204 (s), 1184 (s), 1096 (w), 1072 (w), 1053 (w)
2-(2-Methylene-4-phenyl-but-3-ynyl)-2-prop-2-ynyl-malonic acid diethyl ester

\[ \text{EtO} \underset{\begin{array}{c} \text{O} \\ \text{O} \end{array}}{\text{OEt}} \]

MF: C_{21}H_{22}O_4

MW = 338 g.mol\(^{-1}\)

**Method:** See general procedure 4.1 with (1 equiv., 0.9 mmol, 250 mg) of 2-(2-bromo-allyl)-malonic acid diethyl ester and ethynyl-benzene (2 equiv., 1.8 mmol, 197 \(\mu\)L), followed by general procedure 4.4 using (1 equiv., 0.9 mmol, 270 mg) of the previously prepared coupled product.

**Purification:** Flash column chromatography (silica gel 95:5 PE:AcOEt)/ \(R_f\) (95:5 PE:AcOEt): 0.24.

**Product:** Orange oil.

**Isolated yield:** 66% (for two steps).

**\(^1\)H NMR (\(\delta\) ppm)**

(CDCl\(_3\), 400 MHz) 7.44-7.42 (m, 2H, CH-Ph), 7.31-7.29 (m, 3H, CH-Ph), 5.59 (s, 1H, C=CH\(_2\)), 5.49 (s, 1H, C=CH\(_2\)), 4.26-4.10 (m, 4H, OCH\(_2\)CH\(_3\)), 3.07 (s, 2H, CH\(_2\)), 3.00 (d, \(J = 2.7\)Hz, 2H, CH\(_2\)-propargyl chain), 2.06 (t, \(J = 2.7\)Hz, 1H, CH), 1.23 (t, \(J = 7.2\)Hz, 6H, OCH\(_2\)C\(_3\)).

**\(^13\)C NMR (\(\delta\) ppm)**

(CDCl\(_3\), 100 MHz) 169.4 (Cq x2, C=O), 131.5 (CH-Ph), 128.3 (CH-Ph), 128.3 (CH-Ph), 126.6 (C=CH\(_2\)), 125.5 (Cq), 123.0 (Cq), 89.9 (Cq, C \(\equiv\) C), 89.0 (Cq, C \(\equiv\) C), 79.2 (Cq, C \(\equiv\) C), 71.7 (\(\equiv\)CH), 61.8 (OCH\(_2\)CH\(_3\) x2), 56.6 (Cq), 38.6 (CH\(_2\)), 22.3 (CH\(_2\)), 14.0 (OCH\(_2\)CH\(_3\) x2).

**IR (\(\nu\), cm\(^{-1}\))** (CCl\(_4\) 3312 (w), 2983 (w), 2939 (w), 2908 (w), 1740 (s), 1687 (w), 1599 (w), 1491 (w), 1465 (w), 1444 (w), 1368 (w), 1285 (w), 1247 (w), 1215 (m), 1193 (m), 1097 (w).

**HRMS (EI+, m/z):** Calculated: 338.1518 found: 338.1532.
2-[4-(4-Methoxy-phenyl)-2-methylene-but-3-ynyl]-2-prop-2-ynyl-malonic acid diethyl ester 4.64e

Method: See general procedure 4.1 with (1 equiv., 3.6 mmol, 1.0 g) of 2-(2-bromoallyl)-malonic acid diethyl ester and ethynyltrimethylsilane, followed by general procedure 4.5 using (1 equiv., 2.0 mmol, 580 mg) of the previously prepared coupled product in EtOH, followed by general procedure 4.2 using (1 equiv., 0.63 mmol, 140 mg) of the previously prepared terminal alkyne and (1.1 equiv., 0.69 mmol, 161 mg) of 1-iodo-4-methoxy-benzene, followed by general procedure 4.4 using (1 equiv., 0.42 mmol, 140 mg) of the previously prepared monosubstituted malonate.

Purification: Flash column chromatography (silica gel 95:5 PE: AcOEt)/ Rf (9:1 PE:AcOEt): 0.33.

Product: White solid.

Isolated yield: 25% for four steps.

H NMR (δ, ppm) (CDCl₃, 400 MHz) 7.37 (d, J = 8.6Hz, 2H, CH-Ar), 6.83 (d, J = 8.6Hz, 2H, CH-Ar), 5.54 (s, 1H, C=CH₂), 5.45 (s, 1H, C=CH₂), 4.23-4.12 (m, 4H, OCH₂CH₃), 3.81 (s, 3H, OCH₃), 3.06 (s, 2H, CH₂), 3.00 (d, J = 2.8Hz, 2H, CH₂-propargyl chain), 2.05 (t, J = 2.8Hz, 1H, CH), 1.23 (t, J = 7.0Hz, 6H, OCH₂CH₃).

C NMR (δ, ppm) (CDCl₃, 100 MHz) 169.5 (Cq x2, C=O), 159.7 (Cq), 159.7 (Cq), 133.0 (CH-Ar), 125.8 (C=C), 125.8 (Cq), 115.2 (Cq), 114.0 (CH-Ar), 90.0 (Cq, C=C=C), 87.9 (Cq, C=C=C), 79.4 (Cq, C=C=C), 71.6 (C=C=C), 61.7 (OCH₂CH₃ x2), 56.7 (Cq), 55.3 (OCH₃), 38.7 (CH₂), 22.4 (CH₂), 14.0 (OCH₂CH₃ x2).

IR (ν, cm⁻¹) (CCl₄) 3314 (m), 2983 (m), 2937 (m), 2908 (w), 2838 (w), 1739 (s), 1601 (m), 1571 (w), 1509 (s), 1465 (w), 1442 (w), 1390 (w), 1367 (w), 1289 (s), 1249 (s), 1215 (s), 1193 (s), 1181 (s), 1170 (s), 1096 (w), 1071 (w).

HRMS (EI+, m/z): Calculated: 368.1624 found: 368.1623.
2-[4-(4-Chloro-phenyl)-2-methylene-but-3-ynyl]-2-prop-2-ynyl-malonlic acid diethyl ester

\[
\begin{align*}
\text{MF: } & C_{21}H_{21}O_4Cl \\
\text{MW: } & 372.5 \text{ g.mol}^{-1}
\end{align*}
\]

**Method:**
See general procedure 4.1 with (1 equiv., 3.6 mmol, 1.0 g) of 2-(2-bromoallyl)-malonic acid diethyl ester, followed by general procedure 4.5 using (1 equiv., 2.0 mmol, 580 mg) of the previously prepared coupled product in EtOH, followed by general procedure 4.2 using (1 equiv., 0.63 mmol, 140 mg) of the previously prepared terminal alkyne and (1.1 equiv., 0.69 mmol, 161 mg) of 1-chloro-4-iodo-benzene, followed by general procedure 4.4 using (1 equiv., 0.53 mmol, 151 mg) of the previously prepared monosubstituted malonate.

**Purification:**
Flash column chromatography (silica gel 95:5 PE:AcOEt)/R$_f$ (9:1 PE:AcOEt): 0.50.

**Product:**
White solid.

**Isolated yield:**
27% for four steps.

**$^1$H NMR (δ, ppm)**
(CDCl$_3$, 400 MHz) 7.36 (d, J = 8.6 Hz, 2H, CH-Ar), 7.28 (d, J = 8.6 Hz, 2H, CH-Ar), 5.59 (s, 1H, C=CH$_2$), 5.50 (s, 1H, C=CH$_2$), 4.25-4.09 (m, 4H, OCH$_2$CH$_3$), 3.06 (s, 2H, CH$_2$), 2.98 (d, J = 2.8 Hz, 2H, CH$_2$-propargyl chain), 2.06 (t, J = 2.8 Hz, 1H, CH), 1.23 (t, J = 7.2 Hz, 6H, OCH$_2$C$_3$H$_3$).

**$^{13}$C NMR (δ, ppm)**
(CDCl$_3$, 100 MHz) 169.4 (Cq x2, C=O), 134.4 (Cq), 132.7 (CH-Ar), 128.7 (CH-Ar), 126.9 (C=CH$_2$), 125.4 (Cq), 121.5 (Cq), 90.1 (Cq, C $=$ C), 88.8 (Cq, C $=$ C), 79.2 (Cq, C $=$ C), 71.7 (CH), 61.8 (OCH$_2$CH$_3$ x2), 56.7 (Cq), 38.5 (CH$_2$), 22.4 (CH$_2$), 14.0 (OCH$_2$C$_3$H$_3$ x2)

**IR (ν, cm$^{-1}$) (CCl$_4$)**
3313 (s), 3100 (w), 2933 (s), 2939 (m), 2907 (w), 2873 (w), 1739 (s), 1606 (w), 1591 (w), 1490 (s), 1465 (m), 1445 (m), 1397 (m), 1367 (m), 1321 (m), 1285 (s), 1247 (s), 1215 (s), 1193 (s), 1093 (s), 1071 (s)

**HRMS (EI+, m/z):**
Calculated: 372.1128  found: 372.1121.
2-(2-Methylene-4-phenyl-but-3-ynyl)-2-(3-phenyl-prop-2-ynyl)-malonic acid diethyl ester

MF: C_{27}H_{26}O_{4}
MW = 414 g.mol^{-1}

Method:
To a round bottom flask under an argon atmosphere charged with 2-(2-methylene-4-phenyl-but-3-ynyl)-2-prop-2-ynyl-malonic acid diethyl ester (1 equiv., 0.30 mmol, 100 mg) and Et$_3$N (0.1 M, 3 mL) was sequentially added PhI (1.20 equiv., 0.36 mmol, 40 µL) and PdCl$_2$(PPh$_3$)$_2$ (0.02 equiv., 0.006 mmol, 4.2 mg). The reaction was allowed to stir at room temperature for 5 min., the Cul (0.02 equiv, 0.006 mmol, 1.2 mg) was then added and the reaction was allowed to stir at room temperature overnight. Upon completion (TLC), the reaction was filtered through celite, concentrated under reduced pressure and purified by flash column chromatography.

Reference:

Purification:
Flash column chromatography (silica gel, toluene)/ R$_f$ (toluene): 0.34.

Product:
White solid.

Isolated yield:
31% isolated/ 62% (brsm).

$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz)
7.45–7.43 (m, 2H, CH-Ph), 7.38–7.36 (m, 3H, CH-Ph), 7.30–7.27 (m, 5H, CH-Ph), 5.61 (s, 1H, C=CH$_2$), 5.52 (s, 1H, C=CH$_2$), 4.28–4.12 (m, 4H, OCH$_2$CH$_3$), 3.22 (s, 2H, CH$_2$), 3.13 (s, 2H, CH$_2$), 1.24 (t, J = 7.0Hz, 6H, OCH$_2$CH$_3$).

$^{13}$C NMR (δ, ppm) (CDCl$_3$, 100 MHz)
169.6 (Cq x2, C=O), 131.6 (CH-Ph), 131.5 (CH-Ph), 128.3 (CH-Ph), 128.2 (CH-Ph), 128.2 (CH-Ph), 127.9 (CH-Ph), 126.5 (C=CH$_2$), 125.7 (Cq), 125.7 (Cq), 123.0 (Cq), 89.9 (Cq, C==C), 89.2 (Cq, C==C), 84.8 (Cq, C==C), 83.9 (Cq, C==C), 61.7 (OCH$_2$CH$_3$ x2), 57.0 (Cq), 38.9 (CH$_3$), 23.3 (CH$_3$), 14.0 (OCH$_2$CH$_3$ x2).

IR (ν, cm$^{-1}$) (CCl$_4$)
3059 (w), 2983 (m), 2939 (w), 2907 (w), 1739 (s), 1599 (w), 1491 (m), 1477 (w), 1465 (w), 1444 (m), 1426 (w), 1390 (w), 1367 (w), 1286 (w), 1260 (w), 1215 (m), 1193 (m), 1181 (m), 1097 (w), 1070 (w), 1052 (w), 1030 (w).

HRMS (El+, m/z):
Calculated: 414.1831  found: 414.1843.
Compound 4.64h was synthesized as described in the scheme below:

Scheme B.3.3.2: Synthesis of substrate 4.64h. a Synthesized by Fabien Gagosz, yields unknown. b Following references.  

**2-(2-Methylene-dec-4-ynyl)-2-prop-2-ynyl-malonic acid dimethyl ester** 4.64h

Method : See general procedure 4.4 using (1 equiv., 0.33 mmol, 55 mg) of previously prepared 2-prop-2-ynyl-malonic acid dimethyl ester and previously prepared 1-iodo-dec-4-yn-2-one (1.2 equiv., 0.39 mmol, 108 mg) in the place of propargyl bromide.

Purification : Flash column chromatography (silica gel, 9:1 PE:AcOEt) / Rf (9:1 PE:AcOEt): 0.37.

Product: yellow oil.

Isolated yield: 89%.

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz) 7.32 (s, 1H, C=CH$_2$), 5.00 (s, 1H, C=CH$_2$), 3.76 (s, 6H, OCH$_3$), 2.93 (s, 2H, CH$_2$), 2.84 (d, $J = 2.8$Hz, 2H, CH$_2$-propargyl chain), 2.82 (s, 2H, CH$_2$), 2.20-2.16 (m, 2H, CH$_2$-pentylchain), 2.05 (t, $J = 2.8$Hz, 1H, C=CH), 1.53-1.46 (m, 2H, CH$_2$-pentylchain), 1.38-1.28 (m, 4H, CH$_2$-pentylchain), 0.90 (t, $J = 7.0$Hz, 3H, CH$_3$-pentylchain).

$^{13}$C NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz) 170.4 (Cq x2, C=O), 139.4 (Cq, C=CH$_2$), 116.5 (C=CH$_2$), 83.5 (Cq, C=C), 79.0 (Cq, C=C), 76.3 (Cq, C=C), 71.9 (C=C=CH), 56.5 (Cq), 52.8 (OCH$_3$ x2), 37.2 (CH$_3$), 31.1 (CH$_3$), 28.7 (CH$_3$), 26.8 (CH$_3$), 22.8 (CH$_3$), 22.2 (CH$_3$), 18.7 (CH$_3$), 14.0 (CH$_3$).

IR ($\nu$, cm$^{-1}$) (CCl$_4$) 3314 (m), 2955 (m), 2934 (m), 2861 (w), 2843 (w), 1742 (s), 1647 (w), 1436 (m), 1327 (w), 1292 (m), 1235 (m), 1201 (m), 1179 (m), 1072 (m), 1053 (m).

HRMS (El+, m/z): Calculated: 318.1831 found: 318.1844.

2-(3-Methyl-2-methylene-buty1)-malonic acid diethyl ester

\[ \text{EtO} - \text{CO} \quad \text{OEt} \]

MF: C$_{13}$H$_{22}$O$_4$

MW = 242 g.mol$^{-1}$

Method: See general procedure 4.3 using (1 equiv., 0.54 mmol, 150 mg) of 2-(2-bromo-allyl)-malonic acid diethyl ester and (4 equiv., solution 2 M in THF, 1.08 mL) of $i$PrMgCl.

Purification: Flash column chromatography (silica gel 9:1 PE: AcOEt)/ $R_t$ (9:1 PE:AcOEt): 0.52.

Product: Transparent oil.

Isolated yield: 77%.

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz) 4.81 (s, 1H, C=CH$_2$), 4.70 (s, 1H, C=CH$_2$), 4.19 (q, $J = 7.2$Hz, 4H, OCH$_2$CH$_2$), 3.60 (t, $J = 8.0$Hz, 1H, CH), 2.64 (d, $J = 8.0$Hz, 2H, CH$_2$), 2.25 (hept, $J = 6.8$Hz, 3H, CH$_3$).
(CDCl₃, 400 MHz)  1H, CH-Pr), 1.26 (t, J = 7.2Hz, 6H, OCH₂CH₃), 1.04 (d, J = 6.8Hz, 6H, CH₃-Pr).

**¹³C NMR (δ, ppm)**

(CDCl₃, 100 MHz)

IR (ν, cm⁻¹) (CCl₄)

(THF/toluene, 1.66 mL) of CyMgCl.

**Method:**

See general procedure 4.3 using (1 equiv., 0.54 mmol, 150 mg) of 2-(2-bromo-allyl)-malonic acid diethyl ester and (4 equiv., solution 1.3 M in THF/toluene, 1.66 mL) of CyMgCl.

**Purification:**

Flash column chromatography (silica gel 95:5 PE:AcOEt)/ Rᵥ (9:1 PE:AcOEt): 0.53.

**Product:**

Transparent oil.

**Isolated yield:**

85%.

**¹H NMR (δ, ppm)**

(CDCl₃, 400 MHz)

**¹³C NMR (δ, ppm)**

(CDCl₃, 100 MHz)

IR (ν, cm⁻¹) (CCl₄)

HRMS (EI+, m/z) : Calculated: 242.1518  found: 242.1522.

2-(2-Cyclohexyl-allyl)-malonic acid diethyl ester

![Chemical Structure](image)

MF: C₁₆H₂₆O₄

MW =282 g.mol⁻¹
**HRMS (EI+, m/z):** Calculated: 282.1831  found: 282.1831.

---

**2-(2-Methylene-heptyl)-malonic acid diethyl ester**

![Structure](image)

- **MF:** C<sub>15</sub>H<sub>26</sub>O<sub>4</sub>
- **MW:** 270 g·mol<sup>-1</sup>

**Method:** See **general procedure 4.3** using (1 equiv., 0.54 mmol, 150 mg) of 2-(2-bromo-allyl)-malonic acid diethyl ester and (3 equiv., solution 2M in Et<sub>2</sub>O, 810 µL) of n-C<sub>5</sub>H<sub>11</sub>MgBr.

**Purification:** Flash column chromatography (silica gel 95:5 PE:AcOEt)/ R<sub>f</sub> (9:1 PE:AcOEt): 0.56.

**Product:** Transparent oil.

**Isolated yield:** 62%.

**<sup>1</sup>H NMR (δ, ppm)**

(CDCl<sub>3</sub>, 400 MHz)

- 4.78 (s, 1H, C=CH<sub>2</sub>), 4.74 (s, 1H, C=CH<sub>2</sub>), 4.18 (q, J = 7.2Hz, 4H, OCH<sub>2</sub>CH<sub>3</sub>), 3.57 (t, J = 7.7Hz, 1H, CH), 2.61 (d, J = 7.7Hz, 2H, CH<sub>2</sub>-pentylchain), 1.47-1.39 (m, 2H, CH<sub>2</sub>-pentylchain), 1.35-1.28 (m, 4H, CH<sub>2</sub>-pentylchain), 1.26 (t, J = 7.2Hz, 6H, OCH<sub>2</sub>CH<sub>3</sub>), 0.89 (t, J = 7.0Hz, 3H, CH<sub>3</sub>-pentylchain).

**<sup>13</sup>C NMR (δ, ppm)**

(CDCl<sub>3</sub>, 100 MHz)

- 169.2 (Cq x2, C=O), 146.0 (Cq, C=CH<sub>2</sub>), 110.8 (C=CH<sub>3</sub>), 61.4 (OCH<sub>2</sub>CH<sub>3</sub> x2), 50.7 (CH), 36.0 (CH<sub>2</sub>), 34.8 (CH<sub>2</sub>), 31.5 (CH<sub>2</sub>), 27.3 (CH<sub>2</sub>), 22.5 (CH<sub>3</sub>), 14.1 (CH<sub>3</sub>), 14.0 (CH<sub>3</sub>).

**IR (ν, cm<sup>-1</sup>) (CCl<sub>4</sub>)**

- 3080 (w), 2983 (m), 2960 (m), 2932 (m), 2873 (m), 2860 (m), 1753 (s), 1736 (s), 1647 (w), 1465 (m), 1445 (m), 1391 (m), 1369 (m), 1334 (m), 1234 (m), 1176 (m), 1150 (m), 1097 (m), 1037 (m).

**HRMS (EI+, m/z):** Calculated: 270.1831  found: 270.1827.
2-(2-Phenyl-allyl)-malonic acid diethyl ester

**MF:** C$_{16}$H$_{20}$O$_{4}$

**MW = 276 g mol$^{-1}$**

**Method:** See general procedure 4.3 using (1 equiv., 0.54 mmol, 150 mg) of 2-(2-bromo-allyl)-malonic acid diethyl ester and (4 equiv., solution 1 M in THF, 2.2 mL) of PhMgBr.

**Purification:** Flash column chromatography (silica gel 9:1 PE:AcOEt)/ $R_f$ (9:1 PE:AcOEt): 0.33.

**Product:** Transparent oil.

**Isolated yield:** 84%.

**$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz)**

7.39-7.28 (m, 5H, CH-Ph), 5.30 (s, 1H, C=CH$_2$), 5.13 (s, 1H, C=CH$_2$), 4.15 (q, $J = 7.2$ Hz, 4H, OCH$_2$CH$_3$), 3.50 (t, $J = 7.6$ Hz, 1H, CH), 3.12 (d, $J = 7.6$ Hz, 2H, CH$_2$), 1.24 (t, $J = 7.2$ Hz, 6H, OCH$_2$CH$_3$).

**$^{13}$C NMR (δ, ppm) (CDCl$_3$, 100 MHz)**

168.9 (Cq x2, C=O), 144.9 (Cq), 140.1 (Cq), 128.4 (CH-Ph), 127.7 (CH-Ph), 126.3 (CH-Ph), 114.7 (C=CH$_2$), 61.4 (OCH$_2$CH$_3$ x2), 51.0 (CH), 34.5 (CH$_2$), 14.0 (OCH$_2$CH$_3$ x2).

**IR (ν, cm$^{-1}$) (CC$_4$)**

3467 (w), 3085 (w), 3060 (w), 3025 (w), 2983 (m), 2939 (w), 2907 (w), 2873 (w), 1736 (s), 1631 (w), 1600 (w), 1575 (w), 1495 (m), 1476 (m), 1465 (m), 1446 (m), 1391 (m), 1369 (s), 1331 (s), 1301 (s), 1264 (s), 1230 (s), 1152 (s), 1110 (s), 1097 (s), 1064 (s), 1037 (s).

**HRMS (EI+, m/z):** Calculated: 276.1362  found: 276.1371.
2-(3-Methyl-2-methylene-butyl)-2-prop-2-ynyl-malonic acid diethyl ester

\[
\begin{align*}
\text{MF: } & C_{16}H_{24}O_4 \\
\text{MW: } & 280 \text{g.mol}^{-1}
\end{align*}
\]

Method:
See general procedure 4.4 using (1 equiv., 0.4 mmol, 96 mg) of 2-(3-methyl-2-methylene-butyl)-malonic acid diethyl ester.

Purification:
Flash column chromatography (silica gel 9:1 PE:AcOEt)/ \( R_f \) (9:1 PE:AcOEt): 0.38.

Product:
Transparent oil.

Isolated yield:
99%.

\[ ^1H \text{ NMR} (\delta, \text{ppm}) \] (CDCl\(_3\), 400 MHz)

- 4.93 (s, 1H, C=CH\(_2\))
- 4.83 (s, 1H, C=CH\(_2\))
- 4.24-4.15 (m, 4H, OCH\(_2\)CH\(_3\))
- 2.88 (s, 2H, CH\(_2\))
- 2.82 (d, \( J = 2.8 \text{Hz} \), 2H, CH\(_2\)-propargyl chain)
- 2.07-1.98 (m, 2H, CH\(_2\)-iPr + CH\(_2\)-iPr)
- 1.25 (t, \( J = 7.2 \text{Hz} \), 6H, OCH\(_2\)CH\(_3\))
- 1.01 (d, \( J = 6.8 \text{Hz} \), 6H, CH\(_3\)-Pr).

\[ ^{13}C \text{ NMR} (\delta, \text{ppm}) \]
(CDC\(_1\), 100 MHz)

- 170.2 (Cq x2, C=O)
- 150.6 (Cq, C=CH\(_2\))
- 111.3 (C=CH\(_2\))
- 79.5 (Cq, C\( \equiv \) C)
- 71.6 (C\( \equiv \)CH\(_2\))
- 61.6 (OCH\(_2\)CH\(_3\) x2)
- 56.7 (Cq)
- 36.6 (CH\(_2\))
- 33.6 (CH-Pr)
- 22.5 (CH\(_2\))
- 22.0 (CH\(_3\)-Pr)
- 14.0 (OCH\(_2\)CH\(_3\) x2).

IR (\( \nu, \text{cm}^{-1} \)) (CCl\(_4\))

- 3314 (s)
- 2964 (s)
- 2931 (s)
- 2872 (m)
- 1737 (s)
- 1641 (w)
- 1465 (m)
- 1367 (m)
- 1325 (m)
- 1288 (s)
- 1203 (s)
- 1097 (s)
- 1049 (s)
- 1018 (s).

HRMS (EI+, m/z):
Calculated: 280.1675   found: 280.1680.
2-(2-Cyclohexyl-allyl)-2-prop-2-ynyl-malonic acid diethyl ester

\[
\begin{align*}
\text{MF: } & C_{19}H_{28}O_4 \\
\text{MW: } & 320 \text{ g.mol}^{-1}
\end{align*}
\]

**Method:** See general procedure 4.4 using (1 equiv., 0.46 mmol, 129 mg) of 2-(2-cyclohexyl-allyl)-malonic acid diethyl ester.

**Purification:** Flash column chromatography (silica gel 95:5 PE: AcOEt)/ \( R_f \) (95:5 PE:AcOEt): 0.34.

**Product:** Transparent oil.

**Isolated yield:** 86%.

**\(^1\)H NMR** (\(\delta\), ppm) (CDCl\(_3\), 400 MHz) 
4.89 (s, 1H, C=CH\(_2\)), 4.85 (s, 1H, C=CH\(_2\)), 4.26-4.13 (m, 4H, OCH\(_2\)CH\(_3\)), 2.86 (s, 2H, CH\(_2\)), 2.82 (d, \(J = 2.5\)Hz, 2H, CH\(_2\)-propargyl chain), 2.02 (t, \(J = 2.5\)Hz, 1H, CH), 1.81-1.74 (m, 3H, CH\(_2\)-Cy), 1.67-1.56 (m, 3H, CH\(_2\)-Cy), 1.26 (t, \(J = 7.0\)Hz, 6H, OCH\(_2\)CH\(_3\)), 1.22-1.02 (m, 5H, CH-Cy).

**\(^{13}\)C NMR** (\(\delta\), ppm) (CDCl\(_3\), 100 MHz) 
170.2 (Cq x2, C=O), 149.7 (Cq, C=CH\(_2\)), 112.1 (C=C), 79.5 (Cq, C=C=CH), 71.6 (C=CH), 61.6 (OCH\(_2\)CH\(_3\) x2), 56.7 (Cq), 43.9 (CH-Cy), 36.9 (CH\(_2\)), 32.8 (CH\(_2\)), 26.9 (CH\(_2\)), 26.4 (CH\(_2\)), 22.5 (CH\(_2\)), 14.0 (OCH\(_2\)CH\(_3\) x2).

**IR** (\(\nu\), cm\(^{-1}\)) (CCl\(_4\)) 
3314 (s), 3086 (w), 2983 (s), 2929 (s), 2853 (s), 1737 (s), 1639 (w), 1477 (w), 1464 (w), 1447 (m), 1389 (w), 1367 (w), 1323 (m), 1287 (s), 1259 (s), 1241 (s), 1222 (s), 1201 (s), 1184 (s), 1133 (m), 1096 (m), 1068 (s), 1053 (s), 1017 (s).

Method: See general procedure 4.4 using (1 equiv., 0.30 mmol, 80 mg) of 2-(2-methylene-heptyl)-malonic acid diethyl ester.

Purification: Flash column chromatography (silica gel 95:5 PE:AcOEt)/ Rf (9:1 PE:AcOEt): 0.61.

Product: Transparent oil.

Isolated yield: 86%.

\[^1\text{H NMR}\] (δ, ppm) \((\text{CDCl}_3, 400 \text{ MHz})\) 4.91 (s, 1H, C=CH\(_2\)), 4.91 (s, 1H, C=CH\(_2\)), 4.27-4.13 (m, 4H, OC\(_2\)H\(_2\)CH\(_3\)), 2.83 (s, 2H, CH\(_2\)), 2.82 (d, \(J = 2.7\text{Hz}\), 2H, CH\(_2\)-propargyl chain), 2.02 (t, \(J = 2.7\text{Hz}\), 1H, CH), 1.90 (t, \(J = 7.6\text{Hz}\), 2H, CH\(_2\)-pentyl chain), 1.45-1.38 (m, 2H, CH\(_2\)-pentyl chain), 1.31-1.21 (m, 4H, CH\(_2\)-pentyl chain), 1.26 (t, \(J = 7.2\text{Hz}\), 6H, OCH\(_2\)CH\(_3\)), 0.88 (t, \(J = 7.2\text{Hz}\), 3H, CH\(_3\)-pentyl chain).

\[^{13}\text{C NMR}\] (δ, ppm) \((\text{CDCl}_3, 100 \text{ MHz})\) 170.2 (Cq x2, C=O), 144.3 (Cq, C=CH\(_2\)), 114.7 (C=CH), 79.4 (Cq, C=CH), 71.6 (Cq, C=CH), 61.6 (OC\(_2\)H\(_2\)CH\(_3\) x2), 56.6 (Cq), 37.2 (CH\(_2\)), 36.6 (CH\(_2\)), 31.5 (CH\(_2\)), 27.7 (CH\(_2\)), 22.6 (CH\(_2\)), 22.5 (CH\(_2\)), 14.0 (CH\(_3\) x2), 13.9 (CH\(_3\)).

IR (ν, cm\(^{-1}\)) (CCl\(_4\)) 3314 (m), 3078 (w), 2982 (m), 2960 (m), 2932 (m), 2873 (m), 2860 (m), 1737 (s), 1641 (w), 1465 (w), 1445 (w), 1389 (w), 1367 (w), 1323 (w), 1287 (m), 1260 (m), 1204 (s), 1183 (s), 1097 (m), 1069 (m), 1050 (m), 1017(m).

HRMS (EI+, m/z): Calculated: 308.1988 found: 308.1996.
2-(2-Phenyl-allyl)-2-prop-2-ynyl-malic acid diethyl ester

MF: C_{19}H_{22}O_4

MW = 314 g.mol^{-1}

Method: See general procedure 3.4 using (1 equiv., 0.41 mmol, 114 mg) of 2-(2-phenyl-allyl)-malonic acid diethyl ester.

Purification: Flash column chromatography (silica gel 9:1 PE: AcOEt) \( \text{Rf} \) (9:1 PE:AcOEt): 0.30.

Product: White solid.

Isolated yield: 95%.

\(^1^H\) NMR (\(\delta\), ppm) (CDCl\(_3\), 400 MHz)

7.34-7.23 (m, 5H, CH-Ph), 5.30 (s, 1H, C=CH\(_2\)), 5.29 (s, 1H, C=CH\(_2\)), 3.98-3.90 (m, 2H, OC\(_2\)H\(_2\)CH\(_3\)), 3.80-3.72 (m, 2H, OC\(_2\)H\(_2\)CH\(_3\)), 3.32 (s, 2H, CH\(_2\)), 2.75 (d, \( J = 2.6\) Hz, 2H, CH\(_2\)-propargyl chain), 2.04 (t, \( J = 2.6\) Hz, 1H, \( \equiv \text{CH} \)), 1.13 (t, \( J = 7.2\) Hz, 6H, OCH\(_2\)C\(_6\)H\(_5\)).

\(^{13}\)C NMR (\(\delta\), ppm) (CDCl\(_3\), 100 MHz)

169.5 (Cq x2, C=O), 144.0 (Cq), 141.2 (Cq), 128.0 (CH-Ph), 127.6 (CH-Ph), 126.9 (CH-Ph), 118.8 (C=CH\(_2\)), 79.4 (Cq, C \( \equiv \text{CH} \)), 71.6 (\( \equiv \text{CH} \)), 61.4 (OC\(_2\)H\(_2\)CH\(_3\) x2), 56.4 (Cq), 36.7 (CH\(_2\)), 22.3 (CH\(_2\)), 13.8 (OC\(_2\)H\(_2\)CH\(_3\) x2).

IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))

3314 (m), 3084 (w), 3059 (w), 2983 (m), 2938 (w), 2907 (w), 2873 (w), 1738 (s), 1627 (w), 1600 (w), 1575 (w), 1493 (w), 1475 (w), 1464 (w), 1445 (w), 1425 (w), 1390 (w), 1367 (w), 1322 (w), 1300 (m), 1287 (m), 1241 (m), 1207 (s), 1187 (s), 1096 (w).

HRMS (EI+, m/z) : Calculated: 314.1518 found: 314.1512.
B.3.3.2 Cycloisomerization reaction of 1,6-enynes

The previously prepared 1,6-enynes were submitted to gold catalysis using PPh$_3$ as ancillary ligand and two different counter ions, SbF$_6^-$ and NTf$_2^-$. 

\[
X = \text{C(CO}_2\text{Et)}
\]

**Scheme B.3.3.2**: Gold catalyzed transformations of enynes 4.64 and 4.81-4.84.

**General procedure 4.6.A**: To a NMR tube charged with 1,6-enyne (1 equiv.) and CDCl$_3$ (0.5 mL) was added the catalyst PPh$_3$Au(NCCH$_3$)$_2$SbF$_6$ (0.04 equiv.). The NMR tube was allowed to stand at room temperature while constantly monitored by $^1$H NMR. Upon completion, the content of the NMR tube was transferred to a round bottom flask, concentrated under reduced pressure and purified by flash column chromatography.

**General procedure 4.6.B**: To a NMR tube charged with 1,6-enyne (1 equiv.) and CDCl$_3$ (0.5 mL) at 0°C was added the catalyst PPh$_3$Au(NCCH$_3$)$_2$SbF$_6$ (0.04 equiv.). The NMR tube was allowed to stand at 0°C for an estimated duration of time, after which, the reaction mixture was quenched with Et$_3$N at 0°C before removal the NMR tube from the ice bath. When reaction completion was observed, the content of the NMR tube was transferred to a flask, concentrated under reduced pressure and purified by flash column chromatography.

**General procedure 4.6.C**: To a NMR tube charged with 1,6-enyne (1 equiv.) and CDCl$_3$ (0.5 mL) was added the catalyst PPh$_3$AuNTf$_2$ (0.04 equiv.). The NMR tube was allowed to stand at room temperature while constantly monitored by $^1$H NMR. Upon completion, the content of the NMR tube was transferred to a round bottom flask, concentrated under reduced pressure and purified by flash column chromatography.

**General procedure 4.6.D**: To a NMR tube charged with 1,6-enyne (1 equiv.) and CDCl$_3$ (0.5 mL) at 0°C was added the catalyst PPh$_3$AuNTf$_2$ (0.04 equiv.). The NMR tube was allowed to stand at 0°C for an estimated duration of time, after which, the reaction mixture was
quenched with Et₃N at 0°C before removal the NMR tube from the ice bath. When reaction completion was observed, the content of the NMR tube was transferred to a flask, concentrated under reduced pressure and purified by flash column chromatography.

3-Ethynyl-4-vinyl-cyclopent-3-ene-1,1-dicarboxylic acid diethyl ester 4.73a

\[
\begin{align*}
\text{EtO} & \quad \text{OEt} \\
\end{align*}
\]

\[
\text{MF: C}_{15}\text{H}_{18}\text{O}_4
\]

\[
\text{MW = 262 g.mol}^{-1}
\]

Method: See general procedure 4.6.A using (1 equiv., 0.02 mmol, 6.6 mg) of 2-(2-methylene-4-trimethylsilanyl-but-3-ynyl)-2-prop-2-ynyl-malonic acid diethyl ester

Purification: Flash column chromatography (silica gel 9:1 PE:AcOEt)/ Rf (9:1 PE:AcOEt): 0.35.

Product: Transparent oil.

Isolated yield: 62%.

\(^1\)H NMR (δ, ppm) (CDCl₃, 400 MHz) 6.77 (dd, J = 11.0Hz, 1H, CH=CH₂), 5.29 (d, J = 11.0Hz, 1H, CH=CH₂), 5.24 (d, J = 17.4Hz, 1H, CH=CH₂), 4.21 (q, J = 7.2Hz, 4H, OCH₂CH₃), 3.27 (s, 1H, CH), 3.23 (s, 2H, CH₂), 3.22 (s, 2H, CH₂), 1.26 (t, J = 7.2 Hz, 6H, OCH₂C₃H₇).

\(^13\)C NMR (δ, ppm) (CDCl₃, 100 MHz) 171.3 (Cq x2, C=O), 145.7 (Cq), 130.8 (CH=CH₂), 117.9 (Cq), 117.6 (CH=CH₂), 83.7 (Cq, C ≡ C), 78.7 ( CH), 61.8 (OCH₂CH₃ x2), 57.6 (Cq), 43.8 (CH₂), 39.5 (CH₂), 14.0 (OCH₂C₃H₇ x2).

IR (ν, cm\(^{-1}\)) (CCl₄) 3311 (m), 2982 (m), 2962 (m), 2929 (m), 2856 (w), 1736 (s), 1681 (w), 1465 (w), 1446 (w), 1389 (w), 1367 (w), 1259 (s), 1185 (m), 1096 (m), 1072 (m), 1017 (m).

HRMS (EI+, m/z): Calculated: 262.1205 found: 262.1213.
**3-Hept-1-ynyl-5-methylene-cyclohex-3-ene-1,1-dicarboxylic acid diethyl ester**

![Chemical structure]

MF: C$_{20}$H$_{28}$O$_4$

MW = 332 g.mol$^{-1}$

**Method:**
See general procedure 4.6.C using (1 equiv., 0.02 mmol, 6.6 mg) of 2-(2-methylene-non-3-ynyl)-2-prop-2-ynyl-malonlic acid diethyl ester.

**Purification:**
Flash column chromatography (silica gel 9:1 PE: AcOEt) / $R_f$ (9:1 PE:AcOEt): 0.29.

**Product:**
Transparent oil.

**Isolated yield:**
73%.

**$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz)**
6.35 (s, 1H, C=CH), 4.95 (s, 1H, C=CH$_2$), 4.92 (s, 1H, C=CH$_2$), 4.17 (q, $J$ = 7.2 Hz, 4H, OCH$_2$CH$_3$), 2.82 (s, 2H, CH$_2$), 2.74 (s, 2H, CH$_2$), 2.32 (t, $J$ = 7.2 Hz, 2H, CH$_2$-pentylchain), 1.57-1.50 (m, 2H, CH$_2$-pentylchain), 1.43-1.30 (m, 4H, CH$_2$-pentylchain), 1.23 (t, $J$ = 7.2Hz, 6H, OCH$_2$CH$_3$), 0.90 (t, $J$ = 7.0Hz, 3H, CH$_3$-pentylchain).

**$^{13}$C NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz)**
170.4 (Cq x2, C=O), 138.9 (Cq), 132.7 (C=C), 120.4 (Cq), 114.4 (C=CH$_2$), 92.9 (Cq, C=C=C), 81.1 (Cq, C=C=C), 61.6 (OCH$_2$CH$_3$ x2), 53.8 (Cq), 35.0 (CH$_2$), 35.0 (CH$_3$), 31.1 (CH$_3$), 28.4 (CH$_2$), 22.2 (CH$_3$), 19.5 (CH$_2$), 14.1 (CH$_3$ x2), 14.0 (CH$_3$).

**IR (v, cm$^{-1}$) (CCl$_4$)**
2960 (m), 2933 (m), 2874 (m), 2861 (m), 1738 (s), 1701 (w), 1639 (w), 1466 (w), 1458 (w), 1446 (w), 1419 (w), 1368 (w), 1300 (w), 1253 (m), 1193 (m), 1159 (w), 1098 (w), 1053 (w).

**HRMS (EI+, m/z):**
3-Hept-1-ynyl-5-methylene-cyclohex-3-ene-1,1-dicarboxylic acid diethyl ester and 3-Hept-1-ynyl-4-vinyl-cyclopent-3-ene-1,1-dicarboxylic acid diethyl ester

Method: See general procedure 4.6.A using (1 equiv., 0.03 mmol, 9 mg) of 2-(2-methylene-non-3-ynyl)-2-prop-2-ynyl-malonic acid diethyl ester.

Purification: Flash column chromatography (silica gel 9:1 PE AcOEt)/ R_f (9:1 PE:AcOEt): 0.29.

Product: Transparent oil.

Isolated yield: 58% (ratio 1.9:1).

^1H NMR (δ, ppm) (CDCl₃, 400 MHz)
- **5 membered ring:** 6.74 (dd, J = 11.0Hz, J = 17.0Hz, 1H, CH=CH₂), 5.19 (d, J = 11.0Hz, 1H, CH=CH₂), 5.18 (d, J = 17.0Hz, 1H, CH=CH₂), 4.23-4.15 (m, 4H, OCH₂CH₃), 3.17 (s, 4H, CH₂), 3.27 (t, J = 7.0Hz, 2H, CH₂-pentylchain), 1.57-1.50 (m, 2H, CH₂-pentylchain), 1.43-1.30 (m, 4H, CH₂-pentylchain), 1.25 (t, J = 7.2 Hz, 6H, OCH₂CH₃), 0.90 (t, J = 7.0Hz, 3H, CH₃-pentylchain).

^6-membered ring: 6.35 (s, 1H, C=CH), 4.95 (s, 1H, C=CH₂), 4.92 (s, 1H, C=CH₂), 4.23-4.15 (m, 4H, OCH₂CH₃), 2.82 (s, 2H, CH₂), 2.74 (s, 2H, CH₂), 2.32 (t, J = 7.2 Hz, 2H, CH₂-pentylchain), 1.57-1.50 (m, 2H, CH₂-pentylchain), 1.43-1.30 (m, 4H, CH₂-pentylchain), 1.23 (t, J = 7.2Hz, 6H, OCH₂CH₃), 0.90 (t, J = 7.0Hz, 3H, CH₃-pentylchain).

^13C NMR (δ, ppm) (CDCl₃, 100 MHz)
- **5 and 6 membered rings:** 171.6 (Cq x2, C=O), 170.4 (Cq x2, C=O), 142.1 (Cq), 138.9 (Cq), 132.7 (CH=), 131.2 (CH=), 120.5 (Cq), 119.9 (Cq), 116.1 (CH₂), 114.4 (CH₂), 97.6 (Cq, C=C=C), 92.9 (Cq, C=C=C), 81.2 (Cq, C=C=C), 75.6 (Cq, C=C=C), 61.7 (OCH₂CH₃ x2), 61.6 (OCH₂CH₃ x2), 57.5 (Cq), 53.8 (Cq), 44.3 (CH₂), 39.4 (CH₂), 35.0 (CH₂), 35.0 (CH₂), 31.1 (CH₂ x2), 28.5 (CH₂), 28.4 (CH₂), 22.2 (CH₂ x2), 19.7 (CH₂), 19.5 (CH₂), 14.1 (CH₃ x4), 14.0 (CH₃ x2).

IR (υ, cm⁻¹) (CCl₄)
- 2961 (w), 2933 (w), 2874 (w), 2861 (w), 1737 (s), 1625 (w), 1467 (w), 1458 (w), 1445 (w), 1424 (w), 1367 (w), 1299 (w), 1252 (m), 1163 (m), 1097 (m), 1075 (m), 1045 (m), 1017 (m).


MF: C₂₀H₂₈O₄
MW = 332 g·mol⁻¹
5-Methylene-3-phenylethynyl-cyclohex-3-ene-1,1-dicarboxylic acid diethyl ester

EthO
\[\text{O} \quad \text{O} \quad \text{EtO}\]

MF: C_{21}H_{22}O_4

MW = 338 g mol⁻¹

Method:
See general procedure 4.6.C using (1 equiv., 0.02 mmol, 6.8 mg) of 2-(2-methylene-4-phenyl-but-3-ynyl)-2-prop-2-ynyl-malonic acid diethyl ester.

Purification:
Flash column chromatography (silica gel 9:1 PE AcOEt)/ Rf (95:5 PE:AcOEt): 0.24.

Product:
Transparent oil.

Isolated yield:
100%.

\(^1H\) NMR (δ, ppm) (CDCl₃, 400 MHz)
7.45-7.43 (m, 2H, CH-Ph), 7.31-7.30 (m, 3H, CH-Ph), 6.53 (s, 1H, C=CH), 5.04 (s, 1H, C=CH₂), 5.02 (s, 1H, C=CH₂), 4.20 (q, J = 7.2 Hz, 4H, OCH₂CH₃), 2.87 (s, 4H, CH₂), 1.24 (t, J = 7.2 Hz, 6H, OCH₂CH₃).

\(^13C\) NMR (δ, ppm) (CDCl₃, 100 MHz)
170.3 (Cq x2, C=O), 138.8 (Cq), 134.1 (CH), 131.6 (CH), 128.3 (CH), 128.2 (CH), 123.2 (Cq), 119.7 (Cq), 115.6 (C=CH₂), 91.5 (Cq, C=C=C), 90.0 (Cq, C=C=C), 61.7 (OCH₂CH₃ x2), 53.8 (Cq), 35.0 (CH₂), 34.6 (CH₂), 14.0 (OCH₂CH₃ x2).

IR (ν, cm⁻¹) (CCl₄)
3312 (w), 3059 (w), 2981 (m), 2937 (m), 2908 (m), 1737 (s), 1597 (w), 1487 (m), 1443 (m), 1388 (m), 1366 (m), 1300 (m), 1248 (s), 1188 (s), 1097 (s), 1057 (s), 1016 (s).

HRMS (EI+, m/z):
Calculated: 338.1518 found: 338.1523.
<table>
<thead>
<tr>
<th>Method</th>
<th>See general procedure 4.6.A using (1 equiv., 0.02 mmol, 6.8 mg) of 2-(2-methylene-4-phenyl-but-3-ynyl)-2-prop-2-ynyl-malonic acid diethyl ester.</th>
</tr>
</thead>
<tbody>
<tr>
<td>Purification</td>
<td>Flash column chromatography (silica gel 9:1 PE AcOEt)/ R_f (9:1 PE:AcOEt): 0.32.</td>
</tr>
<tr>
<td>Product</td>
<td>Yellow oil.</td>
</tr>
<tr>
<td>Isolated yield</td>
<td>57% (ratio 2.55:1).</td>
</tr>
</tbody>
</table>

**1H NMR (δ, ppm)**

|                | 5 membered ring: 7.47-7.42 (m, 2H, CH-Ph), 7.33-7.30 (m, 3H, CH-Ph), 6.86 (dd, J = 10.8Hz, J = 17.6Hz, CH=CH₂), 5.29 (d, J = 10.8Hz, 1H, CH=CH₂), 5.25 (d, J = 17.6Hz, 1H, CH=CH₂), 4.25- 4.17 (m, 4H, OCH₂CH₃), 3.31 (s, 2H, CH₂), 3.25 (s, 2H, CH₂), 1.27-1.23 (m, 6H, OCH₂CH₃). |
|                | 6 membered ring: 7.47-7.42 (m, 2H, CH-Ph), 7.33-7.30 (m, 3H, CH-Ph), 6.54 (s, 1H, C=CH), 5.05 (s, 1H, C=CH₂), 5.02 (s, 1H, C=CH₂), 4.25- 4.17 (m, 4H, OCH₂CH₃), 2.87 (s, 4H, CH₂), 1.27-1.23 (m, 6H, OCH₂CH₃). |

**13C NMR (δ, ppm)**

|                | 5 and 6 membered rings: 171.4 (Cq x2, C=O), 170.3 (Cq x2, C=O), 143.9 (Cq), 138.8 (Cq), 134.1 (CH), 131.9 (CH), 131.5 (CH), 131.3 (CH), 131.1 (Cq), 128.3 (CH), 128.3 (CH), 128.2 (CH), 123.2 (Cq), 119.7 (Cq), 119.0 (Cq), 117.1 (+CH₂), 115.6 (+CH₂), 96.1 (Cq, C=CH), 91.5 (Cq, C=CH), 89.9 (Cq, C=CH), 84.4 (Cq, C=CH), 61.8 (OCH₂CH₃ x2), 61.7 (OCH₂CH₃ x2), 57.6 (Cq), 53.8 (Cq), 44.0 (CH₂), 39.6 (CH₂), 35.0 (CH₂), 34.6 (CH₂), 14.0 (OCH₂CH₃ x4). |

**IR (ν, cm⁻¹) (CCl₄)**

|                | 2983 (w), 2963 (w), 2930 (w), 2909 (w), 2856 (w), 1737 (s), 1685 (w), 1599 (w), 1491 (w), 1465 (w), 1444 (w), 1423 (w), 1389 (w), 1368 (w), 1300 (w), 1249 (s), 1186 (m), 1161 (m), 1097 (m). |

**HRMS (EI+, m/z)**

|                | Calculated: 338.1518 found: 338.1524. |
3-(4-Methoxy-phenylethynyl)-5-methylene-cyclohex-3-ene-1,1-dicarboxylic acid diethyl ester

\[
\begin{align*}
\text{MF: } C_{22}H_{24}O_5 \\
\text{MW = 368 g.mol}^{-1}
\end{align*}
\]

**Method:**
See general procedure 4.6.A or C using (1 equiv., 0.02 mmol, 7.4 mg) of 2-[4-(4-methoxy-phenyl)-2-methylene-but-3-ynyl]-2-prop-2-ynyl-malonic acid diethyl ester.

**Purification:**
Flash column chromatography (silica gel 8:2 PE AcOEt) / Rf (9:1 PE:AcOEt): 0.21.

**Product:**
White solid.

**Isolated yield:**
100%.

**\(^1H\) NMR (\(\delta\), ppm)**
(CDCl\(_3\), 400 MHz)
7.37 (d, \(J = 8.8\) Hz, 2H, CH-Ar), 6.84 (d, \(J = 8.8\) Hz, 2H, CH-Ar), 6.50 (s, 1H, C=CH), 5.02 (s, 1H, C=CH\(_2\)), 4.99 (s, 1H, C=CH\(_2\)), 4.19 (q, \(J = 7.2\) Hz, 4H, OCH\(_2\)CH\(_3\)), 3.81 (s, 3H, OCH\(_3\)), 2.86 (s, 4H, CH\(_2\)), 1.24 (t, \(J = 7.2\) Hz, 6H, OCH\(_2\)CH\(_3\)).

**\(^{13}C\) NMR (\(\delta\), ppm)**
(CDCl\(_3\), 100 MHz)
170.3 (Cq x2, C=O), 159.6 (Cq), 139.0 (Cq), 133.4 (C=CH), 133.0 (CH-Ar), 120.0 (Cq), 115.4 (Cq), 115.0 (C=CH\(_2\)), 114.0 (CH-Ar), 91.6 (Cq, CH = C), 88.8 (Cq, C = C), 61.6 (OCH\(_2\)CH\(_3\) x2), 55.3 (OCH\(_3\)), 53.9 (Cq), 35.1 (CH\(_2\)), 34.8 (CH\(_2\)), 14.0 (OCH\(_2\)CH\(_3\) x2).

**IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))**
3472 (w), 2982 (m), 2960 (m), 2932 (m), 2855 (w), 2839 (w), 2198 (w), 1730 (s), 1682 (m), 1604 (m), 1510 (s), 1465 (m), 1443 (w), 1423 (w), 1390 (w), 1367 (w), 1292 (m), 1250 (s), 1182 (s), 1172 (s), 1097 (m), 1036 (s).

**HRMS (EI+, m/z):**
Calculated: 368.1624  found: 368.1633.
5-Methylene-4-phenyl-3-phenylethynyl-cyclohex-3-ene-1,1-dicarboxylic acid diethyl ester and 3-Phenylethynyl-4-(1-phenyl-vinyl)-cyclopent-3-ene-1,1-dicarboxylic acid diethyl ester

\[
\begin{align*}
\text{EtO} & \quad \text{EtO} \\
\text{O} & \quad \text{O} \\
\text{E} & \quad \text{E} \\
\text{t} & \quad \text{Et} \\
\end{align*}
\]

\[
\text{MF: } C_{27}H_{26}O_4 \\
\text{MW = 414 g.mol}^{-1}
\]

**Method:** See general procedure 4.6.A using (1 equiv., 0.03 mmol, 12.2 mg) of 2-(2-methylene-4-phenyl-but-3-ynyl)-2-(3-phenyl-prop-2-ynyl)-malonic acid diethyl ester.

**Purification:** Flash column chromatography (silica gel 9:1 PE:AcOEt) \( R_f \) (9:1 PE:AcOEt): 0.29.

**Product:** Yellow oil.

**Isolated yield:** 91% (ratio 2:1).

**\(^1\)H NMR \((\delta, \text{ppm})\)**

(CDCl\(_3\), 400 MHz)

5 membered ring: 7.47-7.42 (m, 2H, CH-Ph), 7.38-7.27 (m, 8H, CH-Ph), 6.62 (s, 1H, C=CH\(_2\)), 6.51 (s, 1H, C=CH\(_2\)), 4.28-4.13 (m, 4H, OCH\(_2\)CH\(_3\)), 3.18 (s, 4H, CH\(_2\)), 1.18 (t, \( J = 7.2\text{Hz} \), 6H, OCH\(_2\)C\(_3\)H\(_3\)).

6 membered ring: 7.47-7.42 (m, 2H, CH-Ph), 7.38-7.27 (m, 8H, CH-Ph), 6.98 (s, 1H, C=CH\(_2\)), 6.44 (s, 1H, C=CH\(_2\)), 4.28-4.13 (m, 4H, OCH\(_2\)CH\(_3\)), 2.97 (s, 2H, CH\(_2\)), 2.95 (s, 2H, CH\(_2\)), 1.25 (t, \( J = 7.2\text{Hz} \), 6H, OCH\(_2\)C\(_3\)H\(_3\)).

**\(^{13}\)C NMR \((\delta, \text{ppm})\)**

(CDCl\(_3\), 100 MHz)

5 and 6 membered rings: 170.4 (Cq x2, C=O), 170.3 (Cq x2, C=O), 141.4 (Cq), 136.8 (Cq), 136.5 (Cq), 131.7 (CH), 131.5 (Cq), 131.4 (Cq), 131.0 (Cq), 130.3 (Cq), 129.4 (CH), 129.3 (CH), 129.1 (CH), 128.3 (CH x2), 128.2 (CH), 128.1 (CH), 127.0 (Cq), 124.4 (Cq), 123.2 (Cq), 121.4 (CH), 121.3 (CH), 120.0 (C=CH\(_3\)), 119.5 (C=CH\(_3\)), 96.1 (Cq, C Ea C), 92.2 (Cq, C Ea C), 91.8 (Cq, C Ea C), 90.5 (Cq, C Ea C), 61.7 (OCH\(_2\)CH\(_3\) x4), 54.0 (Cq), 54.0 (Cq), 36.8 (CH\(_3\)), 35.5 (CH\(_3\)), 34.9 (CH\(_2\)), 31.0 (CH\(_2\)), 14.1 (OCH\(_2\)C\(_3\)H\(_3\) x2), 14.0 (OCH\(_2\)C\(_3\)H\(_3\)), 13.9 (OCH\(_2\)C\(_3\)H\(_3\)).

**IR \((\nu, \text{cm}^{-1})\)** (CCl\(_4\))

3060 (w), 2982 (w), 2964 (w), 2930 (w), 2910 (w), 2856 (w), 1737 (s), 1598 (w), 1573 (w), 1492 (w), 1478 (w), 1465 (w), 1444 (w), 1424 (w), 1389 (w), 1367 (w), 1301 (w), 1246 (m), 1183 (m), 1097 (w), 1071 (w), 1052 (w).

**HRMS \((\text{EI}^+, \text{m/z})\):** Calculated: 414.1831 found: 414.1842.
5-Methylene-3-oct-2-ynyl-cyclohex-3-ene-1,1-dicarboxylic acid dimethyl ester and 3-Oct-2-ynyl-4-vinyl-cyclopent-3-ene-1,1-dicarboxylic acid dimethyl ester

Method: See general procedure 4.6.A using (1 equiv., 0.02 mmol, 6.4 mg) of 2-(2-methylene-dec-4-ynyl)-2-prop-2-ynyl-malonic acid dimethyl ester.

Purification: Flash column chromatography (silica gel 9:1 PE AcOEt)/ \( R_f \) (9:1 PE:AcOEt): 0.35.

Product: Yellow oil.

Isolated yield: 100% (ratio 3.33:1).

\( ^1H \) NMR (\( \delta, \text{ppm} \)) (CDCl\(_3\), 400 MHz) 5 membered ring: 6.56 (dd, \( J = 11.0 \text{Hz}, J = 17.0 \text{Hz} \), 1H, \( \text{CH}=\text{CH}_2 \)), 5.13 (d, \( J = 11.0 \text{Hz}, 1H, \text{CH}=\text{CH}_2 \)), 5.09 (d, \( J = 17.0 \text{Hz}, 1H, \text{CH}=\text{CH}_2 \)), 3.74 (s, 6H, OCH\(_3\)), 3.23 (s, 2H, CH\(_2\)), 3.18 (s, 2H, CH\(_2\)), 3.09 (s, 2H, CH\(_2\)), 2.15-2.11 (m, 2H, CH\(_2\)-pentylchain), 1.54-1.46 (m, 2H, CH\(_2\)-pentylchain) 1.41-1.25 (m, 4H, CH\(_2\)-pentylchain), 0.89 (t, \( J = 7.0 \text{Hz} \), 3H, CH\(_3\)-pentyl chain).

6 membered ring: 6.19 (s, 1H, C=CH), 4.95 (s 1H, C=CH\(_2\)), 4.93 (s 1H, C=CH\(_2\)), 3.71 (s, 6H, OCH\(_3\)), 2.97 (s, 2H, CH\(_2\)), 2.84 (s, 2H, CH\(_2\)), 2.64 (s, 2H, CH\(_2\)), 2.21-2.16 (m, 2H, CH\(_2\)-pentylchain), 1.54-1.46 (m, 2H, CH\(_2\)-pentylchain), 1.41-1.25 (m, 4H, CH\(_2\)-pentylchain), 0.90 (t, \( J = 6.8 \text{Hz} \), 3H, CH\(_3\)-pentyl chain).

\( ^{13}C \) NMR (\( \delta, \text{ppm} \)) (CDCl\(_3\), 100 MHz) Only 6 membered ring described: 171.2 (Cq x2, C=O), 139.0 (Cq), 134.5 (Cq), 124.5 (C=CH), 112.5 (C=CH\(_2\)), 83.3 (Cq, C=CH\(_2\)), 75.5 (Cq, C=CH\(_2\)), 54.4 (Cq), 52.7 (OCH\(_3\) x2), 35.7 (CH\(_2\)), 34.1 (CH\(_2\)), 31.1 (CH\(_2\)), 28.6 (CH\(_2\)), 26.8 (CH\(_2\)), 22.2 (CH\(_2\)), 18.7 (CH\(_3\)), 13.9 (CH\(_3\)).

IR (\( \nu, \text{cm}^{-1} \)) (CCl\(_4\)) 2956 (m), 2933 (m), 2874 (w), 2861 (w), 1741 (s), 1700 (w), 1436 (m), 1256 (m), 1206 (m), 1179 (m), 1102 (w), 1051 (w).

HRMS (EI+, \( m/z \)) : Calculated: 318.1831 found: 318.1830.
3-Isopropyl-4-vinyl-cyclopent-3-ene-1,1-dicarboxylic acid diethyl ester and 3-Isopropyl-5-methylene-cyclohex-3-ene-1,1-dicarboxylic acid diethyl ester

\[
\text{MF: } C_{18}H_{24}O_4 \\
\text{MW = 280 g.mol}^{-1}
\]

Method: See general procedure 4.6.C using (1 equiv., 0.02 mmol, 5.6 mg) of 2-(3-methyl-2-methylene-butyl)-2-prop-2-ynyl-malonic acid diethyl ester.

Purification: Flash column chromatography (silica gel 95:5 PE:AcOEt) / \( R_f \) (9:1 PE:AcOEt): 0.40.

Product: Transparent oil.

Isolated yield: 100% (ratio 1:0.9 : other minor isomers neglected).

\(^1\text{H NMR} (\delta, \text{ppm})\)

(CDCl\(_3\), 400 MHz)

5 membered ring: 6.61 (dd, \( J = 10.8\text{Hz}, J = 17.3\text{Hz}, 1\text{H, CH}=\text{CH}_2\)), 5.08 (d, \( J = 10.8\text{Hz}, 1\text{H, CH}=\text{CH}_2\)), 5.04 (d, \( J = 17.3\text{Hz}, 1\text{H, CH}=\text{CH}_2\)), 4.23-4.13 (m, 4H, OCH\(_2\)CH\(_3\)), 3.13 (s, 2H, CH\(_2\)), 3.05 (s, 2H, CH\(_2\)), 2.94-2.88 (m, 1H, CH\(_3\)-iPr), 1.27-1.21 (m, 6H, OCH\(_2\)CH\(_3\)), 1.06 (d, \( J = 6.8\text{Hz}, 6\text{H, CH}_3\)-Pr).

6 membered ring: 5.93 (s, 1H, C=CH), 4.84 (s, 1H, C=CH\(_2\)), 4.83 (s, 1H, C=CH\(_2\)), 4.23-4.13 (m, 4H, OCH\(_2\)CH\(_3\)), 2.82 (s, 2H, CH\(_2\)), 2.61 (s, 2H, CH\(_2\)), 2.35-2.30 (m, 1H, CH\(_3\)-Pr), 1.27-1.21 (m, 6H, OCH\(_2\)CH\(_3\)), 1.02 (d, \( J = 6.9\text{Hz}, 6\text{H, CH}_3\)-Pr).

\(^{13}\text{C NMR} (\delta, \text{ppm})\)

(CDCl\(_3\), 100 MHz)

5 and 6 membered rings: 171.2 (Cq x2, C=O), 171.0 (Cq x2, C=O), 144.8 (Cq), 144.6 (Cq), 139.9 (Cq), 129.8 (=CH), 129.6 (Cq), 121.8 (=CH), 113.6 (=CH), 111.3 (=CH), 61.5 (OCH\(_2\)CH\(_3\)), 61.4 (OCH\(_2\)CH\(_3\)), 57.1 (Cq), 54.4 (Cq), 40.4 (CH\(_2\)), 39.9 (CH\(_3\)), 36.0 (CH\(_3\)), 35.0 (CH\(_3\)-Pr), 32.1 (CH\(_3\)), 26.6 (CH\(_3\)-Pr), 21.0 (CH\(_3\)-Pr x2), 20.8 (CH\(_3\)-Pr x2), 14.1 (CH\(_3\) x2), 14.0 (CH\(_3\) x2).

\(\text{IR} (\nu, \text{cm}^{-1})\) (CCl\(_4\))

2964 (m), 2932 (m), 2872 (w), 1735 (s), 1645 (w), 1465 (w), 1446 (w), 1387 (w), 1366 (w), 1299 (m), 1251 (s), 1186 (m), 1097 (m), 1071 (m), 1052 (m), 1018 (m).

HRMS (El+, m/z): Calculated: 280.1675  found: 280.1667.
3-Cyclohexyl-4-vinyl-cyclopent-3-ene-1,1-dicarboxylic acid diethyl ester and 5-Methyl-bicyclohexyliden-5-ene-3,3-dicarboxylic acid diethyl ester

\[
\begin{align*}
\text{MF: } & \text{C}_{19}\text{H}_{28}\text{O}_4 \\
\text{MW = 320 g.mol}^{-1}
\end{align*}
\]

Method: See general procedure 4.6.A using (1 equiv., 0.02 mmol, 6.4 mg) of 2-(2-cyclohexyl-allyl)-2-prop-2-ynyl-malonlic acid diethyl ester.

Purification: Flash column chromatography (silica gel 9:1 PE:AcOEt)/ Rf (PE:AcOEt): 0.27.

Product: Transparent oil.

Isolated yield: 89% (ratio 1:1.4).

\[\text{1}^1\text{H NMR (}\delta, \text{ppm)} \text{ (CDCl}_3\text{, 400 MHz)}
\]

5 membered ring: 6.62 (dd, J = 10.4Hz, J = 17.2Hz, 1H, CH=CH\text{2}), 5.07 (d, J = 10.4Hz, 1H, CH=CH\text{2}), 5.03 (d, J = 17.2Hz, 1H, CH=CH\text{2}), 4.22-4.13 (m, 4H, OCH\text{2}CH\text{3}), 3.13 (s, 2H, CH\text{2}), 3.05 (s, 2H, CH\text{2}), 1.76-1.50 (m, 11H, CH- + CH\text{2}-Cy), 1.27-1.22 (m, 6H, OCH\text{2}CH\text{3}).

6 membered ring: 6.25 (s, 1H, C=CH), 4.22-4.13 (m, 4H, OCH\text{2}CH\text{3}), 2.82 (s, 2H, CH\text{2}), 2.55 (s, 2H, CH\text{2}), 2.26 (t, J = 5.2Hz, 4H, CH\text{2}-Cy), 1.82 (s, 3H, CH\text{3}), 1.76-1.50 (m, 6H, CH\text{2}-Cy), 1.27-1.22 (m, 6H, OCH\text{2}CH\text{3}).

\[\text{1}^3\text{C NMR (}\delta, \text{ppm)} \text{ (CDCl}_3\text{, 100 MHz)}
\]

5 and 6 membered rings: 172.2 (Cq x2, C=O), 171.4 (Cq x2, C=O), 144.1 (Cq), 135.8 (Cq), 132.1 (Cq), 130.0 (Cq), 129.8 (CH=CH\text{2}), 121.0 (Cq), 120.8 (C=\text{CH}), 113.4 (CH=CH\text{2}), 61.5 (OCH\text{2}CH\text{3} x2), 61.3 (OCH\text{2}CH\text{3} x2), 57.3 (Cq), 54.6 (Cq), 41.1 (CH\text{2}), 40.4 (CH\text{2}), 37.4 (CH), 35.9 (CH\text{2}), 33.6 (CH\text{2}), 31.3 (CH\text{2}), 31.1 (CH\text{2}), 30.7 (CH\text{2}), 29.9 (CH\text{2}), 28.3 (CH\text{2}), 28.2 (CH\text{2}), 27.0 (CH\text{2}), 26.4 (CH\text{2}), 26.1 (CH\text{2}), 26.0 (CH\text{2}), 23.9 (CH\text{3}), 14.0 (OCH\text{2}CH\text{3} x4).

\[\text{IR (v, cm}^{-1} \text{)} (\text{CCl}_4)\]

3498 (w), 2930 (s), 2855 (m), 1734 (s), 1464 (w), 1447 (w), 1367 (w), 1259 (s), 1188 (m), 1097 (m), 1071 (m), 1018 (m).

\[\text{HRMS (EI+, m/z):} \text{ Calculated: 320.1988 found: 320.1984.} \]
3-Cyclohexyl-4-vinyl-cyclopent-3-ene-1,1-dicarboxylic acid diethyl ester and 5-Methylene-bicyclohexyl-6-ene-3,3-dicarboxylic acid diethyl ester 4.86a + 4.86c

\[
\text{MF: C}_{19}\text{H}_{28}\text{O}_4
\]
\[
\text{MW} = 320 \text{ g.mol}^{-1}
\]

Method: See general procedure 4.6.B using (1 equiv., 0.02 mmol, 6.4 mg) of 2-(2-cyclohexyl-allyl)-2-prop-2-ynyl-malonic acid diethyl ester.

Purification: Flash column chromatography (silica gel, 9:1 PE :AcOEt)/ Rf (PE: AcOEt): 0.27.

Product: Transparent oil.

Isolated yield: 88% (ratio 1:1.2).

\(^1\)H NMR (δ, ppm) (CDCl\textsubscript{3}, 400 MHz) 5 membered ring: 6.62 (dd, J = 10.8Hz, J = 17.2Hz, 1H, CH=CH\textsubscript{2}), 5.07 (d, J = 10.0Hz, 1H, CH=CH\textsubscript{2}), 5.03 (d, J = 17.2Hz, 1H, CH=CH\textsubscript{2}), 4.22-4.13 (m, 4H, OCH\textsubscript{2}CH\textsubscript{3}), 3.13 (s, 2H, CH\textsubscript{3}), 3.05 (s, 2H, CH\textsubscript{3}), 1.83-1.64 (m, 11H, CH- + CH\textsubscript{2}-Cy), 1.27-1.22 (m, 6H, OCH\textsubscript{2}CH\textsubscript{3}).

6 membered ring: 5.91 (s, 1H, C=CH), 4.83 (s, 1H, C=CH\textsubscript{2}), 4.81 (s, 1H, C=CH\textsubscript{2}), 4.22-4.13 (m, 4H, OCH\textsubscript{2}CH\textsubscript{3}), 2.82 (s, 2H, CH\textsubscript{2}), 2.61 (s, 2H, CH\textsubscript{2}), 1.83-1.64 (m, 11H, CH- + CH\textsubscript{2}-Cy), 1.27-1.22 (m, 6H, OCH\textsubscript{2}CH\textsubscript{3}).

\(^{13}\)C NMR (δ, ppm) (CDCl\textsubscript{3}, 100 MHz) 5 and 6 membered rings: 172.2 (Cq x2, C=O), 171.0 (Cq x2, C=O), 144.2 (Cq), 144.1 (Cq), 140.0 (Cq), 130.0 (=CH), 129.8 (Cq), 122.1 (=CH), 113.4 (=CH\textsubscript{2}), 111.1 (=CH\textsubscript{2}), 61.5 (OCH\textsubscript{2}CH\textsubscript{3} x2), 61.4 (OCH\textsubscript{2}CH\textsubscript{3} x2), 57.3 (Cq), 54.4 (Cq), 45.4 (CH), 41.1(CH\textsubscript{2}), 40.4 (CH\textsubscript{2}), 37.4 (CH), 36.0 (CH\textsubscript{2}), 32.9 (CH\textsubscript{2}), 31.3 (CH\textsubscript{3}), 29.7 (CH\textsubscript{3}), 26.6 (CH\textsubscript{3}), 26.4 (CH\textsubscript{3}), 26.3 (CH\textsubscript{3}), 26.1 (CH\textsubscript{3}), 14.0 (OCH\textsubscript{2}CH\textsubscript{3} x2), 13.9 (OCH\textsubscript{2}CH\textsubscript{3} x2).

IR (ν, cm\textsuperscript{-1}) (CCl\textsubscript{4}) 3498 (w), 2930 (s), 2855 (m), 1734 (s), 1464 (w), 1447 (w), 1367 (w), 1259 (s), 1188 (m), 1097 (m), 1071 (m), 1018 (m).

3-Pentyl-4-vinyl-cyclopent-3-ene,1,1-dicarboxylic acid diethyl ester and 5-Methylene-3-pentyl-cyclohex-3-ene,1,1-dicarboxylic acid diethyl ester

\[
\text{MF: } C_{18}H_{28}O_4 \\
\text{MW = 308 g mol}^{-1}
\]

**Method:** See general procedure 4.6.B or D using (1 equiv., 0.02 mmol, 5.4 mg) of 2-(2-methylene-heptyl)-2-prop-2-ynyl-malonic acid diethyl ester.

**Purification:** Flash column chromatography (silica gel 9:1 PE :AcOEt)/Rf (9:1 PE:AcOEt): 0.36.

**Product:** Transparent oil.

**Isolated yield:** 93% (ratio 1:0.84; other minor isomers neglected).

**\(^1\)H NMR (δ, ppm) (CDCl\(_3\), 400 MHz):**

**5 membered ring:** 6.57 (dd, 1H, J = 11.0 Hz, CH3=CH2), 5.07 (d, J = 11.0 Hz, 1H, CH3=CH2), 5.04 (d, J = 17.0 Hz, 1H, CH3=CH2), 4.22-4.15 (m, 4H, OCH2CH3), 3.15 (s, 2H, CH2), 3.06 (s, 2H, CH2), 2.18 (t, J = 7.6 Hz, 2H, CH2-pentylchain) 1.48-1.37 (m, 2H, CH2-pentyl chain), 1.33-1.20 (m, 10H, CH2-pentylchain and OCH2CH2x2) 0.91-0.86 (m, 3H, CH3-pentylchain).

**6 membered ring:** 5.92 (s, 1H, C=CH), 4.82 (s, 1H, C=CH2), 4.80 (s, 1H, C=CH2), 4.22-4.15 (m, 4H, OCH2CH3), 2.81 (s, 2H, CH2), 2.58 (s, 2H, CH2), 2.07 (t, J = 7.6 Hz, 2H, CH2-pentylchain), 1.48-1.37 (m, 2H, CH2-pentyl chain), 1.33-1.20 (m, 10H, CH2-pentylchain and OCH2CH2x2) 0.91-0.86 (m, 3H, CH3-pentylchain).

**\(^{13}\)C NMR (δ, ppm) (CDCl\(_3\), 100 MHz):**

**5 and 6 membered rings:** 172.2 (Cq x2, C=O), 171.0 (Cq x2, C=O), 139.7 (Cq), 139.7 (Cq), 139.4 (Cq), 131.3 (Cq), 130.1 (CH=CH2), 123.8 (=CH), 113.6 (CH=CH2), 111.0 (=CH2), 61.5 (OCH2CH3 x2), 61.4 (OCH2CH3 x2), 57.2 (Cq), 54.4 (Cq), 44.2 (CH2), 40.5 (CH2), 37.4 (CH2), 35.8 (CH2), 34.3 (CH2), 31.6 (CH2), 31.6 (CH2), 27.9 (CH2), 27.6 (CH2), 27.0 (CH2), 22.5 (CH2), 22.5 (CH2), 14.0 (CH3x4), 14.0 (CH3x2).

**IR (ν, cm\(^{-1}\)) (CCl\(_4\)):** 3314 (w), 2960 (m), 2931 (s), 2873 (m), 2860 (m), 1735 (s), 1465 (w), 1367 (w), 1251 (s), 1186 (m), 1097 (m), 1071 (m), 1017 (m).

**HRMS (EI+, m/z):** Calculated: 308.1988 found: 308.1985.
5-Methylene-3-phenyl-cyclohex-3-ene-1,1-dicarboxylic acid diethyl ester, 5-Methyl-3-phenyl-cyclohexa-2,4-diene-1,1-dicarboxylic acid diethyl ester, 3-Methyl-5-phenyl-cyclohexa-2,4-diene-1,1-dicarboxylic acid diethyl ester and 3-Methyl-5-phenyl-cyclohexa-2,5-diene-1,1-dicarboxylic acid diethyl ester

\[
\begin{align*}
\text{MF: } & C_{19}H_{22}O_4 \\
\text{MW: } & 314 \\
\end{align*}
\]

**Method:**
See general procedure 4.6.A or B or C or D using (1 equiv., 0.02 mmol, 6.3 mg) of 2-((2-phenyl-allyl)-2-prop-2-ynyl-malonie acid diethyl ester.

**Purification:**
Flash column chromatography (silica gel 9:1 PE:AcOEt)/ Rf (9:1 PE:AcOEt): 0.29.

**Product:**
Transparent oil.

**Isolated yield:**
79% (ratio 1: 0.3: 0.3: 0.3).

**\(^1\)H NMR (\(\delta, \text{ppm}\))**

(CDCl\(_3\), 400 MHz) Mixture of four isomers: 7.52-7.43 (m, 8H, CH-Ph), 7.36-7.33 (m, 8H, CH-Ph), 7.31-7.27 (m, 4H, CH-Ph), 6.50 (s, 1H, C=CH first isomer), 6.13 (s, 1H, C=CH), 6.13 (s, 1H, C=CH), 6.04 (s, 2H, C=CH), 5.71 (s, 1H, C=CH), 5.71 (s, 1H, C=CH), 5.06 (s, 1H, C=CH\(_2\) first isomer), 5.02 (s, 1H, C=CH\(_2\) first isomer), 4.24-4.16 (m, 16H, OCH\(_2\)CH\(_3\) all isomers), 3.19 (s, 2H, CH\(_2\)), 3.18 (s, 2H, CH\(_2\)), 3.08 (s, 2H, CH\(_2\) first isomer), 2.94 (s, 2H, CH\(_2\) first isomer), 2.80 (s, 2H, CH\(_2\)), 1.94 (s, 3H, CH\(_3\)), 1.92 (s, 3H, CH\(_3\)), 1.92 (s, 3H, CH\(_3\)), 1.28-1.21 (m, 24H, OCH\(_2\)CH\(_3\)).

**\(^{13}\)C NMR (\(\delta, \text{ppm}\))**

(CDCl\(_3\), 100 MHz) Mixture of four isomers (eight CH, four Cq, one CH\(_3\) and one CH\(_2\) carbons can not be unambiguously assigned): 170.9 (Cq x2, C=O), 170.8 (Cq x2, C=O), 170.7 (Cq x4, C=O), 140.6 (Cq), 139.9 (Cq), 139.8 (Cq), 138.0 (Cq), 136.7 (Cq), 136.3 (Cq), 135.9 (Cq), 134.9 (Cq), 128.4 (CH), 128.4 (CH), 128.4 (CH), 127.6 (CH), 126.0 (CH), 125.9 (CH), 125.7 (CH), 125.7 (CH), 123.2 (CH), 119.6 (Cq), 117.1 (CH), 116.5 (CH), 114.2 (CH\(_2\)), 61.7 (OCH\(_2\)CH\(_3\) x4), 61.6 (OCH\(_2\)CH\(_3\) x4), 55.8 (Cq), 55.6 (Cq), 54.5 (Cq), 35.5 (CH\(_2\)), 34.4 (CH\(_2\)), 33.4 (CH\(_2\)), 32.1 (CH\(_2\)), 23.2 (CH\(_3\)), 21.5 (CH\(_3\)), 14.0 (OCH\(_2\)CH\(_3\) x8).

**IR (\(v, \text{cm}^{-1}\))** (CCl\(_4\))
3474 (w), 3061 (w), 2982 (m), 2931 (m), 2873 (w), 1735 (s), 1600 (w), 1494 (w), 1464 (w), 1446 (m), 1367 (m), 1243 (s), 1186 (s), 1096 (m), 1071 (m), 1054 (m), 1023 (m).

**HRMS (El+, m/z):**
Calculated: 314.1518 found: 314.1530.
B.3.4. Chapter 5: Hydroalkylation of Alkynyl Ethers via a Gold(I)-Catalyzed 1,5-Hydride Shift/Cyclization Sequence

Only experiments performed by this author are presented in this section. These experiments are marked with a red star (*).

B.3.4.1 Synthesis of substrates bearing C(2)-THF and 1,3-dioxolane rings

Most representative alkynes synthesized in this work having C(2)-substituted THF and dioxolane rings were synthesized as depicted in the scheme B.3.4.1. Similar structures corresponding to direct modifications from these previous substrates were synthesized in an analogous manner and are summarized below:

General procedure 5.1\textsuperscript{306}, \textit{NBS promoted cyclisation of secondary homoallylic alcohols}:

To a round bottom flask, charged with homoallyl alcohol (1 equiv.) and DCM (0.2 M) at 0°C, was added \textit{N}-bromosuccinimide (1.5 equiv.) followed by NaHCO\textsubscript{3} (1.5 equiv.). The reaction

mixture was allowed to stir at 0°C, with the temperature being slowly raised to room temperature. Such reactions were normally left overnight. After complete consumption of the starting homoallylic alcohol (TLC), water was added to the mixture, followed by extraction with Et₂O (3x), washed with brine (1x), dried (MgSO₄), and concentrated under reduced pressure. Purification by flash column chromatography afforded pure bromo C(2)-branched THFs in the stated yields.

**General procedure 5.2**

307, *monoalkylation of dimethyl malonate:*

To a round bottom flask, charged with NaH (1.2 equiv.) in toluene (1M to the alkylating agent) was slowly added dimethyl malonate (2 equiv.) at 0°C. The temperature was allowed to warm to room temperature, when the alkylating agent (1 equiv.), dissolved in DMF (1M to the alkylating agent) was introduced, followed by the addition of KI (0.5 equiv. to the alkylating agent). The reaction was then heated up to 100 °C overnight. After the complete consumption of the alkylating agent (TLC), the reaction mixture was cooled to room temperature, quenched with a saturated solution of NH₄Cl, extracted with AcOEt (3x); washed with small amounts of water (5x); dried (MgSO₄) and concentrated under reduced pressure. Purification by flash column chromatography generally afforded the pure monoalkylated malonate.

**General Procedure 5.3, propargylation of monoalkylated dimethyl malonates:**

To a round bottom flask, charged with NaH (1.5 equiv.) and THF (0.125 M) at 0°C, was added the monoalkylated dimethylmalonate (1 equiv.). The reaction mixture was allowed to warm to room temperature. Propargyl bromide (1.2 equiv.) was then added and the reaction mixture allowed to stir at room temperature. After complete consumption of the monoalkylated malonate (TLC), the reaction was quenched with a saturated solution of NH₄Cl, extracted with AcOEt (3x), dried (MgSO₄) and concentrated under reduced pressure. Purification by flash column chromatography afforded dialkylated dimethyl malonates.

General Procedure 5.4, Synthesis of ethyl butynoates derivatives:

To a round bottom flask, charged with diisopropylamine (2 equiv.) and THF (0.250 M) at 0°C was added n-BuLi (1.6M in hexanes, 2 equiv.) and allowed to stir for 15 min. at 0°C. The temperature was then cooled to -78 °C and the alkyne (1 equiv.) dissolved in THF (0.250 M) was added. The reaction mixture was allowed to stir at -78 °C for 1h. The anion was then quenched with ethyl chloroformate (2 equiv.). The reaction mixture was allowed to warm to room temperature, water was added and the reaction mixture extracted with AcOEt (3x), dried (MgSO₄) and concentrated under reduced pressure. Purification by flash column chromatography afforded the desired compounds.

General Procedure 5.5, Sonogashira coupling reaction with propargylated malonates:

To a round bottom flask under an argon atmosphere was sequentially added: alkyne (1 equiv.), piperidine (0.7 M with respect to alkyne), PdCl₂(PPh₃)₂ (35 mg, 0.05 equiv.), CuI (19 mg, 0.1 equiv.), halide (1.2 equiv.) and THF (0.7 M with respect to the alkyne). The reaction mixture was allowed to stir overnight at room temperature. Upon completion (TLC), the reaction was diluted with a saturated solution of NH₄Cl, extracted with Et₂O (3x), dried (MgSO₄), concentrated under reduced pressure and purified by flash column chromatography to afford the title compounds.

General Procedure 5.6 (same as general procedure 1.1): NBS promoted bromination of terminal alkynes: To a round bottom flask charged with the alkyne (1 equiv.) and acetone (1.2 equiv.) at 0 °C. was added N-bromo succinimide (1.2 equiv) and AgNO₃ (0.1 equiv.). The reaction mixture was allowed to warm up to room temperature. The flask was covered with an aluminium foil to avoid light exposure and the reaction was allowed to stir at room temperature until completion (TLC). The reaction was then diluted with H₂O, extracted with Et₂O (3x), dried (MgSO₄) and concentrated under reduced pressure. Purification of the crude mixture by flash column chromatography afforded the bromoalkynes in the given yields.

---

2-Prop-2-ynyl-2-(tetrahydro-furan-2-ylmethyl)-malonic acid dimethyl ester

\[
\text{MeO}^-\text{CO}_2\text{Me}
\]

MF: C_{13}H_{18}O_5

MW = 254 g.mol\(^{-1}\)

**Method:**
See **general procedure 5.2** using (1 equiv., 12 mmol, 1.98 g) of 2-bromomethyl-tetrahydro-furan followed by **general procedure 5.3** using (1 equiv., 4.6 mmol, 1.0 g) of the previously prepared monoalkylated malonate.

**Purification:**
Flash column chromatography (silica gel, 85: 15 PE: AcOEt)/ \(R_f\) (7:3 PE:AcOEt): 0.45.

**Product:**
Transparent oil.

**Isolated yield:**
34% for two steps.

**\(^1\)H NMR** (\(\delta\), ppm)

<table>
<thead>
<tr>
<th>Chem. Shift (ppm)</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>3.98-3.92</td>
<td>(m, 1H, OCH-THF ring)</td>
</tr>
<tr>
<td>3.74</td>
<td>(s, 3H, OCH(_3))</td>
</tr>
<tr>
<td>3.72</td>
<td>(s, 3H, OCH(_3))</td>
</tr>
<tr>
<td>3.70-3.64</td>
<td>(m, 2H, OCH(_2)-THF ring)</td>
</tr>
<tr>
<td>3.05</td>
<td>(dd, (J = 2.7)Hz, (J = 17.3)Hz, 1H, CH(_2)-propargyl chain)</td>
</tr>
<tr>
<td>2.92</td>
<td>(dd, (J = 2.7)Hz, (J = 17.3)Hz, 1H, CH(_2)-propargyl chain)</td>
</tr>
<tr>
<td>2.31</td>
<td>(dd, (J = 3.5)Hz, (J = 14.6)Hz, 1H, CH(_2)CHO)</td>
</tr>
<tr>
<td>2.25</td>
<td>(dd, (J = 9.6)Hz, (J = 14.6)Hz, 1H, CH(_2)CHO)</td>
</tr>
<tr>
<td>2.07-2.02</td>
<td>(m, 1H, CH(_2)-THF ring)</td>
</tr>
<tr>
<td>2.00</td>
<td>(t, (J = 2.7)Hz, 1H, CH(_2)-THF ring)</td>
</tr>
<tr>
<td>1.94-1.76</td>
<td>(m, 2H, CH(_2)-THF ring)</td>
</tr>
<tr>
<td>1.59-1.50</td>
<td>(m, 1H, CH(_2)-THF ring)</td>
</tr>
</tbody>
</table>

**\(^13\)C NMR** (\(\delta\), ppm)

<table>
<thead>
<tr>
<th>Chem. Shift (ppm)</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>170.8 (Cq, C=O)</td>
<td>170.7 (Cq, C=O)</td>
</tr>
<tr>
<td>79.3 (Cq, C=CH)</td>
<td>74.8 (OCH-THF ring)</td>
</tr>
<tr>
<td>71.3 (C=CH)</td>
<td>67.8 (OCH(_2))</td>
</tr>
<tr>
<td>52.8 (OCH(_3))</td>
<td>37.9 (CH(_2))</td>
</tr>
<tr>
<td>32.2 (CH(_3))</td>
<td>25.4 (CH(_2))</td>
</tr>
<tr>
<td>23.0 (CH(_3))</td>
<td>23.0 (CH(_2))</td>
</tr>
</tbody>
</table>

**IR** (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))

<table>
<thead>
<tr>
<th>(\nu) (cm(^{-1}))</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>3314 (s)</td>
<td>2974 (m)</td>
</tr>
<tr>
<td>2953 (s)</td>
<td>2872 (w)</td>
</tr>
<tr>
<td>2843(w)</td>
<td>1743 (s)</td>
</tr>
<tr>
<td>1457 (m)</td>
<td>1436 (s)</td>
</tr>
<tr>
<td>1360 (w)</td>
<td>1323 (w)</td>
</tr>
<tr>
<td>1287 (s)</td>
<td>1220 (s)</td>
</tr>
<tr>
<td>1200 (s)</td>
<td>1183 (s)</td>
</tr>
<tr>
<td>1121 (w)</td>
<td>1086 (s)</td>
</tr>
<tr>
<td>1052 (w)</td>
<td>1015 (w)</td>
</tr>
</tbody>
</table>

**HRMS** (EI+, m/z):
Calculated: 254.1154 found: 254.1163.
2-(3-Phenyl-prop-2-ynyl)-2-(tetrahydro-furan-2-ylmethyl)-malonic acid dimethyl ester

\[
\begin{align*}
\text{MeO} & \quad \text{CO} & \quad \text{OMe} \\
\text{O} & \quad \text{O} & \quad \text{Me}
\end{align*}
\]

MF: C_{19}H_{22}O_{5}

MW = 330 g.mol\(^{-1}\)

Method:
See general procedure 5.5 using (1 equiv., 0.5 mmol, 125 mg) of 2-prop-2-ynyl-2-(tetrahydro-furan-2-ylmethyl)-malonic acid dimethyl ester and (1.1 equiv., 0.55 mmol, 60 µL) of iodo-benzene.

Purification:
Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ \(R_f\) (7:3 PE:AcOEt): 0.44.

Product:
Orange oil.

Isolated yield:
94%.

\(^1\)H NMR (\(\delta, \text{ppm}\)) (CDCl\(_3\), 400 MHz)
7.37-7.34 (m, 2H, CH-Ph), 7.29-7.26 (m, 3H, CH-Ph), 4.05-3.99 (m, 1H, OCH-THF ring), 3.76 (s, 3H, OCH\(_3\)), 3.74 (s, 3H, OCH\(_3\)), 3.76-3.66 (m, 2H, OCH\(_2\)-THF ring), 3.26 (d, \(J = 17.4\)Hz, 1H, CH\(_2\)-propargyl chain), 3.13 (d, \(J = 17.4\)Hz, 1H, CH\(_2\)-propargyl chain), 2.38 (dd, \(J = 3.3\)Hz, \(J = 14.6\)Hz, 1H, CH\(_2\)CHO), 2.30 (dd, \(J = 9.8\)Hz, \(J = 14.6\)Hz, 1H, CH\(_2\)CHO), 2.09-2.01 (m, 1H, CH\(_2\)-THFring), 1.95-1.77 (m, 2H, CH\(_2\)-THFring), 1.60-1.52 (m, 1H, CH\(_2\)-THFring).

\(^{13}\)C NMR (\(\delta, \text{ppm}\)) (CDCl\(_3\), 100 MHz)
171.0 (Cq, C=O), 170.8 (Cq, C=O), 131.6 (CH-Ph), 128.2 (CH-Ph), 127.9 (CH-Ph), 123.3 (Cq-Ph), 84.8 (Cq, C\(\equiv\)C), 83.5 (Cq, C\(\equiv\)C), 74.8 (OCH-THF ring), 67.7 (OCH\(_2\)-THF ring), 56.1 (Cq), 52.8 (OCH\(_3\)), 52.6 (OCH\(_3\)), 38.2 (CH\(_3\)), 32.3 (CH\(_3\)), 25.5 (CH\(_3\)), 23.9 (CH\(_3\)).

IR (\(\nu, \text{cm}^{-1}\)) (CCl\(_4\))
3024 (w), 2974 (w), 2952 (m), 2872 (w), 1757 (m), 1742 (s), 1491 (w), 1436 (m), 1326 (w), 1289 (w), 1260 (w), 1218 (s), 1200 (s), 1182 (s), 1120 (w), 1087 (s), 1051 (w), 1030 (w).

HRMS (EI+, m/z):
calculated: 330.1467 found: 330.1468.
**5-Methoxycarbonyl-5-(tetrahydro-furan-2-ylmethyl)-hex-2-yndedioic acid 1-ethyl ester 6-methyl ester**

![Chemical Structure](image)

**Method:** See general procedure 5.4 using (1 equiv., 1 mmol, 240 mg) of 2-prop-2-ynyl-2-(tetrahydro-furan-2-ylmethyl)-malonic acid dimethyl ester.

**Purification:** Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ Rf (7:3 PE AcOEt): 0.29.

**Product:** Yellow pale oil.

**Isolated yield:** 51%.

**$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz)**

4.19 (q, J = 7.1Hz, 2H, OCH$_2$CH$_3$), 3.96-3.89 (m, 1H, OCH-THF ring), 3.75 (s, 3H, OCH$_3$), 3.73 (s, 3H, OCH$_3$), 3.73-3.64 (m, 2H, OCH$_2$-THF ring), 3.23 (d, J = 17.6Hz, 1H, CH$_2$-propargyl chain), 3.07 (d, J = 17.6Hz, 1H, CH$_2$-propargyl chain), 2.30-2.22 (m, 2H, CH$_2$CHO), 2.08-2.00 (m, 1H, CH$_2$-THF ring), 1.94-1.76 (m, 2H, CH$_2$-THF ring), 1.58-1.51 (m, 1H, CH$_2$-THF ring), 1.28 (t, J = 7.1Hz, 3H, OCH$_2$CH$_3$).

**$^{13}$C NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz)**

170.2 (Cq, C=O), 170.1 (Cq, C=O), 153.3 (Cq, C=O), 83.8 (Cq, C=C=C), 75.5 (Cq, C=C=C), 74.7 (OCH-THF ring), 67.8 (OCH$_2$-THF ring), 61.8 (OCH$_2$CH$_3$), 55.6 (Cq), 52.9 (OCH$_3$), 52.8 (OCH$_3$), 38.1 (CH$_2$), 32.2 (CH$_2$), 25.3 (CH$_2$), 23.2 (CH$_3$), 13.9 (OCH$_2$CH$_3$).

**IR ($\nu$, cm$^{-1}$) (CCl$_4$)**

2979 (m), 2954 (m), 2906 (w), 2873 (w), 2240 (m), 1744 (s), 1716 (s), 1458 (w), 1436 (m), 1366 (w), 1322 (w), 1253 (s), 1220 (s), 1182 (s), 1120 (w), 1086 (s), 1015 (w).

**HRMS (EI+, m/z):**

2-(5-Methyl-tetrahydro-furan-2-ylmethyl)-2-prop-2-ynyl-malonic acid dimethyl ester

**MF:** C_{14}H_{20}O_{5}  
**MW:** 268 g.mol^{-1}

**Method:** See **general procedure 5.1** using (1 equiv., 19 mmol, 1.90 g) of hex-5-en-2-ol followed by **general procedure 5.2** using (1 equiv., 13.5 mmol, 2.42 g) of 2-bromomethyl-5-methyl-tetrahydro-furan, followed by **general procedure 5.3** using (1 equiv., 4.10 mmol, 944 mg) of the previously monoalkylated malonate.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt)/R_{f} (9:1 PE: AcOEt): 0.20.

**Product:** Transparent oil.

**Isolated yield:** 35% for three steps (ratio cis: trans 1:2)

- **{H NMR (δ, ppm)} (CDCl, 400 MHz)**
  - **Major diasteroisomer (trans):** 4.16-4.10 (m, 1H, OCH-THF ring), 3.98-3.87 (m, 1H, OCH-THF ring), 3.73 (s, 3H, OCH$_3$), 3.72 (s, 3H, OCH$_3$), 3.06 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH$_2$-propargyl chain), 2.93 (dd, J = 2.7Hz, J =17.3Hz, 1H, CH$_2$-propargyl chain), 2.38-2.20 (m, 2H, CH$_2$-THF ring), 1.99 (t, J = 2.7Hz, 1H, CH$_2$-THF ring), 1.95-1.87 (m, 1H, OCH$_2$-THF ring), 1.62-1.53 (m, 1H, CH$_2$-THF ring), 1.46-1.35 (m, 1H, CH$_2$-THF ring), 1.13 (d, J = 6.1Hz, 3H, CH$_3$).
  - **Minor diasteroisomer (cis):** 3.98-3.87 (m, 2H, OCH-THF ring x2), 3.73 (s, 3H, OCH$_3$), 3.71 (s, 3H, OCH$_3$), 3.05 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH$_2$-propargyl chain), 2.92 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH$_2$-propargyl chain) 2.38-2.20 (m, 2H, CH$_2$CHO), 1.99 (t, J = 2.7Hz, 1H, CH$_2$-THF ring), 1.95-1.87 (m, 1H, CH$_2$-THFring), 1.62-1.53 (m, 2H, CH$_2$-THFring), 1.46-1.35 (m, 1H, CH$_2$-THFring), 1.15 (d, J = 6.1Hz, 3H, CH$_3$).

- **{C NMR (δ, ppm)} (CDCl, 100 MHz)**
  - **Major diasteroisomer (trans):** 170.7 (Cq, C=O), 170.7 (Cq, C=O), 79.4 (Cq, C=CH), 74.2 (OCH), 74.2 (OCH), 71.2 (C=CH), 55.7 (Cq), 52.8 (OCH$_3$), 52.6 (OCH$_3$), 38.3 (CH$_2$), 33.5 (CH$_2$), 32.7 (CH$_2$), 23.0 (CH$_2$), 21.0 (CH$_3$).
  - **Minor diasteroisomer (cis):** 170.8 (Cq, C=O), 170.7 (Cq, C=O), 77.2 (Cq, C=CH), 75.9 (OCH), 74.8 (OCH), 71.3 (C=CH), 55.7 (Cq), 52.8 (OCH$_3$), 52.6 (OCH$_3$), 38.9 (CH$_2$), 32.6 (CH$_2$), 32.3 (CH$_2$), 22.9 (CH$_2$), 21.3 (CH$_3$).

- **IR (v, cm$^{-1}$) (CCl$_4$)** 3314 (s), 2971 (s), 2953 (s), 2931 (s), 2871 (m), 1758 (s), 1743 (s), 1457 (m), 1437 (s), 1377 (m), 1344 (w), 1323 (m), 1285 (s), 1253 (s), 1222 (s), 1200 (s), 1183 (s), 1151(m), 1120 (m), 1095 (s).

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**HRMS (EI+, m/z)**: Calculated 268.1311 found: 268.1309.

**2-(5-Phenyl-tetrahydro-furan-2-ylmethyl)-2-prop-2-ynyl-malonic acid dimethyl ester** 5.17g

![结构式]

**MF**: C_{19}H_{21}O_{5}

**MW**: 329 g.mol^{-1}

**Method**: See *general procedure 5.1* using (1 equiv., 8.6 mmol, 1.40 g) of 1-phenylpent-4-en-1-ol followed by *general procedure 5.2* using (1 equiv., 7.5 mmol, 1.80 g) of 2-bromomethyl-5-phenyl-tetrahydrofuran followed by *general procedure 5.3* using (1 equiv, 2.3 mmol, 680 mg) of the previously prepared monoalkylated malonate.

**Purification**: Flash column chromatography (silica gel, 8:2 PE:AcOEt)/R_{f}. (8:2 PE:AcOEt): 0.44.

**Product**: Transparent oil.

**Isolated yield**: 31% for three steps (ratio cis : trans 1:2).

**¹H NMR (δ, ppm)**  
(CDCl₃, 400 MHz)

**Major diastereoisomer (trans)**: 7.32-7.20 (m, 5H, CH-Ph), 4.91 (dd, J = 6.4Hz, J = 7.6, 1H, OCH-THF ring), 4.34-4.26 (m, 1H, OCH-THF ring), 3.73 (s, 3H, OCH₃), 3.64 (s, 3H, OCH₃), 3.11 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH₂-propargyl chain), 2.99 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH₂-propargyl chain), 2.40-2.37 (m, 2H, CH₂CHO), 2.36-2.34 (m, 1H, CH₂-THF ring), 2.29-2.17 (m, 1H, CH₂-THF ring), 2.02 (t, J = 2.7Hz, 1H, CH₂-THF ring).

**Minor diastereoisomer (cis)**: 7.32-7.20 (m, 5H, CH-Ph), 4.84 (t, J = 7.2Hz, 1H, OCH-THF ring), 4.10-4.03 (m, 1H, OCH-THF ring), 3.74 (s, 3H, OCH₃), 3.59 (s, 3H, OCH₃), 3.09 (dd, J = 2.7Hz, J = 17.2Hz, 1H, CH₂-propargyl chain), 2.97 (dd, J = 2.7Hz, J = 17.2Hz, 1H, CH₂-propargyl chain), 2.55-2.43 (m, 2H, CH₂CHO), 2.29-2.17 (m, 1H, CH₂-THF ring), 2.15-2.07 (m, 1H, CH₂-THF ring), 2.03 (t, J = 2.7Hz, 1H, CH₂-THF ring).

**¹²C NMR (δ, ppm)**  
(CDCl₃, 100 MHz)

**Major diastereoisomer (trans)**: 170.6 (Cq x2, C=O), 143.4 (Cq, Ph), 128.2 (CH-Ph), 125.3 (CH-Ph), 127.0 (CH-Ph), 79.8 (OCH-THF ring), 79.4 (Cq, CH), 75.1 (OCH-THF ring), 71.3 (CH), 55.8 (Cq), 52.8 (OCH₃), 52.7 (OCH₃), 38.4 (CH₂), 34.7 (CH₂), 32.9 (CH₂), 23.1 (CH₂). **Minor diastereoisomer**
(cis): 170.6 (Cq x2, C=O), 143.1 (Cq, Ph), 128.0 (CH-Ph), 126.9 (CH-Ph), 125.8 (CH-Ph), 81.3 (OCH-THF ring), 79.3 (Cq, C=CH), 75.5 (OCH-THF ring), 71.4 (CH), 55.8 (Cq), 52.8 (OCH3), 52.6 (OCH3), 38.3 (CH2), 34.0 (CH2), 32.1 (CH2), 23.0 (CH2).

IR (v, cm⁻¹) (CCl4) 3433 (w), 3313 (m), 3065 (w), 3031 (w), 2953 (m), 2875 (w), 2843 (w), 1758 (s), 1742 (s), 1692 (w), 1493 (w), 1449 (m), 1437 (m), 1321 (m), 1284 (m), 1251 (m), 1222 (s), 1200 (s), 1183 (s), 1087 (s), 1038 (m), 1027 (m).

HRMS (EI+, m/z): Calculated: 329.1389 found: 329.1395.

2-(5-Methoxycarbonylmethyl-tetrahydro-furan-2-ylmethyl)-2-prop-2-ynyl-malonic acid dimethyl ester 5.17h

Method: See general procedure 5.1 using (1 equiv., 16.3 mmol, 2.58 g) of 3-hydroxy-hept-6-enoic acid methyl ester followed by general procedure 5.2 using (1 equiv., 4.22 mmol, 1.0 g) of (5-bromomethyl-tetrahydro-furan-2-yl)-acetic acid methyl ester followed by general procedure 5.3 using (1 equiv., 0.92 mmol, 265 mg) of the previously prepared monoalkylated malonate.

Purification: Flash column chromatography (silica gel, 7:3 PE:AcOEt) / Rf (7:3 PE:AcOEt) : 0.34.

Product: Transparent oil.

Isolated yield: 15% for three steps (ratio cis: trans: 1:3)

¹H NMR (δ, ppm) (CDCl₃, 400 MHz)

(CDCl₃, 400 MHz)

Major diasteroisomer (trans): 4.30-4.23 (m, 1H, OCH-THFring), 4.13-4.07 (m, 1H, OCH-THFring), 3.72 (s, 3H, OCH3), 3.71 (s, 3H, OCH3), 3.67 (s, 3H, OCH3), 3.02 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH2-propargyl chain), 2.90 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH2-propargyl chain), 2.51 (dd, J = 7.1Hz, J = 14.9Hz, 1H, CH2), 2.38 (dd, J = 6.2Hz, J = 14.9Hz, 1H, CH2), 2.31-2.19 (m, 2H, CH2), 2.16-2.07 (m, 2H, CH2-THFring), 1.99 (t, J = 2.7Hz, 1H, CH=CH), 1.66-1.49 (m, 2H, CH2-THFring). Minor diasteroisomer: (cis): 4.22-4.15 (m, 1H, OCH-THFring), 3.95-3.88 (m, 1H, OCH-THFring), 3.72 (s, 3H, OCH3), 3.70 (s, 3H, OCH3), 3.66 (s, 3H, OCH3), 3.01 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH2-propargyl chain), 2.89 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH2-propargyl chain), 2.55 (dd, J = 6.8Hz, J = 15.0Hz, 1H, CH2), 2.41 (dd, J = 6.2Hz, J = 15.0Hz, 1H, CH2), 2.31-
2.19 (m, 2H, CH$_2$), 2.05-2.02 (m, 2H, CH$_2$-THFring), 1.99 (t, $J = 2.7$ Hz, 1H, CH), 1.66-1.49 (m, 2H, CH$_2$-THFring).

$^{13}$C NMR (δ, ppm)  
*(CDCl$_3$, 100 MHz)*  
**Major diasteroisomer (trans):** 171.5 (Cq, C=O), 170.5 (Cq x2, C=O), 79.3 (Cq, C=CH), 74.9 (OCH-THF ring), 74.6 (OCH-THF ring), 71.2 (C=CH), 55.7 (Cq), 52.8 (OCH$_3$), 52.6 (OCH$_3$), 51.5 (OCH$_3$), 40.7 (CH$_2$), 38.0 (CH$_2$), 32.4 (CH$_3$), 31.4 (CH$_3$), 23.0 (CH$_2$).  
**Minor diasteroisomer (cis):** 171.4 (Cq, C=O), 170.5 (Cq, C=O), 170.4 (Cq, C=O), 79.2 (Cq, C=CH), 75.8 (OCH-THF ring), 75.2 (OCH-THF ring), 71.3 (CH$_2$), 55.6 (Cq), 52.8 (OCH$_3$), 52.6 (OCH$_3$), 51.5 (OCH$_3$), 40.6 (CH$_2$), 38.5 (CH$_2$), 31.9 (CH$_2$), 30.7 (CH$_2$), 22.9 (CH$_2$).

IR (ν, cm$^{-1}$) (CCl$_4$)  
3314 (m), 2953 (m), 2879 (w), 2843 (w), 1743 (s), 1458 (m), 1437 (s), 1377 (w), 1351 (w), 1321 (m), 1285 (s), 1200 (s), 1182 (s), 1084 (s), 1050 (m).

HRMS (El+, m/z) :  
Calculated: 326.1366  found: 326.1368.

**2-(Octahydro-benzofuran-2-ylmethyl)-2-prop-2-ynyl-malonic acid dimethyl ester**  

MF: C$_{17}$H$_{24}$O$_5$  
MW = 308 g.mol$^{-1}$

**Method :** See *general procedure 5.1* using (1 equiv., 7.1 mmol, 1.0 g) of *syn*-2-allylcyclohexanol followed by *general procedure 5.2* using (1 equiv., 5.75 mmol, 1.26 g) of 2-bromomethyl-octahydro-benzofuran, followed by *general procedure 5.3* using (1 equiv., 3.1 mmol, 823 mg) of the previously prepared monoalkylated malonate.

**Purification :** Flash column chromatography (silica gel, 95:5 toluene: AcOEt)/ R$_f$ (8: 2 PE: AcOEt): 0.36.

**Product :** Pale yellow oil.

**Isolated yield :** 32% for three steps (ratio diasteroisomers *cis* : *trans* 1:1.7)

$^1$H NMR (δ, ppm)  
*(CDCl$_3$, 400 MHz)*  
**Major diasteroisomer (trans):** 4.24-4.18 (m, 1H, OCH-THF ring), 3.80 (dd, $J = 3.5$ Hz, $J = 8.6$ Hz, 1H, OCH-THF ring), 3.73 (s, 3H, OCH$_3$), 3.71 (s, 3H, OCH$_3$), 3.06 (dd, $J = 2.7$ Hz, $J = 17.3$ Hz, 1H, CH$_2$-propargyl chain), 2.93 (dd, $J = 2.7$ Hz,
$J = 17.3$ Hz, 1H, CH$_2$-propargyl chain), 2.30 (dd, $J = 8.5$ Hz, $J = 14.5$ Hz, 1H, CH$_2$CHO), 2.24 (dd, $J = 3.5$ Hz, $J = 14.5$ Hz, 1H, CH$_2$CHO), 2.04-1.96 (m, 1H, CH$_2$-octahydro-benzofuran ring), 1.99 (t, $J = 2.7$ Hz, 1H, CH$_2$-octahydro-benzofuran ring), 1.84 (dd, $J = 1.8$ Hz, $J = 7.0$ Hz, $J = 12.4$ Hz, 1H, CH$_2$-octahydro-benzofuran ring), 1.65 (td, $J = 7.0$ Hz, $J = 12.4$ Hz, 1H, CH$_2$-octahydro-benzofuran ring), 1.56-1.31 (m, 7H, CH$_2$-octahydro-benzofuran ring), 1.22-1.12 (m, 1H, octahydro-benzofuran ring).

**Minor diastereoisomer (cis):**
3.96-3.89 (m, 1H, OCH-THF ring), 3.75-3.73 (m, 1H, OCH-THF ring), 3.74 (s, 3H, OCH$_3$), 3.71 (s, 3H, OCH$_3$), 3.08 (dd, $J = 2.7$ Hz, $J = 17.3$ Hz, 1H, CH$_2$-propargyl chain), 2.92 (dd, $J = 2.7$ Hz, $J = 17.3$ Hz, 1H, CH$_2$-propargyl chain), 2.40 (d, $J = 6.7$ Hz, 2H, CH$_2$CHO), 2.17-2.10 (m, 1H, CH$_2$-octahydro-benzofuran ring), 1.99 (t, $J = 2.6$ Hz, 1H, CH$_2$-octahydro-benzofuran ring), 1.79-1.70 (m, 2H, CH$_2$-octahydro-benzofuran ring), 1.56-1.31 (m, 7H, CH$_2$-octahydro-benzofuran ring), 1.22-1.12 (m, 1H, CH$_2$-octahydro-benzofuran ring).

**13C NMR (δ, ppm)**

**Major diastereoisomer (trans):**
170.8 (Cq, C=O), 170.8 (Cq, C=O), 77.8 (CH), 75.8 (OCH-THF ring), 72.5 (OCH-THF ring), 71.1 (Cq, CH), 55.8 (Cq), 52.8 (OCH$_3$), 52.5 (OCH$_3$), 39.7 (CH$_3$), 39.3 (CH$_2$), 38.3 (CH, octahydro-benzofuran ring), 28.1 (CH$_2$), 27.8 (CH$_2$), 24.3 (CH$_2$), 23.0 (CH$_2$), 20.4 (CH$_2$).

**Minor diastereoisomer (cis):**
170.8 (Cq, C=O), 170.7 (Cq, C=O), 79.5 (CH), 75.8 (OCH-THF ring), 73.6 (OCH-THF ring), 71.2 (Cq, CH), 55.9 (Cq), 52.8 (OCH$_3$), 52.6 (OCH$_3$), 39.8 (CH$_3$), 38.4 (CH$_2$), 37.6 (CH), 28.8 (CH$_2$), 28.7 (CH$_2$), 23.8 (CH$_2$), 22.8 (CH$_2$), 21.3 (CH$_2$).

**IR (ν, cm$^{-1}$) (CCl$_4$)**
3314 (s); 3000 (w); 2936 (s); 2856 (m); 1743 (s); 1457 (m); 1436 (s); 1322 (m); 1286 (s); 1252 (s); 1219 (s); 1200 (s); 1183 (s); 1156 (s); 1118 (m); 1091 (s); 1067 (m); 1009 (w).

**HRMS (EI+, m/z):**
Calculated for 308.1624 found: 308.1634.

**2-(Octahydro-benzofuran-2-ylmethyl)-2-prop-2-ynyl-malonate dimethyl ester**

**MF:** C$_{17}$H$_{24}$O$_5$

**MW:** 308 g.mol$^{-1}$

**Method:**
See general procedure 5.1 using (1 equiv., 7.14 mmol, 1.0 g) of anti-2-allyl-cyclohexanol followed by general procedure 5.2 using (1 equiv., 3.9 mmol, 850 mg) of 2-bromomethyl-octahydro-benzofuran followed by general procedure 5.3 using (1 equiv., 1.11 mmol, 300 mg) of the previously prepared monoalkylated malonate.
Purification: Flash column chromatography (silica gel, 95:5 toluene:AcOEt) / Rf (95:5 toluene: AcOEt): 0.30.

Product: Transparent oil.

Isolated yield: 16% for three steps (ratio 93:7 in favor of drawn diastereoisomer).

$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz) Only major diastereoisomer described: 4.18-4.11 (m, 1H, OCH-THF ring), 3.73 (s, 3H, OCH$_3$), 3.70 (s, 3H, OCH$_3$), 3.06 (dd, J = 2.7Hz, J = 17.3Hz, 1H, CH$_2$-propargyl chain), 3.05-2.95 (m, 1H, CHO-THF ring), 2.93 (dd, J = 10.2Hz, J = 14.7Hz, 1H, CH$_2$CHO), 2.31-2.63 (m, 1H, CH$_2$CHO), 2.30 (td, J = 6.6Hz, J = 11.6Hz, 1H, CH), 1.99 (t, J = 2.7Hz, 1H, CH$_2$-octahydro-benzofuran ring), 1.94-1.89 (m, 2H, CH$_2$-octahydro-benzofuran ring), 1.79-1.74 (m, 1H, CH$_2$-octahydro-benzofuran ring), 1.72-1.67 (m, 1H, CH$_2$-octahydro-benzofuran ring), 1.39-1.30 (m, 1H, CH$_2$-octahydro-benzofuran ring), 1.27-1.02 (m, 5H, CH$_2$-octahydro-benzofuran ring).

$^{13}$C NMR (δ, ppm) (CDCl$_3$, 100 MHz) Only major diastereoisomer described: 170.8 (Cq, C=O), 170.7 (Cq, C=O), 81.5 (OCH-THF ring), 79.3 (Cq, C=CH), 73.5 (OCH-THF ring), 71.2 (CH), 55.8 (Cq), 52.8 (OCH$_3$), 52.6 (OCH$_3$), 46.0 (CH), 39.1 (CH$_2$), 38.7 (CH$_2$), 31.2 (CH$_2$), 29.0 (CH$_2$), 25.7 (CH$_2$), 24.3 (CH$_2$), 22.8 (CH$_2$).

IR (ν, cm$^{-1}$) (CCl$_4$) 3314 (m), 2999 (w), 2937 (m), 2858 (m), 1758 (s), 1743 (s), 1456 (m), 1377 (w), 1353 (w), 1284 (m), 1250 (m), 1221 (s), 1200 (s), 1183 (s), 1143 (m), 1109 (w), 1068 (s), 1049 (w)

HRMS (EI+, m/z) : Calculated: 308.1624 Found: 308.1613.

2-[1,3]Dioxolan-2-ylmethyl-2-prop-2-ynyl-malonic acid dimethyl ester 5.17I

Method: See general procedure 5.2 using (1 equiv., 10 mmol, 1.67 g) of 2-bromomethyl-[1,3]dioxolane followed by general procedure 5.3 using (1 equiv., 5 mmol, 1.10 g) of the previously monoalkylated malonate.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt) / Rf (7:3 PE:AcOEt): 0.26.

Product: Transparent solid.
Isolated yield: 42% for two steps.

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz) 4.99 (t, $J = 4.8$Hz, 1H, OCHO-dioxolane ring), 3.93-3.90 (m, 2H, OCH$_2$-dioxolane ring), 3.32-3.79 (m, 2H, OCH$_2$-dioxolane ring), 3.73 (s, 6H, OCH$_3$), 2.98 (d, $J = 2.7$Hz, 2H, CH$_2$-propargyl chain), 2.49 (d, $J = 4.8$Hz, 2H, CH$_2$CHO$_2$), 2.03 (t, $J = 2.7$Hz, 1H, CH$_2$).

$^{13}$C NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz) 170.2 (Cq x2, C=O), 101.6 (OCHO-dioxolane ring), 79.0 (Cq, CH), 71.5 (CH), 64.8 (OCH$_2$-dioxolane ring), 54.6 (Cq), 52.8 (OCH$_3$x2), 35.6 (CH$_2$), 23.6 (CH$_2$).

IR ($\nu$, cm$^{-1}$) (CCl$_4$) 3314 (w), 2954 (w), 2336 (w), 1743 (s), 1436 (w), 1200 (w).

HRMS (EI+, m/z): Calculated: 256.0947 found: 256.0940.

5-Methoxycarbonyl-5-(5-methyl-tetrahydro-furan-2-ylmethyl)-hex-2-ynedioic acid 1-ethyl ester 6-methyl ester

$$\text{MF: C}_{17}\text{H}_{24}\text{O}_7$$

$$\text{MW} = 340 \text{ g.mol}^{-1}$$

Method: See general procedure 5.4 using (1 equiv., 0.82 mmol, 220 mg) of 2-(5-methyl-tetrahydro-furan-2-ylmethyl)-2-prop-2-ynyl-malonic acid dimethyl ester.


Product: Yellow pale oil.

Isolated yield: 48% (ratio cis: trans 1:2).

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz) Major diastereoisomer (trans): 4.19 (q, $J = 7.1$Hz, 2H, OCH$_2$CH$_3$), 4.13-4.07 (m, 1H, OCH-THF ring), 3.99-3.85 (m, 1H, OCH-THF ring), 3.75 (s, 3H, OCH$_3$), 3.73 (s, 3H, OCH$_3$), 3.24 (d, $J = 17.6$Hz, 1H, CH$_2$-propargyl chain), 3.08 (d, $J = 17.6$Hz, 1H, CH$_2$-propargyl chain), 2.36-2.18 (m, 2H, CH$_2$CHO), 2.16-2.09 (m, 1H, CH$_2$-THF ring), 2.07-1.97 (m, 1H, CH$_2$-THF ring), 1.63-1.53 (m, 1H, CH$_2$-THF ring), 1.44-1.34 (m, 1H, CH$_2$-THF ring), 1.29 (t, $J = 7.1$Hz, 3H, OCH$_2$CH$_3$), 1.13 (d, $J = 6.1$Hz, 3H, CH$_3$). Minor diastereoisomer (cis): 4.19 (q, $J = 7.1$Hz,
2H, OCHOCH₃), 4.13-4.07 (m, 1H, OCH-THF ring), 3.79-3.85 (m, 1H, OCH-THF ring), 3.75 (s, 3H, OCH₃), 3.72 (s, 3H, OCH₃), 3.23 (d, J = 17.6Hz, 1H, CH₂-propargyl chain), 3.07 (d, J = 17.6Hz, 1H, CH₂-propargyl chain), 2.36-2.18 (m, 2H, CH₂CHO), 2.07-1.97 (m, 1H, CH₂-THF ring), 1.95-1.87 (m, 1H, CH₂-THF ring), 1.63-1.53 (m, 1H, CH₂-THF ring), 1.44-1.34 (m, 1H, CH₂-THF ring), 1.29 (t, J = 7.1Hz, 3H, OCH₂C₂H₅), 1.15 (d, J = 6.2Hz, 3H, CH₃).

13C NMR (δ, ppm) 
(CDCl₃, 100 MHz) 

Major diasteroisomer (trans): 170.2 (Cq x₂, C=O), 153.3 (Cq, C=O), 83.9 (Cq, C=C=C), 74.4 (OCH₂CH₃), 55.6 (Cq), 53.0 (OCH₃), 52.7 (OCH₃), 38.4 (CH₂), 33.4 (CH₂), 32.8 (CH₂), 23.2 (CH₂), 20.9 (CH₃), 13.9 (OCH₂C₂H₅).

Minor diasteroisomer (cis): 170.3 (Cq x₂, C=O), 153.3 (Cq, C=O), 84.0 (Cq, C=C=C), 76.0 (OCH₂CH₃), 75.5 (Cq, C=C=C), 74.7 (OCH₂CH₃), 61.8 (OCHOCH₃), 55.6 (Cq), 52.9 (OCH₃), 52.7 (OCH₃), 39.1 (CH₂), 32.5 (CH₂), 23.2 (CH₂), 21.2 (CH₂), 13.9 (OCHOCH₃).

IR (υ, cm⁻¹) (CCl₄) 
2954 (m), 2906 (w), 2873 (w), 2843 (w), 2240 (m), 1744 (s), 1716 (s), 1457 (m), 1436 (s), 1376 (m), 1367 (m), 1321 (m), 1253 (s), 1223 (s), 1201 (s), 1183 (s), 1152 (s), 1094 (s), 1016 (m).

HRMS (EI+, m/z): Calculated.: 340.1522 found: 340.1520.

5-(2,3-Dihydro-benzofuran-2-ylmethyl)-5-methoxycarbonyl-hex-2-yneedioic acid 1-ethyl ester 

5.17n

**Method:** See general procedure 5.4 using (1 equiv., 0.71 mmol, 215 mg) of 2-(2,3-dihydro-benzofuran-2-ylmethyl)-2-prop-2-ynyl-malonic acid dimethyl ester.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt)/ Rf (9:1 PE:AcOEt): 0.25

**Product:** Transparent oil.

**Isolated yield:** 53%.

1H NMR (δ, ppm) 
(CDCl₃, 400 MHz) 
7.16 (d, J = 7.5Hz, 1H, CH-Ar), 7.08 (t, J = 7.5Hz, 1H, CH-Ar), 6.83 (t, J = 7.5Hz, 1H, CH-Ar), 6.66 (d, J = 7.5Hz, 1H, CH-Ar), 4.87 (dddd, J = 2.8Hz, J = 6.4Hz, J = 9.1Hz, J = 10.7Hz, 1H, OCH-THF ring), 4.19 (q, J = 7.2Hz, 2H, OCH₂CH₃), 3.79 (s, 3H, OCH₃), 3.75 (s, 3H, OCH₃), 3.40 (dd, J = 9.0Hz, J =
15.6Hz, 1H, CH$_2$CHO), 3.27 (d, J = 17.8Hz, 1H, CH$_2$-propargyl chain), 3.16 (d, J = 17.8Hz, 1H, CH$_2$-propargyl chain), 2.90 (dd, J = 6.4Hz, J = 15.6Hz, 1H, CH$_2$CHO), 2.57 (dd, J = 10.7Hz, J = 15.0Hz, 1H, CH$_2$-THF ring), 2.45 (dd, J = 2.8Hz, J = 15.0Hz, 1H, CH$_2$-THF ring), 1.29 (t, J = 7.1Hz, 3H, OCH$_2$CH$_3$).

$^{13}$C NMR (δ, ppm)

(CDC$_3$, 100 MHz)

169.9 (Cq, C=O), 169.8 (Cq, C=O), 158.8 (Cq, C=O), 153.2 (Cq, Ar), 128.0 (CH-Ar), 126.0 (Cq, Ar), 124.9 (CH-Ar), 120.6 (CH-Ar), 109.5 (CH-Ar), 83.1 (Cq, C = C), 78.6 (OCH-THF ring), 75.9 (Cq, C = C), 62.0 (OCH$_2$CH$_3$), 55.2 (Cq), 53.2 (OCH$_3$), 53.0 (OCH$_3$), 38.8 (CH$_2$), 36.4 (CH$_2$), 23.4 (CH$_2$), 14.0 (OCH$_2$CH$_3$).

IR (ν, cm$^{-1}$) (CCl$_4$)

3035 (w), 2984 (w), 2954 (m), 2907 (w), 2843 (w), 2241 (m), 1745 (s), 1717 (s), 1614 (w), 1599 (w), 1480 (m), 1463 (m), 1436 (m), 1389 (w), 1367 (w), 1323 (w), 1286 (w), 1254 (s), 1230 (s), 1201 (m), 1182 (m), 1094 (m), 1075 (m), 1015 (w).

HRMS (EI+, m/z) : Calculated: 374.1366 Found: 374.1369.

2-(5-Oxo-tetrahydro-furan-2-ylmethyl)-2-prop-2-yny-l-malonic acid dimethyl ester

MF: C$_{13}$H$_{16}$O$_6$

MW = 268 g.mol$^{-1}$

Method : See general procedure 4.1 using (1 equiv., 10 mmol, 1.28 g) of pent-4-enio acid ethyl ester followed by general procedure 5.2 using (1 equiv., 3.5 mmol, 630 mg) of 5-bromomethyl-dihydro-furan-2-one, followed by general procedure 5.3 using the previously prepared monoalkylate malonate.

Purification : Flash column chromatography (silica gel, 1:1 PE: AcOEt)/ R$_f$ (1:1 PE:AcOEt): 0.38.

Product : Transparent oil, cristallizes a white solid in the fridge.

Isolated yield : 6 % for three steps.

$^1$H NMR (δ, ppm)

(CDC$_3$, 400 MHz)

4.66-4.59 (m, 1H, OCH-lactone ring), 3.77 (s, 6H, OCH$_3$), 3.02 (dd, J = 2.7Hz, J = 17.5Hz, 1H, CH$_2$-propargyl chain), 2.92 (dd, J = 2.7Hz, J = 17.5Hz, 1H, CH$_2$-propargyl chain), 2.54-2.48 (m, 3H, CH$_2$-lactone ring + CH$_3$CHO), 2.46-2.34 (m, 2H, CH$_2$-lactone ring), 2.04 (t, J = 2.7Hz, 1H, OCH$_2$CH$_3$), 1.97-1.88 (m, 1H, CH$_2$-lactone ring).
\textbf{\( ^{13}C \) NMR (\( \delta \), ppm)} \hspace{1cm} 176.0 (Cq, C=O), 170.0 (Cq, C=O), 169.9 (Cq, C=O), 78.4 (Cq, C=CH), 76.4 (OCH-lactone ring), 72.0 (CH), 55.2 (Cq), 53.1 (OCH\(_3\)), 53.1 (OCH\(_3\)), 38.5 (CH\(_2\)), 28.6 (CH\(_2\)), 28.2 (CH\(_2\)), 23.4 (CH\(_2\)).

\textbf{IR (\( \nu \), cm\(^{-1}\)) (CCl\(_4\))} \hspace{1cm} 3313 (m), 2954 (m), 2844 (w), 1788 (s), 1743 (s), 1458 (m), 1437 (m), 1354 (w), 1322 (w), 1288 (m), 1263 (m), 1251 (m), 1227 (s), 1203 (s), 1181 (s), 1158 (s), 1075 (s), 1020 (w).

\textbf{HRMS (EI+, m/z) :} \hspace{1cm} Calculated: 268.0947, found: 268.0948.

\begin{center}
\textbf{2-(1-Methanesulfonyl-pyrrolidin-2-ylmethyl)-2-prop-2-ynyl-malonic dimethyl ester} \hspace{1cm} \textbf{acid 5.17q}
\end{center}

\begin{center}
\begin{tikzpicture}
\end{tikzpicture}
\end{center}

\textbf{MF:} C\(_{14}\)H\(_{21}\)O\(_6\)NS

\textbf{MW = 331 g.mol\(^{-1}\)}

\textbf{Method :} \hspace{1cm} See general procedure 5.2 using (1 equiv., 6.5 mmol, 1.66 g) of methanesulfonic acid 1-methanesulfonyl-pyrrolidin-2-ylmethyl ester followed by general procedure 5.3 using (1 equiv., 0.6 mmol, 176 mg) of the previously prepared monoalkylated malonate.

\textbf{Purification :} \hspace{1cm} Flash column chromatography (silica gel, 1:1 PE:AcOEt)/ \( R_f \) (1:1 PE:AcOEt): 0.36.

\textbf{Product :} \hspace{1cm} White solid.

\textbf{Isolated yield :} \hspace{1cm} 9% for two steps.

\textbf{\(^1H\) NMR (\( \delta \), ppm)} \hspace{1cm} 3.99-3.94 (m, 1H, CHN), 3.75 (s, 6H, OCH\(_3\)), 3.41-3.34 (m, 1H, CH\(_2\)N), 3.32-3.26 (m, 1H, CH\(_2\)N), 3.05 (dd, \( J = 2.7Hz \), \( J = 17.5Hz \), 1H, CH\(_2\)-propargyl chain), 2.98 (dd, \( J = 2.7Hz \), \( J = 17.5Hz \), 1H, CH\(_2\)-propargyl chain), 2.80 (s, 3H, SO\(_2\)CH\(_3\)), 2.47 (dd, \( J = 6.5Hz \), \( J = 14.8Hz \), 1H, CH\(_2\)CHN), 2.26 (dd, \( J = 7.4Hz \), \( J = 14.8Hz \), 1H, CH\(_2\)CHN), 2.04 (t, \( J = 2.7Hz \), 1H, C=CH), 2.00-1.92 (m, 3H, CH\(_2\)), 1.79-1.73 (m, 1H, CH\(_2\)).

\textbf{\( ^{13}C\) NMR (\( \delta \), ppm)} \hspace{1cm} 170.3 (Cq x2, C=O), 78.9 (Cq, C=CH), 71.9 (C=CH), 56.4 (SO\(_2\)CH\(_3\)), 55.7 (Cq), 52.9 (OCH\(_3\)), 52.8 (OCH\(_3\)), 47.4 (CH\(_2\)N), 37.4 (CH\(_2\)), 36.6 (CHN), 32.0 (CH\(_2\)), 24.1 (CH\(_2\)), 22.7 (CH\(_2\)).

\textbf{IR (\( \nu \), cm\(^{-1}\)) (CCl\(_4\))} \hspace{1cm} 3313 (m), 2954 (m), 1741 (s), 1438 (w), 1350 (s), 1286 (m), 1202 (s), 1151 (s), 1061 (s).
**HRMS (El+, m/z)**: Calculated: 331.1090  found: 331.1076.

**2-(1-Methanesulfonyl-piperidin-2-ylmethyl)-2-prop-2-ynyl-malonate dimethyl ester**  

```
O
O
MeO

O
O
N
S
Me
Me

MF: C_{15}H_{23}O_6NS

MW = 345 g.mol^{-1}
```

**Method**: See general procedure 5.2 using (1 equiv., 6.3 mmol, 1.61 g) of methanesulfonic acid 1-methanesulfonyl-piperidin-2-ylmethyl ester followed by general procedure 5.3 using (1 equiv., 0.6 mmol, 184 mg) of the previously prepared monoalkylated malonate.

**Purification**: Flash column chromatography (silica gel 1:1 PE:AcOEt)/ R_f (7:3 PE:AcOEt): 0.15.

**Product**: White solid.

**Isolated yield**: 42% for two steps.

**$^1$H NMR ($\delta$, ppm)**  
(CDCl$_3$, 400 MHz)  
4.18-4.15 (m, 1H, CHN-piperidine ring), 3.75 (s, 3H, OCH$_3$), 3.74 (s, 3H, OCH$_3$), 3.57 (dd, J = 3.6Hz, J = 14.6Hz, 1H, CH$_2$N-piperidine ring), 3.16 (dd, J = 2.6Hz, J = 17.6Hz, 1H, CH$_2$-propargyl chain), 3.13 (dd, J = 3.6Hz, J =14.6Hz, 1H, CH$_2$N-piperidine ring), 3.01 (dd, J = 2.6Hz, J = 17.6Hz, 1H, CH$_2$-propargyl chain), 2.89 (s, 3H, SO$_2$CH$_3$), 2.79 (dd, J = 10.4Hz, J = 15.1Hz, 1H, CH$_2$CHN), 2.22 (dd, J = 4.0Hz, J = 15.1Hz, 1H, CH$_2$CHN), 2.01 (t, J = 2.6Hz, 1H, \equiv CH), 1.77-1.69 (m, 2H, CH$_2$-piperidine ring), 1.62-1.50 (m, 4H, CH$_2$-piperidine ring).

**$^{13}$C NMR ($\delta$, ppm)**  
(CDCl$_3$, 100 MHz)  
170.4 (Cq, C=O), 170.0 (Cq, C=O), 79.0 (Cq, C \equiv O), 71.6 ( \equiv CH), 55.5 (Cq), 53.0 (OCH$_3$), 52.9 (OCH$_3$), 48.6 (SO$_2$CH$_3$), 40.7 (CH$_2$N), 40.6 (CHN), 32.0 (CH$_2$), 29.7 (CH$_2$), 24.2 (CH$_3$), 21.9 (CH$_2$), 18.6 (CH$_2$).

**IR (v, cm$^{-1}$) (CCl$_4$)**  
3313 (m), 2953 (m), 1742 (s), 1438 (m), 1340 (s), 1289 (m), 1202 (s), 1148 (s).

**HRMS (El+, m/z)**: Calculated: 345.1246  found: 345.1234.
2-(2-Methoxy-ethyl)-2-prop-2-ynyl-malic acid dimethyl ester

\[
\begin{align*}
\text{MF: } & \text{C}_{11}\text{H}_{16}\text{O}_{5} \\
\text{MW: } & 228 \text{ g.mol}^{-1}
\end{align*}
\]

**Method:**
See general procedure 5.2 using (1 equiv., 10.6 mmol, 970 µL) of 1-chloro-2-methoxy-ethane followed by general procedure 5.3 using (1 equiv., 1.6 mmol, 300 mg) of the previously prepared monoalkylated malonate.

**Purification:**
Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ \( R_f \) (8:2 PE:AcOEt): 0.28.

**Product:**
Transparent oil.

**Isolated yield:**
24% for two steps.

**\(^1\)H NMR (δ, ppm) (CDCl₃, 400 MHz):**
- 3.73 (s, 6H, OCH₃), 3.44 (t, \( J = 6.1 \text{Hz} \), 2H, OCH₂), 3.25 (s, 3H, OCH₃), 2.89 (d, \( J = 2.7 \text{Hz} \), 2H, CH₂-propargyl chain), 2.37 (t, \( J = 6.1 \text{Hz} \), 2H, CH₂), 2.00 (t, \( J = 2.7 \text{Hz} \), 1H, ≡CH).

**\(^{13}\)C NMR (δ, ppm) (CDCl₃, 100 MHz):**
- 170.6 (Cq x2, C=O), 78.9 (Cq, C ≡CH), 71.4 (≡CH), 68.2 (OCH₃), 58.7 (OCH₂), 55.2 (Cq), 52.7 (OCH₃ x2), 32.0 (CH₂), 23.2 (CH₂).

**IR (ν, cm\(^{-1}\)) (CCl₄):**
- 3314 (s), 3028 (w), 2985 (w), 2953 (m), 2928 (m), 2896 (m), 2876 (m), 2833 (m), 2812 (w), 1741 (s), 1481 (w), 1458 (m), 1436 (s), 1389 (w), 1351 (w), 1323 (m), 1289 (s), 1226 (s), 1198 (s), 1184 (s), 1160 (m), 1123 (s), 1102 (s), 1063 (m), 1035 (s), 1004 (w).

**HRMS (El+, m/z):**
Calculated: 228.0998  found: 228.1000.
2-(2-Methoxy-2-phenyl-ethyl)-2-prop-2-ynyl-malonic acid dimethyl ester

\[ \text{MF: } C_{17}H_{20}O_5 \]

\[ \text{MW} = 304 \text{ g.mol}^{-1} \]

**Method:**
See **general procedure 5.2** using (1 equiv., 3.21 mmol, 840 mg) of (2-iodo-2-methoxy-ethyl)-benzene followed by **general procedure 5.3** using (1 equiv., 1.69 mmol, 450 mg) of the previously prepared monoalkylated malonate.

**Purification:**
Flash column chromatography (silica gel, 9:1 PE:AcOEt). / \( R_f \) (8:2 PE:AcOEt): 0.35.

**Product:**
Transparent oil.

**Isolated yield:**
32% for two steps.

**\(^1H\) NMR (\(\delta, \text{ppm}\))**
(CDCl\(_3\), 400 MHz)
- 7.37-7.28 (m, 5H, CH-Ph), 4.24 (dd, \(J = 4.9\)Hz, \(J = 8.2\)Hz, 1H, CHOCH\(_3\)), 3.76 (s, 3H, OCH\(_3\)), 3.73 (s, 3H, OCH\(_3\)), 3.09 (s, 3H, OCH\(_3\)), 3.03 (dd, \(J = 2.6\)Hz, \(J = 17.3\)Hz, 1H, CH\(_2\)-propargyl chain), 2.96 (dd, \(J = 2.7\)Hz, \(J = 17.3\)Hz, 1H, CH\(_2\)-propargyl chain), 2.46-2.43 (m, 2H, CH\(_2\)), 2.02 (t, \(J = 2.7\)Hz, 1H, CH).

**\(^13C\) NMR (\(\delta, \text{ppm}\))**
(CDCl\(_3\), 100 MHz)
- 170.5 (Cq x2, C=O), 141.7 (Cq, Ph), 128.5 (CH-Ph), 127.7 (CH-Ph), 126.4 (CH-Ph), 79.9 (OCH), 79.0 (Cq, CH=C=CH), 71.6 (CH=C=CH), 56.7 (OCH\(_3\)), 55.4 (Cq), 52.7 (OCH\(_3\)), 52.6 (OCH\(_3\)), 41.2 (CH\(_2\)), 23.5 (CH\(_2\)).

**IR (\(\nu, \text{cm}^{-1}\))** (CCl\(_4\))
- 3314 (s), 3088 (w), 3066 (w), 3029 (w), 2999 (w), 2952 (m), 2904 (w), 2888 (w), 2842 (w), 2825 (w), 1741 (s), 1494 (w), 1455 (m), 1436 (s), 1355 (w), 1323 (m), 1289 (s), 1231 (s), 1199 (s), 1182 (s), 1105 (s), 1094 (s), 1065 (s), 1038 (s).

**HRMS (EI+, \text{m/z}):**
Calculated: 304.1311  found: 304.1319.
B.3.4.2 Synthesis of substrates bearing C(3)-THF rings

Alkynes synthesized in this work having C(3)-substituted THF rings were obtained as depicted in the scheme below:

Scheme B.3.4.2: C(3)-linked THF rings synthesized in this project

General procedure 5.7,

To a round bottom flask, charged with allyl alcohol (1 equiv.) and DCM (1M) at -30°C, was added Br2 (1.2 equiv.) dissolved in DCM (4 M). The addition of Br2 is stopped when the orange color persists, which normally means the reaction is over (TLC). The solution is treated with a solution of Na2S2O3 in water, extracted with DCM (3x); dried (MgSO4) and concentrated under reduced pressure. The dibromo compound is then diluted in MeOH (0.4M), followed by the addition of K2CO3 (3 equiv.) and the reaction mixture is allowed to stir overnight at room temperature. Upon completion (TLC), the reaction is filtered, concentrated under reduced pressure and purified by flash column chromatography.

Other related structure synthesized in this project:

\[ \text{General procedure 5.7, } \text{Br}_2 \text{ promoted cyclisation of allylic alcohols:} \]

\[ \text{To a round bottom flask, charged with allyl alcohol (1 equiv.) and DCM (1M) at -30°C, was added Br}_2 \text{ (1.2 equiv.) dissolved in DCM (4 M). The addition of Br}_2 \text{ is stopped when the orange color persists, which normally means the reaction is over (TLC). The solution is treated with a solution of Na}_2\text{S}_2\text{O}_3 \text{ in water, extracted with DCM (3x); dried (MgSO}_4) \text{ and concentrated under reduced pressure. The dibromo compound is then diluted in MeOH (0.4M), followed by the addition of K}_2\text{CO}_3 \text{ (3 equiv.) and the reaction mixture is allowed to stir overnight at room temperature. Upon completion (TLC), the reaction is filtered, concentrated under reduced pressure and purified by flash column chromatography.} \]

\[ \text{Other related structure synthesized in this project:} \]

\[ \text{Scheme B.3.4.2: C(3)-linked THF rings synthesized in this project} \]

\[ \text{General procedure 5.7, } \text{Br}_2 \text{ promoted cyclisation of allylic alcohols:} \]

\[ \text{To a round bottom flask, charged with allyl alcohol (1 equiv.) and DCM (1M) at -30°C, was added Br}_2 \text{ (1.2 equiv.) dissolved in DCM (4 M). The addition of Br}_2 \text{ is stopped when the orange color persists, which normally means the reaction is over (TLC). The solution is treated with a solution of Na}_2\text{S}_2\text{O}_3 \text{ in water, extracted with DCM (3x); dried (MgSO}_4) \text{ and concentrated under reduced pressure. The dibromo compound is then diluted in MeOH (0.4M), followed by the addition of K}_2\text{CO}_3 \text{ (3 equiv.) and the reaction mixture is allowed to stir overnight at room temperature. Upon completion (TLC), the reaction is filtered, concentrated under reduced pressure and purified by flash column chromatography.} \]

\[ \text{Other related structure synthesized in this project:} \]

\[ \text{Scheme B.3.4.2: C(3)-linked THF rings synthesized in this project} \]
2-Prop-2-ynyl-2-(tetrahydro-furan-3-yl)-malonic acid dimethyl ester

Method:

See general procedure 5.2 using (1 equiv., 6 mmol, 1.45 g) of toluene-4-sulfonic acid tetrahydro-furan-3-yl ester followed by general procedure 5.3 using (1 equiv., 1.36 mmol, 275 mg) of the previously prepared monoalkylated malonate.

Purification:

Flash column chromatography (silica gel, 8:2 PE:AcOEt) \( \text{Rf} \) (7:3 PE:AcOEt): 0.43.

Product:

White solid.

Isolated yield:

17% for two steps.

\( ^1H \text{NMR} \) (\( \delta \), ppm) (CDCl\(_3\), 400 MHz)

3.95 (dd, \( J = 8.1 \text{Hz}, J = 9.3 \text{Hz} \)), 1H, OCH\(_2\)-THF ring), 3.86-3.78 (m, 2H, OCH\(_2\)-THF ring), 3.76 (s, 3H, OCH\(_3\)), 3.75 (s, 3H, OCH\(_3\)), 3.66 (dd, \( J = 8.4 \text{Hz}, J = 15.5 \text{Hz} \)), 1H, OCH\(_2\)-THF ring), 3.17-3.09 (m, 1H, CH-THF ring), 2.85 (d, \( J = 2.7 \text{Hz} \)), 2H, CH\(_2\)-propargyl chain), 2.12-2.05 (m, 1H, CH\(_2\)-THF ring), 2.03 (t, \( J = 2.7 \text{Hz}, 1 \text{H}, \equiv \text{CH} \)), 1.86-1.77 (m, 1H, CH\(_2\)-THF ring).

\( ^{13}C \text{NMR} \) (\( \delta \), ppm) (CDCl\(_3\), 100 MHz)

170.2 (Cq, C=O), 169.9 (Cq, C=O), 78.7 (Cq, \( \equiv \text{CH} \)), 71.6 (\( \equiv \text{CH} \)), 69.4 (OCH\(_2\)-THF ring), 67.9 (OCH\(_2\)-THF ring), 58.8 (Cq), 52.8 (OCH\(_3\)), 52.7 (OCH\(_3\)), 41.9 (CH-THF ring), 27.8 (CH\(_3\)), 24.4 (CH\(_2\)).

\( \text{IR} \) (\( \nu \), cm\(^{-1}\)) (CCl\(_4\))

3313 (m), 2954 (m), 2858 (w), 1758 (m), 1736 (s), 1436 (m), 1355 (w), 1276 (m), 1232 (s), 1201 (s), 1182 (m), 1118 (w), 1075 (m), 1054 (w), 1026 (w).

HRMS (EI+, m/z):

Calculated: 240.0998  found: 240.0999.
**Method:**
See **general procedure 5.7** using (1 equiv., 26.3 mmol, 3.0 g) of 2-methyl-hex-5-en-3-ol followed by **general procedure 5.2** using (1 equiv., 3 mmol, 587 mg) of 4-bromo-2-isopropyl-tetrahydro-furan followed by **general procedure 5.3** using (1 equiv., 1.45 mmol, 355 mg) of the previously prepared monoalkylated malonate.

**Purification:**
Flash column chromatography (silica gel, 95:5 toluene:AcOEt) / \( R_f \). (8:2 PE:AcOEt) : 0.48.

**Product:**
Transparent oil.

**Isolated yield:**
4% for three steps (inseparable mixture, ratio diasteroisomers 1:3)

**\(^1\)H NMR (δ, ppm)**

<table>
<thead>
<tr>
<th>Compound</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Major diasteromer (cis):</strong></td>
<td>3.94 (d, ( J = 7.2\text{Hz} ), 2H, OCH- + OCH2- THF ring), 3.75 (s, 3H, OCH3), 3.74 (s, 3H, OCH3), 3.47-3.40 (m, 1H, OCH2-THF ring), 3.21-3.13 (m, 1H, CH-THF ring), 2.99-2.77 (m, 2H, CH2-propargyl chain), 2.08-2.04 (m, 1H-CH-Pr), 2.03 (t, ( J = 2.7\text{Hz} ), 1H, ( \equiv \text{CH} ), 1.71-1.61 (m, 1H, CH2-THF ring), 1.42-1.34 (m, 1H, CH2-THF ring), 0.94 (d, ( J = 6.7\text{Hz} ), 3H, CH3-Pr), 0.85 (d, ( J = 6.7\text{Hz} ), 3H, CH3-Pr). <strong>Minor diasteromer (trans):</strong></td>
</tr>
</tbody>
</table>

**\(^13\)C NMR (δ, ppm)**

<table>
<thead>
<tr>
<th>Compound</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Major diasteroisomer (cis):</strong></td>
<td>170.2 (Cq, C=O), 169.8 (Cq, C=O), 84.9 (OCH-THF ring), 78.7 (Cq, ( \equiv \text{CH} ), 71.6 (( \equiv \text{CH} ), 69.1 (OCH2-THF ring), 58.9 (Cq), 52.8 (OCH3), 52.7 (OCH3), 42.0 (CH), 32.7 (CH), 31.2 (CH2), 24.2 (CH2), 19.3 (CH3), 18.4 (CH3). <strong>Minor diasteroisomer (trans):</strong></td>
</tr>
</tbody>
</table>

**IR (ν, cm⁻¹) (CCl₄)**

3313 (s); 2955 (s); 2873 (s); 2844 (m); 1758 (s); 1736 (s); 1469 (m); 1458 (m); 1448 (m); 1436 (s); 1389 (w); 1366 (w); 1352 (w); 1312 (m); 1282 (s); 1227 (s);
1201 (s); 1185 (s); 1139 (m); 1118 (m); 1078 (s); 1036 (m).

HRMS (EI+, m/z) : Calculated: 282.1467. Found: 282.1459.

2-(5-Pentyl-tetrahydro-furan-3-yl)-2-prop-2-ynyl-malonic acid dimethyl ester 5.36d

Method : See general procedure 5.7 using (1 equiv., 21.1 mmol, 3.0 g) of non-1-en-4-ol followed by general procedure 5.2 using (1 equiv., 9.3 mmol, 2.05 g) of 4-bromo-2-pentyl-tetrahydro-furan followed by general procedure 5.3 using (1 equiv., 3.5 mmol, 963 mg) of the previously prepared monoalkylated malonate.

Purification : Flash column chromatography (silica gel, 9:1 PE:AcOEt)/ Rf (8:2 PE:AcOEt): 0.37.

Product : Transparent oil.

Isolated yield : 11% for three steps (inseparable mixture of diasteroisomers, ratio cis: trans 3:1).

¹H NMR (δ, ppm) (CDCl₃, 400 MHz) Major diasteroisomer (cis): 3.97 (dd, J = 9.7Hz, J = 5.9Hz, 1H, OCH-THF ring), 3.91 (dd, J = 9.7 Hz, J = 8.2Hz, 1H, OCH₂-THF ring), 3.75 (s, 3H, OCH₃), 3.74 (s, 3H, OCH₃), 3.72-3.65 (m, 1H, OCH₂-THF ring), 3.21-3.11 (m, 1H, CH), 2.85 (dd, J = 2.7Hz, J = 17.2Hz, 1H, CH₂-propargyl chain), 2.80 (dd, J = 2.7Hz, J = 17.2Hz, 1H, CH₂-propargyl chain), 2.15-2.09 (m, 1H, CH₂-THF ring), 2.02 (t, J = 2.7Hz, 1H, CH), 1.62-1.52 (m, 1H, CH₂-THF ring), 1.46-1.28 (m, 8H, CH₂-pentyl chain), 0.88 (t, J = 6.7Hz, 3H, CH₃-pentyl chain). Minor diasteroisomer (trans): 4.14 (dd, J = 9.1Hz, J = 7.8Hz, 1H, OCH-THF ring), 3.75 (s, 3H, OCH₃), 3.74 (s, 3H, OCH₃), 3.72-3.65 (m, 2H, OCH₂-THF ring), 3.21-3.11 (m, 1H, CH), 2.88-2.77 (m, 2H, CH₂-propargyl chain), 2.03 (t, J = 2.7Hz, 1H, CH), 1.99-1.94 (m, 1H, CH₂-THF ring), 1.75-1.68 (m, 1H, CH₂-THF ring), 1.46-1.28 (m, 8H, CH₂-pentyl chain), 0.88 (t, J = 6.7Hz, 3H, CH₃-pentyl chain).

¹³C NMR (δ, ppm) (CDCl₃, 100 MHz) Major diasteroisomer (cis): 170.2 (Cq, C=O), 169.8 (Cq, C=O), 79.6 (OCH-THF ring), 78.8 (Cq, C=O), 71.6 (Cq), 68.9 (OCH₂-THF ring), 59.0 (Cq), 52.7 (OCH₃), 52.6 (OCH₃), 42.1 (CH), 35.0 (CH₂), 33.7 (CH₂), 31.9 (CH₂), 25.9 (CH₂), 24.2 (CH₂), 22.5 (CH₂), 13.9 (CH₃-pentyl chain). Minor diasteroisomer (trans): 170.1 (Cq, C=O), 170.0 (Cq, C=O), 79.0 (OCH-THF ring), 78.7 (Cq), 71.5 (Cq), 68.8 (OCH₂-THF ring), 58.6 (Cq), 52.7 (OCH₃), 52.6 (OCH₃), 41.6 (CH), 35.5 (CH₂), 33.2 (CH₂), 31.8 (CH₂), 25.7 (CH₂), 24.4 (CH₂), 22.5 (CH₂), 13.9 (CH₃-pentyl chain).
IR (ν, cm⁻¹) (CCl₄) 3313 (m), 3001 (w), 2932 (m), 2860 (m), 1758 (s), 1736 (s), 1456 (m), 1435 (m), 1379 (w), 1355 (w), 1284 (m), 1230 (s), 1200 (s), 1137 (w), 1117 (w), 1092 (w), 1073 (w), 1051 (w).

HRMS (EI+, m/z): Calculated: 310.1780 found: 310.1765.

2-(5-Phenyl-tetrahydro-furan-3-yl)-2-prop-2-ynyl-malonic acid dimethyl ester

\[
\text{MeO} \quad \text{O} \quad \text{O} \\
\text{O} \quad \text{OMe} \quad \text{MeO} \\
\text{O}
\]

MF: C₁₈H₂₀O₅

MW = 316 g.mol⁻¹

Method: See general procedure 5.7 using (1 equiv., 20 mmol, 3.0 g) of 1-phenyl-but-3-en-1-ol followed by general procedure 5.2 using (1 equiv., 11.9 mmol, 2.7 g) of 4-bromo-2-phenyl-tetrahydro-furan followed by general procedure 5.3 using (1 equiv., 5.03 mmol, 1.4 g) of previously prepared monosubstituted malonate.

Purification: Flash column chromatography (silica gel, 8:2 PE AcOEt)/ Rf (8:2 PE:AcOEt): 0.29.

Product: Yellow pale oil.

Isolated yield: 18% for three steps. (ratio cis: trans 3: 1).

\(^1\text{H NMR}\) (δ, ppm) (CDCl₃, 400 MHz)

Major diasteroisomer (cis): 7.35-7.24 (m, 5H, CH-Ph), 4.79 (dd, J = 5.6Hz, J = 10.2Hz, 1H, OCH-THF ring), 4.19 (dd, J = 6.4Hz, J = 9.4Hz, 1H, OCH₂-THF ring), 4.14 (dd, J = 8.5Hz, J = 9.4Hz, 1H, OCH₂-THF ring), 3.76 (s, CH₃, OCH₃), 3.75 (s, CH₃, OCH₃), 3.40-3.31 (m, 1H, CH), 2.91-2.78 (m, 2H, CH₂-propargyl chain), 2.46 (dd, J = 5.6Hz, J = 8.2Hz, J = 12.6Hz, 1H, CH₂-THF ring), 2.05 (t, J = 2.7Hz, 1H, ===CH), 1.72 (td, J = 10.0Hz, J = 12.6Hz, 1H, CH₂-THFRing).

Minor diasteroisomer (trans): 7.35-7.24 (m, 5H, CH-Ph), 4.86 (dd, J = 6.6Hz, J = 7.7Hz, 1H, OCH-THF ring), 4.34 (dd, J = 7.7Hz, J = 9.2Hz, 1H, OCH₂-THF ring), 3.89 (dd, J = 8.0Hz, J = 9.2Hz, 1H, OCH₂-THF ring), 3.78 (s, 3H, OCH₃), 3.77 (s, 3H, OCH₃), 3.24 (dd, J = 7.7Hz, J = 9.2Hz, 1H, CH₂-propargyl chain), 2.91-2.78 (m, 2H, CH₂-propargyl chain), 2.41-2.34 (m, 1H, CH₂-THF ring), 2.12 (dd, J = 6.6Hz, J = 9.7Hz, J = 13.0Hz, 1H, CH₂-THFRing), 2.03 (t, J = 2.7Hz, 1H, ===CH).

\(^{13}\text{C NMR}\) (δ, ppm) (CDCl₃, 100 MHz)

Major diasteroisomer (cis): 170.1 (Cq, C=O), 169.7 (Cq, C=O), 141.5 (Cq, Ph), 128.3 (CH-Ph), 127.5 (CH-Ph), 125.8 (CH-Ph), 81.0 (OCH-THF ring), 78.6 (Cq, C === CH), 71.7 ( === CH), 69.6 (OCH₂-THF ring), 58.9 (Cq), 52.8 (OCH₃), 52.7 (OCH₃), 42.5 (CH), 36.7 (CH₂), 24.3 (CH₂).

Minor diasteroisomer (trans): 170.0 (Cq, C=O), 169.9 (Cq, C=O), 142.7 (Cq, Ph), 128.3 (CH-Ph), 127.3 (CH-Ph), 125.5 (CH-Ph), 80.0 (OCH-THF ring), 78.5 (Cq, C === CH), 71.8 ( === CH), 69.6 (OCH₂-THF ring), 58.5 (Cq), 52.8 (OCH₃), 52.7 (OCH₃), 41.8 (CH), 35.8 (CH₂), 24.5 (CH₂).
IR (ν, cm⁻¹) (CCl₄) 3313 (m), 3089 (w), 3066 (w), 3032 (w), 3003 (w), 2954 (m), 2869 (w), 2844 (w), 1757 (s), 1736 (s), 1495 (w), 1450 (m), 1436 (s), 1348 (m), 1280 (s), 1226 (s), 1201 (s), 1115 (s), 1086 (s), 1065 (s), 1029 (s).

HRMS (EI+, m/z) : Calculated: 316.1311   found: 316.1309.

5-Methoxycarbonyl-5-(tetrahydro-furan-3-y1)-hex-2-yndioic acid 1-ethyl ester 6-methyl ester

Method : See general procedure 5.4 using (1 equiv., 1 mmol, 240 mmol) of 2-prop-2-ynyl-2-(tetrahydro-furan-2-ylmethyl)-malonic acid dimethyl ester.

Purification : Flash column chromatography (silica gel, 7:3 PE:AcOEt)/ Rₖ (8:2 PE:AcOEt): 0.33.

Product : Yelow pale oil.

Isolated yield : 34 % (76%, brsm).

¹H NMR (δ, ppm) (CDCl₃, 400 MHz) 4.20 (q, J = 7.1Hz, 2H, OCH₂CH₃), 3.93 (dd, J = 8.0Hz, J = 9.4Hz, 1H, OCH₂-THF ring), 3.86-3.80 (m, 2H, OCH₂-THF ring), 3.78 (s, 3H, OCH₃), 3.66 (dd, J = 8.4Hz, J = 15.6Hz, 1H, OCH₂-THF ring), 3.12-3.04 (m, 1H, CH), 3.00 (s, 2H, CH₂-propargyl chain), 2.13-2.05 (m, 1H, CH₂-THFring), 1.86-1.77 (m, 1H, CH₂-THFring), 1.29 (t, J = 7.1Hz, 3H, OCH₂CH₃).

¹³C NMR (δ, ppm) (CDCl₃, 100 MHz) 169.6 (Cq, C=O), 169.4 (Cq, C=O), 153.1 (Cq, C=O), 82.8 (Cq, C=O), 75.8 (Cq, C=O), 69.3 (OCH₂-THF ring), 67.8 (OCH₂-THF ring), 61.9 (OCH₂CH₃), 58.5 (Cq), 53.0 (OCH₃), 52.9 (OCH₃), 42.2 (CH), 27.7 (CH₂), 24.4 (CH₂), 14.0 (OCH₂CH₃).

IR (ν, cm⁻¹) (CCl₄) 3024 (w), 2974 (w), 2952 (m), 2872 (w), 1757 (m), 1742 (s), 1491 (w), 1436 (m), 1326 (w), 1289 (w), 1260 (w), 1218 (s), 1200 (s), 1182 (s), 1120 (w), 1087 (s), 1051 (w), 1030 (w).

HRMS (EI+, m/z) : calculated: 330.1467   found: 330.1468.
5-Methoxycarbonyl-5-(5-pentyl-tetrahydro-furan-3-yl)-hex-2-ynedioic acid 1-ethyl ester 6-methyl ester 5.36g

MF: C$_{20}$H$_{30}$O$_7$

MW = 382 g.mol$^{-1}$

Method:
See general procedure 5.4 using (1 equiv., 1 mmol, 310 mg) of 2-(5-pentyl-tetrahydro-furan-3-yl)-2-prop-2-ynyl-malonic acid dimethyl ester

Purification:
Flash column chromatography (silica gel, 9:1 PE:AcOEt)/R$_f$ (8:2 PE:AcOEt): 0.38.

Product:
Transparent/ pale green oil.

Isolated yield:
45% (73%, brsm, inseparable mixture of diasteroisomers, ratio cis:trans: 3:1)

$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz)

Major diasteroisomer (cis): 4.20 (q, J = 7.0Hz, 2H, OCH$_2$CH$_3$), 3.97 (dd, J = 5.5Hz, J = 9.6Hz, 1H, OCH$_2$-THF ring), 3.90 (dd, J = 8.2Hz, J = 9.6Hz, 1H, OCH$_2$-THF ring), 3.77 (s, 3H, OCH$_3$), 3.75 (s, 3H, OCH$_3$), 3.74-3.64 (m, 1H, OCH$_2$-THF ring), 3.16-3.07 (m, 1H, CH-THF ring), 3.02-2.92 (m, 2H, CH$_2$-propargyl chain), 2.12 (ddd, J = 5.5Hz, J = 8.6Hz, J = 12.5Hz, 1H, CH$_2$-THF ring), 1.57-1.52 (m, 1H, CH$_2$-THF ring), 1.47-1.21 (m, 11H, CH$_2$- + CH$_3$- pentyl chain), 0.88 (t, J = 7.0Hz, 3H, OCH$_2$CH$_3$).

Minor diasteroisomer (trans): 4.20 (q, J = 7.0Hz, 2H, OCH$_2$CH$_3$), 4.13 (dd, J = 7.7Hz, J = 9.2Hz, 1H, OCH$_2$-THF ring), 3.77 (s, 3H, OCH$_3$), 3.76 (s, 3H, OCH$_3$), 3.74-3.64 (m, 2H, OCH$_2$-THF ring), 3.16-3.07 (m, 1H, CH-THF ring), 3.02-2.92 (m, 2H, CH$_2$-propargyl chain), 2.01-1.94 (m, 1H, CH$_2$-THF ring), 1.76-1.68 (m, 1H, CH$_2$-THF ring), 1.47-1.21 (m, 11H, CH$_2$- + CH$_3$- pentyl chain), 0.88 (t, J = 7.0Hz, 3H, OCH$_2$CH$_3$).

$^{13}$C NMR (δ, ppm) (CDCl$_3$, 100 MHz)

Major diasteroisomer (cis): 169.8 (Cq, C=O), 169.4 (Cq, C=O), 153.2 (Cq, C=O), 83.0 (Cq, C==C ), 79.6 (OCH-THF ring), 75.8 (Cq, C==C), 69.0 (OCH$_2$-THF ring), 62.0 (OCH$_2$CH$_3$), 58.8 (Cq), 53.1 (OCH$_3$), 53.0 (OCH$_3$), 42.5 (CH), 35.0 (CH$_2$), 33.7 (CH$_3$), 31.9 (CH$_2$), 25.9 (CH$_2$), 24.3 (CH$_2$), 22.6 (CH$_3$), 14.0 (CH$_3$x2, pentyl chain + OCH$_2$CH$_3$).

Minor diasteroisomer (trans): 169.7 (Cq, C=O), 169.6 (Cq, C=O), 153.2 (Cq, C=O), 82.9 (Cq, C==C), 79.1 (OCH-THF ring), 75.8 (Cq, C==C), 68.8 (OCH$_2$-THF ring), 62.0 (OCH$_2$CH$_3$), 58.4 (Cq), 53.1 (OCH$_3$), 53.0 (OCH$_3$), 42.1
(CH), 35.5 (CH₃), 33.4 (CH₂), 31.9 (CH₂), 25.8 (CH₂), 24.6 (CH₂), 22.6 (CH₂), 14.0 (CH₂x2 pentylen chain + OCH₃CH₃).

**IR** (ν, cm⁻¹) (CCl₄) 2955 (m), 2933 (m), 2873 (w), 2861 (w), 2242 (w), 1737 (s), 1717 (s), 1456 (w), 1435 (m), 1376 (w), 1254 (s), 1231 (s), 1115 (m), 1075 (m).


**2-(3-Phenyl-prop-2-ynyl)-2-(tetrahydro-furan-3-yl)-malonic acid dimethyl ester** 5.36h

\[
\text{MF: } C_{18}H_{20}O_5
\]
\[
\text{MW = 316 g.mol}^{-1}
\]

**Method** : See general procedure 5.5 using (1 equiv., 1 mmol, 240 mg) of 2-prop-2-ynyl-2-(tetrahydro-furan-3-yl)-malonic acid dimethyl ester and (1.1 equiv., 1.10 mmol, 120 µL) of iodo-benzene.

**Purification** : Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ Rf (8:2 PE:AcOEt): 0.41.

**Product** : Yellow oil.

**Isolated yield** : 60%.

**¹H NMR** (δ, ppm) (CDCl₃, 400 MHz)

- 7.37-7.35 (m, 2H, CH-Ph), 7.29-7.27 (m, 3H, CH-Ph), 3.99 (dd, J = 8.0Hz, J = 9.2Hz, 1H, OCH₂-THF ring), 3.88 (dd, J = 6.6Hz, J = 9.2Hz, 1H, OCH₂-THF ring), 3.83 (dt, J = 3.9Hz, J = 8.4Hz, 1H, OCH₂-THF ring), 3.78 (s, 3H, OCH₃), 3.77 (s, 3H, OCH₃), 3.71-3.65 (m, 1H, OCH₂-THF ring), 3.19 (td, J = 8.0Hz, J = 15.6Hz, 1H, CH-THFr ing), 3.07 (s, 2H, CH₂-propargyl chain), 2.17-2.08 (m, 1H, CH₂-THF ring), 1.92-1.83 (m, 1H, CH₂-THF-ring).

**¹³C NMR** (δ, ppm) (CDCl₃, 100 MHz)

- 170.3 (Cq, C=O), 170.1 (Cq, C=O), 131.6 (Cq, Ph), 128.2 (CH-Ph), 128.1 (CH-Ph), 123.0 (CH-Ph), 84.1 (Cq, C==C), 83.8 (Cq, C==C), 69.5 (OCH₂-THF ring), 67.9 (OCH₂-THF ring), 59.2 (Cq), 52.8 (OCH₃), 52.7 (OCH₃), 42.1 (CH), 27.8 (CH₂), 25.3 (CH₂).

**IR** (ν, cm⁻¹) (CCl₄) 3060 (w), 2954 (m), 2860 (w), 1736 (s), 1599 (w), 1573 (w), 1491 (w), 1435 (m), 1354 (w), 1316 (w), 1275 (s), 1231 (s), 1201 (s), 1181 (s), 1161 (s), 1073 (s), 1054 (s), 1027 (m).

339
HRMS (El+, m/z) :  Calculated: 316.1311  found: 316.1305.

2-Pent-4-en-2-ynyl-2-(tetrahydro-furan-3-yl)-malonic acid dimethyl ester 5.36i

\[
\begin{align*}
\text{MeO} & \quad \text{O} \\
\text{O} & \quad \text{Me} \\
\end{align*}
\]

MF: C_{14}H_{18}O_{5}

MW = 266 g.mol\(^{-1}\)

Method : See general procedure 5.5 using (1 equiv., 1 mmol, 240 mg) of 2-prop-2-ynyl-2-(tetrahydro-furan-3-yl)-malonic acid dimethyl ester and (1.2 equiv., 1.20 mmol, solution 1M in THF, 1.2 mL) of iodo-benzene.

Purification : Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ \(R_f\) (8:2 PE:AcOEt): 0.31.

Product : Orange oil.

Isolated yield : 67%.

\(^1\)H NMR (\(\delta\), ppm) (CDCl\(_3\), 400 MHz) 5.72 (tdd, \(J = 2.0Hz, J = 11.0Hz, J = 17.5Hz\), 1H, CH=CH\(_2\)), 5.56 (dd, 2.2Hz, \(J = 17.5Hz\), 1H, CH=CH\(_2\)), 5.42 (dd, \(J = 2.2Hz, J = 11.0Hz\), 1H, CH=CH\(_2\)), 3.96 (dd, \(J = 8.2Hz, J = 9.3Hz\), 1H, OCH\(_2\)-THF ring), 3.83 (dd, \(J = 6.7Hz, J = 9.3Hz\), 1H, OCH\(_2\)-THF ring), 3.80 (dd, \(J = 4.1Hz, J = 8.2Hz\), 1H, OCH\(_2\)-THF ring), 3.76 (s, 3H, OCH\(_3\)), 3.75 (s, 3H, OCH\(_3\)), 3.72-3.64 (m, 1H, OCH\(_2\)-THF ring), 3.11 (td, \(J = 8.0Hz, J = 15.7Hz\), 1H, CH-THF ring), 2.97 (d, \(J = 2.0Hz\), 2H, CH\(_2\)-propargyl chain), 2.12-2.04 (m, 1H, CH\(_2\)-THF ring), 1.87-1.78 (m, 1H, CH\(_2\)-THF ring).

\(^{13}\)C NMR (\(\delta\), ppm) (CDCl\(_3\), 100 MHz) 170.3 (Cq, C=O), 170.0 (Cq, C=O), 126.9 (CH=CH\(_2\)), 116.9 (CH=CH\(_2\)), 84.7 (Cq, C \(\equiv\) C), 82.4 (Cq, C \(\equiv\) C), 69.4 (OCH\(_2\)-THF ring), 67.9 (OCH\(_2\)-THF ring), 59.0 (Cq), 52.7 (OCH\(_3\) x2), 42.0 (CH), 27.8 (CH\(_3\)), 25.2 (CH\(_2\)).

IR (v, cm\(^{-1}\)) (CCl\(_4\)) 2954 (w), 2859 (w), 1737 (s), 1435 (w), 1275 (w), 1232 (m), 1202 (w), 1075 (w)

HRMS (El+, m/z) : Calculated: 266.1154  found: 266.1145.
2-(3-Bromo-prop-2-ynyl)-2-(5-isopropyl-tetrahydro-furan-3-yl)-malonic acid dimethyl ester

MF: C_{16}H_{25}O_{3}Br

MW = 374 g.mol^{-1}

Method: See general procedure 5.6 using (1 equiv., 0.35 mmol, 100 mg) of 2-(5-isopropyl-tetrahydro-furan-3-yl)-2-prop-2-ynyl-malonic acid dimethyl ester.

Purification: Flash column chromatography (silica gel, 8:PE:AcOEt)/ R_{f} (8:2 PE:AcOEt): 0.58.

Product: Transparent oil.

Isolated yield: 97% (inseparable mixture, ratio diastereoisomers cis:trans 3:1).

{\textbf{^1}H NMR (}\delta, \text{ppm}) (CDCl_{3}, 400 MHz)

Major diastereoisomer (cis): 3.93 (d, \(J = 7.0\text{Hz}, 2\text{H, OCH-+ OCH}_2\text{-THF ring}), 3.76 (s, 3H, OCH_{3}), 3.74 (s, 3H, OCH_{3}), 3.48-3.38 (m, 1H, OCH-THF ring), 3.16-3.08 (m, 1H, CH-THF ring), 2.90-2.80 (m, 2H, CH_{2}-propargyl chain), 2.08-2.01 (m, 1H, CH_{2}-THF ring), 1.69-1.61 (m, 1H, CH-Pr), 1.42-1.34 (m, 1H, CH_{2}-THF ring), 0.94 (d, \(J = 6.6\text{Hz}, 3\text{H, CH}_{3}-\text{Pr}), 0.85 (d, \(J = 6.6\text{Hz}, 3\text{H, CH}_{3}-\text{Pr}).

Minor diastereoisomer (trans): 4.13 (t, \(J = 8.6\text{Hz}, 1\text{H, OCH-THF ring}), 3.76 (s, 3H, OCH_{3}), 3.74 (s, 3H, OCH_{3}), 3.63 (t, \(J = 8.7\text{Hz}, 1\text{H, OCH}_{2}-\text{THF ring}), 3.48-3.38 (m, 1H, OCH_{2}-\text{THF ring}), 3.06-3.00 (m, 1H, CH-THF ring), 2.90-2.80 (m, 2H, CH_{2}-propargyl chain), 1.93-1.85 (m, 1H, CH_{2}-THFring), 1.84-1.75 (m, 1H, CH_{2}-THF ring), 1.69-1.61 (m, 1H, CH-Pr), 0.94 (d, \(J = 6.6\text{Hz}, 3\text{H, CH}_{3}-\text{Pr}), 0.85 (d, \(J = 6.6\text{Hz}, 3\text{H, CH}_{3}-\text{Pr}).

{\textbf{^{13}C NMR (}\delta, \text{ppm}) (CDCl_{3}, 100 MHz)

Major diastereoisomer (cis): 170.1 (Cq, C=O), 169.7 (Cq, C=O), 85.0 (OCH-THF ring), 74.8 (Cq, C=CBr), 69.1 (OCH_{2}-THF ring), 58.9 (Cq), 52.8 (OCH_{3}), 52.7 (OCH_{3}), 42.2 (CH), 41.7 (Cq, C=CBr), 32.7 (CH), 31.2 (CH_{2}), 25.4 (CH_{3}), 19.4 (CH_{3}-Pr), 18.4 (CH_{3}-Pr). Minor diastereoisomer (trans): 170.0 (Cq, C=O), 169.9 (Cq, C=O), 84.4 (OCH-THF ring), 74.8 (Cq, C=CBr), 69.1 (OCH_{2}-THF ring), 58.5 (Cq), 52.8 (OCH_{3}), 52.7 (OCH_{3}), 42.0 (CH), 41.9 (Cq,C=CBr), 32.8 (CH), 31.0 (CH_{2}), 25.6 (CH_{2}), 19.0 (CH_{2}-Pr), 18.4 (CH_{3}-Pr).

IR (\nu, \text{cm}^{-1}) (CCl_{4})

2956 (s), 2873 (s), 2772 (w), 2737 (w), 2696 (w), 2222 (w), 1744 (s), 1469 (s), 1436 (s), 1389 (s), 1366 (s), 1352 (s), 1313 (s), 1279 (s), 1222 (s), 1138 (s), 1118 (s), 1078 (s).
HRMS (El+, m/z): Calculated: 374.0729 found: 374.0716.

3-Prop-2-ynyloxy-tetrahydro-furan

\[
\text{MF: C}_{7}\text{H}_{10}\text{O}_{2}
\]

\[
\text{MW} = 126 \text{ g.mol}^{-1}
\]

Method: See general procedure 5.3 using (1 equiv., 5.6 mmol, 500 mg) of tetrahydrofuran-3-ol. Besides what described in the general procedure, it was added (0.2 equiv., 1.13 mmol, 420 mg) of tetrabutyl ammonium iodide.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt)/ R\text{f} (8:2 PE:AcOEt): 0.39.

Product: Orange oil.

Isolated yield: 78%.

\textbf{\textsuperscript{1}H NMR} (\(\delta\), ppm) 
(CDCl\textsubscript{3}, 400 MHz) 
4.37-4.33 (m, 1H, OCH-THF ring), 4.15 (d, \(J = 2.4\)Hz, 2H, CH\textsubscript{2}-propargyl chain), 3.98-3.72 (m, 4H, OCH\textsubscript{2}-THF ring), 2.42 (t, \(J = 2.4\)Hz, 1H, \(==\text{CH}\)), 2.03-2.00 (m, 2H, CH\textsubscript{2}-THF ring).

\textbf{\textsuperscript{13}C NMR} (\(\delta\), ppm) 
(CDCl\textsubscript{3}, 100 MHz) 
79.7 (Cq, C \(==\text{C}\)), 78.7 (OCH), 74.3 (\(==\text{CH}\)), 72.6 (OCH\textsubscript{2}), 67.0 (OCH\textsubscript{2}), 56.3 (OCH\textsubscript{2}), 32.4 (CH\textsubscript{2}-THF ring).

\textbf{IR} (\(\nu\), cm\textsuperscript{-1}) (CCl\textsubscript{4}) 
3312 (s), 2977 (m), 2950 (m), 2858 (m), 1441 (w), 1348 (w), 1269 (w), 1120 (s), 1074 (s), 1010 (w).

\textbf{MS} (Cl, NH\textsubscript{3}, m/z): 
M+H\textsuperscript{+}: 127; M+NH\textsubscript{4}\textsuperscript{+}: 144.
B.3.4.3 Synthesis of benzyl butynyl ether derivatives

Benzyl butynyl ether derivatives were synthesized as it follows:

**Scheme B.3.4.3**: Synthesis of alkynyl ethers with the oxygen atom being part of the 1,5-framework.

**General Procedure 5.8, Benzylation of alkynyl alcohols:**

To a round bottom flask, charged with alkynyl alcohol (1 equiv.) and THF (0.125 M) at 0 °C, was slowly added NaH (1.5 equiv.). The reaction mixture was allowed to warm up to room temperature. Benzyl bromide (1.2 equiv.) was then added, followed by the addition of tetrabutyl ammonium iodide (0.1 equiv.). The reaction mixture was then allowed to stir at room temperature overnight. After the complete consumption of the starting alcohol (TLC), the reaction was quenched with a saturated solution of NH4Cl, extracted with AcOEt (3x); dried (MgSO4), and concentrated under reduced pressure. Purification by flash column chromatography afforded the desired product.

**General procedure 5.9, Synthesis of ethyl pentynoates derivatives:**

To a round bottom flask, charged with alkynyl ether (1 equiv.) and THF (0.125M) at -78°C was added n-BuLi (solution 1.6M in hexanes, 1.5 equiv.) and allowed to stir for 1h at -78°C. The anion was then quenched with ethyl chloroformate (1.5 equiv.). The reaction mixture was allowed to warm up to room temperature, water was added and the reaction mixture
extracted with AcOEt (3x), dried (MgSO₄) and concentrated under reduced pressure. Purification by flash column chromatography afforded the desired compounds.

**5-Benzyloxy-7-phenyl-hept-2-ynoic acid ethyl ester**

\[
\text{MF: } C_{22}H_{24}O_3 \\
\text{MW} = 336 \text{ g.mol}^{-1}
\]

**Method:** See general procedure 5.8 using (1 equiv., 3 mmol, 522 mg) of 1-phenyl-hex-5-yn-3-ol and (1.2 equiv., 3.6 mmol, 430 µL) of benzyl bromide followed by general procedure 5.9 using (1 equiv. 1 mmol, 264 mg) of 4-benzyloxy-6-phenyl-hex-1-yne.

**Purification:** Flash column chromatography (silica gel, 95:5 PE:AcOEt)/Rf (8:2 PE:AcOEt): 0.55.

**Product:** Pale yellow oil.

**Isolated yield:** 35% for two steps.

**1H NMR (δ, ppm)**

(CDCl₃, 400 MHz)

- 7.39-7.28 (m, 7H, CH-Ph), 7.21-7.16 (m, 3H, CH-Ph), 4.66 (d, J = 11.6Hz, 1H, OCH₂), 4.51 (d, J = 11.6Hz, 1H, OCH₂), 4.22 (q, J = 7.1Hz, 2H, OCH₂CH₃), 3.66-3.60 (m, 1H, OCH), 2.84-2.77 (m, 1H, CH₂), 2.70-2.62 (m, 2H, CH₂), 2.58 (dd, J = 6.4Hz, J = 17.2Hz, 1H, CH₂), 2.05-1.92 (m, 2H, CH₂), 1.31 (t, J = 7.1Hz, 3H, OCH₂C₃H₃).

**13C NMR (δ, ppm)**

(CDCl₃, 100 MHz)

- 153.5 (Cq, C=O), 141.5 (Cq, Ph), 138.0 (Cq, Ph), 128.5 (CH-Ph), 128.4 (CH-Ph x2), 127.8 (CH-Ph), 127.7 (CH-Ph), 125.9 (CH-Ph), 85.7 (Cq, C === C), 75.8 (OCH), 74.7 (Cq, C === C), 71.6 (OCH₂CH₃), 61.8 (OCH₂CH₃), 35.9 (CH₂), 31.4 (CH₂), 24.2 (CH₂), 14.0 (OCH₂C₃H₃).

**IR (ν, cm⁻¹)**

(CCl₄)

- 3088 (w), 3066 (w), 3030 (w), 2984 (w), 2939 (w), 2865 (w), 2238 (m), 1715 (s), 1604 (w), 1496 (w), 1455 (w), 1366 (w), 1350 (w), 1300 (w), 1251 (s), 1072 (s), 1028 (m).

**HRMS (El+, m/z):** Calculated: 336.1726  Found: 336.1714.
5-(4-Methoxy-benzylxy)-7-phenyl-hept-2-ynoic acid ethyl ester

**Method:** See general procedure 5.8 using (1 equiv., 5 mmol, 870 mg) of 1-phenyl-hex-5-yn-3-ol and (1.2 equiv., 6 mmol, 820 µL) of p-methoxybenzyl chloride followed by general procedure 5.9 using (1 equiv., 1 mmol, 294 mg) of 1-methoxy-4-(1-phenethyl-but-3-ynloxy)methyl)-benzene.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt) / Rf (8:2 PE:AcOEt): 0.50.

**Product:** Transparent oil.

**Isolated yield:** 36% for two steps.

**1H NMR** (δ, ppm) (CDCl₃, 400 MHz) 7.30-7.26 (m, 4H, CH-Ph + CH-Ar), 7.21-7.15 (m, 3H, CH-Ph), 6.89 (d, J = 8.7Hz, 2H, CH-Ar), 4.58 (d, J = 11.2Hz, 1H, OCH₂Ph), 4.44 (d, J = 11.2Hz, 1H, OCH₂Ph), 4.22 (q, J = 7.1Hz, 2H, OCH₂CH₃), 3.81 (s, 3H, OCH₃), 3.64-3.58 (m, 1H, OCH), 2.83-2.76 (m, 1H, CH₂), 2.67-2.60 (m, 2H, CH₂), 2.55 (dd, J = 6.4Hz, J = 17.2Hz, 1H, CH₂), 1.99-1.93 (m, 2H, CH₂), 1.31 (t, J = 7.1Hz, 3H, OCH₂CH₃).

**13C NMR** (δ, ppm) (CDCl₃, 100 MHz) 159.4 (Cq, C=O), 153.6 (Cq, Ar), 141.6 (Cq, Ar), 130.2 (Cq, Ar), 129.5 (CH-Ar), 128.4 (CH-Ar x2), 125.9 (CH-Ar), 113.9 (CH-Ar), 85.8 (Cq, C ≡ C), 75.4 (OCH), 74.7 (Cq, C ≡ C), 71.3 (OCH₂Ph), 61.8 (OCH₂CH₃), 55.3 (OCH₃), 36.0 (CH₂); 31.4 (CH₂), 24.3 (CH₂), 14.0 (OCH₂CH₃).

**IR** (ν, cm⁻¹) (CCl₄) 3066 (w), 3029 (w), 2984 (w), 2936 (m), 2864 (w), 2837 (w), 2238 (w), 1715 (s), 1613 (w), 1587 (w), 1514 (m), 1497 (w), 1464 (w), 1455 (w), 1366 (w), 1349 (w), 1302 (w), 1250 (s), 1172 (w), 1072 (m), 1040 (w).

**HRMS** (EI+, m/z) : Calculated: 366.1831  Found: 366.1831.
5-Allyloxy-7-phenyl-hept-2-ynoic acid ethyl ester

\[
\text{MF: } C_{18}H_{22}O_3
\]

\[
\text{MW = 286 g.mol}^{-1}
\]

**Method:** See general procedure 5.8 using (1 equiv., 2.6 mmol, 455 mg) of 1-phenyl-hex-5-yne-3-ol and (1.2 equiv., 3 mmol, 260 µL) of allyl bromide followed by general procedure 5.9 using (1 equiv., 1mmol, 214 mg) of (3-allyloxy-hex-5-ylyn)-benzene.

**Purification:** Flash column chromatography (silica gel, 95:5 PE:AcOEt)/ \( R_f \) (8:2 PE:AcOEt): 0.66.

**Product:** Transparent oil.

**Isolated yield:** 63% for two steps.

**\(^1H\) NMR (δ, ppm)**

| (CDCl\(_3, 400\) MHz) | 7.31-7.26 (m, 2H, CH-Ph), 7.21-7.17 (m, 3H, CH-Ph), 5.93 (tdd, J = 5.7Hz, J = 10.4Hz, J = 17.2Hz, 1H, CH\(_2\)CH\(_3\)), 5.30 (ddd, J = 1.4Hz, J = 3.0Hz, J = 17.2Hz, 1H, CH\(_2\)CH\(_3\)), 5.19 (ddd, J = 1.4Hz, J = 3.0Hz, J = 10.4Hz, 1H, CH\(_2\)CH\(_3\)), 4.22 (q, J = 7.2Hz, 2H, OCH\(_2\)CH\(_3\)), 4.12 (tdd, J = 1.4Hz, J = 5.7Hz, J = 12.6Hz, 1H, OCH\(_2\)CH\(_3\)), 3.98 (tdd, J = 1.4Hz, J = 5.7Hz, J = 12.6Hz, 1H, OCH\(_2\)CH\(_3\)), 3.55 (quin, J = 6.2Hz, 1H, OCH), 2.83-2.76 (m, 1H, CH\(_3\)), 2.71-2.62 (m, 1H, CH\(_3\)), 2.61 (dd, J = 6.2Hz, J = 17.2Hz, 1H, CH\(_3\)), 2.53 (dd, J = 6.2Hz, J = 17.2Hz, 1H, CH\(_3\)), 1.95 (dt, J = 6.2Hz, J = 8.0Hz, 2H, CH\(_2\)), 1.30 (t, J = 7.2Hz, 3H, OCH\(_2\)CH\(_3\)). |

**\(^{13}C\) NMR (δ, ppm)**

| (CDCl\(_3, 100\) MHz) | 153.6 (Cq, C=O), 141.6 (Cq, Ph), 134.7 (CH), 128.4 (CH x2), 125.9 (CH), 117.2 (CH=CH\(_2\)), 85.7 (Cq, C=C=C), 75.9 (OCH), 74.7 (Cq, C=C=C), 70.7 (OCH\(_2\)Ph), 61.8 (OCH\(_2\)CH\(_3\)), 36.0 (CH\(_2\)), 31.4 (CH\(_2\)), 24.3 (CH\(_2\)), 14.0 (OCH\(_2\)CH\(_3\)). |

**IR (ν, cm\(^{-1}\)) (CCl\(_4\))**

| 3086 (w), 3066 (w), 3029 (w), 2985 (w), 2938 (w), 2862 (w), 2239 (m), 1715 (s), 1604 (w), 1496 (w), 1455 (w), 1424 (w), 1367 (w), 1342 (w), 1251 (s), 1172 (w), 1135 (w), 1073 (s), 1030 (w). |

**HRMS (EI+, m/z)**

| Calculated: 286.1569 Found: 286.1575. |
**5-Benzylxy-6-methyl-hept-2-ynoic acid ethyl ester**

MF: C_{17}H_{22}O_3

MW = 274 g.mol\(^{-1}\)

**Method:**
See general procedure 5.8 using (1 equiv., 3.6 mmol, 400 mg) of 2-methyl-hex-5-yn-3-ol followed and (1.2 equiv., 4.30 mmol, 510 µL) of benzyl bromide followed by general procedure 5.9 using (1 equiv, 1 mmol, 202 mg) of (1-isopropyl-but-3-ynoxymethyl)-benzene.

**Purification:**
Flash column chromatography (silica gel, 95:5 PE:AcOEt)/ \( R_f \) (8:2 PE:AcOEt): 0.66.

**Product:**
Transparent oil.

**Isolated yield:**
36 % for two steps.

**\(^1\)H NMR (\(\delta\), ppm)**
(CDCl\(_3\), 400 MHz)

7.39-7.27 (m, 5H, CH-Ph), 4.69 (d, \( J = 11.5\)Hz, 1H, OCH\(_2\)Ph), 4.53 (d, \( J = 11.5\)Hz, 1H, OCH\(_2\)Ph), 4.22 (q, \( J = 7.1\)Hz, 2H, OCH\(_2\)CH\(_3\)), 3.40 (q, \( J = 5.8\)Hz, 1H, OCH), 2.57 (d, \( J = 6.0\)Hz, 2H, CH\(_2\)CHO), 1.98 (hept, \( J = 6.2\)Hz, 1H, CH\(_{3}\)iPr), 1.31 (t, \( J = 7.1\)Hz, 3H, OCH\(_2\)CH\(_3\)), 0.96 (d, \( J = 6.2\)Hz, 3H, CH\(_3\)-iPr), 0.95 (d, \( J = 6.2\)Hz, 3H, CH\(_3\)-iPr).

**\(^13\)C NMR (\(\delta\), ppm)**
(CDCl\(_3\), 100 MHz)

153.7 (Cq, C=O), 138.3 (Cq, Ar), 128.3 (CH-Ph), 127.8 (CH-Ph), 127.6 (CH-Ph), 86.8 (Cq, C=C=C), 81.7 (OCH), 74.3 (Cq, C=C=C), 72.5 (OCH\(_2\)Ph), 61.8 (OCH\(_2\)CH\(_3\)), 31.7 (CH\(_3\)-Pr), 21.7 (CH\(_3\)), 18.6 (CH\(_3\)-Pr), 17.5 (CH\(_3\)-Pr), 14.0 (OCH\(_2\)CH\(_3\)).

**IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))**

3091 (w), 3067 (w), 3033 (w), 2964 (m), 2938 (m), 2908 (m), 2875 (m), 2236 (m), 1747 (m), 1713 (s), 1497 (w), 1466 (m), 1455 (m), 1423 (w), 1387 (m), 1366 (m), 1300 (m), 1250 (s), 1205 (w), 1175 (w), 1073 (s), 1028 (m), 1013 (m).

**HRMS (EI+, m/z):**
Calculated: 274.1569  found: 274.1579.
B.3.4.4 Au(I)-Catalyzed transformations

The previously prepared compounds were submitted to gold catalysis as depicted in the scheme below.

**Procedure 5.10.A:** To a solution of alkyne (0.1 mmol, 1 equiv.) in dry nitromethane (500 μL) was added XphosAu(NCCH3)SbF6 (3.8 mg, 0.04 equiv.) The reaction mixture was heated up to reflux (100 °C) being periodically monitored (TLC). Upon completion, the mixture was evaporated and purified by flash column chromatography.
Procedure 5.10.B: To a solution of alkyne (0.1 mmol, 1 equiv.) in dry nitropropane (250 µL) was added XphosAu(NCCH\textsubscript{3})SbF\textsubscript{6} (3.8 mg, 0.04 equiv.) The reaction mixture was heated to reflux (130 °C) being periodically monitored (TLC). Upon completion, the mixture was evaporated and purified by flash column chromatography.

<table>
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<tr>
<th>1-Oxa-spiro[4.5]dec-6-ene-6,9,9-tricarboxylic acid 6-ethyl ester 9,9-dimethyl ester</th>
<th>5.18e</th>
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<tr>
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<td>MF: C\textsubscript{16}H\textsubscript{22}O\textsubscript{7}</td>
</tr>
<tr>
<td></td>
<td>MW = 326 g.mol\textsuperscript{-1}</td>
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</table>

Method: See general procedure 5.10.A using (1 equiv., 0.1 mmol, 33 mg) of 5-methoxycarbonyl-5-(tetrahydro-furan-2-ylmethyl)-hex-2-yne dioic acid 1-ethyl ester 6-methyl ester.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt) / R\textsubscript{f} (7:3 PE AcOEt): 0.43.

Product: Transparent oil.

Isolated yield: 91%.

\textsuperscript{1}H NMR (δ, ppm) (CDCl\textsubscript{3}, 400 MHz)

- 7.09 (dd, J = 2.9Hz, J = 5.1Hz, 1H, CH=); 4.20-4.10 (m, 2H, OCH\textsubscript{2}CH\textsubscript{3}); 3.98 (dt, J = 5.4Hz, J = 7.7Hz, 1H, OCH\textsubscript{2}-THFring); 3.73 (s, 3H, OCH\textsubscript{3}); 3.69 (s, 3H, OCH\textsubscript{3}); 3.03 (dd, J = 1.3Hz, J = 5.1Hz, J = 19.1Hz, 1H, CH\textsubscript{2}CH=); 2.61 (dd, J = 1.3Hz, J = 14.2Hz, 1H, CH\textsubscript{2}); 2.47 (ddd, J = 6.2Hz, J = 9.5Hz, J = 12.4Hz, 1H, CH\textsubscript{2}-THFring); 2.35 (dd, J = 2.9Hz, J = 19.1Hz, 1H, CH\textsubscript{2}CH=); 2.23-2.12 (m, 1H, CH\textsubscript{2}-THFring); 2.02 (d, J = 14.2Hz, 1H, CH\textsubscript{2}); 2.01-1.93 (m, 1H, CH\textsubscript{2}-THFring); 1.64 (ddd, J = 6.2Hz, J = 8.5Hz, J = 12.4Hz, 1H, CH\textsubscript{2}-THFring); 1.28 (t, J = 7.2Hz, 3H, OCH\textsubscript{2}CH\textsubscript{3}).

\textsuperscript{13}C NMR (δ, ppm) (CDCl\textsubscript{3}, 100 MHz)

- 171.7 (Cq, C=O); 170.6 (Cq, C=O); 165.3 (Cq, C=O); 139.6 (Cq, C=CH); 132.6 (Cq, C=CH); 78.4 (Cq, C=O-THF ring); 67.8 (OCH\textsubscript{2}-THF ring); 60.2 (OCH\textsubscript{2}CH\textsubscript{3}); 52.7 (OCH\textsubscript{3}); 52.3 (OCH\textsubscript{3}); 51.2 (Cq); 40.6 (CH\textsubscript{2}); 35.9 (CH\textsubscript{2}); 31.2 (CH\textsubscript{2}); 26.8 (CH\textsubscript{2}); 14.1 (OCH\textsubscript{2}CH\textsubscript{3}).

IR (ν, cm\textsuperscript{-1}) (CCl\textsubscript{4})

- 2981 (s), 2953 (s), 2905 (m), 2885 (m), 2843 (m), 1740 (s), 1645 (w), 1478 (w), 1457 (m), 1437 (s), 1414 (m), 1378 (m), 1357 (m), 1324 (m), 1296 (s), 1254 (s), 1205 (s), 1184 (s), 1171 (s), 1114 (s), 1097 (s), 1074 (s), 1053 (s).

HRMS (EI+, m/z): Calculated: 326.1366 Found: 326.1368.
2-Methyl-1-oxa-spiro[4.5]dec-9-ene,7,7-dicarboxylic acid dimethyl ester and 7-Methyl-10-oxa-tricyclo[5.2.1.01.5]decane-3,3-dicarboxylic acid dimethyl ester

**Method:** See general procedure 5.10 A using (1 equiv., 0.1 mmol, 27 mg) of 2-(5-methyl-tetrahydro-furan-2-ylmethyl)-2-prop-2-ynyl-malonic acid dimethyl ester.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt)/ Rf (7:3 PE:AcOEt): 0.51.

**Product:** Transparent oil.

**Isolated yield:** 82% (inseparable mixture, ratio isomers 5.1:1).

**1H NMR (δ, ppm)**

| Major diastereoisomer: 5.81 (dd, J = 2.7Hz, J = 4.7Hz, J = 9.9Hz, 1H, CH=), 5.58 (tdd, J = 1.5Hz, J = 2.7Hz, J = 9.9Hz, 1H, CH=), 4.00-3.92 (m, 1H, OCH-TFring), 3.71 (s, 3H, OCH3), 3.69 (s, 3H, OCH3), 2.86 (tdd, J = 1.5Hz, J = 4.7Hz, J = 17.8Hz, 1H, CH3CH=), 2.61 (d, J = 13.9Hz, 1H, CH3), 2.16 (td, J = 2.7Hz, J = 17.8Hz, 1H, CH3CH=), 2.10-2.03 (m, 1H, CH2), 1.99 (d, J = 13.9Hz, 1H, CH3), 1.88-1.75 (m, 2H, CH2), 1.56-1.47 (m, 1H, CH2), 1.18 (d, J = 6.1Hz, 3H, CH3). |
| Minor diastereoisomer: 5.81 (dd, J = 2.8Hz, J = 4.6Hz, J = 9.9Hz, 1H, CH=), 5.62 (tdd, J = 1.0Hz, J = 1.8Hz, J = 9.9, 1H, CH=), 4.16-4.08 (m, 1H, OCH-TFring), 3.72 (s, 3H, OCH3), 3.69 (s, 3H, OCH3), 2.80 (tdd, J = 1.5Hz, J = 4.6Hz, J = 17.8Hz, 1H, CH3CH=), 2.52 (td, J = 1.0Hz, J = 13.8Hz, 1H, CH2), 2.18 (td, J = 2.8Hz, J = 17.8Hz, 1H, CH3CH=), 2.16 (d, J = 13.8Hz, 1H, CH3), 2.08-2.01 (m, 1H, CH2), 1.89-1.79 (m, 2H, CH2), 1.67-1.60 (m, 1H, CH2), 1.18 (d, J = 6.1Hz, 3H, CH3). |
| Formal [3+2] product: 3.71 (s, 6H, OCH3 x2), 2.66 (d, J = 15.0Hz, 1H, CH2), 2.53 (d, J = 15.0Hz, 1H, CH2), 2.35 (dd, J = 12.4Hz, J = 7.7Hz, 1H, CH2), 2.21-2.13 (m, 1H, CH2), 2.10-2.01 (m, 3H, CH=CH2), 1.90 -1.75 (m, 3H, CH2), 1.57-1.46 (m, 1H, CH2), 1.17 (d, J = 6.1Hz, 3H, CH3). |

**13C NMR (δ, ppm)**

| Major diastereoisomer: 172.4 (Cq, C=O), 171.2 (Cq, C=O), 131.3 (CH=), 126.1 (CH=), 78.0 (Cq, C=O TFring), 73.9 (CHO-TFring), 52.6 (OCH3), 52.3 (OCH3), 51.8 (Cq), 38.5 (CH2), 38.4 (CH2), 33.5 (CH2), 30.4 (CH2), 21.4 (CH3). |
| Minor diastereoisomer: 172.3 (Cq, C=O), 171.3 (Cq, C=O), 131.1 (CH=), 125.7 (CH=), 78.1 (Cq, C=O TFring), 75.6 (CHO-TFring), 52.6 (OCH3), 52.4 (OCH3), 51.8 (Cq), 40.9 (CH2), 38.9 (CH2), 33.5 (CH2), 30.5 (CH2), 21.9 (CH3). |
| Formal [3+2] product: 173.0 (Cq, C=O), 171.7 (Cq, C=O), 94.3 (Cq, C=O), 85.4 (Cq, C=O), 62.4 (Cq), 52.7 (OCH3), 52.6 (OCH3), 48.3 (CH), 45.9 (CH2). |

MW = 268 g.mol⁻¹
41.0 (CH₂), 37.6 (CH₂), 36.4 (CH₂), 32.9 (CH₂), 21.2 (CH₃).

IR (ν, cm⁻¹) (CCl₄) 3031 (m), 2970 (s), 2952 (s), 2933 (w), 2870 (m), 2842 (w), 1740 (s), 1436 (s), 1395 (m), 1381 (m), 1353 (w), 1301 (m), 1250 (s), 1211 (s), 1198 (s), 1174 (s), 1148 (m), 1097 (s), 1048 (m), 1008 (w).

HRMS (EI+, m/z) : Calculated: 268.1311 Found: 268.1298.

Method: See general procedure 5.10.A using (1 equiv., 0.1 mmol, 33 mg) of 2-(5-phenyl-tetrahydro-furan-2-ylmethyl)-2-prop-2-ynyl-malonic acid dimethyl ester.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt)/RF (8:2 PE: AcOEt): 0.29.

Product: Transparent oil.

Isolated yield: 76% (inseparable mixture, ratio isomers 2.3:1).

¹H NMR (δ, ppm) (CDCl₃, 400 MHz)

**Major diasteroisomer:** 7.37-7.19 (m, 5H, CH-Ph), 5.86 (ddd, J = 2.6Hz, J = 4.8Hz, J = 10.0Hz, 1H, CH=), 5.70 (d, J = 10.0Hz, 1H, CH=), 4.90 (dd, J = 6.8Hz, J = 7.6Hz, 1H, OCH-THF ring), 3.73 (s, 3H, OCH₃), 3.55 (s, 3H, OCH₃), 2.91 (dd, J = 4.6Hz, J = 17.9Hz, 1H, CH₂-CH=), 2.74 (d, J = 14.0Hz, 1H, CH₂), 2.44-2.35 (m, 1H, CH₂-THF ring), 2.21 (td, J = 2.6Hz, J = 17.9Hz, 1H, CH₂-THF ring), 2.13 (d, J = 14.0Hz, 1H, CH₂), 1.97-1.92 (m, 2H, CH₂-THF ring).

**Minor diasteroisomer:** 7.35-7.19 (m, 5H, CH-Ph), 5.86 (ddd, J = 2.8Hz, J = 4.6Hz, J = 10.0Hz, 1H, CH=), 5.76 (d, J = 10.0Hz, 1H, CH=), 5.02 (t, J = 6.8Hz, 1H, OCH-THF ring), 3.73 (s, 3H, OCH₃), 3.49 (s, 3H, OCH₃), 2.81 (dd, J = 4.6Hz, J = 17.7Hz, 1H, CH₂-CH=), 2.76 (d, J = 14.0Hz, 1H, CH₂), 2.40-2.33 (m, 1H, CH₂-THF ring), 2.34 (d, J = 14.0Hz, 1H, CH₂), 2.26 (td, J = 2.8Hz, J = 17.7Hz, 1H, CH₂-THF ring), 2.04-1.92 (m, 3H, CH₂-THF ring).

**Formal [3+2] product:** 7.37-7.19 (m, 5H, CH-Ph), 3.76 (s, 3H, OCH₃), 3.74 (s, 3H, OCH₃), 2.83 (d, J = 14.9Hz, 1H, CH₂), 2.62 (d, J = 14.9Hz, 1H, CH₂), 2.44-2.35 (m, 2H, CH₂), 2.32-2.25 (m, 2H, CH₂), 1.97-1.92 (m, 2H, CH₂), 1.90-
1.81 (m, 2H, CH₂), 1.71 (m, 1H, CH₂).

\(^{13}\)C NMR (δ, ppm)  
(CDCl₃, 100 MHz)  
**Major diasteroisomer:** 172.3 (Cq, C=O), 171.0 (Cq, C=O), 143.0 (Cq, Ph), 130.7 (CH), 128.1 (CH), 127.0 (CH), 126.2 (CH), 125.6 (CH), 125.6 (CH), 79.5 (OCH-THFring), 78.5 (Cq, C-O THFring), 52.6 (OCH₃), 52.4 (OCH₃), 52.3 (Cq), 38.8 (CH₂), 38.5 (CH₂), 34.9 (CH₂), 30.5 (CH₂).

**Minor diasteroisomer:** 172.2 (Cq, C=O), 171.1 (Cq, C=O), 142.9 (Cq, Ph), 130.6 (CHPh), 128.0 (CHPh), 127.2 (CHPh), 126.1 (CHPh), 125.7 (CHPh), 81.2 (OCH₃), 78.9 (Cq, C-O THFring), 52.6 (OCH₃), 52.5 (OCH₃), 52.0 (Cq), 40.3 (CH₂), 38.4 (CH₂), 34.6 (CH₂), 30.5 (CH₂).

**Formal 3+2 product:** 172.9 (Cq, C=O), 171.7 (Cq, C=O), 142.5 (Cq, Ph), 128.1 (CHPh), 126.9 (CHPh), 125.0 (CHPh), 94.5 (Cq), 88.7 (Cq), 62.5 (Cq), 52.8 (OCH₃), 52.6 (OCH₃), 48.1 (CH), 47.6 (CH₂), 41.0 (CH₂), 37.8 (CH₂), 36.8 (CH₂), 32.4 (CH₂).

IR (ν, cm⁻¹) (CCl₄)  
3066 (w), 3031 (w), 2952 (m), 2871 (w), 1740 (s), 1494 (w), 1448 (m), 1435 (w), 1299 (w), 1252 (m), 1217 (w), 1198 (w), 1180 (w), 1095 (w), 1050 (w), 1027 (w).

HRMS (EI+, m/z):  
Calculated: 330.1467  
Found: 330.1463.

**Product:**  
Transparent oil.

**Isolated yield:** 82% (inseparable mixture, ratio isomers 1.5:1).

\(^{1}H\) NMR (δ, ppm)  
(CDCl₃, 400 MHz)  
**Major diasteroisomer:** 5.82 (ddd, J = 2.5Hz, J = 4.7Hz, J = 9.9Hz, 1H, CH=), 5.54 (d, J = 9.9Hz, 1H, CH=), 4.30-4.23 (m, 1H, OCH-THFring), 3.71 (s, 3H, OCH₃), 3.69 (s, 3H, OCH₃), 3.66 (s, 3H, OCH₃), 2.85 (dd, J = 4.7Hz, J = 17.9Hz, 1H, CH₂CH=), 2.60 (d, J = 14.5Hz, 1H, CH₂), 2.58 (d, J = 15.4Hz, 1H, CH₂), 2.41 (dd, J = 7.2Hz, J = 15.4Hz, 1H, CH₂), 2.23-2.13 (m, 1H, CH₂-THFring), 2.16 (dt, J = 2.5Hz, J = 17.9Hz, 1H, CH=), 2.03 (d, J = 14.5Hz,
1H, CH₃), 1.83 (t, J = 7.2Hz, 2H, CH₂-THFring), 1.71-1.62 (m, 1H, CH₂-THFring).

**Minor diasteroisomer:** 5.81 (ddd, J = 3.0Hz, J = 4.4Hz, J = 10.0Hz, 1H, CH=), 5.62 (d, J = 10.0Hz, 1H, CH=), 4.41-4.34 (m, 1H, OCH-THFring), 3.72 (s, 3H, OCH₃), 3.66 (s, 3H, OCH₃), 2.76 (dd, J = 4.4Hz, J = 17.8Hz, 1H, CH₂CH=), 2.62 (dd, J = 6.1Hz, J = 15.3Hz, 1H, CH₂), 2.49 (dd, J = 14.1Hz, 1H, CH₂), 2.41 (dd, J = 7.4Hz, J = 15.3Hz, 1H, CH₂), 2.24-2.17 (m, 3H, CH₂-THF ring + CH₂CH=), 1.88-1.82 (m, 2H, CH₂-THF ring), 1.80-1.69 (m, 1H, CH₂-THF ring).

**Formal [3+2] product:** 3.70 (s, 3H, OCH₃), 3.67 (s, 3H, OCH₃), 3.65 (s, 3H, OCH₃), 2.63-2.55 (m, 1H, CH₂), 2.43-2.32 (m, 1H, CH₂), 2.23-2.12 (m, 5H, CH₂), 1.86-1.63 (m, 6H, CH₂).

**13C NMR (δ, ppm)**

<table>
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<tr>
<th>Component</th>
<th>Assignments</th>
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<td>Major diasteroisomer:</td>
<td>172.3 (Cq, C=O), 171.5 (Cq, C=O), 171.1 (Cq, C=O), 130.7 (CH=), 126.5 (CH=), 78.5 (Cq, C-O THF ring), 74.2 (OCH-THF ring), 52.7 (OCH₃), 52.4 (OCH₃), 51.7 (Cq), 51.5 (OCH₃), 40.9 (CH₂), 38.5 (CH₂), 38.1 (CH₂), 31.5 (CH₂), 30.4 (CH₂).</td>
</tr>
<tr>
<td>Minor diasteroisomer:</td>
<td>172.2 (Cq, C=O), 171.4 (Cq, C=O), 171.2 (Cq, C=O), 130.4 (CH=), 125.9 (CH=), 78.7 (Cq, C-O THF ring), 75.4 (OCH-THF ring), 52.6 (OCH₃), 52.5 (OCH₃), 51.9 (Cq), 51.5 (OCH₃), 41.2 (CH₂), 40.5 (CH₂), 38.1(CH₂), 31.4 (CH₂), 30.4 (CH₂).</td>
</tr>
<tr>
<td>Formal [3+2] product:</td>
<td>172.8 (Cq, C=O), 171.4 (Cq, C=O), 170.7 (Cq, C=O), 94.3 (Cq), 84.6 (Cq), 62.3 (Cq), 52.3 (OCH₃), 51.5 (OCH₃), 47.7 (CH), 44.1 (CH₂), 40.7 (CH₂), 37.4 (CH₂), 35.1 (CH₂), 32.3 (CH₂), 29.6 (CH₂).</td>
</tr>
</tbody>
</table>

**IR (ν, cm⁻¹)**

<table>
<thead>
<tr>
<th>Compound</th>
<th>ν (cm⁻¹)</th>
</tr>
</thead>
<tbody>
<tr>
<td>CCl₄</td>
<td>3031 (w), 2953 (m), 2843 (w), 1741 (s, C=O), 1436 (m), 1382 (w), 1302 (w), 1251 (s), 1200 (s), 1175 (s), 1146 (w), 1097 (m), 1052 (m).</td>
</tr>
</tbody>
</table>

**HRMS (EI+, m/z):** Calculated: 326.1366 Found: 326.1357.

**Name not found (Beilstein):** 5.18i + 5.20i

**Method:** See general procedure 5.10.A using (1 equiv., 0.1 mmol, 31 mg) of 2-(octahydro-benzofuran-2-ylmethyl)-2-prop-2-ynyl-malonic acid dimethyl ester.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt) Rf (8:2 PE:AcOEt): 0.40.

**Product:** Transparent oil.
Isolated yield: 74% (inseparable mixture, ratio isomers 2.1:1).

Mass spectrometry:
HRMS (EI+, m/z): Calculated: 308.1624 Found: 308.1620.

**1H NMR** (δ, ppm) (CDCl₃, 400 MHz)

**Major diasteroisomer:** 5.79 (ddd, J = 2.6Hz, J = 4.6Hz, J = 9.9Hz, 1H, CH=), 5.70 (d, J = 10.0Hz, 1H, CH=), 3.84 (q, J = 3.7Hz, 1H, OCH-THF ring), 3.70 (s, 3H, OCH₃), 3.68 (s, 3H, OCH₃), 2.86 (dd, J = 4.6Hz, J = 17.9Hz, 1H, CH₂CH=), 2.63 (d, J = 14.0Hz, 1H, CH₃), 2.14 (td, J = 2.5Hz, J = 17.9Hz, 1H, CH₂CH=), 2.09-2.02 (m, 1H, CH-octahydro-benzofuran ring), 1.97 (d, J = 14.0Hz, 1H, CH₂), 1.92-1.85 (m, 2H, CH₂-octahydro-benzofuran ring), 1.77-1.41 (m, 8H, CH₂-octahydro-benzofuran ring).

**Minor diasteroisomer:** 5.73 (ddd, J = 2.0Hz, J = 4.0Hz, J = 10.0Hz, 1H, CH=), 5.68 (d, J = 10.0Hz, 1H, CH=), 3.95 (dd, J = 5.2Hz, J = 10.6Hz, 1H, OCH-THF ring), 3.72 (s, 3H, OCH₃), 3.70 (s, 3H, OCH₃), 2.78 (td, J = 4.0Hz, J = 17.7Hz, 1H, CH₂CH=), 2.65 (d, J = 13.8Hz, 1H, CH₂), 2.24-2.17 (m, 1H, CH-octahydro-benzofuran ring), 2.17 (td, J = 2.0Hz, J = 17.7Hz, 1H, CH₂CH=), 1.88 (dd, J = 7.2Hz, J = 12.8Hz, 1H, CH₂-octahydro-benzofuran ring), 1.76 (dd, J = 6.3Hz, J = 12.8Hz, 1H, CH₂-octahydro-benzofuran ring), 1.66-1.44 (m, 8H, CH₂-octahydro-benzofuran ring).

**Formal [3+2] product:** 3.71 (s, 3H, OCH₃), 3.70 (s, 3H, OCH₃), 2.64 (d, J = 15.3Hz, 1H, CH₂), 2.55 (d, J = 15.3Hz, 1H, CH₂), 2.32 (q, J = 4.1Hz, 1H, CH), 2.09-2.02 (m, 2H, CH₂CH₂), 1.77-1.41 (m, 4H, CH₂), 1.38-1.28 (m, 5H, CH₂), 1.20-1.03 (m, 4H, CH₂).

**13C NMR** (δ, ppm) (CDCl₃, 100 MHz)

**Major diasteroisomer:** 172.4 (Cq, C=O), 171.3 (Cq, C=O), 132.3 (CH=), 125.6 (CH=), 76.5 (Cq, C-O THF ring), 75.4 (OCH-THF ring), 52.5 (OCH₃), 52.3 (OCH₃), 51.9 (Cq), 45.2 (CH₂), 39.3 (CH₂), 38.9 (CH), 30.1 (CH₂), 28.5 (CH₂), 28.2 (CH₂), 24.2 (CH₂), 20.5 (CH₂).

**Minor diasteroisomer:** 172.3 (Cq, C=O), 171.4 (Cq, C=O), 131.7 (CH=), 124.0 (CH=), 77.2 (Cq, C-O THF ring), 76.9 (OCH-THF ring), 52.6 (OCH₃), 52.3 (OCH₃), 52.1 (Cq), 43.5 (CH₂), 41.8 (CH₂), 37.9 (CH), 30.2 (CH₂), 29.2 (CH₂), 27.8 (CH₂), 23.1 (CH₂), 21.7 (CH₂).

**Formal [3+2] product:** 173.2 (Cq, C=O), 171.8 (Cq, C=O), 94.6 (Cq, C-O), 85.7 (Cq, C-O), 62.4 (Cq), 52.8 (OCH₃), 52.6 (OCH₃), 48.2 (CH), 45.4 (CH₂), 41.8 (CH), 41.4 (CH₂), 41.0 (CH₂), 37.6 (CH₂), 33.5 (CH₂), 31.3 (CH₂), 24.9 (CH₂), 23.0 (CH₂).

**IR** (ν, cm⁻¹) (CCl₄): 3031 (w), 2935 (m), 2855 (m), 1740 (s), 1435 (m), 1395 (w), 1380 (w), 1364 (w), 1322 (w), 1300 (w), 1249 (s), 1201 (m), 1178 (m), 1156 (w), 1144 (w), 1097 (w), 1081 (w), 1047 (w), 1017 (w).

**HRMS** (EI+, m/z): Calculated: 308.1624 Found: 308.1620.
Name not found (Beilstein)

\[ \text{MF: } C_{17}H_{24}O_5 \]
\[ \text{MW = 308 g.mol}^{-1} \]

Method: See general procedure 5.10.A using (1 equiv. 0.1 mmol, 31 mg) of 2-(octahydro-benzofuran-2-ylmethyl)-2-prop-2-ynyl-malic acid dimethyl ester.

Purification: Flash column chromatography (silica gel, 9:1 PE:AcOEt)/ \( R_f \) (8:2 PE: AcOEt): 0.34.

Product: Transparent oil.

Isolated yield: 50% (only two diasteroisomers, ratio 2:1).

\(^1\text{H NMR} \) (\( \delta \text{, ppm} \)) (CDCl\(_3\), 400 MHz)

**Major diasteroisomer:** 5.76 (ddd, \( J = 2.5\text{Hz}, J = 4.8\text{Hz}, J = 10.0\text{Hz}, 1\text{H, CH}= \)), 5.59 (d, \( J = 10.0\text{Hz}, 1\text{H, CH}= \)), 3.78-3.07 (m, 1H, OCH-THF ring), 3.71 (s, 3H, OCH\(_3\)), 3.68 (s, 3H, OCH\(_3\)), 3.04 (td, \( J = 3.4\text{Hz}, J = 10.0\text{Hz}, 1\text{H, CH} \)), 2.88 (ddd, \( J = 4.8\text{Hz}, J = 17.8\text{Hz}, 1\text{H, CH}= \)), 2.60 (d, \( J = 14.0\text{Hz}, 1\text{H, CH} \)), 2.17 (dt, \( J = 2.5\text{Hz}, J = 17.8\text{Hz}, 1\text{H, CH}= \)), 2.14 (d, \( J = 14.0\text{Hz}, 1\text{H, CH} \)).

**Minor diasteroisomer:** 5.82 (ddd, \( J = 3.2\text{Hz}, J = 4.0\text{Hz}, J = 9.8\text{Hz}, 1\text{H, CH}= \)), 5.66 (d, \( J = 9.8\text{Hz}, 1\text{H, CH}= \)), 3.78-3.07 (m, 1H, CHO-THF ring), 3.71 (s, 3H, OCH\(_3\)), 3.67 (s, 3H, OCH\(_3\)), 3.18 (td, \( J = 3.8\text{Hz}, J = 10.0\text{Hz}, 1\text{H, CH} \)), 2.77 (dd, \( J = 4.4\text{Hz}, J = 17.8\text{Hz}, 1\text{H, CH}= \)), 2.53 (d, \( J = 13.8\text{Hz}, 1\text{H, CH} \)), 2.24 (dt, \( J = 2.6\text{Hz}, J = 17.8\text{Hz}, 1\text{H, CH} \)), 2.15 (d, \( J = 13.8\text{Hz}, 1\text{H, CH} \)).

\(^{13}\text{C NMR} \) (\( \delta \text{, ppm} \)) (CDCl\(_3\), 100 MHz)

**Major diasteroisomer:** 172.4 (Cq, C=O), 171.3 (Cq, C=O), 131.2 (CH=), 124.9 (CH=), 81.8 (OCH-THF ring), 76.6 (Cq, O-THF ring), 52.6 (OCH\(_3\)), 52.3 (OCH\(_3\)), 52.0 (Cq), 45.5 (CH), 44.4 (CH\(_2\)), 40.0 (CH\(_2\)), 31.3 (CH\(_2\)), 30.3 (CH\(_2\)), 29.1 (CH\(_2\)), 25.7 (CH\(_2\)), 24.3 (CH\(_2\)).

**Minor diasteroisomer:** 172.2 (Cq, C=O), 171.3 (Cq, C=O), 131.9 (CH=), 125.8 (CH=), 83.3 (OCH-THF ring), 72.2 (Cq, O-THF ring), 52.5 (OCH\(_3\)), 52.3 (OCH\(_3\)), 51.9 (Cq), 45.3 (CH), 44.5 (CH\(_2\)), 41.2 (CH\(_2\)), 31.9 (CH\(_2\)), 30.4 (CH\(_2\)), 29.0 (CH\(_2\)), 25.8 (CH\(_2\)), 24.3 (CH\(_2\)).

IR (\( \nu \text{, cm}^{-1} \)) (CCl\(_4\))

3030 (w), 2935 (m), 2858 (w), 1741 (s), 1436 (m), 1384 (w), 1351 (w), 1325 (w), 1299 (w), 1247 (m), 1202 (w), 1181 (w), 1141 (w), 1110 (w), 1081 (w), 1060 (w).

HRMS (EI+, m/z) : Calculated: 308.1624  Found: 308.1629.
2-Methyl-1-oxa-spiro[4.5]dec-6-ene-6,9,9-tricarboxylic acid 6-ethyl ester 9,9-dimethyl ester

\[ \text{MeO}_2\text{C} - \text{CO}_2\text{Me} - \text{EtO}_2\text{C} \]

\[ \text{MF: C}_{17}\text{H}_{24}\text{O}_7 \]

\[ \text{MW} = 340 \text{ g.mol}^{-1} \]

Method: See general procedure 5.10.A using (1 equiv., 0.1 mmol, 34 mg) of 5-methoxycarbonyl-5-(5-methyl-tetrahydro-furan-2-ylmethyl)-hex-2-ynedioic acid 1-ethyl ester 6-methyl ester.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt)/Rf (7:3PE:AcOEt): 0.39.

Product: Transparent oil.

Isolated yield: 78%.

\(^1\)H NMR (\(\delta\), ppm) (CDCl\(_3\), 400 MHz)

**Major diasteroisomer:** 7.10 (dd, \(J = 3.2\text{Hz}, J = 4.9\text{Hz}, 1\text{H, CH}=\)), 4.47-4.39 (m, 1H, OCH-THF ring), 4.27-4.14 (m, 2H, OCH\(_2\)CH\(_3\)), 3.77 (s, 3H, OCH\(_3\)), 3.74 (s, 3H, OCH\(_3\)), 2.99 (ddd, \(J = 1.3\text{Hz}, J = 4.9\text{Hz}, J = 19.0\text{Hz}, 1\text{H, CH}=\)), 2.78-2.66 (m, 1H, CH\(_2\)-THF ring), 2.61 (dd, \(J = 1.3\text{Hz}, J = 14.0\text{Hz}, 1\text{H, CH}_2\)), 2.43 (dd, \(J = 3.1\text{Hz}, J = 19.0\text{Hz}, 1\text{H, CH}_2\)), 2.20-2.14 (m, 1H, CH\(_2\)-THF ring), 1.77-1.65 (m, 2H, CH\(_2\)-THF ring), 1.33 (t, \(J = 7.1\text{Hz}, 3\text{H, OCH}_2\text{C}_3\)), 1.24 (d, \(J = 6.1\text{Hz}, 3\text{H, CH}_3\)).

**Minor diasteroisomer:** 7.13-7.09 (m, 1H, CH=), 4.27-4.14 (m, 2H, OCH\(_2\)CH\(_3\)), 4.02-3.95 (m, 1H, OCH-THF ring), 3.77 (s, 3H, OCH\(_3\)), 3.74 (s, 3H, OCH\(_3\)), 3.16 (ddd, \(J = 1.7\text{Hz}, J = 5.3\text{Hz}, J = 19.1\text{Hz}, 1\text{H, CH}=\)), 2.76 (d, \(J = 14.1\text{Hz}, 1\text{H, CH}_2\)), 2.33 (dd, \(J = 2.7\text{Hz}, J = 19.2\text{Hz}, 1\text{H, CH}_2\)), 2.06-2.02 (m, 1H, CH\(_2\)-THFring), 2.01 (d, \(J = 14.1\text{Hz}, 1\text{H, CH}_2\)), 1.93-1.85 (m, 1H, CH\(_2\)-THF ring), 1.77-1.65 (m, 2H, CH\(_2\)-THF ring), 1.33 (t, \(J = 7.1\text{Hz}, 3\text{H, OCH}_2\text{C}_3\)), 1.27 (d, \(J = 6.0\text{Hz}, 3\text{H, CH}_3\)).

\(^{13}\)C NMR (\(\delta\), ppm) (CDCl\(_3\), 100 MHz)

**Major diasteroisomer:** 171.8 (Cq, C=O), 170.7 (Cq, C=O), 165.5 (Cq, C=O), 139.0 (C=C\(_\text{H}\)), 133.3 (Cq, C=C\(_\text{H}\)), 78.8 (Cq, C-O-THFring), 77.1 (OCH-THFring), 60.2 (OCH\(_2\)CH\(_3\)), 52.6 (OCH\(_3\)), 52.5 (OCH\(_3\)), 51.1 (Cq), 43.1 (CH\(_2\)), 36.5 (CH\(_3\)), 34.4 (CH\(_2\)), 31.3 (CH\(_3\)), 21.8 (CH\(_3\)), 14.1 (OCH\(_2\)C\(_3\)).

**Minor diasteroisomer:** 171.9 (Cq, C=O), 170.5 (Cq, C=O), 165.3 (Cq, C=O), 139.6 (C=C\(_\text{H}\)), 132.8 (Cq, C=C\(_\text{H}\)), 78.2 (Cq, C-O-THFring), 74.4 (OCH-THFring), 67.6 (OCH\(_2\)CH\(_3\)), 52.7 (OCH\(_3\)), 52.3 (OCH\(_3\)), 51.1 (Cq), 45.9 (CH\(_2\)), 40.5 (CH\(_3\)), 37.4(CH\(_3\)), 34.6 (CH\(_3\)), 20.3 (CH\(_3\)), 14.1 (OCH\(_2\)C\(_3\)).

IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))

2971 (s), 2954 (s), 2931 (s), 2905 (m), 2870 (m), 2842 (w), 1740 (s), 1644 (w), 1477 (w), 1458 (s), 1436 (s), 1414 (m), 1375 (m), 1357 (m), 1324 (m), 1299 (s), 5.18m
1258 (s), 1211 (s), 1174 (s), 1140 (m), 1096 (s), 1070 (s), 1038 (s), 1009 (m).

HRMS (EI+, m/z):  

4-Methylene-hexahydro-cyclopenta[b]furan-6,6-dicarboxylic acid dimethyl ester and 2,3,3a,7a-Tetrahydro-6H-benzofuran-7,7-dicarboxylic acid dimethyl ester  

\[
\begin{align*}
4\text{-Methylene-hexahydro-cyclopenta[b]furan-6,6-dicarboxylic acid dimethyl ester} \\
5.37b + 5.38b
\end{align*}
\]

MF: C_{12}H_{16}O_{5}
MW = 240 g.mol^{-1}

Method: See general procedure 5.10.A using (1 equiv., 0.1 mmol, 24 mg) of 2-prop-2-yny-2-(tetrahydro-furan-3-yl)-malonic acid dimethyl ester

Purification: Flash column chromatography (silica gel, 8:2 PE: AcOEt) / R_{f} (7:3 PE:AcOEt): 0.39.

Product: Transparent oil.

Isolated yield: 95 % (ratio 6:1).

\(^1\)H NMR (δ, ppm)  
(CDCl\(_3\), 400 MHz)  
\textbf{5-exo product:} 5.19 (d, J = 2.6Hz, 1H, C=CH\(_2\)), 5.11 (s, 1H, C=CH\(_2\)), 4.71 (d, J = 7.2Hz, 1H, OCH-THF ring), 3.91 (td, J = 8.0Hz, J = 3.3Hz, 1H, OCH\(_2\)-THF ring), 3.74 (s, 3H, OCH\(_3\)), 3.71 (s, 3H, OCH\(_3\)), 3.65 (td, J = 8.0Hz, J = 5.8Hz, 1H, OCH\(_2\)-THF ring), 3.45 (dd, J = 16.5Hz, J = 8.5Hz, 1H, CH\(_3\)), 3.20 (dq, J = 16.2Hz, J = 1.0Hz, 1H, CH\(_3\)), 2.64 (d, J = 16.2Hz, 1H, CH\(_3\)), 2.03-1.95 (m, 1H, CH\(_2\)-THF ring), 1.57-1.48 (m, 1H, CH\(_2\)-THF ring).  
\textbf{6-endo product:} 5.69 (ddd, J = 10.3Hz, J = 5.3Hz, J = 2.3Hz, 1H, =CH\(_2\)), 5.58 (dt, J = 10.3Hz, J = 3.2Hz, 1H, =CH\(_2\)), 4.78-4.76 (m, 1H, OCH-THF ring), 3.90-3.72 (m, 2H, OCH\(_2\)-THF ring), 3.74 (s, 3H, OCH\(_3\)), 3.72 (s, 3H, OCH\(_3\)), 3.12 (dd, J = 17.5Hz, J = 8.8Hz, 1H, CH\(_3\)), 2.696-2.59 (m, 2H, CH\(_2\)), 1.89-1.87 (m, 1H, CH\(_2\)), 1.80-1.68 (m, 1H, CH\(_2\)).

\(^{13}\)C NMR (δ, ppm)  
(CDCl\(_3\), 100 MHz)  
\textbf{5-exo product:} 171.8 (Cq, C=O), 170.0 (Cq, C=O), 147.0 (Cq, C=CH\(_2\)), 113.1 (C=CH\(_2\)), 84.6 (OCH-THF ring), 68.5 (OCH\(_2\)-THF ring), 61.9 (Cq), 52.9 (OCH\(_3\)), 52.8 (OCH\(_3\)), 48.0 (CH), 38.0 (CH\(_2\)), 29.8 (CH\(_2\)).  
\textbf{6-endo product:} 171.3 (Cq, C=O), 170.8 (Cq, C=O), 127.9 (CH\(_=\)), 125.9 (CH\(_=\)), 74.4 (OCH-THF ring), 65.8 (OCH\(_2\)-THF ring), 56.2 (Cq), 52.6 (OCH\(_3\)), 52.3 (OCH\(_3\)), 40.4 (CH), 26.8 (CH\(_2\)), 25.9 (CH\(_2\)).

357
IR (v, cm⁻¹) (CCl₄)  2981 (w), 2953 (m), 2855 (w), 1736 (s), 1435 (s), 1242 (m), 1160 (s), 1063 (s).

HRMS (EI+, m/z):  Calculated: 240.0998  Found: 240.0994.

2-Isopropyl-6-methylene-hexahydro-cyclopenta[b]furan-4,4-dicarboxylic acid dimethyl ester  5.37c

\[
\text{MF: C}_{15}\text{H}_{22}\text{O}_5
\]

\[
\text{MW = 282 g.mol}^{-1}
\]

**Method**:  See general procedure 5.10.A using (1 equiv., 0.1 mmol, 28 mg) of 2-(5-isopropyl-tetrahydro-furan-3-yl)-2-prop-2-ynyl-malonic acid dimethyl ester.

**Purification**:  Flash column chromatography  (silica gel, 8:2 PE:AcOEt)/ Rf. (8:2 PE: AcOEt): 0.58.

**Product**:  Transparent oil.

**Isolated yield**:  93% (inseparable mixture, ratio diasteroisomers 1:5).

**¹H NMR (δ, ppm)**

(CDCl₃, 400 MHz)

Major diasteroisomer  5.17 (s, 1H, C=CH₂), 5.07 (s, 1H, C=CH₂), 4.65 (d, J = 7.3Hz, 1H, OCH-THF ring), 3.73 (s, 3H, OCH₃), 3.69 (s, 3H, OCH₃), 3.50-3.44 (m, 2H, OCH-THF ring + CH), 3.20 (d, J = 15.8Hz, 1H, CH₂-THF ring), 1.95-1.89 (m, 1H, CH₂-THF ring), 1.73-1.66 (m, 1H, CH-Pr), 1.15 (q, J = 11.2Hz, 1H, CH₂-THF ring), 0.95 (d, J = 6.7Hz, 3H, CH₃-Pr), 0.84 (d, J = 6.7Hz, 3H, CH₃-Pr).

Minor diasteroisomer:  5.17 (s, 1H, C=CH₂), 5.09 (s, 1H, C=CH₂), 4.83 (d, J = 6.9Hz, 1H, OCH-THF ring), 3.73 (s, 3H, OCH₃), 3.69 (s, 3H, OCH₃), 3.50-3.44 (m, 2H, OCH-THF ring + CH), 3.18 (d, J = 16.0Hz, 1H, CH₂-THF ring), 2.66 (d, J = 16.0Hz, 1H, CH₂-THF ring), 1.81-1.75 (m, 1H, CH₂-THF ring), 1.73-1.66 (m, 1H, CH-Pr), 1.54-1.48 (m, 1H, CH₂-THF ring), 0.92 (d, J = 6.7Hz, 3H, CH₃-Pr), 0.82 (d, J = 6.7Hz, 3H, CH₃-Pr).

**¹³C NMR (δ, ppm)**

(CDCl₃, 100 MHz)

Major diasteroisomer:  171.7 (Cq, C=O), 170.0 (Cq, C=O), 146.8 (Cq, C=CH₂), 113.0 (C=CH₂), 86.0 (OCH-THF ring), 84.0 (OCH-THF ring), 61.7 (Cq), 52.8 (OCH₃), 52.5 (OCH₃), 48.1 (CH), 37.3 (CH₂), 33.0 (CH₂), 32.4 (CH-Pr), 19.6 (CH₃-Pr), 18.4 (CH₃-Pr).

Minor diasteroisomer:  171.8 (Cq, C=O), 170.2 (Cq, C=O), 147.8 (Cq, C=CH₂), 112.5 (C=CH₂), 84.7 (OCH-THF ring), 84.3 (OCH-THF ring), 62.1 (Cq), 52.7 (OCH₃), 52.5 (OCH₃), 47.5 (CH), 38.7 (CH₂), 32.3 (CH-Pr), 31.7 (CH₃), 19.1 (CH₃-Pr), 18.2 (CH₃-Pr).

**IR (v, cm⁻¹)** (CCl₄)

3082 (w), 3030 (w), 2956 (s), 2909 (s), 2873 (s), 2843 (s), 1744 (s), 1673 (w), 1469 (s), 1435 (s), 1450 (s), 1406 (w), 1389 (m), 1366 (m), 1331 (w), 1271 (s), 1238 (s), 1193 (s), 1161(s), 1113 (s), 1079 (s), 1032 (s).

358
**HRMS (El+, m/z):**
Calculated: 282.1467    Found: 282.1462.

**6-Methylene-2-pentyl-hexahydro-cyclopenta[b]furan-4,4-dicarboxylic acid dimethyl ester**

![Chemical Structure Image]

**MF:** C₁₇H₂₆O₅

**MW:** 310 g/mol

**Method:**
See general procedure 5.10.A using (1 equiv., 0.1 mmol, 31 mg) of 2-(5-pentyl-tetrahydro-furan-3-yl)-2-prop-2-ynyl-malonic acid dimethyl ester.

**Purification:**
Flash column chromatography (silica gel, 9:1 PE:AcOEt) / Rf (8:2 PE:AcOEt): 0.53.

**Product:**
Transparent oil.

**Isolated yield:**
88% (inseparable mixture, ratio diasteroisomers 1:5).

**¹H NMR (δ, ppm)**
(CDCl₃, 400 MHz)

**Major diasteroisomer:** 5.17 (s, 1H, C=CH₂), 5.07 (s, 1H, C=CH₂), 4.63 (d, J = 7.3Hz, 1H, OCH-THF ring), 3.76-3.66 (m, 1H, OCH-THF ring), 3.73 (s, 3H, OCH₃), 3.69 (s, 3H, OCH₃), 3.51-3.45 (m, 1H, CH-THF ring), 3.21 (d, J = 15.7Hz, 1H, CH₂), 2.58 (d, J = 15.7Hz, 1H, CH₂), 2.01-1.95 (m, 1H, CH₂-THF ring), 1.71-1.20 (m, 8H, CH₂-pentyl chain), 1.09 (q, J = 11.2, 1H, CH₂-THF ring), 0.86 (t, J = 6.8Hz, 3H, CH₃-pentyl chain).

**Minor diasteroisomer:** 5.17 (s, 1H, C=CH₂), 5.09 (s, 1H, C=CH₂), 4.82 (d, J = 6.8Hz, 1H, OCH-THF ring), 3.76-3.66 (m, 1H, OCH-THF ring), 3.73 (s, 3H, OCH₃), 3.70 (s, 3H, OCH₃), 3.51-3.45 (m, 1H, CH), 3.17 (d, J = 16.0Hz, 1H, CH₂), 2.65 (d, J = 16.0Hz, 1H, CH₂), 1.71-1.20 (m, 8H, CH₂-pentyl chain), 1.68-1.57 (m, 1H, CH₂-THF ring), 1.14-1.05 (q, J = 11.2, 1H, CH₂-THF ring), 0.88 (m, 3H, CH₃-pentyl chain).

**¹³C NMR (δ, ppm)**
(CDCl₃, 100 MHz)

**Major diasteroisomer:** 171.7 (Cq, C=O), 170.0 (Cq, C=O), 146.7 (Cq, C=CH₂), 113.1 (C=CH₂), 84.1 (OCH-THF ring), 80.7 (OCH-THF ring), 61.8 (Cq), 52.8 (OCH₃), 52.5 (OCH₃), 48.3 (CH), 37.4 (CH₂), 35.7 (CH₂), 34.6 (CH₂), 31.9 (CH₂), 25.9 (CH₂), 22.5 (CH₂), 13.9 (CH₃).

**Minor diasteroisomer:** 171.8 (Cq, C=O), 170.1 (Cq, C=O), 147.7 (Cq, C=CH₂), 112.6 (C=CH₂), 84.1 (OCH-THF ring), 79.5 (OCH-THF ring), 62.1 (Cq), 52.8 (OCH₃), 52.4 (OCH₃), 47.4 (CH), 38.7 (CH₂), 34.9 (CH₂), 34.4 (CH₂), 31.8 (CH₂), 25.7 (CH₂), 22.5 (CH₂), 13.9 (CH₃).

**IR (ν, cm⁻¹)** (CCl₄)
3082 (w), 3029 (w), 2955 (s), 2932 (s), 2860 (s), 1736 (s), 1673 (w), 1545 (s), 1435 (s), 1379 (w), 1273 (s), 1238 (s), 1194 (s), 1160 (s), 1092 (s), 1078 (s), 1031 (s).
**HRMS (EI+, m/z)**: Calculated: 310.1780  Found: 310.1767.

### 6-Methylene-2-phenyl-hexahydro-cyclopenta[b]furan-4,4-dicarboxylic acid dimethyl ester

![Chemical Structure](image)

**MF:** $C_{18}H_{20}O_5$  
**MW:** 316 g.mol$^{-1}$

**Method:** See general procedure 5.10.A using (1 equiv., 0.1 mmol, 32 mg) of 2-(5-phenyl-tetrahydro-furan-3-yl)-2-prop-2-ynyl-malonlic acid dimethyl ester.

**Purification:** Flash column chromatography (silica gel, 85:15 PE:AcOEt)/ $R_f$ (8:2 PE:AcOEt): 0.56.

**Product:** Transparent oil.

**Isolated yield:** 29% (inseparable mixture, ratio diasteroisomers 1:3).

**$^1H$ NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz)**

<table>
<thead>
<tr>
<th>Chemical Shift</th>
<th>Description</th>
</tr>
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<tr>
<td>7.34-7.28</td>
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<tr>
<td>5.27</td>
<td>(s, 1H, C=CH$_2$)</td>
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<td>5.16</td>
<td>(s, 1H, C=CH$_2$)</td>
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<td>4.85</td>
<td>(d, $J = 7.5$Hz, 1H, OCH-THF ring)</td>
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<tr>
<td>4.82</td>
<td>(dd, $J = 4.6$Hz, $J = 11.1$Hz, 1H, OCH-THF ring)</td>
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<tr>
<td>3.73</td>
<td>(s, 3H, OCH$_3$)</td>
</tr>
<tr>
<td>3.70</td>
<td>(s, 3H, OCH$_3$)</td>
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<tr>
<td>3.69-3.63</td>
<td>(m, 1H, CH)</td>
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<tr>
<td>3.35</td>
<td>(d, $J = 16.0$Hz, 1H, CH$_2$)</td>
</tr>
<tr>
<td>2.69</td>
<td>(d, $J = 16.0$Hz, 1H, CH$_2$)</td>
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<td>2.30</td>
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<tr>
<td>1.49</td>
<td>(q, $J = 11.9$Hz, 1H, CH$_2$-THF ring)</td>
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**$^{13}C$ NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz)**

<table>
<thead>
<tr>
<th>Chemical Shift</th>
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<tr>
<td>171.7</td>
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</tr>
<tr>
<td>169.9</td>
<td>(Cq, C=O)</td>
</tr>
<tr>
<td>146.5</td>
<td>(Cq)</td>
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<td>140.3</td>
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<td>128.4</td>
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<td>127.8</td>
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<td>126.0</td>
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<tr>
<td>113.6</td>
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<tr>
<td>84.6</td>
<td>(OCH-THF ring)</td>
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<tr>
<td>82.2</td>
<td>(OCH-THF ring)</td>
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<td>61.8</td>
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<tr>
<td>52.9</td>
<td>(OCH$_3$)</td>
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<tr>
<td>52.6</td>
<td>(OCH$_3$)</td>
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<tr>
<td>48.8</td>
<td>(CH)</td>
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<td>38.8</td>
<td>(CH$_2$)</td>
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<td>37.4</td>
<td>(CH$_3$)</td>
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**IR ($\nu$, cm$^{-1}$) (CCl$_4$)**

<table>
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<th>Frequency</th>
<th>Description</th>
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<tr>
<td>3067</td>
<td>(w)</td>
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<tr>
<td>3032</td>
<td>(w)</td>
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<td>1459</td>
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<tr>
<td>1450</td>
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<td>(s)</td>
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<td>1193</td>
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<tr>
<td>1159</td>
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<tr>
<td>1092</td>
<td>(w)</td>
</tr>
<tr>
<td>1057</td>
<td>(w)</td>
</tr>
<tr>
<td>1029</td>
<td>(w)</td>
</tr>
</tbody>
</table>

**HRMS (EI+, m/z)**: Calculated: 316.1311  Found: 316.1308.
4-[1-Ethoxycarbonyl-meth-(Z)-ylidene]-hexahydro-cyclopenta[b]furan-6,6-dicarboxylic acid dimethyl ester and 2,3,3a,7a-Tetrahydro-6H-benzofuran-4,7,7-tricarboxylic acid 4-ethyl ester 7,7-dimethyl ester

\[ \text{MF: } C_{15}H_{20}O_7 \]
\[ \text{MW = 312 g.mol}^{-1} \]

Method: See general procedure 5.10.A using (1 equiv., 0.11 mmol, 34 mg) of 5-methoxycarbonyl-5-(tetrahydro-furan-3-yl)-hex-2-yne dioic acid 1-ethyl ester 6-methyl ester.

Purification: Flash column chromatography (silica gel, 6:4 PE:AcOEt)/ \( R_f \) (7:3 PE:AcOEt): 0.21.

Product: Transparent oil.

Isolated yield: 81% (inseparable mixture of regioisomers, ratio 1:1).

\(^{1}\text{H NMR} \)\((\delta, \text{ ppm})\) (CDCl\(_3\), 400 MHz)

**5-exo product:** 5.84 (s, 1H, CH=), 5.50 (d, \( J = 7.5 \text{Hz} \), 1H, OCH-THF ring), 4.25-4.13 (m, 2H, OCH\(_2\)CH\(_3\)), 3.95-3.90 (m, 1H, OCH\(_2\)-THF ring), 3.75 (s, 3H, OCH\(_3\)), 3.70 (s, 3H, OCH\(_3\)), 3.70-3.66 (m, 1H, OCH\(_2\)-THF ring), 3.49 (q, \( J = 8.4 \text{Hz} \), 1H, CH-THF ring), 3.34 (d, \( J = 17.0 \text{Hz} \), 1H, CH\(_2\)), 2.72 (d, \( J = 17.0 \text{Hz} \), 1H, CH\(_2\)), 2.07-1.99 (m, 1H, CH\(_2\)-THF ring), 1.58-1.49 (m, 1H, CH\(_2\)-THF ring), 1.30-1.24 (m, 3H, OCH\(_2\)C\(_2\)H\(_5\)).

**6-endo products:** 6.85 (dd, \( J = 4.9 \text{Hz} \), \( J = 3.8 \text{Hz} \), 1H, CH=), 5.11 (d, \( J = 7.8 \text{Hz} \), 1H, OCH-THF ring), 4.25-4.13 (m, 2H, OCH\(_2\)CH\(_3\)), 3.83-3.77 (m, 2H, OCH\(_2\)-THF ring), 3.74 (s, 3H, OCH\(_3\)), 3.70 (s, 3H, OCH\(_3\)), 3.26 (q, \( J = 9.0 \text{Hz} \), 1H, CH-THF ring), 2.82-2.81 (m, 2H, CH\(_2\)), 1.90-1.83 (m, 1H, CH\(_2\)-THF ring), 1.74-1.64 (m, 1H, CH\(_2\)-THF ring), 1.30-1.24 (m, 3H, OCH\(_2\)C\(_2\)H\(_5\)).

\(^{13}\text{C NMR} \)\((\delta, \text{ ppm})\) (CDCl\(_3\), 100 MHz)

**5-exo and 6-endoproducts:** 171.3 (Cq, C=O), 170.6 (Cq, C=O), 170.2 (Cq, C=O), 169.6 (Cq, C=O), 166.8 (Cq, C=O), 164.9 (Cq, C=O), 157.4 (Cq, C=CH), 135.6 (C=CH), 131.5 (Cq, C=CH), 117.7 (C=CH), 80.7 (OCH-THF ring), 73.2 (OCH-THF ring), 69.0 (OCH\(_2\)-THF ring), 66.1 (OCH\(_2\)-THF ring), 60.7 (OCH\(_2\)CH\(_3\)), 60.6 (Cq), 60.2 (OCH\(_2\)CH\(_3\)), 56.2 (Cq), 53.0 (OCH\(_3\)), 52.9 (OCH\(_3\)), 52.8 (OCH\(_3\)), 52.7 (OCH\(_3\)), 48.1 (CH-THF ring), 40.8 (CH-THF ring), 39.4 (CH\(_3\)), 30.2 (CH\(_3\)), 27.4 (CH\(_3\)), 26.7 (CH\(_3\)), 14.2 (OCH\(_2\)C\(_2\)H\(_5\)), 14.1 (OCH\(_2\)CH\(_3\)).

IR \((\nu, \text{ cm}^{-1})\) (CCl\(_4\)) 2981 (s), 2954 (s), 2905 (m), 2858 (m), 1740 (s), 1676 (m), 1631 (m), 1477 (w), 1436 (s), 1372 (m), 1350 (w), 1254 (s), 1213 (s), 1182 (s), 1161 (s), 1132 (s), 1079 (s), 1065 (s), 1040 (s).

HRMS (El+, m/z): Calculated: 312.1209 Found: 312.1221.
2-Pentyl-3,3a,5,7a-tetrahydro-2H-benzofuran-4,4,7-tricarboxylic acid 7-ethyl ester 4,4-dimethyl ester and 6-[1-Ethoxycarbonyl-methylene]-2-pentyl-hexahydro-cyclopenta[b]furan-4,4-dicarboxylic acid dimethyl ester

\[
\text{MF: } C_{20}H_{30}O_7 \\
\text{MW = 382 g.mol}^{-1}
\]

Method: See general procedure 5.10.A using (1 equiv., 0.1 mmol, 38 mg) of 5-methoxycarbonyl-5-(5-pentyl-tetrahydro-furan-3-y1)-hex-2-yne dioic acid 1-ethyl ester 6-methyl ester

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt) \( R_f \) (8:2 PE:AcOEt): 0.36.

Product: Transparent oil.

Isolated yield: 90 % (inseparable mixture, ratio isomers major 6-endo: minor 6-endo: major 5-exo: minor 5-exo: 0.5: 0.3:1: 0.15).

\(^1\)H NMR (\( \delta, \text{ ppm} \))

(CDCl\(_3\), 400 MHz)

**Major 6-endo**: 6.75 (dd, \( J = 1.8Hz, J = 5.0Hz, 1H, C=CH \)), 5.05 (d, \( J = 8.0Hz, 1H, \text{OCH-THF ring} \)), 4.24-4.13 (m, 2H, OCH\(_2\)CH\(_3\)), 3.83-3.77 (m, 1H, OCH-THF ring), 3.75 (s, 3H, OCH\(_3\)), 3.70 (s, 3H, OCH\(_3\)), 3.36-3.27 (m, 1H, CH-THF ring), 2.81 (d, \( J = 1.8Hz, 2H, \text{CH}_{2} \)), 1.84 (ddd, \( J = 5.0Hz, J = 7.0Hz, J = 11.9Hz, 1H, \text{CH}_{2}-\text{THF ring} \)), 1.62-1.53 (m, 1H, CH\(_2\)-THF ring), 1.46-1.20 (m, 11H, CH\(_2\)-pentyl chain + OCH\(_2\)CH\(_3\)), 0.87-0.83 (m, 3H, CH\(_3\)-pentyl chain).

**Minor 6-endo**: 6.87 (dd, \( J = 3.7Hz, J = 4.8Hz, 1H, C=CH \)), 5.17 (d, \( J = 7.4Hz, 1H, \text{OCH-THF ring} \)), 4.24-4.13 (m, 2H, OCH\(_2\)CH\(_3\)), 3.83-3.77 (m, 1H, OCH-THF ring), 3.75 (s, 3H, OCH\(_3\)), 3.70 (s, 3H, OCH\(_3\)), 3.36-3.27 (m, 1H, CH-THF ring), 2.81 (d, \( J = 1.8Hz, 2H, \text{CH}_{2} \)), 1.76-1.53 (m, 2H, CH\(_2\)-THF), 1.46-1.20 (m, 11H, CH\(_2\)-pentyl chain + OCH\(_2\)CH\(_3\)), 0.87-0.83 (m, 3H, CH\(_3\)-pentyl chain).

**Major 5-exo**: 5.81 (s, 1H, C=CH), 5.44 (d, \( J = 7.8Hz, 1H, \text{OCH-THF ring} \)), 4.24-4.13 (m, 2H, OCH\(_2\)CH\(_3\)), 3.99-3.89 (m, 1H, OCH-THF ring), 3.75 (s, 3H, OCH\(_3\)), 3.70 (s, 3H, OCH\(_3\)), 3.50 (dd, \( J = 7.8Hz, J = 18.1Hz, 1H, \text{CH-THF ring} \)), 3.36 (d, \( J = 16.7Hz, 1H, \text{CH}_{2} \)), 2.67 (d, \( J = 16.7Hz, 1H, \text{CH}_{2} \)), 2.04 (ddd, \( J = 4.0Hz, J = 8.6Hz, J = 12.3Hz, 1H, \text{CH}_{2}-\text{THF ring} \)), 1.46-1.20 (m, 11H, CH\(_2\)-pentyl chain + OCH\(_2\)CH\(_3\)), 1.09 (dd, \( J = 10.8Hz, J = 18.1Hz, 1H, \text{CH}_{2}-\text{THF ring} \)), 0.87-0.83 (m, 3H, CH\(_3\)-pentyl chain).

**Minor 5-exo**: neglected, many pics are not distinguishable in the mixture.

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**13C NMR** (δ, ppm)  
Major/minor 6-endo and major 5-exo diasteroisomers: 171.3 (Cq, C=O), 170.6 (Cq, C=O), 170.5 (Cq, C=O), 170.3 (Cq, C=O), 170.2 (Cq, C=O), 169.7 (Cq, C=O), 166.0 (Cq, C=O), 164.9 (Cq, C=O), 157.9 (Cq), 157.1 (Cq, C=O), 136.2 (CH), 133.9 (CH), 132.7 (Cq), 131.7 (Cq), 117.7 (CH), 80.8 (OCH), 80.0 (OCH), 78.5 (OCH), 76.7 (OCH), 73.5 (OCH), 72.7 (OCH), 60.5 (OCH₂CH₃), 60.4 (OCH₂CH₃), 60.1 (OCH₂CH₃), 56.4 (Cq), 56.3 (Cq), 56.1 (Cq), 53.0 (OCH₂x2), 52.9 (OCH₂x2), 52.7 (OCH₂x2), 48.4 (CH), 41.4 (CH), 40.4 (CH), 40.2 (CH), 38.8 (CH₂), 36.0 (CH₂), 35.5 (CH₂), 34.7 (CH₂), 34.4 (CH₂), 33.6 (CH₂), 32.3 (CH₂), 31.9 (CH₂), 31.6 (CH₂), 31.7 (CH₂), 28.7 (CH₂), 26.4 (CH₂), 25.7 (CH₂), 25.6 (CH₂), 25.5 (CH₂), 22.6 (CH₂), 22.5 (CH₂), 14.1 (CH₂x2), 14.0 (CH₂x2), 13.9 (CH₂x2).

**IR** (ν, cm⁻¹) (CCl₄)  
2955 (s), 2932 (s), 2861 (s), 1736 (s), 1677 (m), 1661 (m), 1456 (s), 1436 (s), 1372 (s), 1355 (m), 1330 (m), 1254 (s), 1213 (s), 1160 (s), 1133 (s), 1079 (s), 1039 (s).

**HRMS** (EI+, m/z):  

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**6-Phenethyl-2-phenyl-5,6-dihydro-2H-pyran-3-carboxylic acid ethyl ester 5.44a**

![6-Phenethyl-2-phenyl-5,6-dihydro-2H-pyran-3-carboxylic acid ethyl ester 5.44a](image)

**MF:** C₂₂H₂₄O₃  
**MW:** 336 g.mol⁻¹

**Method:** See general procedure 5.10.A using (1 equiv., 0.1 mmol, 34 mg) of 5-benzyloxy-7-phenyl-hept-2-ynoic acid ethyl ester.

**Purification:**  
Flash column chromatography (silica gel, 95:5 PE:AcOEt)/ Rf (9:1 PE:AcOEt): 0.39.

**Product:**  
Transparent oil.

**Isolated yield:** 65% (ratio of diasteroisomers 16:1 in favor of the drawn molecule).

**1H NMR** (δ, ppm)  
(CDCl₃, 400 MHz)  
7.37-7.18 (m, 10H, CH-Ph), 7.12 (d, J = 6.1Hz, 1H, CH=), 5.39 (s, 1H, OCH), 4.05-3.96 (m, 1H, OCH₂CH₃), 3.95-3.87 (m, 1H, OCH₂CH₃), 3.60-3.54 (m, 1H, OCH), 2.83-2.69 (m, 2H, CH₂), 2.36-2.20 (m, 2H, CH₂), 2.02-1.93 (m, 1H, CH₂), 1.86-1.78 (m, 1H, CH₂), 1.00 (t, J = 7.1Hz, 3H, OCH₂CH₃).

**13C NMR** (δ, ppm)  
(CDCl₃, 100 MHz)  
165.6 (Cq, C=O), 141.7 (Cq), 140.6 (Cq), 137.2 (CH=), 133.8 (Cq), 128.4 (CH-Ph), 128.3 (CH-Ph), 128.2 (CH-Ph), 127.9 (CH-Ph), 125.7 (CH-Ph), 78.0 (OCH), 72.3 (OCH), 60.0 (OCH₂CH₃), 36.8 (CH₂), 31.5 (CH₂), 31.4 (CH₂), 13.8 (OCH₂CH₃).
(CDCl$_3$, 100 MHz)

**IR (ν, cm$^{-1}$) (CCl$_4$)**

3088 (m), 3066 (m), 3031 (s), 2982 (s), 2930 (s), 2859 (s), 1721 (s), 1653 (m), 1617 (m), 1604 (m), 1496 (m), 1477 (w), 1455 (s), 1432 (w), 1420 (w), 1370 (s), 1353 (s), 1337 (s), 1322 (m), 1290 (s), 1254 (s), 1172 (m), 1161 (m), 1137 (m), 1103 (s), 1066 (s), 1045 (s).

**HRMS (EI+, m/z)**: Calculated: 336.1726 Found: 336.1734.

6-Isopropyl-2-phenyl-5,6-dihydro-2$\text{H}$-pyran-3-carboxylic acid ethyl ester

5.44d

![Structural formula](image)

**MF**: C$_{17}$H$_{22}$O$_3$

**MW**: 274 g.mol$^{-1}$

**Method**: See general procedure 5.10.A using (1 equiv., 0.1 mmol, 27 mg) of 5-benzyloxy-6-methyl-hept-2-ynoic acid ethyl ester.

**Purification**: Flash column chromatography (silica gel, 9:1 PE:AcOEt)/$R_f$ (8:2 PE:AcOEt): 0.54.

**Product**: Transparent oil.

**Isolated yield**: 74 % (ratio diasteroisomers > 25:1 in favor of the drawn structure)

**$^1$H NMR (δ, ppm)**

(CDCl$_3$, 400 MHz)

7.39-7.30 (m, 5H, CH-Ph), 7.19-7.17 (m, 1H, C=CH), 5.44 (dd, $J = 3.7$Hz, $J = 4.7$Hz, 1H, OCH), 4.05 (qd, $J = 7.1$Hz, $J = 10.8$Hz, 1H, OCH$_2$CH$_3$), 3.96 (qd, $J = 7.1$Hz, $J = 10.8$Hz, 1H, OCH$_2$CH$_3$), 3.36 (ddd, $J = 4.0$Hz, $J = 6.9$Hz, $J = 9.3$Hz, 1H, OCH), 2.40-2.26 (m, 2H, CH$_2$), 1.89-1.77 (m, 1H, CH$_3$-Pr), 1.06 (t, $J = 7.1$Hz, 3H, OCH$_2$CH$_3$), 1.02 (d, $J = 6.7$Hz, 3H, CH$_3$-Pr), 0.97 (d, $J = 6.8$Hz, 3H, CH$_3$-Pr).

**$^{13}$C NMR (δ, ppm)**

(CDCl$_3$, 100 MHz)

165.7 (Cq, C=O), 140.9 (Cq), 137.5 (C=CH), 133.9 (Cq), 128.2 (CH-Ph), 128.1 (CH-Ph), 127.8 (CH-Ph), 78.4 (OCH), 78.2 (OCH), 60.1 (OCH$_2$CH$_3$), 32.5 (CH$_3$-Pr), 28.4 (CH$_2$), 18.8 (CH$_3$-Pr), 17.9 (CH$_3$-Pr), 13.8(OCH$_2$CH$_3$).

**IR (ν, cm$^{-1}$) (CCl$_4$)**

3090 (w), 3067 (w), 3035 (w), 2962 (s), 2932 (s), 2907 (s), 2874 (s), 2847 (s), 1746 (s), 1721 (s), 1655 (s), 1616 (w), 1496 (m), 1470 (s), 1455 (s), 1421 (m), 1388 (s), 1369 (s), 1353 (s), 1322 (s), 1292 (s), 1246 (s), 1206 (m), 1173 (m), 1145 (m), 1129 (s), 1101 (s), 1055 (s), 1030 (s).

**HRMS (EI+, m/z)**: Calculated: 274.1569 Found: 274.1573.
B.3.5 Chapter 6: Formation of cinnoline derivatives by a gold(I) mediated synthesis of cinnoline derivatives

All molecules synthesized for in this work are described below. This project was developed individually.

B.3.5.1 Synthesis of hydrazine derivatives

Hydrazines derivatives were synthesized as it follows:

Procedure 6.1A

\[
\text{CICO}_2\text{Me} \rightarrow \text{Y} \quad \text{DCM, 0 °C to rt}
\]

Procedure 6.2

\[
\text{Y} \quad \text{MnO}_2 \rightarrow \text{DCM, rt}
\]

Procedure 6.1B

\[
\text{NaOH, H}_2\text{O} \rightarrow \text{then pyr, CICO}_2\text{Me} \rightarrow \text{Y} \quad \text{DCM, 0 °C to rt}
\]

Procedure 6.3

\[
\text{R}^1\text{MgX} \rightarrow \text{THF -78 °C}
\]

Procedure 6.4

\[
\text{NaH, } \equiv \equiv \text{Br} \rightarrow \text{Y} \quad \text{TBAI, DMF, rt}
\]

Procedure 6.5

\[
\text{Pd(PPh}_3)_4 \rightarrow \text{(cat), R}_2\text{X \rightarrow Y} \quad \text{DCM, r.t.}
\]

6.11q, \( R^1 = \text{Me}, R^2 = \text{Ph}, Y = \text{H} \)

6.11r, \( R^1 = \text{Me}, R^2 = \equiv \equiv, Y = \text{H} \)

6.11a, \( R^1 = \text{Ph}, Y = \text{H} \)

6.11b, \( R^1 = n\text{-C}_3\text{H}_7, Y = \text{H} \)

6.11c, \( R^1 = \text{Pr}, Y = \text{H} \)

6.11d, \( R^1 = \text{CH}_2\text{Ph}, Y = \text{H} \)

6.11e, \( R^1 = \text{Bu}, Y = \text{H} \)

6.11f, \( R^1 = \text{Me}, Y = \text{p-OMe} \)

6.11g, \( R^1 = \text{Me}, Y = \text{p-Cl} \)

6.11h, \( R^1 = \text{Me}, Y = \text{p-F} \)

6.11i, \( R^1 = \text{Me}, Y = \text{p-CO}_2\text{Et} \)

6.11j, \( R^1 = \text{Me}, Y = \text{p-CN} \)

6.11k, \( R^1 = \text{Me}, Y = \text{p-CF}_3 \)

6.11l, \( R^1 = \text{Me}, Y = \text{p-NO}_2 \)

6.11m, \( R^1 = \text{Me}, Y = m\text{-Me} \)

6.11n, \( R^1 = \text{Me}, Y = m\text{-Cl} \)

6.11o, \( R^1 = \text{Me}, Y = o\text{-Me} \)

6.11p, \( R^1 = \text{Me}, Y = o\text{-Cl} \)

6.11q, \( R^1 = \text{Ph}, Y = \text{p-OME} \)

6.11r, \( R^1 = \text{Ph}, Y = \text{p-F} \)

6.11s, \( R^1 = \text{Ph}, Y = \text{p-CO}_2\text{Et} \)

Other structures synthesized in this project:

Scheme B.3.5.1: Synthesis of hydrazine derivatives 6.11a-u.
General Procedure 6.1.A\textsuperscript{310}, Protection of hydrazines in free form: To a solution of free hydrazine (1 equiv.) and pyridine (1 equiv.) in DCM (0.6 M) in a round bottom flask at 0 °C was slowly added methyl chloroformate (1 equiv.). The reaction was allowed to warm up and to stir at room temperature until complete consumption of the starting hydrazine (TLC, reaction usually takes 30 min.). The solution was then concentrated under reduced pressure (usually left long time to evaporate pyridine) and used in the next step without purification.

General Procedure 6.1.B\textsuperscript{311}, Protection of hydrazine.HCl: To a solution of the salt hydrazine.HCl (1 equiv.) in H\textsubscript{2}O (0.6 M) in a round bottom flask at room temperature was added NaOH (1 equiv.). To the resulting mixture was added pyridine (4 equiv.) and the temperature was cooled to 0 °C, followed by the slowly addition of methyl chloroformate (1 equiv.). The reaction was allowed to warm up and to stir at room temperature until the complete consumption of the starting hydrazine (TLC, reaction usually takes 30 min.). The reaction was extracted with AcOEt (3x), dried (MgSO\textsubscript{4}) and concentrated under reduced pressure (usually left long time to evaporate pyridine). The protected hydrazine was used in the next step without purification.

General Procedure 6.2\textsuperscript{312}, Oxidation of hydrazides: To a solution of the hydrazide previously prepared (1 equiv.) in DCM (0.85 M) in a round bottom flask at room temperature was added MnO\textsubscript{2} (5 equiv.). The solution was allowed to stir at room temperature until the complete consumption of the starting hydrazide (TLC, reaction usually takes 30 min.). The solution was then filtered through a short pad of celite and concentrated under reduced pressure. The oxidation products were used in the next step without purification.

**General Procedure 6.3**, **Addition of Grignard reagents**: To a solution of the oxidized hydrazide previously prepared (1 equiv.) in THF (0.25 M) in a round bottom flask under an argon atmosphere at -78 °C was added the Grignard reagent (1.5 equiv.) and the reaction was allowed to stir at -78 °C until the complete consumption of the starting hydrazide (TLC, reaction usually takes 30 min.). The reaction was then quenched with H₂O at -78 °C and allowed to warm up to room temperature. The solution was extracted with AcOEt (3x), dried (MgSO₄) and concentrated under reduced pressure. The alkylated hydrazine was used in the next step without purification.

**General Procedure 6.4, N-Alkylation with propargyl bromide.** To a solution of a an alkylated hydrazine (1 equiv.) in DMF (0.125 M) in a round bottom flask at room temperature was added NaH (1.5 equiv.) followed by the addition of propargyl bromide (1.5 equiv., 80 % w/w solution in toluene) and tetrabutil ammonium iodide (0.1 equiv.). The reaction was allowed to stir at room temperature until the complete consumption of the starting hydrazine (TLC, reaction usually left overnight). The reaction was then quenched with a saturated solution of NH₄Cl, extracted with AcOEt (3x), washed with water (5x), dried (MgSO₄) and concentrated under reduced pressure. Purification of the crude mixture by flash column chromatography furnished the fully substituted hydrazine in the stated yields.

**General Procedure 6.5, Sonogashira coupling reaction of propargylated hydrazines**: To a solution of propargylated hydrazine (1 equiv.) and halide (1.5 equiv.) in DMF (0.10 M) in a round bottom flask under an argon atmosphere at room temperature was added piperidine (10 equiv.), CuI (0.1 equiv.) and Pd(PPh₃)₄ (0.05 equiv.). The reaction was allowed to stir at room temperature overnight. Upon reaction completion (TLC), the reaction was quenched with a saturated solution of NH₄Cl, extracted with AcOEt (3x), washed with H₂O (5x), dried (MgSO₄) and concentrated under reduced pressure. Purification by flash column chromatography afforded the desired products in the stated yields.

---

Method: See general procedure 6.3 using (1 equiv., 3 mmol, 470 µL) of diethyl azodicarboxylate and (1 equiv., 3 mmol, 1.0 M solution THF, 3.0 mL) of PhMgBr followed by general procedure 6.4 (1 equiv., 3 mmol, 750 mg) of the previously prepared compound.

Purification: Flash column chromatography (silica gel, 9:1 PE:AcOEt).

Product: Pale yellow oil.

Isolated yield: 80% for two steps.

\(^1^H\text{ NMR}\) (δ, ppm): 7.50 (d, J = 7.9Hz, 1H, CH-Ph), 7.45 (d, J = 7.9Hz, 1H, CH-Ph), 7.33 (t, J = 7.9Hz, 2H, CH-Ph), 7.19 (t, J = 7.9Hz, 1H, CH-Ph), 4.46-4.16 (m, 6H, CH\(_2\)-propargyl chain + OCH\(_2\)CH\(_3\)), 2.22 (t, J = 2.4Hz, 0.3H, CH\(_\text{rotamere}\)), 2.19 (t, J = 2.4Hz, 0.7H, CH\(_\text{rotamere}\)), 1.34-1.22 (m, 6H, OCH\(_2\)C\(_\text{H}_3\)).

\(^{13}C\text{ NMR}\) (δ, ppm): 155.6 (Cq, C=O), 155.2 (Cq, C=O\(_\text{rotamere}\)), 154.3 (Cq, C=O\(_\text{rotamere}\)), 153.9 (Cq, C=O), 140.8 (Cq, Ph\(_\text{rotamere}\)), 140.5 (Cq, Ph), 128.5 (CH-Ph), 126.4 (CH-Ph), 126.0 (CH-Ph\(_\text{rotamere}\)), 123.4 (CH-Ph), 73.5 (C\(_\text{rotamere}\)), 73.3 (Cq, C\(_\text{rotamere}\)), 63.1 (OCH\(_2\)CH\(_3\)\(_\text{rotamere}\)), 62.9 (OCH\(_2\)CH\(_3\)), 62.8 (OCH\(_2\)CH\(_3\)), 40.2 (CH\(_2\)-propargyl chain\(_\text{rotamere}\)), 39.3 (CH\(_2\)-propargyl chain), 14.4 (OCH\(_2\)CH\(_3\)), 14.3 (OCH\(_2\)CH\(_3\)).

IR (ν, cm\(^{-1}\)) (CCl\(_4\)): 3314 (m), 3068 (w), 3046 (w), 2983 (m), 2935 (w), 2913 (w), 2872 (w), 1725 (s), 1598 (m), 1492 (m), 1482 (m), 1465 (m), 1444 (m), 1403 (s), 1373 (s), 1305 (s), 1275 (s), 1238 (s), 1200 (s), 1174 (s), 1133 (s), 1096 (s), 1061 (s), 1030 (s).

HRMS (EI+, m/z): Calculated: 290.1267 Found 290.1270.
$N,N'$-Diphenyl-$N$-prop-2-ynyl-hydrazinecarboxylic acid methyl ester

\[
\text{MF: } C_{17}H_{18}N_2O_2
\]

\[
\text{MW} = 280 \text{ g.mol}^{-1}
\]

**Method:** See general procedure 6.2 using (1 equiv., 3 mmol, 500 mg) of $N$-phenyl-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 3 mmol, 492 mg) of the previously prepared compound and (1.5 equiv., 4.5 mmol, 1M solution in THF, 4.5 mL) of PhMgBr followed by general procedure 6.4 using (1 equiv., 3 mmol, 721 mg) of $N,N'$-diphenyl-hydrazinecarboxylic acid methyl ester.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:** Yellow oil.

**Isolated yield:** 67% for three steps.

**$^1H$ NMR (δ, ppm)** (CDCl$_3$, 400 MHz)

- 7.31-7.27 (m, 4H, CH$_{-}$Ph), 7.16-7.14 (m, 4H, CH$_{-}$Ph), 7.03 (t, $J = 7.3$Hz, 2H, CH$_{-}$Ph), 4.40 (br s, 2H, CH$_2$-propargyl chain), 3.82 (br s, 0.8H, OCH$_3$ rotamere), 3.69 (br s, 2.2H, OCH$_3$), 2.26 (br s, 1H, $\equiv$CH).

**$^{13}$C NMR (δ, ppm)** (CDCl$_3$, 100 MHz)

- 156.9 (Cq, C=O), 144.2 (Cq, Ph), 129.1 (CH$_{-}$Ph), 122.9 (CH$_{-}$Ph), 119.8 (CH$_{-}$Ph rotamere), 119.4 (CH$_{-}$Ph), 78.2 (Cq, $\equiv$CH), 73.3 (C$\equiv$CH), 53.6 (OCH$_3$), 38.9 (CH$_2$).

**IR (ν, cm$^{-1}$) (CCl$_4$)**

- 3313 (m), 3070 (w), 3042 (w), 2956 (w), 1718 (s), 1592 (s), 1497 (s), 1446 (s), 1377 (m), 1330 (m), 1314 (m), 1275 (m), 1239 (m), 1196 (w), 1178 (w), 1127 (w), 1079 (w), 1031 (w).

**HRMS (EI+, m/z)**

- Calculated: 280.1212  Found: 280.1225.
**Method:**

See **general procedure 6.2** using (1 equiv., 3.4 mmol, 570 mg) of \(N\)-phenyl-hydrazinecarboxylic acid methyl ester followed by **general procedure 6.3** using (1 equiv., 1.7 mmol, 280 mg) of the previously prepared compound and (1.5 equiv., 4.5 mmol, 1M solution in THF, 2.6 mL) of \(n\)-C\(_5\)H\(_{11}\)MgBr followed by **general procedure 6.4** using (1 equiv., 1.31 mmol, 310 mg) of \(N\)-pentyl-\(N\)-phenyl-hydrazinecarboxylic acid methyl ester.

**Purification:**
Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:**
Pale yellow oil.

**Isolated yield:**
64% for three steps.

\(^1\)H NMR (\(\delta\), ppm)

<table>
<thead>
<tr>
<th>Chemical Shift (ppm)</th>
<th>Multiplicity</th>
<th>Assignment</th>
</tr>
</thead>
<tbody>
<tr>
<td>7.27-7.23 (m, 2H, CH-Ph)</td>
<td>6.84 (t, J = 7.3Hz, 1H, CH-Ph)</td>
<td>6.74-6.68 (m, 2H, CH-Ph), 4.55 (br d, J = 17.5Hz, 0.7H, NCH(_2)-propargyl chain), 4.45 (br d, J = 17.5Hz, 0.3H, NCH(_2)-propargyl chain rotamere), 4.01 (br d, J = 17.5Hz, 1H, NCH(_2)-propargyl chain), 3.81 (br s, 0.8H, OCH(_3) rotamere), 3.70 (br s, 2.2H, OCH(_3)), 3.60-3.53 (m, 1H, NCH(_2)-pentyl chain), 3.44-3.37 (m, 1H, NCH(_2)-pentyl chain), 2.29 (br s, 1H, CH(_3)-pentyl chain), 1.77 (br s, 2H, CH(_2)-pentyl chain), 1.39-1.36 (m, 4H, CH(_2)-pentyl chain), 0.93 (t, J = 6.7Hz, 3H, CH(_3)-pentyl chain).</td>
</tr>
</tbody>
</table>

\(^{13}\)C NMR (\(\delta\), ppm)

<table>
<thead>
<tr>
<th>Chemical Shift (ppm)</th>
<th>Multiplicity</th>
<th>Assignment</th>
</tr>
</thead>
<tbody>
<tr>
<td>157.1 (Cq, C=O), 147.2 (Cq, Ph), 129.2 (CH-Ph), 119.2 (CH-Ph), 112.4 (CH-Ph), 78.9 (C(\equiv)CH), 72.5 (C(\equiv)CH), 53.4 (OCH(_3)), 52.8 (NCH(_2)), 40.4 (NCH(_2) rotamere), 39.0 (NCH(_2)), 28.3 (CH(_2)-pentyl chain), 27.2 (CH(_2)-pentyl chain), 22.5 (CH(_2)-pentyl chain), 14.1 (CH(_3)-pentyl chain)</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

IR (\(\text{v, cm}^{-1}\)) (CCl\(_4\))

<table>
<thead>
<tr>
<th>Frequency (cm(^{-1}))</th>
<th>Assignment</th>
</tr>
</thead>
<tbody>
<tr>
<td>3313 (m), 3095 (w), 3068 (w), 3030 (w), 2957 (m), 2932 (m), 2873 (w), 1716 (s), 1599 (s), 1499 (s), 1446 (s), 1415 (w), 1378 (s), 1337 (m), 1273 (s), 1238 (s), 1194 (m), 1130 (m), 1091 (w), 1070 (w), 1034 (w)</td>
<td></td>
</tr>
</tbody>
</table>

HRMS (El+, m/z):
Calculated: 274.1681  Found: 274.1694.
**N'-Isopropyl-N'-phenyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester 6.11c**

\[
\text{MF: } C_{14}H_{18}O_2N_2
\]

\[
\text{MW = 246 g.mol}^{-1}
\]

**Method:**
See general procedure 6.2 using (1 equiv., 2 mmol, 332 mg) of N-phenyl-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 2 mmol, 328 mg) of the previously prepared compound and (1.5 equiv., 3 mmol, solution 2M in THF, 1.5 mL) of iPrCl followed by general procedure 6.3 using (1 equiv., 2 mmol, 450 mg) of N'-isopropyl-N'-phenyl-hydrazinecarboxylic acid methyl ester.

**Purification:**
Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:**
Yellow oil.

**Isolated yield:**
47% for three steps.

**\(^1\)H NMR (δ, ppm)** (CDCl\(_3\), 400 MHz)
- 7.26 (dd, \(J = 7.3\) Hz, 2H, CH-Ph), 6.82 (t, \(J = 7.3\) Hz, 1H, CH-Ph), 6.70 (d, \(J = 8.5\) Hz, 0.5H, CH-Ph\(_{\text{rotamere}}\)), 6.65 (d, \(J = 8.5\) Hz, 1.5H, CH-Ph), 4.61 (dd, \(J = 2.4\) Hz, \(J = 17.4\) Hz, 0.7H, NCH\(_2\)), 4.50 (br d, \(J = 17.4\) Hz, 0.3H, NCH\(_2\) \(_{\text{rotamere}}\)), 3.90 (br d, \(J = 17.4\) Hz, 0.3H, NCH\(_2\) \(_{\text{rotamere}}\)), 3.83 (dd, \(J = 2.4\) Hz, \(J = 17.4\) Hz, 0.7H, NCH\(_2\) \(_{\text{rotamere}}\)), 3.84 (s, 0.7H, OCH\(_3\) \(_{\text{rotamere}}\)), 3.71 (s, 2.3H, OCH\(_3\)), 2.38 (t, \(J = 2.4\) Hz, 1H, CH\(_3\)-iPr), 1.36 (d, \(J = 6.6\) Hz, 3H, CH\(_3\)-iPr), 1.34 (d, \(J = 6.6\) Hz, 3H, CH\(_3\)-iPr).

**\(^{13}\)C NMR (δ, ppm)** (CDCl\(_3\), 100 MHz)
- 157.8 (Cq, C=O), 156.5 (Cq, C=O \(_{\text{rotamere}}\)), 145.5 (Cq, Ph), 129.3 (CH-Ph), 129.2 (CH-Ph \(_{\text{rotamere}}\)), 118.7 (CH-Ph \(_{\text{rotamere}}\)), 118.6 (CH-Ph), 112.7 (CH-Ph \(_{\text{rotamere}}\)), 112.2 (CH-Ph), 79.0 (Cq, C=CH \(_{\text{rotamere}}\)), 78.7 (Cq, C=CH), 73.2 (C=CH), 73.1 (C=CH \(_{\text{rotamere}}\)), 53.4 (OCH\(_3\) \(_{\text{rotamere}}\)), 53.3 (OCH\(_3\)), 50.4 (CH\(_3\)-iPr \(_{\text{rotamere}}\)), 50.1 (CH\(_3\)-iPr), 42.1 (NCH\(_2\) \(_{\text{rotamere}}\)), 41.2 (NCH\(_2\)), 20.2 (CH\(_3\)-iPr \(_{\text{rotamere}}\)), 19.8 (CH\(_3\)-iPr), 19.0 (CH\(_3\)-iPr \(_{\text{rotamere}}\)), 18.8 (CH\(_3\)-iPr).

**IR (ν, cm\(^{-1}\))** (CCl\(_4\))
- 3313 (m), 3096 (w), 3068 (w), 3028 (w), 2978 (w), 2955 (w), 2876 (w), 1737 (s), 1716 (s), 1596 (s), 1498 (s), 1447 (s), 1413 (w), 1381 (s), 1369 (s), 1333 (m), 1295 (s), 1274 (s), 1238 (s), 1194 (m), 1171 (w), 1141 (m), 1116 (s), 1039 (w).

**HRMS (El+, m/z):**
- Calculated: 246.1368  Found: 246.1359.
**N-Benzyl-N-phenyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester**  

\[ \text{MF: } C_{18}H_{18}O_2N_2 \]

\[ \text{MW} = 294 \text{ g.mol}^{-1} \]

**Method:**

To a round bottom flask under an argon atmosphere at -78 °C charged with N-phenyl-hydrazinecarboxylic acid methyl ester (1 equiv., 3 mmol, 498 mg) and THF (0.2 M, 15 mL) was added \( n \)-BuLi (2.5 equiv., 7.5 mmol, 2.5 M in hexanes, 3 mL) and the reaction was allowed to stir at -78 °C for 20 min. Then benzyl bromide (1 equiv., 3 mmol, 360 µL) was added. The temperature was allowed to warm up and to stir at room temperature for 2h. Then propargyl bromide (1.1 equiv., 3.3 mmol, solution 80% w/w in toluene, 360 µL) was added. The reaction was allowed to stir at room temperature overnight. Upon completion (TLC), the reaction was quenched with \( \text{H}_2\text{O} \), extracted with AcOEt (3x), dried (MgSO\(_4\)) and concentrated under reduced pressure. Purification by flash column chromatography afforded the pure product.

**Reference:**


**Purification:**

Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:**

Orange solid.

**Isolated yield:**

20%.

**\(^1\)H NMR** (δ, ppm)  
(CDCl\(_3\), 400 MHz)  
7.43 (br d, \( J = 5.9 \text{Hz} \), 2H, CH-Ph), 7.34 (t, \( J = 7.4 \text{Hz} \), 2H, CH-Ph), 7.27 (t, \( J = 7.4 \text{Hz} \), 2H, CH-Ph), 7.21 (dd, \( J = 7.4 \text{Hz}, J = 8.8 \text{Hz} \), 2H, CH-Ph), 6.85 (t, \( J = 7.3 \text{Hz}, 1 \text{H}, \text{CH-Ph} \), 6.80-6.73 (m, 1H, CH-Ph), 4.86 (br d, \( J = 15.5 \text{Hz}, 1 \text{H}, \text{NCH}_2 \)), 4.64 (br d, \( J = 15.5 \text{Hz}, 1 \text{H}, \text{NCH}_2 \)), 4.37 (br d, \( J = 16.7 \text{Hz}, 1 \text{H}, \text{NCH}_2 \)), 4.19 (d, \( J = 16.7 \text{Hz}, 1 \text{H}, \text{NCH}_2 \)), 3.81 (br s, 0.8H, OCH\(_3\) rotamere), 3.70 (br s, 2.2H, OCH\(_3\)), 2.20 (br s, 1H, \( \equiv \text{CH} \)).

**\(^{13}\)C NMR** (δ, ppm)  
(CDCl\(_3\), 100 MHz)  
156.7 (Cq, C=O), 155.2 (Cq, C=O rotamere), 147.6 (Cq, Ph), 137.5 (Cq, Ph), 129.0 (CH-Ph), 128.5 (CH-Ph), 127.4 (CH-Ph), 127.2 (CH-Ph), 119.7 (CH-Ph), 112.9 (CH-Ph), 78.6 (Cq, \( \equiv \text{C}=\text{CH} \)), 72.8 (C \( \equiv \equiv \text{CH} \)), 58.1 (NCH\(_2\)), 53.4 (OCH\(_3\)), 40.9 (NCH\(_2\) rotamere), 39.4 (NCH\(_2\)).

**IR** (ν, cm\(^{-1}\)) (CCl\(_4\))  
3313 (m), 3091 (w), 3067 (w), 3031 (w), 2956 (w), 2855 (w), 1718 (s), 1600 (s), 1499 (s), 1446 (s), 1376 (s), 1335 (m), 1268 (m), 1237 (s), 1194 (m), 1137 (m), 1110 (w), 1089 (w), 1062 (w), 1030 (w).

**HRMS** (EI+, m/z):  
Calculated: 294.1368  
Found: 294.1372.
**N'-phenyl-N'-tert-butyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester**  

\[
\text{MF: } C_{15}H_{20}O_2N_2
\]

\[
\text{MW} = 283 \text{ g.mol}^{-1}
\]

**Method:**

See **general procedure 6.2** using (1 equiv., 1.71 mmol, 250 mg) of \(N\)-tert-butyl-hydrazinecarboxylic acid methyl ester followed by **general procedure 6.3** using (1 equiv., 1.7 mmol, 246 mg) of the previously prepared compound and (1.5 equiv., 2.56 mmol, solution 1.0 M in THF, 2.6 mL) of PhMgBr followed by **general procedure 6.4** using (1 equiv., 1.1 mmol, 244 mg) of \(N\)-phenyl-\(N\)-tert-butyl-hydrazinecarboxylic acid methyl ester.

**Purification:**

Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:**

Yellow pale oil.

**Isolated yield:**

66% for three steps.

\(\text{^1H NMR (} \delta \text{, ppm)}\)

(CDC\(_3\), 400 MHz) 7.24-7.20 (m, 2H, CH-Ar), 6.90-6.84 (br s, 3H, CH-Ar), 4.52 (br d, \(J = 17.4\text{Hz}, 1\text{H}, \text{NCH}_2\)), 3.87 (br d, \(J = 17.4\text{Hz}, 1\text{H}, \text{NCH}_2\)), 3.75 (br s, 3H, OCH\(_3\)), 2.33 (br s, 1H, \(==\text{CH}\)), 1.53 (s, 9H, CH\(_3\)-tBu).

**N'-(4-Methoxy-phenyl)-N'-methyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester**  

\[
\text{MF: } C_{13}H_{16}O_2N_2
\]

\[
\text{MW} = 248 \text{ g.mol}^{-1}
\]

**Method:**

See **general procedure 6.2** using (1 equiv., 1.5 mmol, 300 mg) of \(N\)-(4-methoxy-phenyl)-hydrazinecarboxylic acid methyl ester followed by **general procedure 6.3** using (1 equiv., 1.5 mmol, 297 mg) of the previously prepared compound and (1.5 equiv., 2.25 mmol, solution 1.4 M in 3:1 toluene:THF, 1.6 mL) of MeMgBr followed by **general procedure 6.4** using (1 equiv., 1.36 mmol, 285 mg) of \(N\)-(4-methoxy-phenyl)-\(N\)-methyl-hydrazinecarboxylic acid methyl ester.
ester.

**Purification:** Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product:** Orange oil.

**Isolated yield:** 60% for three steps.

**$^1$H NMR ($\delta$, ppm)**

(CDCl$_3$, 400 MHz)

- 6.83 (br d, $J = 9.1$Hz, 2H, CH-Ar)
- 6.68 (br s, 2H, CH-Ar)
- 4.54 (br d, $J = 17.0$Hz, 1H, NCH$_2$)
- 4.03 (br d, $J = 17.0$Hz, 1H, NCH$_2$)
- 3.76 (br s, 3H, OCH$_3$)
- 3.72 (br s, 3H, OCH$_3$)
- 3.21 (br s, 3H, NCH$_3$)
- 2.28 (br s, 1H, CH$_3$)

**$^{13}$C NMR ($\delta$, ppm)**

(CDCl$_3$, 100 MHz)

- 157.0 (Cq, C=O)
- 153.4 (Cq, Ar)
- 142.1 (Cq, Ar)
- 114.7 (CH-Ar)
- 113.5 (CH-Ar)
- 79.2 (Cq, C=CH)
- 72.3 (C=CH)
- 55.6 (OCH$_3$)
- 53.4 (OCH$_3$)
- 39.6 (NCH$_3$)
- 37.9 (NCH$_3$)

**IR ($\nu$, cm$^{-1}$) (CCl$_4$)**

- 3313 (m)
- 2999 (w)
- 2955 (m)
- 2906 (w)
- 2833 (w)
- 1716 (s)
- 1510 (m)
- 1464 (m)
- 1446 (s)
- 1374 (s)
- 1273 (m)
- 1246 (s)
- 1194 (m)
- 1181 (m)
- 1156 (m)
- 1119 (m)
- 1071 (m)
- 1042 (m)

**HRMS (EI+, m/z):**

Calculated: 248.1161  Found: 248.1157.

---

$N$-(4-Chloro-phenyl)-$N$-methyl-$N$-prop-2-ynyl-hydrazinecarboxylic acid methyl ester

**MF:** C$_{12}$H$_{13}$O$_2$N$_2$Cl

**MW:** 252.5 g.mol$^{-1}$

**Method:** See general procedure 6.2 using (1 equiv., 1.5 mmol, 300 mg) of $N$-(4-chloro-phenyl)-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 1.5 mmol, 298 mg) of the previously prepared compound and (1.5 equiv., 2.25 mmol, solution 1.4 M in 3:1 toluene:THF, 1.6 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 1.5 mmol, 330 mg) of $N$-(4-chloro-phenyl)-$N$-methyl-hydrazinecarboxylic acid methyl ester.

**Purification:** Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product:** Orange oil.

**Isolated yield:** 71% for three steps.
$^{1}H$ NMR (δ, ppm)  
(CDCl$_3$, 400 MHz)  
7.19 (br d, $J = 8.7$Hz, 2H, CH-Ar), 6.64 (br s, 2H, CH-Ar), 4.54 (br d, $J = 15.6$Hz, 1H, NCH$_2$), 4.09 (br d, $J = 15.6$Hz, 1H, NCH$_2$), 3.70 (br s, 3H, OCH$_3$), 3.22 (br s, 3H, NCH$_3$), 2.29 (br s, 1H, CH).

$^{13}C$ NMR (δ, ppm)  
(CDCl$_3$, 100 MHz)  
156.4 (Cq, C=O), 146.8 (Cq, Ar), 129.1 (CH Ar), 124.2 (Cq, Ar), 113.2 (CH Ar), 78.6 (Cq, C CH), 72.8 (C CH), 53.5 (OCH$_3$), 39.3 (NCH$_3$), 38.2 (NCH$_2$).

IR ( ν, cm$^{-1}$) (CCl$_4$)  
3312 (m), 3095 (w), 3003 (w), 2957 (m), 2903 (w), 2818 (w), 1721 (s), 1599 (s), 1495 (s), 1446 (s), 1419 (m), 1375 (s), 1347 (m), 1324 (s), 1270 (s), 1237 (s), 1194 (s), 1184 (s), 1156 (s), 1119 (s), 1099 (m), 1069 (m).

HRMS (EI+, m/z)  
Calculated: 252.0666  
Found: 252.0657.

$N'$-(4-Fluoro-phenyl)-$N'$-methyl-$N$-prop-2-ynyl-hydrazinecarboxylic acid methyl ester  

![Chemical Structure](image)

MF: C$_{12}$H$_{13}$O$_{2}$N$_{2}$F  
MW = 236 g.mol$^{-1}$

**Method:**  
See general procedure 6.2 using (1 equiv., 2 mmol, 368 mg) of $N'$-(4-fluoro-phenyl)-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 2 mmol, 364 mg) of the previously prepared compound and (1.5 equiv.; 3 mmol, solution 1.4 M in 3:1 toluene:THF, 2.14 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 2 mmol, 396 mg) of $N'$-(4-fluoro-phenyl)-$N'$-methyl-hydrazinecarboxylic acid methyl ester

**Purification:**  
Flash column chromatography (silica gel, 8:2 PE: AcOEt).

**Product:**  
Orange oil.

**Isolated yield:**  
99% for three steps.

$^{1}H$ NMR (δ, ppm)  
(CDCl$_3$, 400 MHz)  
6.98 (t, $J = 8.7$Hz, 2H, CH-Ar), 6.70 (br s, 2H, CH-Ar), 4.57 (br d, $J = 16.8$Hz, 1H, NCH$_2$), 4.12 (br d, $J = 16.8$Hz, 1H, NCH$_2$), 3.73 (br s, 3H, OCH$_3$), 3.25 (s, 3H, NCH$_3$), 2.34 (br s, 1H, CH).

$^{13}C$ NMR (δ, ppm)  
(CDCl$_3$, 100 MHz)  
156.7 (d, $J_{C,F} = 237.5$Hz, Cq, Ar), 156.5 (Cq, C=O), 144.4 (Cq, Ar), 115.50 (d, $J_{C,F} = 22.5$Hz, CH-Ar), 113.0 (br s, CH-Ar), 78.7 (Cq, CH), 72.6 (C CH), 53.3 (OCH$_3$), 39.4 (NCH$_3$), 38.0 (NCH$_2$).
IR (ν, cm⁻¹) (CCl₄): 3312 (m), 3058 (w), 3003 (w), 2957 (w), 2901 (w), 1720 (s), 1614 (w), 1509 (s), 1446 (s), 1418 (w), 1375 (s), 1322 (m), 1272 (m), 1231 (s), 1195 (m), 1158 (m), 1118 (s), 1069 (m).

HRMS (EI+, m/z): Calculated: 236.0961 Found: 236.0969.

4-(N-Methoxycarbonyl-N-methyl-N-prop-2-ynyl-hydrazino)-benzoic acid ethyl ester

\[
\text{MF: C}_{15}\text{H}_{18}\text{O}_{4}\text{N}_2 \\
\text{MW = 290 g.mol}^{-1}
\]

Method: See general procedure 6.2 using (1 equiv., 1.26 mmol, 300 mg) of 4-(N-methoxycarbonyl-hydrazino)-benzoic acid ethyl ester followed by general procedure 6.3 using (1 equiv., 1.26 mmol, 297 mg) of the previously prepared compound and (1.2 equiv., 1.89 mmol, solution 1.4M in 3:1 toluene:THF, 1.10 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 1.26 mmol, 318 mg) of 4-(N-methoxycarbonyl-N-methyl-hydrazino)-benzoic acid ethyl ester.

Purification: Flash column chromatography (silica gel, 7:3 PE:AcOEt).

Product: Orange/yellow oil.

Isolated yield: 71% for three steps.

\(^1\)H NMR (δ, ppm) (CDCl₃, 400 MHz): 7.94 (d, J = 8.4Hz, 2H, CH-Ar), 6.68 (br d, J = 8.4Hz, 2H, CH-Ar), 4.61 (br d, J = 17.1Hz, 1H, NCH₂), 4.33 (q, J = 7.1Hz, 2H, OCH₂CH₃), 4.11 (d, J = 17.1Hz, 1H, NCH₂), 3.70 (br s, 3H, OCH₃), 3.31 (s, 3H, NCH₃), 2.29 (br s, 1H, CH), 1.36 (t, J = 7.1Hz, 3H, OCH₂CH₃).

\(^13\)C NMR (δ, ppm) (CDCl₃, 100 MHz): 166.5 (Cq, C=O), 156.3 (Cq, C=O), 151.5 (Cq, Ar), 131.4 (CH-Ar), 120.9 (Cq, Ar), 110.9 (CH-Ar), 78.3 (Cq, C≡CH), 73.0 (C≡CH), 60.4 (OCH₂CH₃), 53.7 (OCH₃), 39.3 (NCH₂), 38.1 (NCH₃), 14.4 (OCH₂CH₃).

IR (ν, cm⁻¹) (CCl₄): 3312 (m), 2982 (m), 2957 (m), 2906 (w), 1713 (s), 1608 (s), 1558 (w), 1555 (w), 1550 (m), 1547 (m), 1535 (w), 1464 (m), 1446 (s), 1428 (w), 1368 (s), 1333 (s), 1276 (s), 1126 (s), 1182 (s), 1156 (m), 1108 (s), 1071 (m), 1021 (m).

HRMS (EI+, m/z): Calculated: 290.1267 Found: 290.1259.
**N-(4-Cyano-phenyl)-N'-methyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester**

\[
\text{MF: } C_{13}H_{13}O_2N_3 \\
\text{MW = 243 g.mol}^{-1}
\]

**Method:**
See **general procedure 6.2** using (1 equiv., 1.6 mmol, 300 mg) of \(N\)-(4-cyano-phenyl)-hydrazinecarboxylic acid methyl ester followed by **general procedure 6.3** using (1 equiv., 1.6 mmol, 328 mg) of the previously prepared compound and (1.5 equiv., 2.4 mmol, solution 1.4 M in 3:1 toluene:THF, 1.7 mL) of MeMgBr followed by **general procedure 6.4** using (1 equiv., 1.6 mmol, 325 mg) of \(N\)-(4-cyano-phenyl)-\(N\)-methyl-hydrazinecarboxylic acid methyl ester.

**Purification:**
Flash column chromatography (silica gel, 7:3 PE:AcOEt).

**Product:**
Orange oil.

**Isolated yield:**
87% for three steps.

**\(^1\)H NMR (\(\delta\), ppm)**

(CDCl\(_3\), 400 MHz)

7.52 (d, \(J = 8.5\)Hz, 2H, CH-Ar), 6.72 (br d, \(J = 8.5\)Hz, 2H, CH-Ar), 4.56 (br d, \(J = 17.2\)Hz, 1H, NCH\(_2\)), 4.14 (d, \(J = 17.2\)Hz, 1H, NCH\(_2\)), 3.72 (br s, 3H, OCH\(_3\)), 3.30 (s, 3H, NCH\(_3\)), 2.30 (br s, 1H, \(\equiv\)CH).

**\(^{13}\)C NMR (\(\delta\), ppm)**

(CDCl\(_3\), 100 MHz)

155.9 (Cq, C=O), 151.2 (Cq, Ar), 133.6 (CH-Ar), 119.8 (Cq, Ar), 111.7 (CH-Ar), 101.4 (Cq, CN), 77.9 (Cq, \(\equiv\)CH), 73.3 (C \(\equiv\) CH), 53.8 (OCH\(_3\)), 39.0 (NCH\(_3\)), 38.1 (NCH\(_2\)).

**IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))**
3311 (m), 3005 (w), 2957 (w), 2908 (w), 2825 (w), 2225 (m), 1727 (s), 1609 (s), 1514 (s), 1446 (s), 1376 (s), 1345 (s), 1275 (m), 1238 (s), 1195 (w), 1179 (w), 1159 (w), 1119 (w), 1070 (w).

**HRMS (El+, m/z):**
Calculated: 243.1008 Found: 243.1008.
**N-Methyl-N-prop-2-ynyl-N-(4-trifluoromethyl-phenyl)-hydrazinecarboxylic acid methyl ester**

MF: C_{13}H_{13}O_{2}N_{2}F_{3}

MW = 286 g/mol

**Method:**
See general procedure 6.2 using (1 equiv., 1.28 mmol, 300 mg) of \(N\)-(4-trifluoromethyl-phenyl)-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 1.28 mmol, 297 mg) of the previously prepared compound and (1.5 equiv., 1.92, solution 1.4M in 3:1 toluene:THF, 1.3 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 1.28 mmol, 317 mg) of \(N\)-methyl-\(N\)-(4-trifluoromethyl-phenyl)-hydrazinecarboxylic acid methyl ester.

**Purification:**
Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product:**
Orange oil.

**Isolated yield:**
77% for three steps.

**\(^1H\) NMR** (\(\delta, \text{ppm}\))

\((\text{CDCl}_3, 400 \text{ MHz})\)

- 7.45 (d, \(J = 8.8\text{Hz}, 2\text{H}, \text{CH-Ar}\)), 6.75 (br s, 2H, CH-Ar), 4.46 (br d, \(J = 17.8\text{Hz}, 1\text{H}, \text{NCH}_2\)), 4.13 (d, \(J = 17.8\text{Hz}, 1\text{H}, \text{NCH}_2\)), 3.71 (br s, 3H, OCH\(_3\)), 3.29 (s, 3H, NCH\(_3\)), 2.30 (br s, 1H, CH).

**\(^{13}C\) NMR** (\(\delta, \text{ppm}\))

\((\text{CDCl}_3, 100 \text{ MHz})\)

- 156.2 (Cq, C=O), 150.4 (q, \(J_{C-F} = 21.0\text{Hz}, \text{Cq, Ar}\)), 126.7 (q, \(J_{C-F} = 270.6 \text{Hz}, \text{Cq, CF}_3\)), 126.0 (q, \(J = 3.8\text{Hz}, \text{CH-Ar}\)), 121.1 (Cq, Ar), 111.2 (CH-Ar), 78.3 (Cq, \(C \equiv CH\)), 73.0 (C \(C \equiv CH\)), 53.7 (OCH\(_3\)), 39.2 (NCH\(_3\)), 38.2 (NCH\(_2\)).

**IR** (\(\nu, \text{cm}^{-1}\)) (CCl\(_4\))

- 3312 (s), 3004 (w), 2957 (m), 2907 (w), 2823 (w), 1724 (s), 1619 (s), 1585 (w), 1523 (s), 1446 (s), 1377 (s), 1327 (s), 1276 (s), 1237 (s), 1194 (s), 1166 (s), 1123 (s), 1076 (s), 1065 (s), 1005 (s).

**HRMS** (EI+, m/z):

- Calculated: 286.0929
- Found: 286.0925.
**N-Methyl-N-(4-nitro-phenyl)-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester**

![Chemical Structure Image]

**Method:**
See **general procedure 6.2** using (1 equiv., 1.18 mmol, 250 mg) of N-(4-nitro-phenyl)-hydrazinecarboxylic acid methyl ester followed by **general procedure 6.3** using (1 equiv., 1.18 mmol, 246 mg) of the previously prepared compound and (1.5 equiv., 1.77 mmol, solution 1.4 M in 3:1 toluene:THF, 1.3 mL) of MeMgBr followed by **general procedure 6.4** using (1 equiv., 1.0 mmol, 230 mg) of N-methyl-N-(4-nitro-phenyl)-hydrazinecarboxylic acid methyl ester.

**Purification:**
Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product:**
Orange oil.

**Isolated yield:**
71% for three steps.

**$^1$H NMR (δ, ppm)**

(CDCl$_3$, 400 MHz)

8.12 (d, $J = 9.0$Hz, 2H, CH-Ar), 6.70 (d, $J = 9.0$Hz, 2H, CH-Ar), 4.54 (br d, $J = 17.3$Hz, 1H, NCH$_2$), 4.18 (d, $J = 17.3$Hz, 1H, NCH$_3$), 3.72 (br s, 3H, OCH$_3$), 3.33 (s, 3H, NCH$_3$), 2.31 (br s, 1H, CH).

**$^{13}$C NMR (δ, ppm)**

(CDCl$_3$, 100 MHz)

155.5 (Cq, C=O), 153.0 (Cq, Ar), 139.8 (Cq, Ar), 125.8 (CH-Ar), 110.8 (CH-Ar), 77.7 (Cq, C === CH), 73.5 (C === CH), 53.8 (OCH$_3$), 39.1 (NCH$_3$), 38.2 (NCH$_2$).

**IR (ν, cm$^{-1}$) (CCl$_4$)**

3313 (m), 3090 (w), 3004 (w), 2958 (m), 2934 (w), 2910(w), 2855 (w), 2827 (w), 2662 (w), 1725 (s), 1599 (s), 1511 (s), 1446 (s), 1421 (m), 1376 (s), 1335 (s), 1276 (s), 1238 (s), 1193 (s), 1158 (m), 1110 (s), 1070 (m).

**HRMS (EI+, m/z):**
Calculated: 263.0906   Found: 263.0912.
Method: See general procedure 6.2 using (1 equiv., 1.56 mmol, 280 mg) of N-m-tolyl-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 1.56 mmol, 278 mg) of the previously prepared compound with (1.5 equiv., 2.34 mmol, solution 1.4 M in 3:1 toluene:THF, 1.7 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 1.53 mmol, 296 mg) of N-methyl-N'-m-tolyl-hydrazinecarboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 9:1 PE:AcOEt).

Product: Yellow oil.

Isolated yield: 73% for three steps.

\[ ^1H\text{ NMR} (\delta, \text{ppm}) \]
\[
\begin{align*}
7.14 (t, J = 7.9Hz, 1H, CH-Ar), & 6.67 (d, J = 7.9Hz, 1H, CH-Ar), & 6.50 (br s, 2H, CH-Ar), & 4.62 (br d, J = 16.8Hz, 1H, NCH\_3), & 4.02 (br d, J = 16.8Hz, 1H, NCH\_3), \\
3.71 (br s, 3H, OCH\_3), & 3.25 (s, 3H, NCH\_3), & 3.23 (s, 3H, NCH\_3), & 2.32 (s, 3H, Ph-CH\_3), & 2.28 (br s, 1H, CH). \\
\end{align*}
\]

\[ ^{13}C\text{ NMR} (\delta, \text{ppm}) \]
\[
\begin{align*}
156.8 (C, C=O), & 147.8 (C, Ar), & 139.1 (C, Ar), & 129.1 (CH-Ar), & 120.1 (CH-Ar), \\
112.5 (CH-Ar), & 109.1 (CH-Ar), & 79.0 (C, CH), & 72.4 (C=CH), & 53.5 \\
(OCH\_3), & 39.4 (NCH\_3), & 38.0 (NCH\_3), & 21.8 (Ph-CH\_3). \\
\end{align*}
\]

\[ \text{IR} (\nu, \text{cm}^{-1}) (\text{CCl}_4) \]
\[
3313 (m), 3043 (w), 3001 (w), 2956 (m), 2922 (w), 2826 (w), 1717 (s), 1605 (s), 1587 (s), 1493 (s), 1447 (s), 1376 (s), 1321 (m), 1275 (m), 1237 (s), 1194 (m), 1173 (w), 1154 (w), 1120 (m), 1097 (w), 1083 (w), 1068 (w). \\
\]

**N-(3-Chloro-phenyl)-N'-methyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester**

![Chemical Structure](image)

MF: C_{12}H_{13}O_{2}N_{2}Cl  
MW = 252.5 g.mol^{-1}

**Method:**  
See general procedure 6.2 using (1 equiv., 5.6 mmol, 1 g) of \( N-(3\)-chloro-phenyl)-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 2 mmol, 397 mg) of the previously prepared compound and (1.5 equiv., 3 mmol, solution 1.4 M in 3:1 toluene:THF, 2.14mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 2 mmol, 450 mg) of \( N-(3\)-chloro-phenyl)-\( N \)-methyl-hydrazinecarboxylic acid methyl ester.

**Purification:**  
Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product:**  
Orange oil.

**Isolated yield:**  
70% for three steps.

**\( ^{1}H \) NMR (\( \delta \), ppm) (CDCl₃, 400 MHz)**  
7.16 (t, \( J = 8.1\) Hz, 1H, CH-Ar), (dd, \( J = 1.0\) Hz, \( J = 7.8\) Hz, 1H, CH-Ar), 6.69 (br s, 1H, CH-Ar), 6.58 (br s, 1H, CH-Ar), 4.59 (d, \( J = 16.6\) Hz, 1H, NCH₃), 4.08 (d, \( J = 16.6\) Hz, 1H, NCH₃), 3.71 (br s, 3H, OCH₃), 3.24 (s, 3H, NCH₃), 2.30 (br s, 1H, CH).

**\( ^{13}C \) NMR (\( \delta \), ppm) (CDCl₃, 100 MHz)**  
156.4 (Cq, C=O), 149.2 (Cq, Ar), 135.2 (Cq, Ar), 130.2 (CH-Ar), 119.1 (CH-Ar), 112.0 (CH-Ar), 110.1 (CH-Ar), 78.5 (Cq, C=CH), 72.8 (C=CH), 53.6 (OCH₃), 39.3 (NCH₃), 38.2 (NCH₃).

**IR (\( \nu \), cm\(^{-1}\)) (CCl₄)**  
3312 (m), 3077 (w), 3004 (w), 2957 (m), 2908 (w), 2818 (w), 1721 (s), 1596 (s), 1573 (m), 1487 (s), 1446 (s), 1419 (m), 1375 (s), 1347 (m), 1296 (m), 1266 (s), 1237 (s), 1195 (m), 1157 (m), 1123 (s), 1106 (s), 1083 (s).

**HRMS (El+, m/z):**  
Calculated: 252.0666  found: 252.0656.
**N'-Methyl-N-prop-2-ynyl-N-o-tolyl-hydrazinecarboxylic acid methyl ester**

**MF:** C_{13}H_{16}N_{2}O_{2}

**MW:** 232 g mol^{-1}

**Method:** See general procedure 6.2 using (1 equiv., 1.56 mmol, 280 mg) of N-o-tolyl-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 1.56 mmol, 278 mg) of the previously prepared compound and (1.5 equiv., 2.34 mmol, solution 1.4 M in 3:1 toluene:THF, 1.7 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 1.56 mmol, 304 mg) of N-methyl-N'-o-tolyl-hydrazinecarboxylic acid methyl ester.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:** Yellow oil.

**Isolated yield:** 72% for three steps.

**^1H NMR** (δ, ppm) (CDCl₃, 400 MHz)

- 7.19-7.12 (m, 2H, CH-Ar), 7.06 (br s, 1H, CH-Ar), 6.99 (t, J = 7.4Hz, 1H, CH-Ar), 4.00 (br s, 2H, NCH₂), 3.84 (s, 3H, OCH₃), 3.22 (s, 3H, NCH₃), 2.21 (br s, 1H, CH), 2.19 (s, 3H, Ph-C₃H₃).

**^13C NMR** (δ, ppm) (CDCl₃, 100 MHz)

- 155.7 (Cq, C=O), 145.8 (Cq, Ar), 131.8 (CH-Ar), 131.3 (Cq, Ar), 126.3 (CH-Ar), 123.7 (CH-Ar), 118.4 (CH-Ar), 79.8 (Cq, CH), 71.8 (Cq, CH), 53.3 (OCH₃), 40.3 (NCH₃), 35.0 (NCH₃), 18.9 (Ph-C₃H₃).

**IR** (ν, cm⁻¹) (CCl₄)

- 3313 (m), 3072 (w), 3025 (w), 2996 (w), 2956 (m), 2897 (w), 2806 (w), 1713 (s), 1600, 1495 (w), 1449 (s), 1417 (m), 1384 (s), 1337 (m), 1304 (m), 1270 (s), 1238 (s), 1193 (m), 1146 (m), 1130 (m), 1111 (s), 1071 (w), 1056 (w), 1036 (w).

**HRMS** (EI+, m/z): Calculated: 232.1212  Found: 232.1215.

382
**N-(2-Chloro-phenyl)-N-methyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester**

![Chemical Structure]

**MF:** C_{12}H_{13}O_{2}N_{2}Cl  
**MW:** 252.5 g.mol^{-1}

**Method:**  See general procedure 6.2 using (1 equiv., 2 mmol, 400 mg) of N-(2-chloro-phenyl)-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 2 mmol, 397 mg) of the previously prepared compound and (1.5 equiv., 3 mmol, solution 1.4 M in 3:1 toluene:THF, 2.15 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 0.84 mmol, 180 mg) of N-(2-chloro-phenyl)-N-methyl-hydrazinecarboxylic acid methyl ester.

**Purification:**  Flash column chromatography (silica gel, 9:1 PE, AcOEt).

**Product:**  Orange oil.

**Isolated yield:**  72% for three steps.

**\(^1\)H NMR (\(\delta\), ppm)**

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<th>Compound</th>
<th>ppm</th>
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<tr>
<td>7.30 (d, (J = 7.7)Hz, 1H, CH-Ph), 7.20 (t, (J = 7.7)Hz, 1H, CH-Ph), 7.07 (br s, 1H, CH-Ph), 6.95 (t, (J = 7.7)Hz, 1H, CH-Ph), 4.50 (br s, 1H, NCH(_2)), 4.31 (br s, 1H, NCH(_2)), 3.75 (s, 3H, OCH(_3)), 3.29 (br s, 3H, NCH(_3)), 2.26 (br s, 1H, (==)CH).</td>
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**\(^13\)C NMR (\(\delta\), ppm)**

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<tr>
<td>156.4 (Cq, C=O), 155.0 (Cq, C=O_rotamere), 145.5 (Cq, Ar), 144.2 (Cq, Ar_rotamere), 131.2 (CH-Ar), 130.8 (CH-Ar_rotamere), 127.0 (CH-Ar), 126.2 (Cq, Ar), 125.8 (Cq, Ar_rotamere), 123.8 (CH-Ar), 120.5 (CH-Ar), 119.6 (CH-Ar_rotamere), 79.8 (Cq, CH), 79.5 (Cq, CH_rotamere), 72.1 (CH), 53.1 (OCH(_3)), 40.7 (NCH(_3)), 40.1 (NCH(_2)), 39.5 (NCH(_3)_rotamere), 37.3 (NCH(_2)_rotamere).</td>
<td></td>
</tr>
</tbody>
</table>

**IR (\(\nu\), \text{cm}^{-1}) (CCl\(_4\))**

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<th>(\text{cm}^{-1})</th>
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<td>3068 (w)</td>
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<td>3000 (w)</td>
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<td>2956 (w)</td>
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<td>2899 (w)</td>
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<td>1239 (s)</td>
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<td>1206 (s)</td>
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<td>1194 (w)</td>
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<td>1071 (s)</td>
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<td>1043 (m)</td>
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**HRMS (El+, m/z):**

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<th>Compound</th>
<th>m/z</th>
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<td></td>
</tr>
<tr>
<td>Calculated: 252.0666</td>
<td>Found: 252.0669.</td>
</tr>
</tbody>
</table>
**N-Methyl-N-phenyl-N-(3-phenyl-prop-2-ynyl)-hydrazinecarboxylic acid methyl ester**

![Chemical structure](image)

**MF:** C_{18}H_{18}O_{2}N_{2}

**MW:** 294 g.mol^{-1}

**Method:**
See **general procedure 6.2** using (1 equiv., 10 mmol, 1.66 g) of N-phenyl-hydrazinecarboxylic acid methyl ester followed by **general procedure 6.3** using (1 equiv., 10 mmol, 1.64 mmol) of the previously prepared compound and (1.5 equiv., 15 mmol, solution 1.4 M in 3:1 toluene:THF, 10.7 mL) of MeMgBr followed by **general procedure 6.4** using (1 equiv., 10 mmol, 1.8 g) of N-methyl-N-phenyl-hydrazinecarboxylic acid methyl ester followed by **general procedure 6.5** using (1 equiv., 1 mmol, 218 mg) of N-methyl-N-phenyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester and (1.5 equiv., 1.5 mmol, 170 µL) of iodo benzene.

**Purification:**
Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:**
Orange oil.

**Isolated yield:**
65% for four steps.

**¹H NMR (δ, ppm)**
(CDCl₃, 400 MHz)
7.38-7.36 (m, 2H, CH-Ph), 7.31-7.28 (m, 5H, CH-Ph), 6.86 (t, J = 7.3Hz, 1H, CH-Ph), 6.76 (br s, 2H, CH-Ph), 4.83 (d, J = 17.3Hz, 1H, NCH₂), 4.33 (d, J = 17.3Hz, 1H, NCH₂), 3.81 (br s, 0.7H, OCH₃ rotamere), 3.72 (br s, 2.3H, OCH₃), 3.31 (s, 3H, NCH₃).

**¹³C NMR (δ, ppm)**
(CDCl₃, 100 MHz)
156.9 (Cq, C=O), 147.9 (Cq, Ph), 131.6 (CH-Ph), 129.2 (CH-Ph), 128.3 (CH-Ph), 128.2 (CH-Ph), 122.6 (Cq, Ph), 119.0 (CH-Ph), 111.8 (CH-Ph), 84.2 (Cq, C ≡ C), 77.2 (Cq, C ≡ C), 53.5 (OCH₃), 39.4 (NCH₂), 38.6 (NCH₂).

**IR (ν, cm⁻¹) (CCl₄)**
3066 (w), 3033 (w), 3002 (w), 2956 (m), 2899 (w), 2817 (w), 1716 (s), 1600 (s), 1500 (s), 1491 (s), 1446 (s), 1416 (w), 1376 (s), 1349 (m), 1323 (m), 1269 (s), 1235 (s), 1194 (m), 1157 (m), 1119 (s), 1089 (m), 1066 (m), 1030 (m), 1005 (m).

**HRMS (EI+, m/z):**
Calculated: 294.1368  Found: 294.1370.
**N-Methyl-N-pent-4-en-2-ynyl-N-phenyl-hydrazinecarboxylic acid methyl ester**

![Chemical Structure](image)

**MF:** C_{14}H_{16}O_2N_2

**MW:** 244 g.mol⁻¹

**Method:**
See general procedure 6.2 using (1 equiv, 10 mmol, 1.66 g) of N-phenyl-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 10 mmol, 1.64 mmol) of the previously prepared compound and (1.5 equiv., 15 mmol, solution 1.4 M in 3:1 toluene:THF, 10.7 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 10 mmol, 1.8 g) of N-methyl-N-phenyl-hydrazinecarboxylic acid methyl ester followed by general procedure 6.5 using (1 equiv., 1 mmol, 218 mg) of N-methyl-N-phenyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester and (2 equiv., 2 mmol, solution 1M in THF, 2 mL) of vinyl bromide.

**Purification:**
Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:**
Orange oil.

**Isolated yield:**
61% for four steps.

**¹H NMR (δ, ppm)**
(CDCl₃, 400 MHz)
7.26 (t, J = 8.0Hz, 2H, CH-Ph), 6.84 (t, J = 7.3Hz, 1H, CH-Ph), 6.71 (br s, 2H, CH-Ph), 5.77 (dd, J = 11.0Hz, 11.0Hz, 1H, CH=CH₂), 5.59 (d, J = 17.3Hz, 1H, CH=C=CH₂), 5.47 (d, J = 11.0Hz, 1H, CH=CH₂), 4.72 (br d, J = 17.6Hz, 1H, NCH₂), 4.19 (d, J = 17.6Hz, 1H, NCH₂), 3.70 (br s, 3H, OCH₃), 3.26 (s, 3H, NCH₃).

**¹³C NMR (δ, ppm)**
(CDCl₃, 100 MHz)
156.8 (Cq, C=O), 147.8 (Cq, Ph), 129.2 (CH), 127.4 (CH=CH₂), 119.0 (CH), 116.6 (CH), 111.8 (CH), 84.9 (Cq, C=C), 82.9 (Cq, C=C), 53.5 (OCH₃), 39.3 (NCH₂), 38.5 (NCH₂).

**IR (ν, cm⁻¹) (CCl₄)**
3097 (w), 3031 (w), 2956 (w), 2817 (w), 1716 (s), 1600 (s), 1500 (s), 1446 (s), 1414 (w), 1376 (s), 1348 (m), 1323 (m), 1272 (m), 1236 (m), 1194 (w), 1148 (w), 1118 (m), 1089 (w), 1065 (w), 1031 (w).

**HRMS (EI+, m/z):**
Calculated: 244.1212 Found: 244.1212.
**N-Hex-5-en-2-ynyl-\(N\)-methyl-\(N\)-phenyl-hydrazinecarboxylic acid methyl ester**

![Chemical Structure](Image)

**MF:** C\(_{15}\)H\(_{18}\)O\(_2\)N\(_2\)

**MW:** 258 g.mol\(^{-1}\)

**Method:**
See **general procedure 6.2** using (1 equiv., 10 mmol, 1.66 g) of \(N\)-phenyl-hydrazinecarboxylic acid methyl ester followed by **general procedure 6.3** using (1 equiv., 10 mmol, 1.64 mmol) of the previously prepared compound and (1.5 equiv., 15 mmol, solution 1.4 M in 3:1 toluene:THF, 10.7 mL) of MeMgBr followed by **general procedure 6.4** using (1 equiv., 10 mmol, 1.8 g) of \(N\)-methyl-\(N\)-phenyl-hydrazinecarboxylic acid methyl ester.

Next, to a solution of propargylated hydrazine (1 equiv., 1 mmol, 218 mg) in DMF (1 M, 1 mL) under an argon atmosphere at room temperature was added K\(_2\)CO\(_3\) (2.8 equiv., 2.8 mmol, 386 mg), tetrabutylammonium bromide (0.15 equiv., 0.15 mmol, 48 mg) and CuI (0.1 equiv., 0.1 mmol, 19 mg). The reaction was allowed to stir at room temperature for 30 min, allyl bromide (5 equiv., 5 mmol, 430 µL) and NaI (0.8 equiv., 0.8 mmol, 120 mg) were added and the reaction was allowed to stir at room temperature for 60 h. The reaction was then diluted with a saturated solution of NH\(_4\)Cl, extracted with AcOEt (3x), washed with H\(_2\)O (5x), dried (MgSO\(_4\)) and purified by flash column chromatography.

**Purification:**
Flash column chromatography (silica gel, 9:1 PE: AcOEt).

**Product:**
Yellow oil.

**Isolated yield:**
33% for four steps.

**\(^1H\) NMR (δ, ppm)**
(CDCl\(_3\), 400 MHz)

- 7.27-7.23 (m, 2H, CH-Ph), 6.83 (t, J = 7.3Hz, 1H, CH-Ph), 6.71 (br s, 2H, CH-Ph), 5.77 (ddd, J = 5.2Hz, J = 10.2Hz, J = 16.3Hz, 1H, CH=CH\(_2\)), 5.27 (dd, J = 1.0Hz, J = 16.3Hz, 1H, CH=CH\(_2\)), 5.09 (dd, J = 1.0Hz, J = 10.2Hz, 1H, CH=CH\(_2\)), 4.61 (br d, J = 17.4Hz, 1H, NCH\(_3\)), 4.09 (br d, J = 17.4Hz, 1H, NCH\(_3\)), 3.69 (br s, 3H, OCH\(_3\)), 3.25 (s, 3H, NCH\(_3\)), 3.00 (br d, J = 5.2Hz, 2H, CH\(_2\)-allyl chain).

**\(^13C\) NMR (δ, ppm)**
(CDCl\(_3\), 100 MHz)

- 156.9 (Cq, C=O), 148.0 (Cq, Ph), 132.2 (CH), 129.2 (CH), 118.9 (CH), 116.2 (CH=CH\(_2\)), 111.8 (CH), 81.2 (Cq, C \(\equiv\) C), 77.2 (Cq, C \(\equiv\) C), 53.4 (OCH\(_3\)), 39.3 (NCH\(_3\)), 38.3 (NCH\(_3\)), 23.0 (CH\(_2\)-allyl chain).

**IR (ν, cm\(^{-1}\)) (CCl\(_4\))**

- 3093 (w), 3068 (w), 3031 (w), 2986 (w), 2956 (w), 2897 (w), 2816 (w), 1717 (s), 1643 (w), 1600 (s), 1500 (s), 1446 (s), 1419 (m), 1377 (s), 1349 (s), 1324 (s), 1274 (s), 1237 (s), 1194 (s), 1158 (m), 1117 (s), 1089 (m), 1065 (m), 1031 (m).
**HRMS (EI+, m/z)**: Calculated: 258.1368 Found: 258.1369.

**N’ methyl-N’-phenyl-N-prop-2-ynyl-[(4-methylphenyl)sulfonyl]hydrazine** 6.11t

![Chemical structure](image)

**MF:** C₁₇H₁₈O₂N₂S

**MW:** 314 g.mol⁻¹

**Method:** See general procedure 6.2 using (1 equiv., 1 mmol, 262 mg) of N-phenyl-N’-tosyl hydrazine followed by general procedure 6.3 using (1 equiv., 1 mmol, 260 mg) of the previously prepared compound and (1.5 equiv., 1.5 mmol, solution 1.4 M in 3:1 toluene:THF, 1.10 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 1 mmol, 276 mg) of the previously methylated compound.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:** Orange oil.

**Isolated yield:** 32% for three steps (last compound is not stable on SiO₂, it decomposes losing Ts group).

**¹H NMR (δ, ppm)** (CDCl₃, 400 MHz)

<table>
<thead>
<tr>
<th>7.85 (d, J = 8.2Hz, 2H, CH-Ts)</th>
<th>7.29 (d, J = 8.2Hz, 2H, CH-Ts)</th>
<th>7.19 (dd, J = 7.4Hz, J = 8.7Hz, 2H, CH-Ph)</th>
<th>6.87-6.82 (m, 3H, CH-Ph)</th>
<th>4.51 (br d, J = 17.5Hz, 1H, NCH₂)</th>
<th>4.08 (br d, J = 17.5Hz, NCH₂)</th>
<th>3.00 (s, 3H, NCH₃)</th>
<th>2.42 (s, 3H, Ph-CH₃)</th>
<th>2.23 (t, J = 2.5Hz, 1H, CH).</th>
</tr>
</thead>
</table>

**¹³C NMR (δ, ppm)** (CDCl₃, 100 MHz)

| 147.9 (Cq, Ar) | 144.2 (Cq, Ar) | 135.6 (Cq, Ar) | 129.5 (CH-Ar) | 128.9 (CH-Ar) | 128.4 (CH-Ar) | 120.3 (CH-Ar) | 113.6 (CH-Ar) | 78.2 (Cq, CH) | 73.9 (CH) | 37.8 (NCH₃) | 36.0 (NCH₃) | 21.6 (Ph-CH₃) |
|----------------|----------------|----------------|--------------|--------------|--------------|--------------|--------------|--------------|........|...............|................|................|

**IR (ν, cm⁻¹)** (CCl₄)

| 3312 (m) | 3068 (w) | 3032 (w) | 2925 (w) | 2816 (w) | 1600 (s) | 1498 (s) | 1470 (w) | 1455 (w) | 1417 (w) | 1362 (s) | 1305 (w) | 1292 (w) | 1264 (w) | 1185 (m) | 1168 (s) | 1118 (m) | 1096 (s) | 1034 (w) | 1019 (w) |
|-----------|----------|----------|----------|----------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|--------|

**HRMS (EI+, m/z):** Calculated: 314.1089 Found: 314.1080.
**N-allyl-N-methyl-N-phenyl-hydrazinecarboxylic acid methyl ester**

\[ \text{MF: } C_{12}H_{16}O_2N_2 \]

\[ \text{MW = 220 g.mol}^{-1} \]

**Method:** See general procedure 6.2 using (1 equiv., 1.9 mmol, 315 mg) of \( N \)-phenyl-hydrazinecarboxylic acid methyl ester followed by general procedure 6.3 using (1 equiv., 1.9 mmol, 312 mg) of the previously prepared compound and (1.5 equiv., 2.85 mmol, solution 1.4 M in 3:1 toluene:THF, 2.0 mL) of MeMgBr followed by general procedure 6.4 using (1 equiv., 1.9 mmol, 350 mg) of \( N \)-methyl-\( N \)-phenyl-hydrazinecarboxylic acid methyl ester and allyl bromide (1.5 equiv., 2.85 mmol, 250 \( \mu \)L) instead of propargyl bromide.

**Purification:** Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product:** Orange oil.

**Isolated yield:** 70% for three steps.

**\( ^1H \) NMR (\( \delta \), ppm) (CDCl\(_3\), 400 MHz)**

- 7.27-7.23 (m, 2H, CH-Ph), 6.82 (t, \( J = 7.3\)Hz, 1H, CH-Ph), 6.66 (br d, \( J = 7.3\)Hz, 2H, CH-Ph), 6.03-5.93 (m, 1H, CH\(_{=CH_2}\)), 5.24 (dd, \( J = 1.2\)Hz, \( J = 17.1\)Hz, 1H, CH\(_{=CH_2}\)), 5.17 (dd, \( J = 1.2\)Hz, \( J = 10.1\)Hz, 1H, CH\(_{=CH_2}\)), 4.39 (br s, 1H, NCH\(_2\)), 3.68 (br s, 3H, OCH\(_3\)), 3.51 (s, 3H, NCH\(_3\)).

**\( ^{13}C \) NMR (\( \delta \), ppm) (CDCl\(_3\), 100 MHz)**

- 162.7 (Cq, C=O), 157.2 (Cq, Ph), 148.1 (CH), 133.3 (CH), 129.2 (CH), 118.8 (CH\(_{=CH_2}\)), 111.7 (CH), 53.2 (OCH\(_3\)), 51.8 (NCH\(_2\)), 39.7 (NCH\(_3\)).

**IR (\( \nu \), cm\(^{-1}\)) (CCl\(_4\))**

- 3070 (w), 3031 (w), 2997 (w), 2955 (m), 2899 (w), 2816 (w), 1713 (s), 1644 (w), 1600 (s), 1500 (s), 1447 (s), 1418 (w), 1378 (s), 1323 (s), 1277 (s), 1240 (s), 1194 (s), 1157 (s), 1113 (s), 1084 (m), 1056 (m), 1031 (m).

**HRMS (EI+, m/z):**

- Calculated: 220.1212   Found: 220.1210.
**N-(4-Methoxy-phenyl)-N-phenyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester**

\[
\text{MF: } \text{C}_{18}\text{H}_{18}\text{O}_{3}\text{N}_2
\]

\[
\text{MW} = 310 \text{ g.mol}^{-1}
\]

**Method:**

See *general procedure 6.2* using (1 equiv., 1.5 mmol, 300 mg) of *N-(4-methoxy-phenyl)-hydrazinecarboxylic acid methyl ester* followed by *general procedure 6.3* using (1 equiv., 1.5 mmol, 291 mg) of the previously prepared compound and (1.5 equiv., 2.25 mmol, solution 1M in THF, 2.25 mL) of PhMgBr followed by *general procedure 6.4* using (1 equiv., 1.5 mmol, 416 mg) of *N-(4-methoxy-phenyl)-N-phenyl-hydrazinecarboxylic acid methyl ester*.

**Purification:**

Flash column chromatography (silica gel, 85:15 PE: AcOEt).

**Product:**

Orange oil.

**Isolated yield:**

82% for three steps.

**$^1H$ NMR ($\delta$, ppm)**

(CDC$_3$, 400 MHz)

7.36-7.29 (br s, 2H, CH-Ar), 7.22 (t, $J = 7.9$Hz, 2H, CH-Ar), 6.90-6.85 (m, 5H, CH-Ar), 4.38 (br s, 2H, NCH$_2$), 3.82 (br s, 3H, OCH$_3$), 3.72 (br s, 3H, OCH$_3$), 2.25 (br s, CH).

**$^{13}$C NMR ($\delta$, ppm)**

(CDC$_3$, 100 MHz)

157.2 (Cq, C=O), 156.9 (Cq, C=O$_{\text{rotamere}}$), 145.7 (Cq, Ar), 136.3 (Cq, Ar), 133.30 (Cq, Ar), 128.9 (CH-Ar), 126.4 (CH-Ar$_{\text{rotamere}}$), 125.6 (CH-Ar), 120.4 (CH-Ar), 115.1 (CH-Ar), 114.5 (CH-Ar), 78.4 (Cq, $\equiv$CH), 73.1 ($\equiv$CH), 55.5 (OCH$_3$), 53.6 (OCH$_3$), 39.8 (NCH$_2$$_{\text{rotamere}}$), 38.7 (NCH$_2$).

**IR (\nu, cm$^{-1}$) (CCl$_4$)**

3313 (m), 3092 (w), 3066 (w), 3042 (w), 3003 (m), 2955 (m), 2934 (m), 2909 (w), 2836 (w), 1725 (s), 1599 (s), 1494 (s), 1445 (s), 1416 (m), 1378 (s), 1275 (s), 1196 (s), 1181 (s), 1126 (s), 1082 (m), 1040 (s).

**HRMS (EI+, m/z):**

Calculated: 310.1318  Found: 310.1317.
\[ \text{N-(4-Fluoro-phenyl)-N'-phenyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester} \]

\[
\text{MF: C}_{17}\text{H}_{15}\text{O}_{2}\text{N}_{2}\text{F} \]

\[
\text{MW = 298 g mol}^{-1} \]

**Method:**

See *general procedure 6.2* using (1 equiv., 1.6 mmol, 300 mg) of \(N\)-(4-fluoro-phenyl)-hydrazinecarboxylic acid methyl ester followed by *general procedure 6.3* using (1 equiv., 1.6 mmol, 297 mg) and (1.5 equiv., 2.45 mmol, solution 1M in THF, 2.5 mL) of PhMgBr followed by *general procedure 6.4* using (1 equiv., 1.6 mmol, 423 mg) of \(N\)-(4-fluoro-phenyl)-\(N\)-phenyl-hydrazinecarboxylic acid methyl ester.

**Purification:**

Flash column chromatography (silica gel, 9:1 PE: AcOEt).

**Product:**

Yellow oil.

**Isolated yield:**

77% for three steps.

\(\text{\textsuperscript{1}H NMR (\delta, ppm)}\)

\(\text{(CDCl}_3, \text{ 400 MHz)}\)

7.32-7.25 (m, 4H, CH\(_{-}\)Ar), 7.07-7.00 (m, 5H, CH\(_{-}\)Ar), 4.49 (br d, \(J = 17.5\text{Hz}, 1\text{H}, \text{NCH\(_2\)}\)), 4.36 (br d, \(J = 17.5\text{Hz}, 1\text{H}, \text{NCH\(_2\)}\)), 3.85 (br s, 0.9H, CH\(_3\) rotamere), 3.74 (br s, 2.1 H, OCH\(_3\)), 2.30 (br s, 1H, \(\equiv\text{CH}\)).

\(\text{\textsuperscript{13}C NMR (\delta, ppm)}\)

\(\text{(CDCl}_3, \text{ 100 MHz)}\)

160.6 (Cq, C=O), 157.49 (d, \(J_{C-F} = 141.0\text{Hz}, \text{Cq, CF-Ar}\)), 144.5 (Cq, Ar), 140.1 (Cq, Ar), 129.1 (CH-Ar), 124.1 (CH-Ar), 123.3 (CH-Ar rotamere), 122.0 (CH-Ar), 117.2 (CH-Ar), 115.9 (d, \(J_{C-F} = 22.6\text{Hz}, \text{CH-Ar}\)), 78.2 (Cq, \(\equiv\text{CH}\)), 73.4 (C \equiv\text{CH}), 53.7 (OCH\(_3\)), 39.8 (NCH\(_3\) rotamere), 38.7 (NCH\(_2\)).

**IR (\(\nu, \text{cm}^{-1}\)) (CCl\(_4\))**

3312 (m), 3043 (w), 2956 (w), 1720 (s), 1597 (m), 1506 (s), 1496 (s), 1446 (s), 1376 (m), 1317 (m), 1272 (m), 1232 (s), 1197 (m), 1174 (m), 1156 (m), 1126 (m), 1032 (w).

**HRMS (El+, m/z):**

Calculated: 298.1118 Found: 298.1118.
4-(N-Methoxycarbonyl-N-phenyl-N-prop-2-ynyl-hydrazino)-benzoic acid ethyl ester

MF: C₂₀H₂₀O₄N₂

MW = 352 g.mol⁻¹

Method: See general procedure 6.2 using (1 equiv., 1.26 mmol, 300 mg) of 4-(N-methoxycarbonyl-hydrazino)-benzoic acid ethyl ester followed by general procedure 6.3 using (1 equiv., 1.26 mmol, 297 mg) of the previously prepared compound and (1.5 equiv., 1.89 mmol, solution 1M in THF, 1.9 mL) of PhMgBr followed by general procedure 6.4 using (1 equiv., 1.26 mmol, 406 mg) of 4-(N-methoxycarbonyl-N-phenyl-hydrazino)-benzoic acid ethyl ester.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt).

Product: Orange oil.

Isolated yield: 72% for three steps.

¹H NMR (δ, ppm) (CDCl₃, 400 MHz)

7.91 (d, J = 8.8Hz, 2H, CH-Ar), 7.40-7.30 (m, 4H, CH-Ph), 7.22 (t, J = 7.3Hz, 1H, CH-Ph), 6.97 (d, J = 8.8Hz, 2H, CH-Ar), 4.54 (br d, J = 17.8Hz, 0.7H, NCH₂), 4.45 (br d, J = 17.8Hz, 0.3H, NCH₂ rotamere), 4.33 (q, J = 7.2Hz, 2H, OCH₂CH₃), 4.28 (d, J = 17.8Hz, 1H, NCH₂), 3.83 (br s, 1H, OCH₃ rotamere), 3.71 (br s, 2H, OCH₃), 2.25 (br s, 1H, — CH), 1.36 (t, J = 7.2Hz, 3H, OCH₂CH₃).

¹³C NMR (δ, ppm) (CDCl₃, 100 MHz)

166.3 (Cq, C=O), 156.4 (Cq, C=O), 149.2 (Cq, Ar), 142.0 (Cq, Ar), 130.9 (CH-Ar), 129.5 (CH-Ar), 125.9 (CH-Ar), 124.4 (CH-Ar rotamere), 123.6 (CH-Ar), 122.7 (Cq, Ar), 114.6 (CH- Ar), 77.7 (Cq, — CH), 73.7 (C — CH), 60.5 (OCH₂CH₃), 53.8 (OCH₃), 39.6 (NCH₂ rotamere), 38.5 (NCH₂), 14.4 (OCH₂CH₃).

IR (ν, cm⁻¹) (CCl₄)

3312 (m), 2982 (w), 2957 (w), 2907 (w), 1716 (s), 1608 (s), 1594 (m), 1509 (m), 1492 (m), 1446 (m), 1425 (w), 1367 (m), 1313 (m), 1273 (s), 1238 (m), 1177 (s), 1107 (s), 1026 (s).

HRMS (EI+, m/z) : Calculated: 352.1423 Found: 352.1423.
B.3.5.2 Au(I)-catalyzed Friedel-Crafts reaction

Previous synthesized substrates were submitted to gold catalysis as it is shown in the scheme below.

**General Procedure 6.6, Gold catalyzed Friedel-Crafts reaction:** To a reaction vessel (a NMR tube in our case) was added the hydrazine derivative (1 equiv.), the gold catalyst XphosAu(NCCH₃)SbF₆ (0.04 equiv.) and CD₃NO₂ (0.2 M). The reaction was heated up to reflux (100 °C). Upon completion (NMR), the reaction mixture was transferred to a round bottom flask and concentrated under reduced pressure. Flash column chromatography afforded the hydroarylated product in the stated yields.

Scheme B.3.5.2: Gold catalyzed transformation towards tetrahydrocinnolines 6.12a-x.

6.12a, R¹ = Ph, Y = H
6.12b, R¹ = n-C₅H₁₁, Y = H
6.12c, R¹ = CH₂Ph, Y = H
6.12d, R¹ = Pr, Y = H
6.12e, R¹ = ²Bu, Y = H
6.12f, R¹ = Me, Y = p-OMe
6.12g, R¹ = Me, Y = p-Cl
6.12h, R¹ = Me, Y = p-F
6.12i, R¹ = Me, Y = p-CO₂Et
6.12j, R¹ = Me, Y = p-CN
6.12k, R¹ = Me, Y = p-CF₃
6.12ma, R¹ = Me, Y = m-Cl
6.12na, R¹ = Me, Y = p-OMe
6.12oa, R¹ = Ph, Y = p-CF₃
6.12pa, R¹ = Me, Y = o-Cl
6.12va, R¹ = Ph, Y = p-OMe
6.12wa, R¹ = Ph, Y = p-CO₂Et
6.12xa, R¹ = Ph, Y = p-F

*: Other regioisomers found

4-Methylene-1-phenyl-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl 6.12a

MF: C₁₇H₁₆O₂N₂

MW = 280 g.mol⁻¹
Method: See general procedure 6.6 using (1 equiv., 0.1 mmol, 28 mg) of \(N,\ N\)-diphenyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt).

Product: White solid.

Isolated yield: 100%.

\(^1\)H NMR (\(\delta, \text{ppm}\))
- (CDCl\(_3\), 400 MHz)
  - 7.80 (d, \(J = 7.9\text{Hz}, 1\text{H, CH-Ar}\), 7.33-7.28 (m, 4\text{H, CH-Ar}), 7.20-7.14 (m, 3\text{H, CH-Ar}), 7.07 (t, \(J = 7.3\text{Hz}, 1\text{H, CH-Ar}\), 5.74 (s, 1\text{H, C=CH}_2), 5.11 (s, 1\text{H, C=CH}_2), 4.84 (br s, 1\text{H, NCH}_2), 4.07 (br s, 1\text{H, NCH}_2), 3.83 (s, 3\text{H, OCH}_3).

\(^13\)C NMR (\(\delta, \text{ppm}\))
- (CDCl\(_3\), 100 MHz)
  - 158.0 (Cq, C=O), 146.6 (Cq), 140.2 (Cq), 134.8 (Cq), 129.0 (CH-Ar), 128.3 (CH-Ar), 125.3 (Cq), 124.8 (CH-Ar), 124.0 (CH-Ar), 123.1 (CH-Ar), 122.9 (CH-Ar), 118.1 (CH-Ar), 108.3 (C=CH\(_2\)), 53.6 (OCH\(_3\)), 47.3 (NCH\(_2\)).

IR (\(\nu, \text{cm}^{-1}\)) (CCl\(_4\))
- 3069 (w), 3025 (w), 2945 (w), 2848 (w), 1740 (s), 1644 (w), 1590 (w), 1572 (w), 1490 (s), 1400 (s), 1384 (m), 1363 (s), 1350 (s), 1245 (s), 1200 (m), 1130 (m), 1100 (w).

HRMS (EI+, m/z): Calculated: 280.1212 Found: 280.1215.

---

4-Methylene-1-pentyl-3,4-dihydro-1\(H\)-cinnoline-2-carboxylic acid methyl ester 6.12b

\[
\begin{align*}
&\text{MF: } C_{16}H_{22}O_2N_2 \\
&MW = 274 \text{ g.mol}^{-1}
\end{align*}
\]

Method: See general procedure 6.6 using (1 equiv., 0.1 mmol, 27 mg) of \(N\)-pentyl-\(N\)-phenyl-\(N\)-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt).

Product: Yellow oil.

Isolated yield: 100%.
\textbf{\( ^1\text{H NMR} \) (\( \delta, \text{ppm} \))}

\((\text{CDCl}_3, 400 \text{ MHz})\)

7.58 (dd, \( J = 1.1\text{Hz}, 1\text{H, CH-Ar} \)), 7.37-7.18 (m, 1H, CH-Ar), 6.92-6.82 (m, 2H, CH-Ar), 5.56 (s, 1H, C=CH\(_2\)), 4.98 (s, 1H, C=CH\(_2\)), 4.69 (br s, 1H, NCH\(_2\)), 3.80 (br s, 1H, NCH\(_2\)), 3.70 (s, 3H, OCH\(_3\)), 3.44 (br s, 1H, NCH\(_2\)), 3.32 (br s, 1H, NCH\(_2\)), 1.68-1.58 (m, 2H, CH\(_2\)-pentyl chain), 1.38-1.36 (m, 4H, CH\(_2\)-pentyl chain), 0.94-0.91 (m, 3H, CH\(_3\)-pentyl chain).

\textbf{\( ^{13}\text{C NMR} \) (\( \delta, \text{ppm} \))}

\((\text{CDCl}_3, 100 \text{ MHz})\)

158.0 (Cq, C=O), 145.1 (Cq), 135.3 (Cq), 129.0 (CH-Ar), 124.5 (CH-Ar), 121.9 (Cq), 120.6 (CH-Ar), 117.5 (CH-Ar), 107.1 (C=CH\(_2\)), 55.6 (NCH\(_2\)), 53.2 (OCH\(_3\)), 45.3 (NCH\(_2\)), 29.3 (CH\(_2\)-pentyl chain), 27.3 (CH\(_2\)-pentyl chain), 22.6 (CH\(_2\)-pentyl chain).

\textbf{IR} (\( \nu, \text{cm}^{-1} \)) (\( \text{CCl}_4 \))

3072 (w), 2957 (m), 2931 (m), 2873 (w), 2860 (w), 1730 (s), 1636 (w), 1601 (w), 1570 (w), 1499 (w), 1483 (m), 1378 (m), 1268 (m), 1245 (m), 1206 (m), 1136 (w), 1122 (w), 1096 (w), 998 (w).

\textbf{HRMS} (EI+, m/z) : 

Calculated: 274.1681  
Found: 274.1690.

\begin{center}
\begin{tabular}{|c|c|}
\hline
1-Benzyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl & 6.12c \\\nester &  \\
\hline
\end{tabular}
\end{center}

\begin{center}
\includegraphics[width=0.5\textwidth]{structure.png}
\end{center}

MF: C\(_{18}\)H\(_{18}\)N\(_2\)O\(_2\)

MW = 294 g.mol\(^{-1}\)

\textbf{Method :}

See \textit{general procedure 6.6} using (1 equiv., 0.1 mmol, 29 mg) of \(N\)-benzyl-\(N\)-phenyl-\(N\)-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.

\textbf{Purification :}

Flash column chromatography  (silica gel, 9:1 PE:AcOEt).

\textbf{Product :}

Yellow oil.

\textbf{Isolated yield :}

87%.

\textbf{\( ^1\text{H NMR} \) (\( \delta, \text{ppm} \))}

\((\text{CDCl}_3, 400 \text{ MHz})\)

7.59 (d, \( J = 7.7\text{Hz}, 1\text{H, CH-Ph} \)), 7.37-7.21 (m, 6H, CH-Ph), 7.02 (d, \( J = 7.7\text{Hz}, 1\text{H, CH-Ph} \)), 6.91 (t, \( J = 7.7\text{Hz}, 1\text{H, CH-Ph} \)), 5.54 (s, 1H, C=CH\(_2\)), 4.92 (s, 1H, C=CH\(_2\)), 4.74 (br s, 1H, NCH\(_2\)), 4.56 (br s, 2H, NCH\(_2\)), 3.64 (s, 3H, OCH\(_3\)), 3.51 (br s, 1H, NCH\(_2\)).

\textbf{\( ^{13}\text{C NMR} \) (\( \delta, \text{ppm} \))}

\((\text{CDCl}_3, 100 \text{ MHz})\)

157.6 (Cq, C=O), 144.2 (Cq, Ar), 136.9 (Cq, Ar), 135.2 (Cq, Ar), 129.0 (CH-Ph), 128.7 (CH-Ph), 128.3 (CH-Ph), 127.6 (CH-Ph), 124.7 (CH-Ph), 122.0 (Cq, C=CH\(_2\)), 120.7 (CH-Ph), 117.0 (CH-Ph), 107.2 (C=CH\(_2\)), 59.2 (NCH\(_2\)), 53.1 (OCH\(_3\)), 46.4 (NCH\(_2\)).
**IR** (ν, cm⁻¹) (CCl₄) 3067 (w), 3033 (w), 2955 (w), 2929 (w), 2855 (w), 1716 (s), 1637 (w), 1603 (w), 1569 (w), 1542 (w), 1484 (m), 1454 (s), 1377 (m), 1356 (m), 1329 (m), 1314 (m), 1246 (s), 1206 (s), 1164 (w), 1140 (w), 1119 (w), 1098 (w), 1080 (w), 1063 (w), 1044 (w), 1029 (w).

**HRMS** (EI+, m/z) : Calculated: 294.1368  Found: 294.1379.

<table>
<thead>
<tr>
<th>1-Isopropyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester</th>
<th>6.12d</th>
</tr>
</thead>
<tbody>
<tr>
<td><img src="image" alt="Chemical Structure" /></td>
<td>MF: C₂₆H₂₈N₂O₂</td>
</tr>
<tr>
<td>MW = 246 g·mol⁻¹</td>
<td></td>
</tr>
</tbody>
</table>

**Method** : See general procedure 6.6 (1 equiv., 0.1 mmol, 25 mg) of N-isopropyl-N-phenyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.

**Purification** : Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product** : Yellow pale oil.

**Isolated yield** : 96%.

**¹H NMR** (δ, ppm) (CDCl₃, 400 MHz)

<table>
<thead>
<tr>
<th>Chemical Shift (ppm)</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>7.58 (dd, J = 8.0 Hz, 1H, CH-Ph)</td>
<td>1H, CH-Ph</td>
</tr>
<tr>
<td>7.20 (ddd, J = 8.0Hz, 1H, CH-Ph)</td>
<td>6.99 (d, J = 8.0Hz, 1H, CH-Ph)</td>
</tr>
<tr>
<td>6.87 (t, J = 8.0Hz, 1H, CH-Ph)</td>
<td>5.54 (s, 1H, C=CH₂)</td>
</tr>
<tr>
<td>4.80 (br d, J = 15.0Hz, 1H, NCH₂)</td>
<td>4.09 (hept, J = 6.6Hz, 1H, CH-Pr)</td>
</tr>
<tr>
<td>3.75 (d, J = 15.0Hz, 1H, NCH₂)</td>
<td>3.69 (s, 3H, OCH₃)</td>
</tr>
<tr>
<td>1.32 (d, J = 6.6Hz, 3H, CH₃-Pr)</td>
<td>1.15 (d, J = 6.6Hz, 3H, CH₃-Pr)</td>
</tr>
</tbody>
</table>

**¹³C NMR** (δ, ppm) (CDCl₃, 100 MHz)

<table>
<thead>
<tr>
<th>Chemical Shift (ppm)</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>159.0 (Cq, C=O)</td>
<td>143.9 (Cq)</td>
</tr>
<tr>
<td>135.4 (Cq)</td>
<td>128.9 (CH-Ph)</td>
</tr>
<tr>
<td>124.6 (CH-Ph)</td>
<td>122.4 (Cq)</td>
</tr>
<tr>
<td>120.4 (CH-Ph)</td>
<td>118.0 (CH-Ph)</td>
</tr>
<tr>
<td>107.1 (C=CH₂)</td>
<td>56.0 (CH-Pr)</td>
</tr>
<tr>
<td>53.2 (OCH₃)</td>
<td>48.3 (NCH₂)</td>
</tr>
<tr>
<td>20.7 (CH₃-Pr)</td>
<td>19.8 (CH₃-Pr)</td>
</tr>
</tbody>
</table>

**IR** (ν, cm⁻¹) (CCl₄) 3072 (w), 2976 (m), 2955 (m), 1712 (s), 1636 (w), 1604 (w), 1568 (w), 1481 (m), 1456 (m), 1446 (s), 1384 (m), 1362 (m), 1313 (w), 1300 (w), 1280 (w), 1238 (m), 1204 (m), 1172 (w), 1132 (m), 1091 (w), 1043 (w).

**HRMS** (EI+, m/z) : Calculated: 246.1368  Found: 246.1374.
6-Methoxy-1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester

```
MeO

<table>
<thead>
<tr>
<th>MF: C_{13}H_{16}N_{2}O_{3}</th>
</tr>
</thead>
<tbody>
<tr>
<td>MW = 248 g.mol^{-1}</td>
</tr>
</tbody>
</table>
```

**Method:**
See general procedure 6.6 using (1 equiv., 0.11 mmol, 27 mg) of \(N\)-(4-methoxy-phenyl)-\(N\)-methyl-\(N\)-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.

**Purification:**
Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product:**
Yellow solid.

**Isolated yield:**
83 %.

\[ ^1\text{H} \text{NMR} \] (\(\delta\), ppm)
(CDCl\(_3\), 400 MHz)
7.08 (d, \(J = 2.8\)Hz, 1H, CH-Ar), 6.91 (d, \(J = 8.9\)Hz, 1H, CH-Ar), 6.83 (dd, \(J = 2.8\)Hz, \(J = 8.9\)Hz, 1H, CH-Ar), 5.55 (s, 1H, C=CH\(_2\)), 5.03 (s, 1H, C=CH\(_2\)), 4.61 (br s, 1H, NCH\(_2\)), 3.98 (br s, 1H, NCH\(_2\)), 3.78 (s, 3H, OCH\(_3\)), 3.72 (s, 3H, OCH\(_3\)), 3.10 (s, 3H, NCH\(_3\)).

\[ ^{13}\text{C} \text{NMR} \] (\(\delta\), ppm)
(CDCl\(_3\), 100 MHz)
156.7 (Cq, C=O), 154.4 (Cq, Ar), 139.8 (Cq, Ar), 135.2 (Cq, Ar), 123.3 (Cq, Ar), 120.0 (CH-Ar), 116.1 (CH-Ar), 108.3 (CH-Ar), 107.8 (C=C=CH\(_2\)), 55.6 (OCH\(_3\)), 55.5 (OCH\(_3\)), 53.2 (OCH\(_3\)), 43.4 (NCH\(_3\)).

**IR** (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))
3001 (w), 2955 (w), 1711 (s), 1613 (w), 1571 (w), 1558 (w), 1542 (w), 1493 (s), 1454 (s), 1388 (w), 1295 (w), 1234 (s), 1118 (m), 1048 (w).

**HRMS** (EI+, m/z):
Calculated: 248.1161 Found: 248.1162.

---

6-Chloro-1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester

```
Cl

<table>
<thead>
<tr>
<th>MF: C_{12}H_{13}N_{2}O_{3}Cl</th>
</tr>
</thead>
<tbody>
<tr>
<td>MW = 252 g.mol^{-1}</td>
</tr>
</tbody>
</table>
```

---

396
Method: See general procedure 6.6 using (1 equiv., 0.08 mmol, 20 mg) of \( N'-(4\text{-chlorophenyl})-N'\text{-methyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.} \)

Purification: Flash column chromatography (silica gel, 7:3 PE:AcOEt).

Product: White solid.

Isolated yield: 100%.

\(^1\text{H NMR}\) (\(\delta, \text{ppm}\))

\[\text{(CDCl}_3, 400 \text{ MHz)}\]

7.52 (d, \(J = 2.1\text{Hz, 1H, CH-Ar}\)), 7.16 (dd, \(J = 2.1\text{Hz, } J = 8.8\text{Hz, 1H, CH-Ar}\)), 6.81 (d, \(J = 8.8\text{Hz, 1H, CH-Ar}\)), 5.55 (s, 1H, C=CH\(_2\)), 5.03 (s, 1H, C=CH\(_2\)), 4.63 (br s, 1H, NCH\(_2\)), 3.78 (br s, 1H, NCH\(_2\)), 3.72 (s, 3H, OCH\(_3\)), 3.18 (NCH\(_3\)).

\(^{13}\text{C NMR}\) (\(\delta, \text{ppm}\))

\[\text{(CDCl}_3, 100 \text{ MHz)}\]

157.0 (Cq, C=O), 143.5 (Cq), 134.3 (Cq), 128.9 (CH-Ar), 125.8 (Cq), 124.3 (CH-Ar), 123.0 (Cq), 117.9 (CH-Ar), 108.4 (C=CH\(_2\)), 53.4 (OCH\(_3\)), 45.5 (NCH\(_3\)), 42.2 (NCH\(_3\)).

IR (\(\nu, \text{cm}^{-1}\)) (CCl\(_4\)) 2956 (w), 1720 (s), 1484 (s), 1448 (s), 1375 (s), 1308 (w), 1244 (m), 1212 (s), 1150 (w), 1119 (w).

HRMS (EI+, m/z): Calculated: 252.0666 Found: 252.0667.

---

6-Fluoro-1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester

\[
\begin{array}{c}
\text{F} \\
\text{N} \\
\text{N} \\
\text{O} \\
\text{O} \\
\text{C} \\
\text{H}_13 \\
\text{O}_2 \\
\text{N}_2 \\
\text{F} \\
\text{MF: C}_{12}\text{H}_{13}\text{O}_2\text{N}_2\text{F} \\
\text{MW = 236 g.mol}^{-1}
\end{array}
\]

Method: See general procedure 6.6 using (1 equiv., 0.1 mmol, 24 mg) of \( N'-(4\text{-fluorophenyl})-N'\text{-methyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.} \)

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt).

Product: Yellow pale solid.

Isolated yield: 79%.

\(^1\text{H NMR}\) (\(\delta, \text{ppm}\))

7.25 (dd, \(J_{H-H} = 2.6\text{Hz, } J_{H-F} = 9.0\text{Hz, 1H, CH-Ar}\)), 6.94 (dd, \(J_{H-H} = 2.6\text{Hz, } J_{H-F} = 7.8\text{Hz, } J_{H-H} = 9.0\text{Hz, 1H, CH-Ar}\)), 6.88 (dd, \(J_{H-F} = 5.0\text{Hz, } J_{H-H} = 9.0\text{Hz, 1H, CH-Ar}\)), 5.54 (s, 1H, C=CH\(_2\)), 5.06 (s, 1H, C=CH\(_2\)), 4.64 (br s, 1H, NCH\(_2\)), 3.88 (br
(CDCl$_3$, 400 MHz) s, 1H, NCH$_2$), 3.73 (s, 3H, OCH$_3$), 3.14 (s, 3H, NCH$_3$).

$^{13}$C NMR (δ, ppm) 157.6 (d, $J_{C-F}$ = 239.7Hz, Cq, C-F), 156.8 (Cq, C=O), 141.71 (d, $J$ = 1.7Hz, Cq, Ar), 134.6 (Cq), 123.5 (Cq), 119.32 (d, $J_{C-F}$ = 8.0Hz, CH-Ar), 116.20 (d, $J_{C-F}$ = 23.2Hz, CH-Ar), 108.6 (C=C$_{H_2}$), 53.3 (OCH$_3$), 47.7 (NCH$_2$), 43.0 (NCH$_3$).

IR ($\nu$, cm$^{-1}$) (CCl$_4$) 2956 (w), 1724 (s), 1635 (w), 1581 (w), 1509 (w), 1491 (s), 1452 (s), 1383 (w), 1292 (w), 1243 (m), 1212 (s), 1133 (w), 1114 (w), 1080 (w).

HRMS (EI+, m/z) : Calculated: 236.0961   Found: 236.0962.

1-Methyl-4-methylene-3,4-dihydro-1H-cinnoline-2,6-dicarboxylic acid 6-ethyl ester 2-methyl ester

Method : See general procedure 6.6 using (1 equiv., 0.1 mmol, 29 mg) of 4-($N$-methoxycarbonyl-$N$-methyl-$N'$-prop-2-ynyl-hydrazino)-benzoic acid ethyl ester

Purification : Flash column chromatography (silica gel, 7:3 PE:AcOEt).

Product : Yellow oil.

Isolated yield : 86% (with ~17% inseparable impurity based on $^1$H NMR).

$^1$H NMR (δ, ppm) 8.24 (d, $J$ = 1.8Hz, 1H, CH-Ar), 7.87 (dd, $J$ = 1.8Hz, $J$ = 8.7Hz, 1H, CH-Ar), 6.76 (d, $J$ = 8.7Hz, 1H, CH-Ar), 5.66 (s, 1H, C=CH$_2$), 5.00 (s, 1H, C=CH$_2$), 4.69 (br s, 1H, NCH$_2$), 4.34 (q, $J$ = 7.1Hz, 2H, OCH$_2$CH$_3$), 3.80 (br s, 1H, NCH$_2$), 3.72 (s, 3H, OCH$_3$), 3.30 (s, 3H, NCH$_3$), 1.37 (t, $J$=7.1Hz, 3H, OCH$_2$CH$_3$).

$^{13}$C NMR (δ, ppm) 166.5 (Cq, C=O), 157.5 (Cq, C=O), 147.4 (Cq), 130.4 (CH-Ar), 126.7 (CH-Ar), 121.0 (Cq), 119.9 (Cq), 113.4 (CH-Ph), 110.8 (Cq), 108.1 (C=C$_{H_2}$), 60.6 (OCH$_2$CH$_3$), 53.5 (OCH$_3$), 41.0 (NCH$_2$), 38.1 (NCH$_3$), 14.4 (OCH$_2$CH$_3$).

IR ($\nu$, cm$^{-1}$) (CCl$_4$) 2982 (w), 2957 (w), 1715 (s), 1608 (m), 1559 (w), 1499 (w), 1446 (m), 1367 (w), 1308 (m), 1276 (m), 1258 (s), 1216 (s), 1175 (w), 1154 (m), 1110 (m), 1025 (w).

HRMS (EI+, m/z) : Calculated: 290.1267 Found: 290.1265.
6-Cyano-1-methyl-4-methylene-3,4-dihydro-1\textit{H}-cinnoline-2-carboxylic acid methyl ester  

MF: C_{13}H_{13}N_{3}O_{2}  

MW = 243 g.mol\(^{-1}\)

Method: See general procedure 6.6 using (1 equiv., 0.1 mmol, 24 mg) of \textit{N}-(4-cyano-phenyl)-\textit{N}-methyl-\textit{N}-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 7:3 PE:AcOEt).

Product: Yellow solid.

Isolated yield: 63%.

\(^1\)H NMR (\(\delta\), ppm)  
(CDCl\(_3\), 400 MHz)  
7.79 (d, \(J = 1.8\)Hz, 1H, CH-Ar), 7.44 (dd, \(J = 1.8\)Hz, \(J = 8.6\)Hz, 1H, CH-Ar), 6.75 (d, \(J = 8.6\)Hz, 1H, CH-Ar), 5.57 (s, 1H, C=CH\(_2\)), 5.03 (s, 1H, C=CH\(_2\)), 4.65 (br s, 1H, NCH\(_2\)), 3.79 (br s, 1H, NCH\(_2\)), 3.73 (s, 3H, OCH\(_3\)), 3.31 (s, 3H, NCH\(_3\)).

\(^{13}\)C NMR (\(\delta\), ppm)  
(CDCl\(_3\), 100 MHz)  
157.5 (Cq, C=O), 146.6 (Cq), 132.4 (CH-Ar), 129.1 (CH-Ar), 120.5 (Cq), 119.6 (Cq), 113.5 (CH-Ar), 111.6 (Cq), 109.0 (C=CH\(_2\)), 101.5 (Cq, CN), 53.6 (OCH\(_3\)), 47.7 (NCH\(_2\)), 40.6 (NCH\(_3\)).

IR (\(\nu\), \text{cm}\(^{-1}\)) (CCl\(_4\))  
2957 (w), 2929 (w), 2226 (m, CN), 1732 (s, C=O), 1635 (w), 1608 (m), 1497 (m), 1446 (m), 1371 (w), 1325 (m), 1279 (w), 1243 (m), 1217 (m), 1194 (w), 1159 (w), 1117 (w), 1085 (w), 1064 (w).

HRMS (EI+, \text{m/z}) :  
Calculated: 243.1008  
Found: 243.1002.

---

1-Methyl-4-methylene-6-trifluoromethyl-3,4-dihydro-1\textit{H}-cinnoline-2-carboxylic acid methyl ester  

MF: C_{13}H_{13}O_{2}N_{2}F_{3}  

MW = 286 g.mol\(^{-1}\)
Method: See general procedure 6.6 using (1 equiv., 0.1 mmol, 29 mg) of N-methyl-N-prop-2-ynyl-N-(4-trifluoromethyl-phenyl)-hydrazinecarboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt).

Product: White solid.

Isolated yield: 76%.

$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz) 7.78 (d, $J_{\text{H-H}} = 1.6$ Hz, 1H, CH-Ar), 7.43 (dd, $J_{\text{H-H}} = 1.6$ Hz, $J_{\text{H-H}} = 8.6$ Hz, 1H, CH-Ar), 6.84 (d, $J_{\text{H-H}} = 8.6$ Hz, 1H, CH-Ar), 5.61 (s, 1H, C=CH$_2$), 5.04 (s, 1H, C=CH$_2$), 4.66 (br s, 1H, NCH$_2$), 3.80 (br s, 1H, NCH$_2$), 3.73 (s, 3H, OCH$_3$), 3.28 (s, 3H, NCH$_3$).

$^{13}$C NMR (δ, ppm) (CDCl$_3$, 100 MHz) 157.5 (Cq, C=O), 152.0 (q, $J_{\text{C-F}} = 21.0$ Hz, Cq, Ar), 146.5 (Cq), 134.4 (Cq), 125.8 (q, $J_{\text{C-F}} = 2.8$ Hz, CH-Ar), 122.0 (q, $J_{\text{C-F}} = 3.7$ Hz, CH-Ar), 120.5 (Cq), 114.2 (q, $J_{\text{C-F}} = 256$ Hz, Cq, CF$_3$), 114.5 (CH-Ar), 108.5 (C=CH$_2$), 53.5 (OCH$_3$), 46.7 (NCH$_2$), 41.3 (NCH$_3$).

IR (ν, cm$^{-1}$) (CCl$_4$) 3003 (w), 2957 (m), 2930 (m), 2856 (w), 1728 (s), 1622 (s), 1576 (w), 1542 (s), 1509 (m), 1445 (s), 1334 (s), 1284 (s), 1213 (s), 1171 (s), 1150 (s), 1126 (s), 1085 (s), 1008 (m).

HRMS (EI+, m/z): Calculated: 286.0934 Found: 286.0934.

1,5-Dimethyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester

MF: C$_{13}$H$_{16}$N$_2$O$_2$

MW = 232 g.mol$^{-1}$

Method: See general procedure 6.6 using (1 equiv., 0.1 mmol, 23 mg) of N-methyl-N-prop-2-ynyl-N-m-tolyl-hydrazinecarboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 9:1 PE:AcOEt).

Product: Transparent oil.

Isolated yield: 48%.
**1H NMR (δ, ppm)**

(CDCl₃, 400 MHz)

7.11 (t, J = 8.0Hz, 1H, CH-Ar), 6.78 (d, J = 8.0Hz, 2H, CH-Ar), 5.41 (s, 1H, C=CH₂), 5.31 (br s, 1H, C=CH₂), 4.52 (br s, 1H, NCH₂), 3.83 (br s, 1H, NCH₂), 3.74 (s, 3H, OCH₃), 3.19 (s, 3H, NCH₃), 2.47 (s, 3H, CH₃).

**13C NMR (δ, ppm)**

(CDCl₃, 100 MHz)

Mixture of regioisomers: 157.0 (Cq, C=O), 146.1 (Cq), 136.8 (Cq), 135.0 (Cq), 127.7 (CH-Ar), 124.0 (CH-Ar), 121.8 (Cq), 114.9 (C=CH₂), 114.5 (CH-Ar), 53.2 (OCH₃), 46.7 (NCH₂), 43.5 (NCH₃), 23.2 (CH₃).

**IR (κ, cm⁻¹) (CCl₄)**

3068 (w), 2956 (s), 2857 (s), 2810 (w), 1716 (s), 1633 (m), 1593 (m), 1574 (m), 1468 (s), 1445 (s), 1378 (s), 1330 (s), 1310 (s), 1260 (s), 1234 (s), 1210 (s), 1132 (s), 1083 (s), 1032 (w).

**HRMS (El+, m/z)**


---

7-Chloro-1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester and 5-Chloro-1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester

**Method:**

See general procedure 6.6 using (1 equiv., 0.1 mmol, 25 mg) of N-(3-chlorophenyl)-N-methyl-N-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.

**Purification:**

Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product:**

Yellow oil.

**Isolated yield:**

96% (mixture inseparable of regioisomers 1:1).
43.2 (NCH$_2$), 41.7 (NCH$_3$), 39.2 (NCH$_3$), 38.5 (NCH$_3$).

IR (ν, cm$^{-1}$) (CCl$_4$) 3001 (w), 2956 (m), 2928 (w), 2813 (w), 1721 (s), 1634 (w), 1595 (s), 1561 (m), 1486 (s), 1447 (s), 1374 (s), 1214 (s), 1156 (s), 1118 (s), 1103 (s), 1076 (s), 1062 (s), 1043 (m), 1003 (s), 963 (m), 827 (s), 758 (s), 691 (s), 630 (s).

HRMS (EI+, m/z)  Calculated: 252.0666  Found: 252.0653.

1,8-Dimethyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester and 1,9-Dimethyl-1,3-dihydro-benzo[c][1,2]diazepine-2-carboxylic acid methyl ester

6.12oa + 6.12ob

MF: C$_{13}$H$_{16}$N$_2$O$_2$

MW = 232 g.mol$^{-1}$

Method: See general procedure 6.6 using (1 equiv., 0.11 mmol, 25 mg) of N-methyl-N-prop-2-ynyl-N-α-tolyl-hydrazinecarboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 8:2 PE:AcOEt).

Product: Yellow oil.

Isolated yield: 81% (mixture inseparable of regioisomers, ratio 6.5:1).

$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz)

6-exo product: 7.51 (d, J = 7.6 Hz, 1H, CH-Ar), 7.10 (d, J = 7.6 Hz, 1H, CH-Ar), 7.03-6.97 (m, 1H, CH$_2$), 5.62 (s, 1H, C=CH$_2$), 5.07 (s, 1H, C=CH$_2$), 4.78 (br d, J = 14.8 Hz, 1H, NCH$_2$), 4.33 (br d, J = 14.8 Hz, 1H, NCH$_3$), 3.71 (s, 3H, OCH$_3$), 2.93 (s, 3H, NCH$_3$), 2.37 (s, 3H, CH$_3$).

7-endo-product: 7.08-7.07 (m, 1H, CH-Ar), 7.03-6.96 (m, 2H, CH-Ar), 6.39 (d, J = 12.4 Hz, 1H, CH$_=\$), 5.85 (ddd, J = 2.1 Hz, J = 4.8 Hz, J = 12.4 Hz, 1H, CH$_=\$), 5.00 (br d, J = 14.8 Hz, 1H, NCH$_3$), 4.26 (br d, J = 14.8 Hz, 1H, NCH$_3$), 3.65 (s, 3H, OCH$_3$), 2.93 (s, 3H, NCH$_3$), 2.49 (CH$_3$).

$^{13}$C NMR (δ, ppm) (CDCl$_3$, 100 MHz)

6-endo product: 156.3 (Cq, C=O), 145.4 (Cq), 135.4 (Cq), 131.0 (CH-Ar), 130.5 (Cq), 125.4 (Cq), 123.9 (CH-Ar), 121.8 (CH-Ar), 108.1 (C=CH$_2$), 53.1 (OCH$_3$), 42.2 (NCH$_2$), 41.7 (NCH$_3$), 17.9 (CH$_3$).

7-endo product: 157.0 (Cq, C=O), 147.0 (Cq, Ar), 133.9 (Cq, Ar), 130.4 (CH), 130.4 (CH), 128.9 (CH), 127.7 (CH), 125.0 (CH), 53.0 (OCH$_3$), 45.1 (NCH$_2$), 38.7 (NCH$_3$), 18.4 (CH$_3$).

IR (ν, cm$^{-1}$) (CCl$_4$) 2956 (w), 2824 (w), 2857 (w), 1707 (s), 1630 (w), 1593 (w), 1558 (m), 1542 (m), 1452 (s), 1390 (m), 1313 (w), 1279 (w), 1263 (w), 1250 (m), 1228 (w), 1197 (w), 1139 (w), 1110 (w), 1089 (w), 1035 (w).
**HRMS** (EI+, m/z) : Calculated: 232.1208  Found: 232.1208.

6-Methoxy-4-methylene-1-phenyl-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester and 1-(4-Methoxy-phenyl)-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester

\[
\begin{align*}
&\text{MF: } C_{18}H_{18}O_3N_2 \\
&\text{MW} = 310 \text{ g mol}^{-1}
\end{align*}
\]

**Method** : See general procedure 6.6 using (1 equiv., 0.1 mmol, 31 mg) of \(N\)-(4-methoxy-phenyl)-\(N\)-phenyl-\(N\)-prop-2ynyl-hydrazinecarboxylic acid methyl ester.

**Purification** : Flash column chromatography (silica gel, 85:15 PE:AcOEt).

**Product** : Dark green oil.

**Isolated yield** : 84% (inseparable mixture of regioisomers, ratio: 1:2).

**\(^1\)H NMR** (\(\delta\) ppm) (CDCl\(_3\), 400 MHz)

**Major regioisomer**:
- 7.72 (d, \(J = 7.7\) Hz, 1H, CH-Ar)
- 7.24-6.96 (m, 5H, CH-Ar)
- 6.82 (d, \(J = 9.0\) Hz, 2H, CH-Ar)
- 5.67 (s, 1H, C=CH\(_2\))
- 5.05 (s, 1H, C=CH\(_2\))
- 4.69 (br s, 2H, NCH\(_2\))
- 3.78 (s, 6H, OCH\(_3\))

**Minor regioisomer**:
- 7.24-6.96 (m, 8H, CH-Ar)
- 5.65 (s, 1H, C=CH\(_2\))
- 5.10 (s, 1H, C=CH\(_2\))
- 4.69 (br s, 2H, NCH\(_2\))
- 3.84 (s, 3H, OCH\(_3\))
- 3.77 (s, 3H, OCH\(_3\))

**\(^13\)C NMR** (\(\delta\) ppm) (CDCl\(_3\), 100 MHz)

**Major and minor regioisomers**:
- 157.9 (Cq)
- 157.8 (Cq)
- 156.4 (Cq)
- 156.3 (Cq)
- 147.0 (Cq)
- 141.2 (Cq)
- 140.6 (Cq)
- 137.7 (Cq)
- 134.8 (Cq)
- 134.7 (Cq)
- 133.7 (Cq)
- 132.8 (Cq)
- 129.0 (CH-Ar)
- 128.4 (CH-Ar)
- 125.0 (CH-Ar)
- 124.6 (CH-Ar)
- 123.1 (CH-Ar)
- 122.3 (CH-Ar)
- 122.2 (CH-Ar)
- 121.9 (CH-Ar)
- 117.2 (CH-Ar)
- 115.1 (CH-Ar)
- 114.2 (CH-Ar)
- 109.2 (C=CH\(_2\))
- 108.7 (CH-Ar)
- 108.1 (C=CH\(_2\))
- 55.5 (OCH\(_3\))
- 55.4 (OCH\(_3\))
- 53.5 (OCH\(_3\))
- 53.6 (OCH\(_3\))
- 46.6 (NCH\(_2\))
- 46.3 (NCH\(_2\))

**IR** (\(\nu\), cm\(^{-1}\)) (CCl\(_4\))

- 3072 (w)
- 3001 (m)
- 2955 (m)
- 2934 (m)
- 2910 (m)
- 2836 (m)
- 1717 (s)
- 1632 (w)
- 1598 (m)
- 1571 (w)
- 1558 (s)
- 1555 (s)
- 1551 (s)
- 1547 (s)
- 1543 (s)
- 1508 (s)
- 1492 (s)
- 1450 (s)
- 1378 (s)
- 1297 (s)
- 1246 (s)
- 1181 (s)
- 1131 (s)
- 1041 (s)

**HRMS** (EI+, m/z) : Calculated: 310.1318  Found: 310.1309.
4-Methylene-1-phenyl-3,4-dihydro-1H-cinnoline-2,6-dicarboxylic acid 6-ethyl ester 2-methyl ester and 1-(4-Ethoxycarbonyl-phenyl)-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester

\[
\text{MF: C}_{20}\text{H}_{20}\text{N}_{2}\text{O}_{4}
\]

\[
\text{MW = 352 g.mol}^{-1}
\]

\[
\text{EtO}_2\text{C} + \text{CO}_2\text{Et} \rightarrow \text{C}_20\text{H}_{20}\text{N}_2\text{O}_4
\]

**Method:** See general procedure 6.6 using (1 equiv., 0.1 mmol, 35 mg) of 4-(N-methoxycarbonyl-N-phenyl-N-prop-2-ynyl-hydrazino)benzoic acid ethyl ester.

**Purification:** Flash column chromatography (silica gel, 8:2 PE: AcOEt).

**Product:** Yellow pale oil.

**Isolated yield:** 97% (inseparable mixture of regioisomers, ratio 1:1.5).

**\(^1\text{H NMR (\(\delta\), ppm)}\)**

**Minor product:** 8.43 (d, \(J = 1.8\text{Hz}, 1\text{H, CH-}\text{Ar}\)), 7.85 (dd, \(J = 1.8\text{Hz}, J = 8.6\text{Hz}, 1\text{H, CH-}\text{Ar}\)), 7.32-7.08 (m, 6H, CH-\text{Ar}), 5.81 (s, 1H, C=CH\_2), 5.13 (s, 1H, C=CH\_2), 4.87 (br s, 1H, NCH\_2), 4.38 (q, \(J = 7.2\text{Hz}, 2\text{H, OCH}_2\text{CH}_3\)), 4.37 (br s, 1H, NCH\_2), 3.76 (s, 3H, OCH\_3), 1.39 (t, \(J = 7.2\text{Hz}, 3\text{H, OCH}_2\text{C}_\text{H}_3\)).

**Major product:** 7.93 (d, \(J = 9.0\text{Hz}, 2\text{H, CH-}\text{Ar}\)), 7.76 (d, \(J = 9.9\text{Hz}, 1\text{H, CH-}\text{Ar}\)), 7.32-7.08 (m, 5H, CH-\text{Ar}), 5.70 (s, 1H, C=CH\_2), 5.08 (s, 1H, C=CH\_2), 4.34 (q, \(J = 7.2\text{Hz}, 2\text{H, OCH}_2\text{CH}_3\)), 4.06 (br d, \(J = 14.9\text{Hz}, 1\text{H, NCH}_2\)), 3.76 (s, 3H, OCH\_3), 3.73 (br s, 1H, NCH\_2), 1.37 (t, \(J = 7.1\text{Hz}, 3\text{H, OCH}_2\text{C}_\text{H}_3\)).

**\(^{13}\text{C NMR (\(\delta\), ppm)}\)**

**Major and minor products:** 166.2 (Cq, C=O), 166.0 (Cq, C=O), 158.1 (Cq, C=O), 157.8 (Cq, C=O), 149.6 (Cq), 145.5 (Cq), 143.9 (Cq), 138.9 (Cq), 134.3 (Cq), 133.9 (Cq), 130.9 (CH-\text{Ar}), 129.2 (CH-\text{Ar}), 129.1 (CH-\text{Ar}), 128.4 (CH-\text{Ar}), 126.8 (CH-\text{Ar}), 126.2 (Cq), 126.0 (Cq), 125.0 (CH-\text{Ar}), 124.1 (CH-\text{Ar}), 123.9 (CH-\text{Ar}), 123.2 (CH-\text{Ar}), 121.1 (CH-\text{Ar}), 119.8 (CH-\text{Ar}), 115.5 (CH-\text{Ar}), 113.2 (Cq), 112.7 (Cq), 109.4 (C=CH\_2), 109.0 (C=CH\_2), 60.9 (OCH\_2CH\_3), 60.6 (OCH\_2CH\_3), 53.8 (OCH\_3), 53.7 (OCH\_3), 48.5 (NCH\_2), 47.9 (NCH\_2), 14.4 (OCH\_2CH\_3), 14.3 (OCH\_2CH\_3).

**IR (\(\nu\), cm\(^{-1}\)) (CCl\_4):** 3070 (w), 3031 (w), 2982 (m), 2957 (m), 2932 (m), 2873 (w), 2855 (w), 1717 (s), 1631 (m), 1599 (s), 1509 (s), 1494 (s), 1482 (s), 1448 (s), 1367 (s), 1311 (s), 1271 (s), 1215 (s), 1175 (s), 1109 (s), 1024 (s).

**HRMS (EI+, m/z):** Calculated: 352.1423 Found: 352.1438.
6-Fluoro-4-methylene-1-phenyl-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester and 1-(4-Fluoro-phenyl)-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester

\[
\text{MF: C}_{17}\text{H}_{15}\text{O}_2\text{N}_2\text{F} \\
\text{MW} = 298 \text{ g.mol}^{-1}
\]

Method: See general procedure 6.6 using (1 equiv., 0.1 mmol, 30 mg) of \(N\)-(4-fluoro-phenyl)-\(N\)-phenyl-\(N\)-prop-2-ynyl-hydrazinecarboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 9:1 PE:AcOEt).

Product: White solid.

Isolated yield: 100% (inseparable mixture of regioisomers, ratio 1: 6.5).

\(^1\text{H NMR (\delta, ppm)}\) Only major regioisomer described: 7.79 (d, \(J_{HH} = 7.9\text{Hz, 1H, CH-Ar}\), 7.27 (t, \(J_{HH} = 8.4\text{Hz, 1H, CH-Ar}\), 7.19-7.11 (m, 4H, CH-Ar), 7.00 (dd, \(J_{HF} = 8.3\text{Hz, 1H, CH-Ar}\), 6.90 Hz, 1H, CH-Ar), 5.74 (s, 1H, C=CH\(\text{CH}_3\)), 5.11 (s, 1H, C=CH\(\text{CH}_3\)), 4.78 (br s, 1H, NCH\(_2\)), 4.05 (br s, 1H, NCH\(_2\)), 3.83 (s, 3H, OCH\(_3\)).

\(^{13}\text{C NMR (\delta, ppm)}\) Only major regioisomer described: 159.1 (d, \(J_{CF} = 242.4\text{Hz, Cq, C-F}\)), 157.9 (Cq, C=O), 143.1 (Cq), 143.0 (Cq), 140.3 (Cq), 134.5 (Cq), 129.1 (CH-Ar), 128.5 (CH-Ar), 124.8 (CH-Ar), 123.9 (CH-Ar), 122.5 (CH-Ar), 115.6 (d, \(J_{CF} = 22.6\text{Hz, CH-Ar}\)), 108.5 (C=CH\(\text{CH}_3\)), 53.6 (OCH\(_3\)), 46.7 (NCH\(_2\)).

\(\text{IR (\nu, \text{cm}^{-1}) (CCl}_4\) 3074 (w), 3033 (w), 3000 (w), 2957 (m), 2930 (m), 2856 (w), 1721 (s, C=O), 1633 (m), 1599 (m), 1572 (m), 1505 (s), 1484 (s), 1449 (s), 1376 (s), 1328 (s), 1308 (s), 1265 (s), 1231 (s), 1158 (s), 1130 (s), 1106 (m), 1036 (w).

HRMS (El+, m/z): Calculated: 298.1118  Found: 298.1108.
B.3.5.3 Synthesis of cinnoline derivatives via Au(I) catalysis

The external double bond from the hydroarylated products can be isomerized to the internal position to produce substituted cinnoline derivatives (Scheme B.3.5.3).

Scheme B.3.5.3: Isomerization to dihydrocinnolines 6.13a-k.

General Procedure 6.7, Isomerization of the double bond: To a reaction vessel (a NMR tube in our case) was added the hydroarylated derivative (1 equiv.), PTSA.H₂O (0.05 equiv.) and CDCl₃ (0.2 M). The reaction was heated up to reflux (60 °C). Upon completion (NMR), the reaction mixture was diluted in DCM, transferred to a separating funnel, washed with a saturated solution of NaHCO₃, extracted with DCM (3x), dried (MgSO₄), concentrated under reduced pressure and purified by flash column chromatography to afford the desired isomerized products in the stated yields.

4-Methyl-1-phenyl-1H-cinnoline-2-carboxylic acid methyl ester 6.13a

MF: C₁₇H₁₅O₂N₂

MW = 280 g.mol⁻¹
Method: See general procedure 6.7 using (1 equiv., 0.1 mmol, 28 mg) of 4-methylene-1-phenyl-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 95:5 PE:AcOEt).

Product: White solid.

Isolated yield: 76%.

$^1$H NMR ($\delta$, ppm) (CDCl$_3$, 400 MHz) 7.40-7.30 (m, 4H, Ar), 7.19 (dd, $J = 7.4$Hz, $J = 8.4$Hz, 2H, CH-Ar), 6.98 (t, $J = 7.4$Hz, 1H, CH-Ar), 6.89 (d, $J = 8.4$Hz, 2H, CH-Ar), 6.77 (br s, 1H, =CHN), 3.88 (s, 3H, OCH$_3$), 2.04 (d, $J = 1.4$Hz, 1H, CH$_3$).

$^{13}$C NMR ($\delta$, ppm) (CDCl$_3$, 100 MHz) 155.0 (Cq, C=O), 148.8 (Cq), 138.9 (Cq), 128.6 (CH), 127.7 (CH), 126.4 (CH), 125.2 (CH), 124.0 (Cq), 123.3 (CH), 122.6 (CH), 120.1 (Cq), 116.2 (CH), 114.2 (CH), 53.6 (OCH$_3$), 14.7 (CH$_3$).

IR (v, cm$^{-1}$) (CCl$_4$) 3072 (w), 3030 (w), 2955 (w), 2923 (w), 2859 (w), 1747 (m), 1718 (s), 1646 (w), 1594 (w), 1568 (w), 1492 (s), 1450 (s), 1443 (s), 1386 (m), 1361 (s), 1351 (s), 1242 (s), 1197 (m), 1177 (w), 1159 (w), 1138 (m), 1110 (w), 1073 (w), 1032 (w).

HRMS (EI+, m/z): Calculated: 280.1212 Found: 280.1216.

---

4-Methyl-1-pentyl-1H-cinnoline-2-carboxylic acid methyl ester 6.13b

MF: C$_{16}$H$_{22}$O$_2$N$_2$

MW = 274 g.mol$^{-1}$

Method: See general procedure 6.7 using (1 equiv., 0.1 mmol, 27 mg) of 4-methylene-1-pentyl-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 7:3 PE:AcOEt).

Product: Yellow oil.

Isolated yield: 91%.
$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz) 7.25-7.21 (m, 2H, CH-Ar), 7.19-7.09 (m, 2H, CH-Ar), 6.62 (br s, 1H, NCH=), 3.80 (s, 3H, OCH$_3$), 3.09 (br s, 1H, NCH$_2$), 2.82 (br s, 1H, NCH$_2$), 2.07 (d, $J = 1.2$Hz, 3H, CH$_3$-pentyl chain), 1.38-1.26 (m, 6H, CH$_2$-pentyl chain), 0.88 (t, $J = 6.9$Hz, 3H, CH$_3$-pentyl chain).

$^{13}$C NMR (δ, ppm) (CDCl$_3$, 100 MHz) 154.7 (Cq, C=O), 144.2 (Cq), 128.0 (CH), 127.2 (Cq), 125.3 (CH), 122.7 (CH), 122.4 (CH), 121.2 (CH), 118.2 (Cq), 58.2 (NCH$_2$-pentyl chain), 53.1 (OCH$_3$), 29.2 (CH$_2$-pentyl chain), 26.8 (CH$_2$-pentyl chain), 22.5 (CH$_2$-pentyl chain), 14.8 (CH$_3$), 14.0 (CH$_3$).

IR (ν, cm$^{-1}$) (CCl$_4$) 3032 (w), 2957 (m), 2929 (m), 2858 (w), 1744 (s), 1711 (s), 1649 (w), 1600 (w), 1499 (w), 1484 (w), 1455 (s), 1443 (s), 1387 (w), 1363 (s), 1334 (m), 1263 (s), 1245 (m), 1194 (w), 1135 (w), 1109 (w), 1016 (w).

HRMS (EI+, m/z) : Calculated: 274.1681  Found: 274.1684.

**1-Benzyl-4-methyl-1$H$-cinnoline-2-carboxylic acid methyl ester**

![Chemical structure](image)

MF: C$_{18}$H$_{18}$N$_2$O$_2$

MW = 294 g.mol$^{-1}$

**Method :** See general procedure 6.7 using (1 equiv., 0.08 mmol, 23 mg) of 1-benzyl-4-methylene-3,4-dihydro-1$H$-cinnoline-2-carboxylic acid methyl ester.

**Purification :** Flash column chromatography (silica gel, 9:1 PE:AcOEt).

**Product :** Transparent oil.

**Isolated yield :** 81%.

$^1$H NMR (δ, ppm) (CDCl$_3$, 400 MHz) 7.27-7.22 (m, 6H, CH-Ph), 7.16-7.14 (m, 2H, CH-Ph), 6.96-6.94 (m, 1H, CH-Ph), 6.66 (br s, 1H, NCH=), 4.14 (br s, 1H, NCH$_2$), 3.99 (br s, 1H, NCH$_3$), 3.59 (br s, 3H, OCH$_3$), 2.01 (s, 3H, CH$_3$).

$^{13}$C NMR (δ, ppm) (CDCl$_3$, 100 MHz) 154.4 (Cq, C=O), 143.9 (Cq), 135.9 (Cq), 130.2 (CH-Ar), 128.4 (Cq), 127.9 (CH-Ar), 127.8 (CH-Ar), 127.7 (CH-Ar), 125.7 (CH -Ar), 123.0 (CH -Ar), 122.4 (CH-Ar), 121.6 (NCH=), 117.8 (Cq), 62.3 (NCH$_3$), 53.0 (OCH$_3$), 14.8 (CH$_3$).

IR (ν, cm$^{-1}$) (CCl$_4$) 3068 (w), 3033 (w), 2954 (w), 2923 (w), 2856 (w), 1712 (s), 1650 (w), 1603 (w), 1496 (w), 1484 (w), 1455 (s), 1442 (s), 1387 (s), 1365 (s), 1351 (s), 1258 (m), 1246 (m), 1195 (m), 1158 (w), 1137 (m), 1114 (w), 1080 (w), 1019 (w).
HRMS (EI+, m/z) : Calculated: 294.1368 Found: 294.1378.

1-Isopropyl-4-methyl-1H-cinnoline-2-carboxylic acid methyl ester  6.13d

\[
\begin{array}{c}
\text{MF: C}_{14}\text{H}_{18}\text{N}_{2}\text{O}_{2} \\
\text{MW} = 246 \text{ g.mol}^{-1}
\end{array}
\]

Method : See general procedure 6.7 using (1 equiv., 0.1 mmol, 25 mg) of 1-isopropyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester.

Purification : Flash column chromatography (silica gel, 9:1 PE:AcOEt).

Product : Yellow pale oil.

Isolated yield : 75%.

\(^1\)H NMR (δ, ppm) (CDCl\(_3\), 400 MHz) 7.24-7.11 (m, 4H, CH-Ph), 6.68 (br s, 1H, NCH=), 3.78 (s, 3H, OCH\(_3\)), 3.31 (hept, J = 6.4Hz, 1H, CH-Pr), 2.06 (s, 3H, CH\(_3\)), 1.07 (d, J = 6.4Hz, 3H, CH\(_3\)-'Pr), 1.04 (d, J = 6.4Hz, 3H, CH\(_3\)-'Pr).

\(^13\)C NMR (δ, ppm) (CDCl\(_3\), 100 MHz) 155.5 (Cq, C=O), 142.6 (Cq), 128.7 (Cq), 127.4 (CH), 125.3 (CH), 124.3 (CH), 122.8 (CH), 122.2 (CH), 118.9 (Cq), 57.0 (CH-Pr), 53.1 (OCH\(_3\)), 20.6 (CH\(_3\)-Pr), 19.8 (CH\(_3\)-Pr), 14.8 (CH\(_3\)).

IR (v, cm\(^{-1}\)) (CCl\(_4\)) 3070 (w), 3032 (w), 2974 (s), 2954 (s), 2925 (s), 2859 (m), 1746 (s), 1708 (s), 1650 (m), 1601 (w), 1482 (m), 1455 (s), 1442 (s), 1386 (s), 1350 (s), 1327 (s), 1257 (s), 1244 (s), 1234 (s), 1193 (m), 1173 (m), 1159 (m), 1133 (s), 1104 (m), 1072 (w), 1034 (w), 1016 (s).

HRMS (EI+, m/z) : Calculated: 246.1368 Found: 246.1373.
6-Methoxy-1,4-dimethyl-1H-cinnoline-2-carboxylic acid methyl ester 6.13f

![Structure of 6-Methoxy-1,4-dimethyl-1H-cinnoline-2-carboxylic acid methyl ester]

**MF:** C_{13}H_{16}N_{2}O_{3}

**MW:** 248 g.mol^{-1}

**Method:**
See general procedure 6.7 using (1 equiv., 0.08 mmol, 21 mg) of 6-methoxy-1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester.

**Purification:**
Flash column chromatography (silica gel, 1:1 PE: AcOEt).

**Product:**
White solid.

**Isolated yield:**
88%.

\[1^H\text{ NMR (δ, ppm)}\]
(CDCl\textsubscript{3}, 400 MHz)
7.05 (br s, 1H, CH-Ar), 6.74-6.72 (m, 2H, CH-Ar), 6.67 (br s, 1H, NCH=), 3.83 (br s, 3H, OCH\textsubscript{3}), 3.79 (s, 3H, OCH\textsubscript{3}), 2.76 (s, 3H, NCH\textsubscript{3}), 2.06 (s, 3H, CH\textsubscript{3}).

\[13^C\text{ NMR (δ, ppm)}\]
(CDCl\textsubscript{3}, 100 MHz)
157.4 (Cq, Ar), 153.8 (Cq, C=O), 138.5 (Cq, Ar), 128.7 (Cq, Ar), 123.2 (CH), 120.9 (CH), 116.8 (Cq), 112.8 (CH), 108.2 (CH), 55.5 (OCH\textsubscript{3}), 53.4 (OCH\textsubscript{3}), 45.5 (NCH\textsubscript{3}), 14.9 (CH\textsubscript{3}).

\[\text{IR (ν, cm}^{-1})\] (CCl\textsubscript{4})
3076 (w), 2998 (w), 2955 (m), 2922 (w), 2861 (w), 2835 (w), 2783 (w), 1737 (s), 1712 (s), 1649 (m), 1607 (m), 1573 (m), 1487 (s), 1444 (s), 1404 (m), 1388 (s), 1361 (s), 1311 (m), 1265 (s), 1246 (s), 1209 (m), 1193 (m), 1179 (m), 1146 (m), 1125 (s), 1113 (s), 1064 (m), 1048 (s), 1012 (m).

**HRMS (El+, m/z):**
Calculated: 248.1161 Found: 248.1156.

---

6-Chloro-1,4-dimethyl-1H-cinnoline-2-carboxylic acid methyl ester 6.13g

![Structure of 6-Chloro-1,4-dimethyl-1H-cinnoline-2-carboxylic acid methyl ester]

**MF:** C_{12}H_{13}O_{3}N_{2}Cl

**MW:** 252 g.mol^{-1}
Method: See general procedure 6.7 using (1 equiv., 0.07 mmol, 19 mg) of 6-chloro-1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 7:3 PE: AcOEt).

Product: White solid.

Isolated yield: 95%.

\(^1\)H NMR (\(\delta\), ppm) (CDCl\(_3\), 400 MHz) 7.16-7.14 (m, 2H, CH-Ar), 7.03 (br s, 1H, CH-Ar), 6.68 (br s, 1H, NCH=), 3.84 (s, 3H, OCH\(_3\)), 2.79 (s, 3H, NCH\(_3\)), 2.04 (s, 3H, CH\(_3\)).

\(^13\)C NMR (\(\delta\), ppm) (CDCl\(_3\), 100 MHz) 153.5 (Cq, C=O), 143.7 (Cq), 131.0 (Cq), 129.2 (Cq), 127.9 (CH), 123.6 (CH), 122.4 (CH), 121.7 (CH), 116.0 (Cq), 53.4 (OCH\(_3\)), 45.5 (NCH\(_3\)), 14.8 (CH\(_3\)).

IR (\(\nu\), cm\(^{-1}\)) (CCl\(_4\)) 2956 (w), 1741 (m), 1716 (s), 1558 (w), 1477 (m), 1444 (s), 1416 (w), 1403 (w), 1388 (w), 1358 (s), 1262 (m), 1243 (m), 1194 (w), 1123 (m), 1092 (m).

HRMS (El+, m/z) Calculated: 252.0666 Found: 252.0663.

6-Fluoro-1,4-dimethyl-1H-cinnoline-2-carboxylic acid methyl ester 6.13h

\[\text{MF: } C_{12}H_{13}O_2N_2F\]

\[\text{MW = 236 g.mol}^{-1}\]

Method: See general procedure 6.7 using (1 equiv., 0.07 mmol, 16 mg) of 6-fluoro-1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester.

Purification: Flash column chromatography (silica gel, 6:4 PE:AcOEt).

Product: White solid.

Isolated yield: 75%.

\(^1\)H NMR (\(\delta\), ppm) (CDCl\(_3\), 400 MHz) 7.07 (br s; 1H, CH-Ar), 6.90-6.85 (m, 2H, CH-Ar), 6.70 (br s, 1H, NCH=), 3.84 (s, 3H, OCH\(_3\)), 2.78 (s, 3H, NCH\(_3\)), 2.05 (s, 3H, CH\(_3\)).
**13C NMR (δ, ppm)**  
(CDCl₃, 100 MHz)  
160.49 (d, J₇₈ = 242.9 Hz, Cq, C=O), 153.0 (Cq, C=O), 141.0 (Cq), 129.5 (Cq), 123.8 (CH), 121.5 (CH), 115.8 (Cq), 114.5 (d, J₇₈ = 24.0 Hz, CH-Ar), 109.1 (d, J₇₈ = 24.0 Hz, CH-Ar), 53.5 (OCH₃), 45.5 (NCH₃), 14.8 (CH₃).

**IR (ν, cm⁻¹) (CCl₄)**  
3073 (w), 2997 (w), 2956 (w), 2923 (w), 2861 (w), 1744 (s), 1714 (s), 1650 (w), 1610 (w), 1581 (w), 1509 (w), 1496 (s), 1483 (s), 1445 (s), 1388 (m), 1359 (s), 1337 (s), 1265 (s), 1244 (s), 1194 (m), 1183 (m), 1143 (w), 1122 (s), 1111 (s), 1143 (w), 1030 (w), 1013 (w).

**HRMS (El+, m/z)**  
Calculated: 236.0961  
Found: 236.0959.

### 1,4-Dimethyl-1H-cinnoline-2,6-dicarboxylic acid 6-ethyl ester 2-methyl ester 6.13i

![Chemical Structure](image)

**MF:** C₁₅H₁₈N₂O₄  
**MW:** 290 g·mol⁻¹

**Method:**  
See **general procedure 6.7** using (1 equiv., 0.07 mmol, 16 mg) of 1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2,6-dicarboxylic acid 6-ethyl ester 2-methyl ester.

**Purification:**  
Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product:**  
White solid.

**Isolated yield:**  
81%.

**¹H NMR (δ, ppm)**  
(CDCl₃, 400 MHz)  
7.89-7.85 (m, 2H, CH-Ar), 7.10 (d, J = 8.0 Hz, 1H, CH-Ar), 6.65 (br s, 1H, NCH₃), 4.37 (q, J = 7.1 Hz, 2H, OCH₂CH₃), 3.84 (s, 3H, OCH₃), 2.89 (s, 3H, OCH₂CH₃), 2.10 (s, 3H, CH₃), 1.39 (t, J = 7.1 Hz, 3H, OCH₂CH₃).

**¹²C NMR (δ, ppm)**  
(CDCl₃, 100 MHz)  
166.1 (Cq, C=O), 149.6 (Cq, C=O), 131.3 (CH), 129.7 (CH), 129.1 (Cq), 127.3 (Cq), 123.8 (CH), 121.6 (CH), 116.9 (Cq), 110.8 (Cq), 61.0 (OCH₂CH₃), 53.4 (OCH₃), 45.4 (NCH₃), 15.0 (OCH₂CH₃), 14.3 (CH₃).

**IR (ν, cm⁻¹) (CCl₄)**  
3073 (w), 2980 (m), 2956 (m), 2926 (m), 2859 (m), 1717 (s), 1654 (w), 1607 (m), 1572 (w), 1488 (m), 1444 (s), 1389 (s), 1358 (s), 1279 (s), 1249 (s), 1226 (s), 1194 (m), 1181 (m), 1149 (m), 1106 (s), 1088 (m), 1028 (m).

**HRMS (El+, m/z)**  
Calculated: 290.1267  
Found: 290.1261.
6-Cyano-1,4-dimethyl-1H-cinnoline-2-carboxylic acid methyl ester

\[
\text{MF: C}_{13}\text{H}_{13}\text{N}_{3}\text{O}_{2} \\
\text{MW} = 243 \text{ g.mol}^{-1}
\]

**Method:**

See general procedure 6.7 using (1 equiv., 0.06 mmol, 15 mg) of 6-cyano-1-methyl-4-methylene-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester.

**Purification:**

Flash column chromatography (silica gel, 7:3 PE:AcOEt).

**Product:**

White solid.

**Isolated yield:**

73%.

**\(^{1}\text{H NMR}\) (δ, ppm)**

(CDCl\(_3\), 400 MHz)

7.48-7.42 (m, 2H, CH-Ar), 7.10 (d, \(J = 8.2\text{Hz}, 1\text{H, CH-Ar}\)), 6.68 (br s, 1H, NCH=), 3.84 (s, 3H, OCH\(_3\)), 2.91 (s, 3H, NCH\(_3\)), 2.05 (s, 3H, CH\(_3\)).

**\(^{13}\text{C NMR}\) (δ, ppm)**

(CDCl\(_3\), 100 MHz)

153.5 (Cq, C=O), 149.4 (Cq), 132.0 (CH), 128.3 (Cq), 126.2 (CH), 123.0 (CH), 122.3 (CH), 118.8 (Cq), 115.7 (Cq), 108.6 (Cq, CN), 53.6 (OCH\(_3\)), 45.5 (NCH\(_3\)), 14.7 (CH\(_3\)).

**IR (ν, cm\(^{-1}\)) (CCl\(_4\))**

3000 (w), 2956 (m), 2925 (m), 2230 (m), 1742 (s), 1718 (s), 1648 (m), 1608 (m), 1488 (s), 1444 (s), 1419 (m), 1406 (m), 1389 (m), 1359 (s), 1286 (s), 1263 (s), 1248 (s), 1195 (s), 1120 (s), 1070 (m), 1032 (m), 1013 (m).

**HRMS (El+, m/z):**


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1,4-Dimethyl-6-trifluoromethyl-1H-cinnoline-2-carboxylic acid methyl ester

\[
\text{MF: C}_{13}\text{H}_{13}\text{O}_{2}\text{N}_{2}\text{F}_{3} \\
\text{MW} = 286 \text{ g.mol}^{-1}
\]
**Method:**  See general procedure 6.7 using (1 equiv., 0.05 mmol, 15 mg) of 1-methyl-4-methylene-6-trifluoromethyl-3,4-dihydro-1H-cinnoline-2-carboxylic acid methyl ester.

**Purification:**  Flash column chromatography (silica gel, 8:2 PE:AcOEt).

**Product:**  White solid.

**Isolated yield:**  67%.

**$^1$H NMR ($\delta$, ppm)**  (CDCl$_3$, 400 MHz)

<table>
<thead>
<tr>
<th>Peak Description</th>
<th>Chemical Shift (ppm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>7.46-7.40 (m, 2H, CH-Ar)</td>
<td>7.17 (d, $J = 7.7\text{Hz}$, 1H, CH-Ar)</td>
</tr>
<tr>
<td>6.70 (br s, 1H, NCH=)</td>
<td>3.84 (s, 3H, OCH$_3$)</td>
</tr>
<tr>
<td>2.87 (s, 3H, NCH$_3$)</td>
<td>2.09 (s, 3H, CH$_3$)</td>
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</table>

**$^{13}$C NMR ($\delta$, ppm)**  (CDCl$_3$, 100 MHz)

<table>
<thead>
<tr>
<th>Peak Description</th>
<th>Chemical Shift (ppm)</th>
</tr>
</thead>
<tbody>
<tr>
<td>155.3 (q, $J_{C-F} = 3.5\text{Hz}$, CH-Ar)</td>
<td>124.0 (q, $J_{C-F} = 272.0\text{Hz}$, Cq, CF$_3$)</td>
</tr>
<tr>
<td>122.2 (NCH=)</td>
<td>119.4 (CH-Ar, q, $J_{C-F} = 3.3\text{Hz}$)</td>
</tr>
<tr>
<td>116.1 (Cq)</td>
<td>111.2 (Cq), 53.5 (OCH$_3$), 45.5 (NCH$_3$), 14.8 (CH$_3$)</td>
</tr>
</tbody>
</table>

**IR ($\nu$, cm$^{-1}$) (CCl$_4$)**

<table>
<thead>
<tr>
<th>Peak Description</th>
<th>Chemical Shift (cm$^{-1}$)</th>
</tr>
</thead>
<tbody>
<tr>
<td>3421 (w)</td>
<td>2957 (m), 2926 (m), 2856 (m), 2716 (s), 1648 (w), 1620 (s), 1579 (w), 1494 (w), 1444 (s), 1389 (s), 1357 (s), 1318 (s), 1284 (s), 1235 (s), 1194 (s), 1169 (s), 1130 (s), 1077 (s)</td>
</tr>
</tbody>
</table>

**HRMS (EI+, m/z):**

Calculated: 286.0929  Found: 286.0928.